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ELECTRICAL RESISTIVITY OF CHROMIUM, COBALT, IRON, AND NICKEL

By
T. K. Chu and C. Y. Ho

CINDAS Report 60

June 1982

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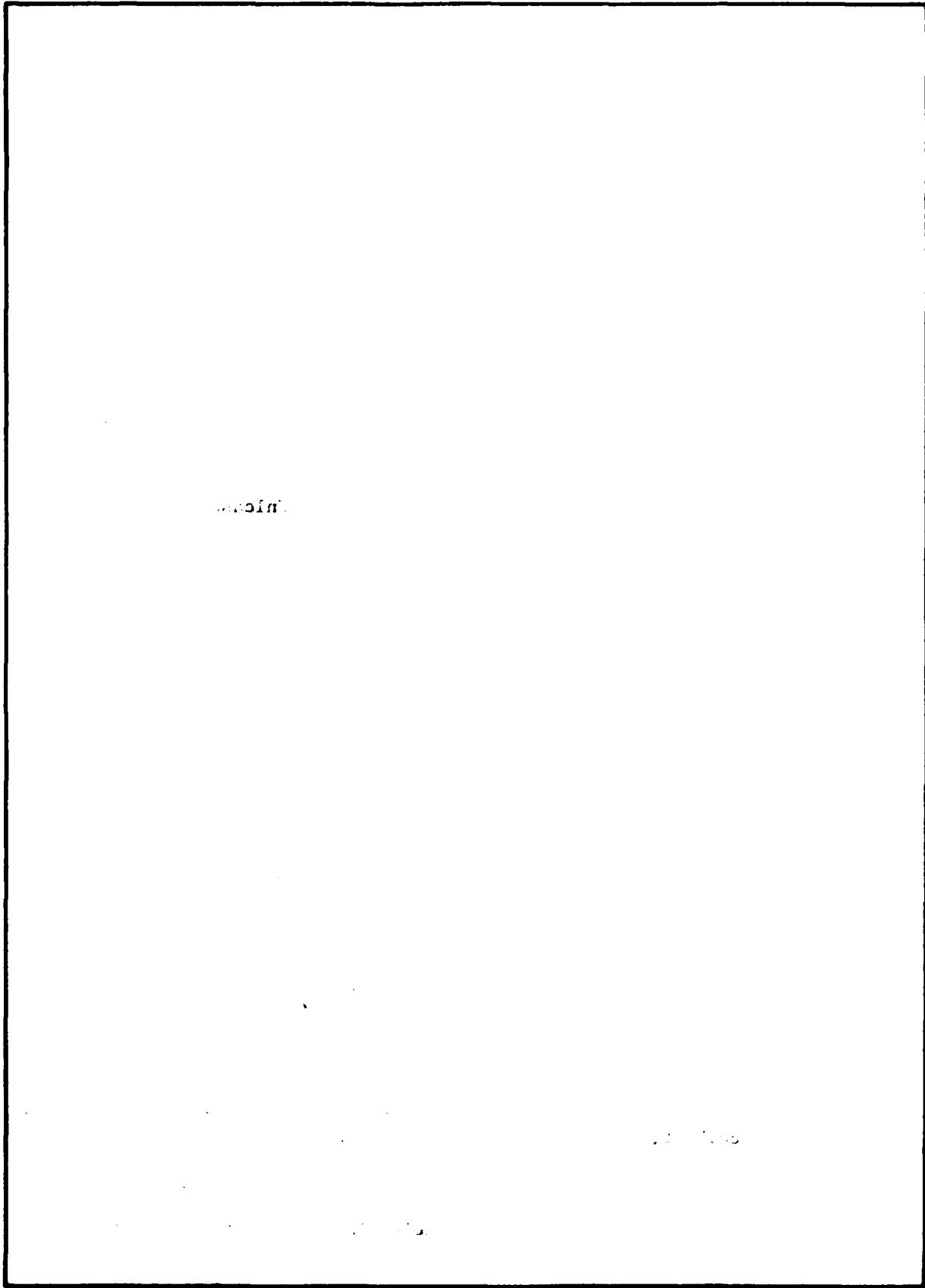
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PREFACE

This technical report was prepared by the Center for Information and Numerical Data Analysis and Synthesis (CINDAS), Purdue University, West Lafayette, Indiana, under the auspices of the Office of Standard Reference Data of the National Bureau of Standards (NBS), Department of Commerce, Washington, D.C.

This report represents the most exhaustive compilation and critical evaluation of the recorded world knowledge on the electrical resistivity of chromium, cobalt, iron, and nickel, and is one of a series of technical reports on the electrical resistivity of selected elements. The literature search and data compilation have been done in a most extensive and detailed manner, making it possible for all users of the subject to have access to the original data without having to duplicate the laborious and costly process of literature search and data extraction. Also, for the active researchers in the field, a detailed discussion is presented for each material, reviewing the available data and information, giving details of data analysis and synthesis, and discussing the considerations involved in arriving at the final recommended values.

It is hoped that this work will prove useful not only to the engineers and scientists in the field but also to other engineering research and development programs and for industrial applications, as it provides a wealth of knowledge heretofore unknown or inaccessible to many. In particular, it is thought that the critical evaluation, analysis and synthesis, and reference data generation constitute a unique aspect of this work.

Although this report is primarily the result of financial support and interest of the NBS Office of Standard Reference Data, the extensive documentary activity essential to this work was supported by the Defense Logistics Agency of the Department of Defense. Thanks are due Dr. H. J. White, Jr., of the NBS Office of Standard Reference Data for his guidance, cooperation, and sympathetic understanding during the course of this work.

ABSTRACT

This work compiles, reviews, and discusses the available data and information on the electrical resistivity of chromium, cobalt, iron, and nickel and presents the recommended values resulting from critical evaluation, correlation, analysis, and synthesis of the available data and information. The recommended values presented are uncorrected and also corrected for the thermal expansion of the material and cover the temperature range from 1 K to above the melting point into the molten state. The estimated uncertainties in most of the recommended values are about $\pm 5\%$.

Key words: Chromium; cobalt; conductivity; critical evaluation; data analysis; data compilation; data synthesis; electrical conductivity; electrical resistivity; elements; iron; metals; nickel; recommended values; resistivity.

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NOMENCLATURE

A	Constant in eqs (3b) and (8)
c	Impurity concentration
C	Constant in eq (3a)
e	Base of natural logarithm
\hbar	Planck constant divided by 2π
k	Boltzmann constant
L	Length of specimen at T
L_0	Length of specimen at T_0
ΔL	$\Delta L = L - L_0$
M	Atomic weight
T	Temperature
T_0	Reference temperature
x	$x = \hbar\omega/kT$
α	Constant in eqs (7) and (8)
Δ	Deviation from the Matthiessen's rule
θ_D	Debye temperature
θ_R	Characteristic temperature for intrinsic electrical resistivity
ρ	Electrical resistivity
ρ_0	Residual electrical resistivity
ρ_e	Electrical resistivity due to electron-electron scattering
ρ_i	Intrinsic electrical resistivity
ω	Phonon angular frequency

1. INTRODUCTION

The principal objective of this project was to exhaustively compile, critically evaluate, analyze, and synthesize all the available data and information on the electrical resistivity of a large number of selected elements and to generate recommended values over a full range of temperature from 1 K to the melting point and beyond. The results on the electrical resistivity of chromium, cobalt, iron, and nickel are presented in this work, which is one in a series of similar works on the electrical resistivity of selected elements, some published [1-3]¹. The comprehensive study of the electrical resistivity of the elements at the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) has been a continuation of a similar extensive work on the thermal conductivity of the elements [4].

The general background information on this work is given in Section 2, which includes a brief introduction to the theory of the electrical resistivity of metals and a detailed explanation of the specifics and conventions used in the presentation of the data and information.

The experimental data and information and the recommended values for the electrical resistivity of the four elements are presented in Section 3. In the discussion of the electrical resistivity of each element, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties in the recommended values are stated. Recommended values uncorrected and corrected for the thermal expansion of the material are both presented in this section. The values cover the temperature range from 1 K to above the melting point.

The last three sections are for acknowledgments, appendices, and references. There are two appendices given. The first appendix presents a logical organization of the methods for the measurement of electrical resistivity. The methods are designated with respective code letters and the same code letters are used in the "Method Used" column of the Table of Measurement Information for indicating

¹Numbers in brackets indicate literature references listed in Section 6.

the experimental methods used by the various authors. The second appendix presents conversion factors for the units of electrical resistivity, which may be used to convert easily the electrical resistivity values in the SI units given in this work to values in any of the several other units listed.

2. GENERAL BACKGROUND

2.1. Theoretical Background

It was found experimentally by Matthiessen [5,6] that the increase in the electrical resistivity of a metal due to the presence of a small amount of another metal in solid solution is independent of the temperature. According to this Matthiessen's rule, the total electrical resistivity of an impure metal may therefore be separated into two additive contributions and written in the form

$$\rho(c,T) = \rho_0(c) + \rho_i(T), \quad (1)$$

where ρ_0 is the residual resistivity caused by the scattering of electrons by impurity atoms and lattice defects and is temperature-independent but dependent on the impurity concentration, c , and ρ_i is the temperature-dependent intrinsic resistivity arising from the scattering of electrons by lattice waves, or phonons.

In reality, however, deviations from Matthiessen's rule do occur. Thus, in general the electrical resistivity of an impure metal is given by

$$\rho(c,T) = \rho_0(c) + \rho_i(T) + \Delta(c,T), \quad (2)$$

where Δ is the deviation from the Matthiessen's rule.

The intrinsic electrical resistivity which is due to scattering of electrons by phonons may be approximated by the Bloch-Grüneisen formula [7,8]:

$$\rho_i = \frac{C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2} \quad (3a)$$

$$= A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2}, \quad (3b)$$

where C is a constant characteristic of the metal and proportional to the square of the electron-phonon interaction constant, M is the atomic weight, θ_R is a characteristic temperature of the metal which characterizes its intrinsic electrical resistivity in the same way as the Debye temperature,

θ_D , characterizes its lattice specific heat, and $A \equiv C/M\theta_R$. The dimensionless variable of integration $x = \hbar\omega/kT$, where \hbar is the Planck constant divided by 2π , ω is the phonon angular frequency, and k is the Boltzmann constant. The derivation of eq (3) is based on the simplifying assumptions that the Fermi surface is spherical, that the conduction electrons can be treated as free in the first approximation, that the spectrum of lattice vibrations is that of the Debye model, that the phonon distribution is essentially undisturbed by the scattering processes, and that electron-phonon Umklapp processes can be ignored. Consequently, it is perhaps most reasonable to expect the Bloch-Grüneisen formula to agree with experiment in the case of monovalent metals. Nevertheless, the intrinsic resistivity of many metals can be well represented by eq (3) over a wide temperature range by a suitable choice of θ_R and C , though no single values of θ_R can fits the data at all temperatures.

At low temperatures ($T \leq \theta_R/20$), eq (3a) reduced to

$$\rho_i = \frac{124.4C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5, \quad (4)$$

while at high temperatures ($T > \theta_R$), to a good approximation, it reduces to

$$\rho_i \approx \frac{C}{4M\theta_R} \left(\frac{T}{\theta_R} \right). \quad (5)$$

Thus it agrees with the experimental facts that at very low temperatures the intrinsic electrical resistivity (after subtracting ρ_0 from ρ) of most metallic elements is proportional to T^5 , and at high temperatures the resistivity of most metals increases approximately linearly with temperature.

In separating the electrical resistivity into its components, the temperature dependent part sometimes includes the electrical resistivity due to electron-electron scattering, ρ_e ; indeed, this is thought to be the dominant temperature-dependent term in transition metals at low temperatures. That is,

$$\rho = \rho_0 + \rho_e + \rho_i(T) \quad (6)$$

As in the case of the scattering of electrons by phonons, electron-electron collisions are of two types: normal processes in which the total wave vector is conserved, and Umklapp processes in which the total wave vectors before and after the collision differ by a reciprocal lattice vector. On the other hand, unlike electron-phonon Umklapp processes which are frozen out at low temperatures

if the Fermi surface is everywhere clear of the zone boundary, electron-electron Umklapp processes are not frozen out at low temperatures. Normal processes, involving the collision between two s-band conduction electrons, do not contribute directly to the electrical resistivity because they do not change the total momentum and thus have no effect on the current. Normal processes involving the scattering of an s-band conduction electron by a non-conducting d-band electron do contribute to the electrical resistivity, and are thought to be the dominant temperature-dependent resistive processes in transition elements and their alloys at very low temperatures, since their resistivities show the T^2 temperature dependence expected for electron-electron scattering rather than the T^5 temperature dependence expected for the intrinsic resistivity. This temperature dependence of the electrical resistivity due to electron-electron scattering:

$$\rho_e = \alpha T^2 \quad (7)$$

comes about through the double application of the exclusion principle in the scattering processes; it applies to both the initial states and final states. In eq (7), α is a constant.

Umklapp processes between two conduction electrons do contribute to the electrical resistivity. Because these processes involve a reciprocal lattice vector, the wave functions of the electrons involved cannot be regarded as simple plane waves, but must be treated as true Bloch functions having the periodicity of the lattice. The results of this are to introduce into the expression for the resistivity the square of an interference factor. Apparently this factor is quite small, as the low temperature electrical resistivity of most ordinary metals does not show the T^2 temperature dependence expected for such a resistive mechanism.

Substituting eqs (7) and (3b) into eq (6) yields

$$\rho = \rho_0 + \alpha T^2 + A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2} \quad (8)$$

Equation (8) has been used frequently in analyzing the experimental data and in generating the recommended values for the electrical resistivity of cobalt, iron, and nickel at low temperatures.

2.2. Presentation of Data and Information

In each of the subsections in Section 3, electrical resistivity data and information for each element are presented in the following order:

- (1) A discussion text,
- (2) A table of recommended values,
- (3) A figure presenting recommended values and experimental data as a function of temperature in log-log scale,
- (4) A figure presenting recommended values and experimental data as a function of temperature in linear scale,
- (5) A table giving measurement information on the experimental data presented in the figures, and
- (6) A comparable table tabulating experimental data of all the data sets presented in the figures and/or listed in the tables.

In the discussion text on the electrical resistivity of each alloy system, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties of the recommended values are stated.

The recommended values are for well-annealed high-purity specimens of the respective elements; however, those values for low temperatures are applicable only to the particular specimens having residual electrical resistivities as given at 1 K in the tables.

Recommended values uncorrected and corrected for the thermal expansion of the element are both given in the table. The uncorrected and corrected values are related by the following equation:

$$\rho_{\text{corrected}}(T) = \left(1 + \frac{\Delta L(T)}{L_0}\right) \rho_{\text{uncorrected}}(T), \quad (9)$$

where $\Delta L = L - L_0$, and L and L_0 are the lengths of the specimen at any temperature T and at a reference temperature T_0 , respectively. The thermal expansion correction amounts roughly to about -0.2% to -0.7% at very low temperatures, zero at room temperature, about 0.3% to 0.7% at 500 K, and about 2% near the melting point of the element.

The recommended values in some cases are given with more significant figures than warranted, which is merely for tabular smoothness or for the convenience of internal comparison. Hence, the number of significant figures given in the table has no bearing on the degree of accuracy or uncertainty in the values; the uncertainty in the values is always explicitly stated.

In the figures, a data set consisting of a single data point is denoted by a number enclosed by a square, and a curve that connects a set of two or more data points is denoted by a ringed number. These data set numbers correspond to those listed in the accompanying tables providing measurement information and tabulating numerical data for each of the data sets. When several sets of data are too close together to be distinguishable, some of the data sets, though listed and tabulated in the tables, are omitted from the figure for the sake of clarity. The data set numbers of those data sets omitted from the figure are asterisked in both tables providing the measurement information and tabulating the experimental data. If only part of the data points of a data set are omitted from the figure, only those data points omitted are asterisked in the table tabulating the experimental data.

The tables providing the measurement information contain for each set of experimental data the following information: data set number, reference number, author(s), year of publication, experimental method used for the measurement, temperature range covered by the data, name and specimen designation, specimen composition, specification and characterization, and information on measurement conditions, which are contained in the original paper. The experimental methods used for the measurement of the electrical resistivity are indicated in the column headed "Method Used" in the table by the following code letters:

- A Direct-current potentiometer method
- B Direct-current bridge method
- C Alternating-current potentiometer method
- D Alternating-current bridge method
- G Galvanometer amplifier method
- R Rotating magnetic field method
- V Voltmeter and ammeter direct reading method

- This symbol means either that the method described by the author is not sufficient for assigning a specific code letter or that the use of a code letter would not convey enough of the information reported in the research document, and therefore the method used is described briefly in the last column of the table.

Details of these and other methods for the measurement of electrical resistivity may be found in the literature references given in Appendix 5.1, which presents a complete scheme for the classification and organization of the methods.

In the tables tabulating the experimental data, all the original data reported in different units have been converted to have the same units: the SI units $10^{-8} \Omega \text{ m}$. The recommended values generated are also given in the same units. Conversion factors for the units of electrical resistivity, which may be used to convert the electrical resistivity values in the SI units given in this work to values in other units, are given in Appendix 5.2.

3. ELECTRICAL RESISTIVITY DATA AND INFORMATION

3.1. Chromium

There are 163 sets of experimental data available for the electrical resistivity of chromium. The information on specimen characterization and measurement condition for each of the data sets is given in table 2. The data are tabulated in table 3 and shown partially in figures 1 and 2.

Chromium undergoes an antiferromagnetic-paramagnetic transition at about 312 K; it is not surprising that more than one third of the data sets deal with the behavior of the electrical resistivity at temperatures in the vicinity of the transition. In general, the purities of the chromium specimens studied are not as high as those of other more common elements, such as iron and nickel. Judging from the impurity analyses reported, a purity of 99.99% appears to be the highest available at the present time. It is therefore not unexpected that the reported residual resistivity ratios are not very high. Indeed, the highest for a polycrystalline specimen is 380 given by Laubitz and Matsumura [9] (data sets 53-61). Their specimen was the same as that of Moore et al. [10] (data sets 50-52), who reported a residual resistance ratio of 280. This specimen was prepared by compacting (apparently small) crystals, and by hot extending. The former authors, in addition, carried out extended annealing periods: four days at 1100 K and one day at 1200 K. The reported purity of this specimen was not particularly high; 99.98⁺% with major impurities of 0.0070% C and 0.0030% Fe. For comparison, the cast specimen of Meaden et al. [11-13] (data sets 69-75), had a reported purity of 99.999% with major impurities of 0.0010% C and 0.0080% O. However, the residual resistance ratio of this specimen is only 178. After annealing at >1273 K for 75 hours (of which 50 hours is at 1473 K), the residual resistance ratio increases to 295 (data sets 76, 77). It is apparent that, in order to obtain a true indication of the residual resistivity of a chromium specimen, prolonged annealing at temperatures in excess of 1000 K is quite essential. The present recommendation for the electrical resistivity of chromium at low temperatures is only for chromium having a particular residual resistivity, which is based on the residual resistance ratio reported by Laubitz and Matsumura [9] for data set 53.

There are only a few data sets which give the electrical resistivity of chromium from liquid-helium temperature to room temperature in reasonably small temperature intervals: Harper et al. [14] (data set 17), Goff [15,16] (data sets 79,80), and Araj's et al. [17-20] (data sets 94-102). The data of Harper

et al. had been analyzed by White and Woods [21] who found the temperature dependent part of the resistivity proportional to $T^{3.2}$ for temperatures below 100 K. A similar analysis on the data sets 7, 79-80, and 99-100 substantiated the finding of White and Woods; the exponent was found to be 3.23, with an uncertainty of ± 0.20 . In addition, selected data points from those reported by Chiu et al. [22] (data set 46), and by Moore et al. [10] (data sets 47-50) in the temperature range 80-100 K are also in agreement with this finding. With a coefficient of $5.756 \times 10^{-15} \Omega \text{ m K}^{-3.23}$, the experimental data of $\rho - \rho_0$ predominantly stay within $0.1 \times 10^{-8} \Omega \text{ m}$ of the calculated values at the higher end of this temperature range, and within $0.002 \times 10^{-8} \Omega \text{ m}$ at lower temperatures ($< 20 \text{ K}$). (It is interesting to note that the specimen of Araj's et al. [17,18] (data sets 94-99) is for a single crystal specimen with residual resistivity of $1.06 \times 10^{-8} \Omega \text{ m}$.) The electrical resistivity values below 100 K were therefore obtained by the relation

$$\rho(10^{-8} \Omega \text{ m}) = \rho_0 + 5.756 \times 10^{-7} T^{3.23} \quad (10)$$

At temperatures above 100 K, the rate of increase of the temperature dependent part of the resistivity becomes slower with increasing temperature. The discrepancies between the data sets also become larger with increasing temperature, indicating that the deviation from Mattheissen's Rule becomes important. From studies of the electrical resistivities of chromium alloys (see, e.g., Araj's et al. [20], deVries [23], Cox and Lucke [24], Taylor [25], and Muheim and Müller [26]), it was found that impurities not only affect the values of the resistivity, but also the Néel temperature. Since the electrical resistivities of chromium and dilute chromium alloys generally show a local maximum at temperatures slightly below the Néel temperature, it is not unexpected that the data sets show greater discrepancies as the Néel temperature is approached. Furthermore, depending on the type of impurity, the electrical resistivity of a chromium alloy can be lower than that of the pure chromium at temperatures immediately below the Néel temperature of pure chromium (see, e.g., Taylor [25], Susuki [27] and deVries [23]). The recommended values in the temperature range from 100 K to the Néel temperature are based on the data of Moore et al. [10] (data set 50), and in the vicinity of the Néel temperature they are based on the above data and that of Laubitz and Matsumura [9] (data set 53). As it is mentioned previously, the same specimen was used in both of these two measurements: the latter authors annealed the specimen at a higher temperature for

long periods of time. The difference between the resistivity values of these data sets at ~ 300 K is $\sim 0.4\%$ or $\sim 0.05 \times 10^{-8} \Omega \text{m}$. This difference is higher than the difference of $\sim 0.01 \times 10^{-8} \Omega \text{m}$ in their residual resistivities (calculated from the reported residual resistance ratios); but is still within the limits given by the reported measurement inaccuracies. The data of Meaden et al. [13, 28] (data sets 76-78) for a specimen with a residual resistance ratio of 295 show slightly weaker temperature dependence: the values being $\sim 10\%$ above and $\sim 2\%$ below those of data set 50 at ~ 100 K and ~ 300 K, respectively.

The behavior of the electrical resistivity of chromium in the vicinity of the Néel temperature has been studied quite extensively: it goes through a broad maximum at approximately 4 K below the Néel temperature and decreases rapidly as the Néel temperature is approached. The temperature derivative of the electrical resistivity then goes through a spike-like minimum. The ensuing minimum in electrical resistivity value occurs at a few tenths of a degree above the Néel temperature. The position of the minimum in the temperature derivative has been associated with the Néel temperature. However it has been proposed recently, from theoretical calculations, that the temperature derivative of the electrical resistivity should follow a power law relation: $(T_N - T)T_N^{-1}$ (see, e.g., Suezaki and Mori [29], Alexander et al. [30]). The recent publication by Rapp et al. [31] showed that the power law was only applicable in the temperature range from $T_N - 8.5$ K to $T_N - 0.5$ K, and the temperature derivative of the measured resistance was at a minimum at about 0.18 degree below T_N determined by a fit to the power law relation. The simultaneous measurements of electrical resistivity and sublattice magnetization by neutron diffraction method on a single crystalline iodide chromium specimen by Ishikawa et al. [32] (data set 158), showed that the minimum in the electrical resistivity occurred at about 0.5 degree above the Néel temperature. They also found that there was some residual ordering above the transition. This residual ordering persisted till ~ 315 K, and was attributed by the authors to the strain introduced in spot welding the specimen. However, this interpretation appears to be in conflict with the observation of Stebler [33] (data sets 112-115), who reported considerable hysteresis across the Néel transition. Stebler attributed the hysteresis to (thermal) strain, but failed to observe appreciable residual ordering (again with neutron diffraction method) in his specimen. It is apparent that the critical phenomenon of antiferromagnetic-paramagnetic transition in chromium is

a complex one and is still subject to further investigations. Experimentally, the accurate determination of the electrical resistivity in the close vicinity of the Néel temperature poses considerable difficulties. This is due to the rapid change in the temperature derivative of the electrical resistivity over a narrow temperature range, while the value of the electrical resistivity itself changes only slightly. Thus, even for the same specimen, Laubitz and Matsumura [9] (data set 53) found that the position of the resistivity minimum was at 311.7 K, but from the data of Moore et al. [34] (data set 63), it appeared that the minimum occurred at around 312.3 K. For comparison, Matsumoto and Mitsui [35] stated that the minimum was at 312.0 K for a specimen with residual resistivity ratio comparable to that of the above (350 instead of 380). For these reasons, it is concluded that a recommendation for the detailed variation of the electrical resistivity of chromium at the close vicinity ($\sim \pm 0.5$ degree K) is beyond the scope of the present study. The position of the resistivity minimum is tentatively taken at 311.7 K following Laubitz and Matsumura [9], as these authors apparently made their measurements at very small temperature intervals. This temperature is within ± 0.2 degree of those determined from the data of Anderson et al. [36] (data set 87), Stebler [33] (data sets 114, 115) and Trego and Mackintosh [37] (data set 116). The Néel temperature of chromium is tentatively taken as 311.5 K, as determined by specific heat measurements.

Even though hysteresis across the Néel transition is not generally mentioned by most authors, it has been reported by some: Mitsui and Tomizuka [38] (data sets 33, 34), and Stebler [33] (data sets 112, 113). If the hysteresis is caused by strain, as suggested by Stebler [33], it should disappear in specimens that have been annealed for a sufficiently long period of time at high temperatures and are heated up and/or cooled down through the transition at a sufficiently slow rate during the measurements.

For chromium, there is another transition occurring at about 120 K: the spin-flip transition. At this temperature, the polarizations of the spin-density waves, which give rise to the antiferromagnetism in chromium, changes from longitudinal (at lower temperatures) to transverse (at higher temperatures). Arajs [39], upon reanalyzing the earlier data of Arajs and Dunmyre [19] (data set 100), concluded that there is a change in the temperature coefficient of the resistivity, which was first reported by Matsumoto et al. [35]. Meaden et al. [12,13] (data sets 71-77) reported a step-type anomaly, in addition to a change of slope.

However, the slope changes reported are quite different: from $T^{2.8}$ to $T^{2.0}$ according to Arajs [39] and from $T^{2.45}$ to $T^{2.25}$ according to Meaden et al. [12,13]. In addition, the step-type anomaly, also reported by Kostina et al. [40] (data sets 134, 138) for single crystalline specimen, is in the opposite direction (a decrease instead of an increase in value) to that reported by Meaden et al. [12,13]. Most other authors did not report any unusual behavior of the electrical resistivity at this temperature, and Moore et al. [10] stated that the spin-flip transition did not have a noticeable effect on the electrical resistivity. Muir and Ström-Olsen [41] also did not detect any change of the temperature coefficient of the measured resistance of their single domain specimen at the spin-flip temperature. In generating the present recommended values, it is assumed that the electrical resistivity is not affected by this transition, and the values and their temperature derivative are continuous through the transition.

At temperatures above the Néel transition, the electrical resistivity of chromium varies smoothly with temperature. Among the available data sets, there are three from independent sources that agree well with one another: Moore et al. [34] (data set 64), Arajs et al. [20] (data set 102), and Cox and Lucke [24] (data set 103). The agreement between data sets 64 and 102 is within 1% from 400 to 1000 K, and between data sets 64 and 103 is within 2.5% from 400 to 1300 K. The recommended values from the Néel temperature to 1300 K are based on these three data sets, with more weight given to that of Moore et al. [34] (data set 64) since the specimen of this data set is the same as that of data sets 50 and 53 upon which the recommended values at lower temperatures are based. It should be noted here that the resistivity values reported by Moore et al. [34] had been corrected for thermal expansion. However, these authors did not report the method by which the correction was applied. Therefore, the comparison mentioned above was carried out after account had been taken of the effect of thermal expansion, using the recommended thermal expansion values of Touloukian et al. [42, p. 61].

There are a number of data sets for temperatures above 1300 K. The agreement between them is not good: the spread among them is about $20 \times 10^{-8} \Omega m$ at 1500 K and $30 \times 10^{-8} \Omega m$ at 1900 K. Even though there are five data sets (26, 27, 29, 82, 83) from essentially two groups of workers that show agreement within $\pm 4 \times 10^{-8} \Omega m$ at 1700 K, these data are considered not reliable. Those by Anderson et al. [43] (data sets 82, 83) give values that are much too low at lower temperatures, and

those by Baum et al. [44,45] (data sets 26, 27), and by Levin et al. [46] (data set 29) show slopes that are considered too low. In addition, the room temperature value given by data set 26 is much too high (by $\sim 4.5 \times 10^{-8} \Omega \text{m}$ than the recommended value). The recommended values from 1300 to 1700 K were derived by extrapolation of the recommended values for lower temperatures with a temperature dependence that was based roughly on the data by Powell and Tye [47] (data set 106) and by Anderson et al. [43] (data set 84). In this temperature range, both of these data sets show a slight curvature toward the temperature axis, and are more or less parallel to each other, even though data set 106 is for a 99.985% pure electrodeposited specimen and data set 84 is apparently for a single crystal-line specimen. The slight curvature also appears to be evident in data sets 26, 27, and 29. The recommended values for temperatures from 1700 K to the melting point are based on numerical extrapolation of the values for 1300 to 1700 K. At 1700 K the recommended value is higher by $\sim 11\%$ (or $\sim 9 \times 10^{-8} \Omega \text{m}$) than the data of Baum et al. [44,45] (data set 26, 27), and at 2100 K it is higher by $\sim 12\%$ (or $\sim 12 \times 10^{-8} \Omega \text{m}$). Anderson et al. [43] (data sets 82, 83) reported sudden increase in electrical resistivity values at ~ 1900 K, which they attributed to the evaporation of sample material. There was no evidence of such behavior from the data of Baum et al. [44,45] (data sets 26, 27) and those of Levin et al. [46] (data set 29). Neither was evident from the data of Grube and Knabe [48] (data sets 6-8) which were apparently for specimens that were either porous and/or less pure.

There are only three data sets on the electrical resistivity of chromium in the molten state: by Baum et al. [44,45] (data sets 26, 27) and by Levin et al. [46] (data set 29). As it is mentioned in the last paragraph, the electrical resistivity values of these data for lower temperatures appeared to be questionable. The recommended value for the electrical resistivity of molten chromium at the melting point was obtained by multiplying the recommended value for solid chromium at the melting point by the ratio of the electrical resistivity values (101.5 and $108.1 \times 10^{-8} \Omega \text{m}$, respectively, for the solid and the molten states) reported explicitly in the text by Baum et al. [45] (data set 27). For temperatures above the melting point, the recommended values were calculated according to a linear dependence based on data sets 26 and 29.

The recommended values both uncorrected and corrected for thermal expansion of the material are presented in table 1, while only the uncorrected values are

shown in figures 1 and 2 along with the experimental data. The values are applicable to chromium of purity 99.98% or higher; however, those values for temperatures below 100 K are applicable only to chromium having a residual resistivity of $0.0306 \times 10^{-8} \Omega\text{m}$. The estimated uncertainty in the recommended values is about $\pm 5\%$ up to 1300 K. The uncertainty increases with temperature at higher temperatures and is estimated to be $\pm 10\%$ immediately below the melting point, and $\pm 15\%$ for the molten state.

From the available data, it appears that the low-temperature resistivity of chromium of lower purity can be obtained by the use of the Mattheissen's rule if the residual resistivity of a specimen does not exceed $\sim 0.2 \times 10^{-8} \Omega\text{m}$. Thus, using the recommended values and the Mattheissen's rule, the data for the specimen CrB of Moore et al. [10] (data set 47) would be reproduced to within $\sim \pm 5\%$. For the data of Chiu et al. [22] (data set 46), it was $\sim \pm 10\%$. Generally, this method underestimates the resistivity values. And even though the derivation from Mattheissen's rule can be negative for some dilute chromium alloys, it is not likely to occur for chromium of reasonable purity.

The recommended values uncorrected for thermal expansion given in table 1 can be represented approximately by the following expressions to within $\pm 0.5\%$.

1-90 K:

$$\rho = 0.0306 + 5.756 \times 10^{-7} T^{3.23} \quad (11)$$

90-293 K:

$$\rho = 0.398 - 2.950 \times 10^{-2} T + 5.112 \times 10^{-4} T^2 - 9.218 \times 10^{-7} T^3 \quad (12)$$

293-305 K:

$$\rho = 250.125 - 2.65115 T + 9.68307 \times 10^{-3} T^2 - 1.16108 \times 10^{-5} T^3 \quad (13)$$

305-311 K:

$$\rho = 1.4614467 \times 10^4 - 1.4360559 \times 10^2 T + 4.7073008 \times 10^{-1} T^2 - 5.142874 \times 10^{-4} T^3 \quad (14)$$

312-400 K:

$$\rho = 27.036 - 1.5301 \times 10^{-1} T + 4.5057 \times 10^{-4} T^2 - 3.4505 \times 10^{-7} T^3 \quad (15)$$

400-1300 K:

$$\rho = 4.457 + 1.3084 \times 10^{-2} T + 4.9046 \times 10^{-5} T^2 - 3.0031 \times 10^{-8} T^3 + 8.653 \times 10^{-12} T^4 \quad (16)$$

1300-2133 K:

$$\rho = -49.515 + 1.11856 \times 10^{-1}T - 2.3954 \times 10^{-5}T^2 + 3.4937 \times 10^{-9}T^3 \quad (17)$$

2133-2300 K:

$$\rho = 14.54 + 0.050 T \quad (18)$$

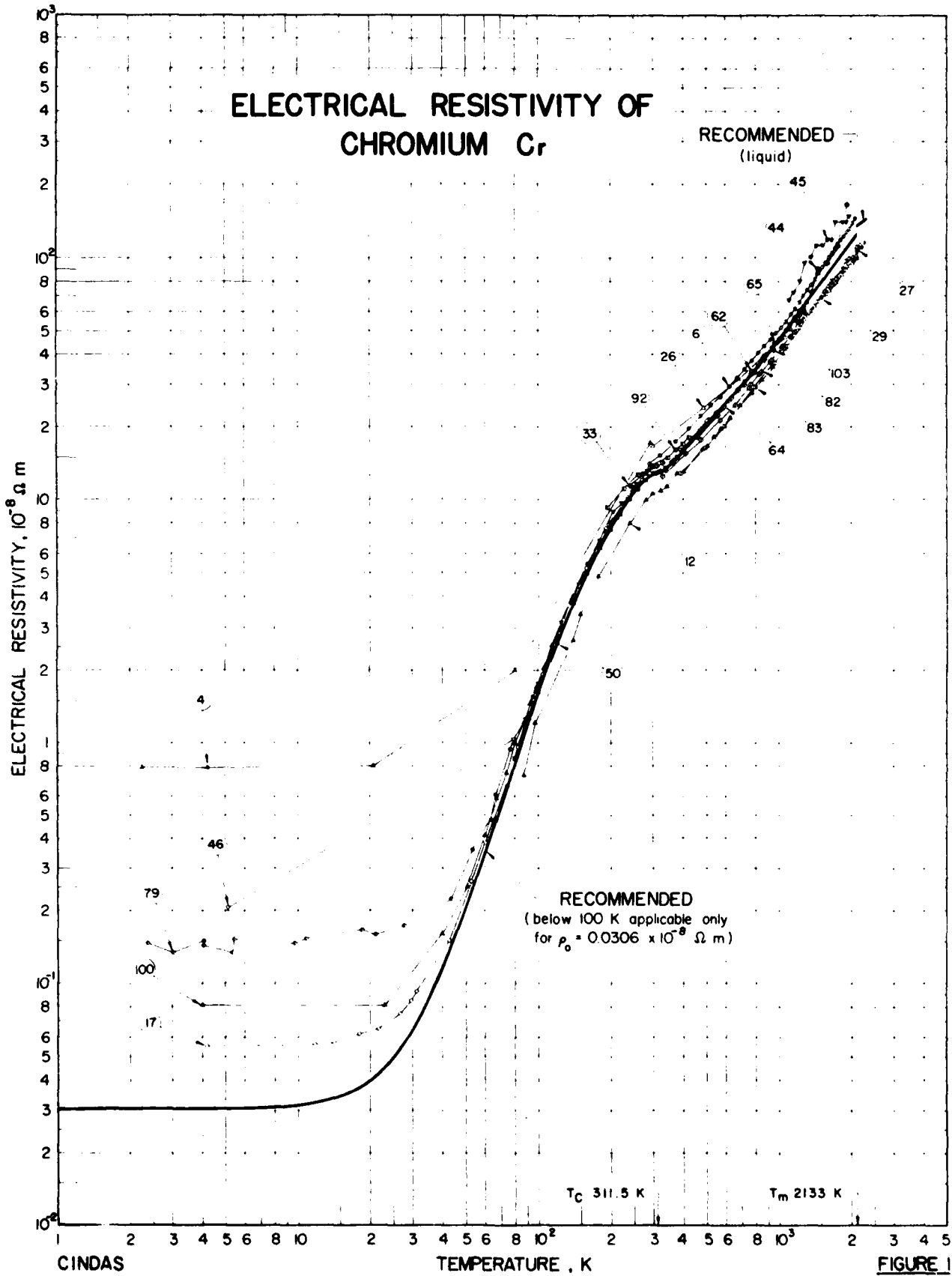
It should be emphasized that these expressions do not necessarily suggest any theoretical justification, and should be treated, most appropriately, as numerical aids only. It should also be understood that giving these expressions does not imply a recommendation for the temperature derivative of the electrical resistivity.

TABLE 1. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF CHROMIUM^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega m$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.0306	0.0306	316	12.785	12.786
4	0.0307	0.0307	320	12.905	12.907
7	0.0309	0.0309	350	13.888	13.891
10	0.0316	0.0316	400	15.84	15.86
15	0.0342	0.0342	500	20.05	20.08
20	0.0398	0.0398	600	24.60	24.66
25	0.0495	0.0495	700	29.43	29.53
30	0.0646	0.0645	800	34.48	34.64
35	0.0865	0.0864	900	39.75	39.98
40	0.117	0.117	1000	45.21	45.52
50	0.207	0.207	1100	50.89	51.30
60	0.349	0.349	1200	56.84	57.37
70	0.555	0.554	1300	63.09 ^b	63.78 ^b
80	0.838	0.837	1400	69.72 ^b	70.56 ^b
90	1.212	1.21	1500	76.16 ^b	77.18 ^b
100	1.64	1.64	1600	82.44 ^b	83.68 ^b
150	4.36	4.36	1700	88.58 ^b	90.05 ^b
200	7.57	7.57	1800	94.59 ^b	96.33 ^b
250	10.57	10.57	1900	100.5 ^b	102.5 ^b
273	11.69	11.69	2000	106.3 ^b	108.7 ^b
293	12.45	12.45	2100	112.1 ^b	114.8 ^b
300	12.650	12.650	2133	114.0 ^b (s)	116.8 ^b (s)
306	12.760	12.761	2133	121.4 ^b (l)	124.4 ^b (l)
308	12.779	12.780	2200	124.8 ^b	127.8 ^b
309	12.779	12.780	2300	129.8 ^b	132.8 ^b
310	12.769	12.770			
311	12.739	12.740			
311.7	12.670	12.671			
312	12.673	12.674			
314	12.728	12.729			

^a The values are for chromium of purity 99.98% or higher, but those below 100 K are applicable only to chromium having a residual resistivity of $0.0306 \times 10^{-8} \Omega m$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation.

^b Provisional value.



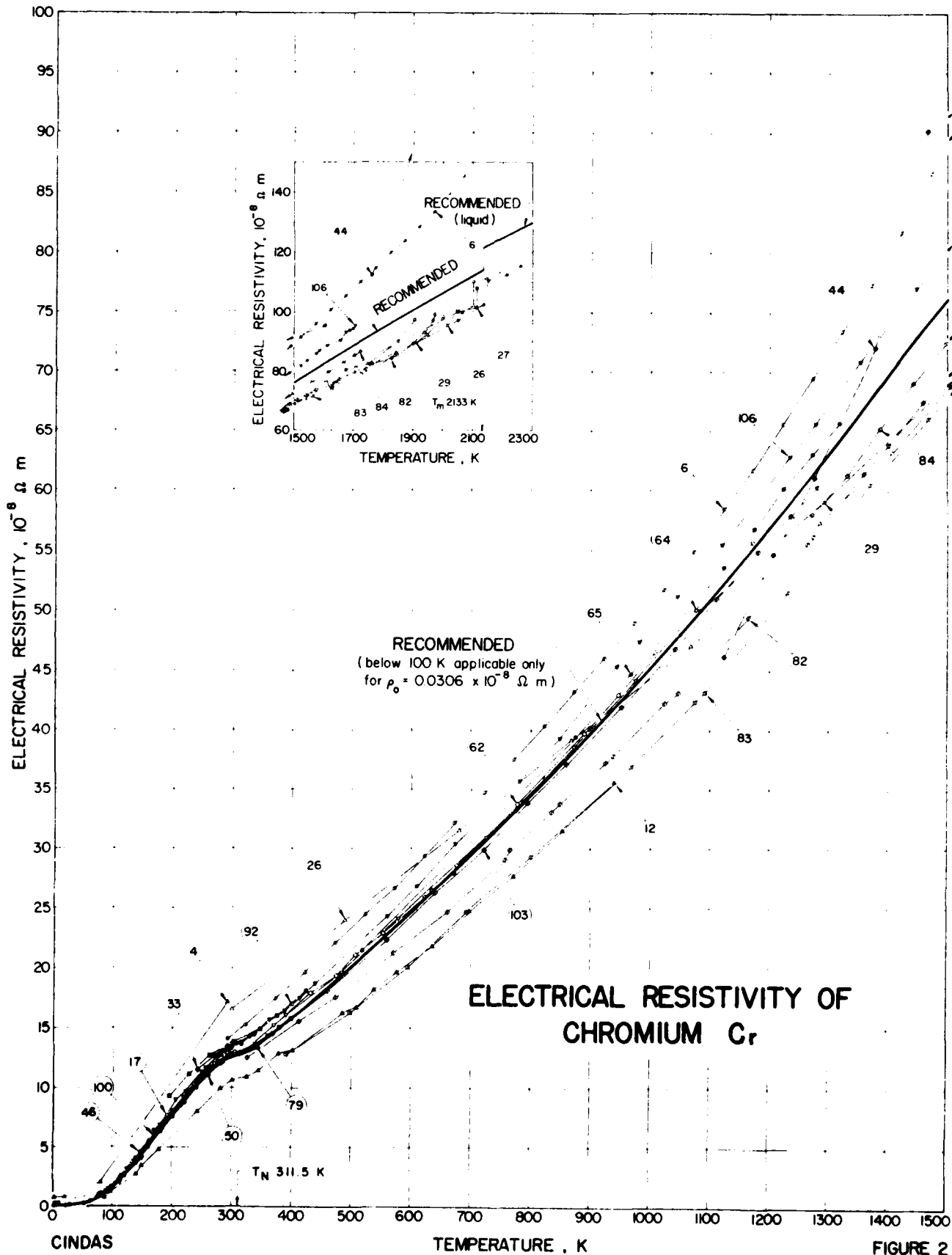


TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1*	49	McLennan, J.C. and Niven, C.D.	1927	B	2.4-290	Unaged-I	Trace of Cu; strip sample obtained from General Electric Co. of England; electrolytically deposited sheet ground down to 3.81 cm (1.5 in.) long and 0.475 cm (0.187 in.) thick; unannealed; data uncorrected for thermal expansion.
2*	49	McLennan, J.C. and Niven, C.D.	1927	B	2.2-290	Unaged-II	Trace of Cu; strip sample; material obtained from General Electric Co. of England; electrolytically deposited sheet dissolved in acid with aid of electrical potential; unannealed; data uncorrected for thermal expansion.
3*	49	McLennan, J.C. and Niven, C.D.	1927	B	20.6-292	Aged-I	Similar to the above specimen; annealed for 1 h at comparatively low temperature, cooled to room temperature, annealed for 2 h at much higher temperature; data uncorrected for thermal expansion.
4	50	McLennan, J.C., Niven, C.D., and Wilhelm, J.O.	1928	B	2.3-293		The above specimen with measurements extended to lower temperatures (there is an apparent discrepancy between values at 80 K between data set and the above).
5*	48	Grube, G. and Knabe, R.	1936	A	373-1673		"Pure Cr."
6	48	Grube, G. and Knabe, R.	1936	A	293-2073		Electrolytic Cr, sintered in H ₂ atmosphere at 1673 K; density 6.95 x 10 ³ kg cm ⁻³ .
7*	48	Grube, G. and Knabe, R.	1936	A	1643-1973	B	Electrolytic Cr, remelted and measured in an Ar atmosphere.
8*	48	Grube, G. and Knabe, R.	1936	A	1643-1933	B	The above specimen measured with decreasing temperature.
9*	48	Grube, G. and Knabe, R.	1936	A	1553-1973	C	Electrolytical Cr, remelted and measured in an Ar atmosphere.
10*	48	Grube, G. and Knabe, R.	1936	A	1563-1953	C	The above specimen measured with decreasing temperature.
11*	51	Potter, H.H.	1941	V	87-1064		"99.99" pure; 1 cm long; annealed at 873 K.
12	51	Potter, H.H.	1941	V	87-941		Similar to the above; outgassed just below the melting point by electron bombardment.
13*	14	Harper, A.F.A., Kemp, W.R.G., Klemens, P.G., Tainsh, R.J., and White, G.K.	1957	G	4.2	Sample 1	99.998 pure; supplied by Dr. H.L. Wain of the Aeronautical Res. Lab., Commonwealth Dept. of Supply; 3 mm in diam and 8 cm long; cold worked; temperature 4.2 K assumed.
14*	14	Harper, A.F.A., et al.	1957	G	4.2	Sample 2	The above specimen annealed in vacuum at 1323 K for 4 h.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
15*	14	Harper, A.F.A., Kemp, W.R.G., Klemens, P.G., Taineh, R.J., and White, G.K.	1957	G	4.2	Sample 3	Similar to the above; partially recrystallized after cold worked.
16*	14	Harper, A.F.A., et al.	1957	G	4.2	Sample 4	The above specimen; annealed.
17	14	Harper, A.F.A., et al.	1957	G	4.2-320	Sample 5	Similar to the above; cold worked and fully recrystallized.
18*	52	De Morton, M.E.	1958		200-344		0.013 O ₂ , 0.0015 C, 0.0005 Al, Fe, N ₂ and Si each, 0.00027 H ₂ (3N1/100 gm), 0.0002 Cu, Mg, and Pb each, and 0.00001 Ag; metallic impurities spectrographically determined; 0.073 cm diam and 20.7 cm long; arc-melted electrolytic Cr ingot 1.5 in. diam; extruded to 0.5 in. diam; swaged to 0.2 in. diam; drawn at 573 K to 0.027 in. diam; given a 3% reduction at 423 K; total reduction 98%.
19*	52	De Morton, M.E.	1958		200-346		The above specimen annealed at 973 K for 15 min. under a vacuum of 2×10^{-4} mmHg; "completely recrystallized".
20*	53	Newmann, M.M. and Stevens, K.W.H.	1959	A	94-390		Spectrographically standardized chromium supplied by Johnson and Matthey Co.; contains dissolved oxygen; 2 cm long; machined; sealed in evacuated quartz tube and annealed at 1200 K for 1 month.
21*	54	Sabine, T.M. and Svenson, A.C.	1968	D	301-328		0.024 O ₂ and 0.015 N ₂ ; supplied by Aeronautic Res. Lab., Melfourne; hot extruded (1373 K); 0.5 in. in diam and 3 in. long; measured with an AC bridge at 50 Hz; error of measurement 1%.
22*	54	Sabine, T.M. and Svenson, A.C.	1968	D	291-328		Similar to the above, annealed at 1373 K.
23*	54	Sabine, T.M. and Svenson, A.C.	1968	D	289-328		Similar to the above, annealed at 1673 K; fine grained.
24*	54	Sabine, T.M. and Svenson, A.C.	1968	D	289-328		Similar to the above, annealed at 1673 K; course grained; grain diam 0.25 in. approximately.
25*	55	Hamaguchi, Y. and Kunitomi, N.	1964		295-810		99.9 pure; cut by diamond saw; measured in vacuum.
26	44	Baum, B.A., Gel'd, P.V., and Sachil'nikov, S.I.	1964	R	300-2194		99.98 pure.
27	45	Baum, B.A., Gel'd, P.V., and Sachil'nikov, S.I.	1963	R	1673-2113		>99.98 pure (specimen appears to be the same as the above); resistivity values at melting point (reported at 2113 K approximately) from text.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
28*	56	Pavars, I.A., Baum, B.A., and Gel'd, P.V.	1969	R	2013		99.98 pure, doubly refined electrolytic chromium; melted in a fusion induction furnace in an argon atmosphere of 500 mmHg.
29	46	Levin, E.S., Gel'd, P.V., and Ayushins, G.D.	1973	R	1084-2261		99.98 pure, doubly refined electrolytic.
30*	57	Levin, E.S.	1971	R	1973		99.98 pure, doubly refined.
31*	58	Fine, M.E., Greiner, E.S., and Ellis, W.C.	1951	A	78-401		99.8 pure; prepared by cold pressing sintered electrolytic powder compact; annealed at 1673 K in purified helium.
32*	58	Fine, M.E., et al.	1951	A	275-377		99.998 pure, from electrochemical analysis by E.K. Jaycox, no Fe or Ni detected; electroformed from an aqueous solution by R.A. Ehrhardt and G. Brittrich, using the method of Brennar, Burhead and Jennings, NBS J. of Res., 40, 31, (1948); vacuum annealed at 1273 K with specimen packed in chromium powder.
33	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-331		0.08 N and 0.03 O, others not detectable spectrographically; supplied by Prof. R. Street of Monash Univ., Melbourne; 0.5 mm wide, 0.5 mm thick and 58 mm long approximately; manufactured from ARL chromium ingot; ground and formed; annealed and recrystallized at 1523 K for 2 h; resistivity values calculated from reported $\rho(T)/\rho(273 K)$ and reported $\rho(273 K) = 12.7 \times 10^{-9} \Omega m$.
34*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-322		The above measured with decreasing temperature.
35*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-330		The above measured under a pressure of 0.98 kbar.
36*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-326		The above with decreasing temperature.
37*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-331		The above measured under a pressure of 1.96 kbar.
38*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-329		The above with decreasing temperature.
39*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-331		The above measured under a pressure of 2.94 kbar.
40*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-326		The above measured under a pressure of 5.30 kbar.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
41*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-326		The above with decreasing temperature.
42*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-325		The above measured under a pressure of 7.16 kbar.
43*	38	Mitsui, T. and Tomizuka, C.T.	1965	A	243-326		The above measured under a pressure of 7.85 kbar.
44	59	Barykin, B.M., Gordon, V.G., Levinov, B.M., Rekov, A.I., and Spiridonov, E.G.	1974	V	1124-1938		99.9 ⁺ pure chromium powder; average particle size 5-10 μ m; pressed and sintered at 1973 K; density 7.0 g cm ⁻³ ; porosity 1.5%.
45*	59	Barykin, B.M., et al.		V	1120-1976		Similar to the above.
46	22	Chiu, C.H., Jericho, N.M., and March, R.H.	1971	V	5.1-313		0.0012 O ₂ and Fe each, 0.0010 Si, 0.00009 N ₂ , 0.00003 Al, Ca and Ni each, 0.00002 H ₂ and 0.00001 Cu, Mg, and Mn each (at.%); "10 ⁻² cm ² diameter," and 5 cm long; spark machined; measurement error 2%.
47*	10	Moore, J.P., William, R.K., and McElroy, D.L.	1968	A	80-400	CrB	99.98 ⁺ pure; 0.0060 C, 0.0028 N, <0.0020 Ga, 0.0009 H, 0.0008 Mo, 0.0006 O, 0.0005 Cu, S, and Si each, 0.0004 K, 0.0003 Co, Fe, and U each, <0.0003 Pt, <0.0002 Sr and Zn each, 0.0001 B, <0.0001 Hg, Pd, Rn, Ru, Sb, Te, Tl, U, W, and Zr each, <0.00008 Cu and Pb each, 0.00006 Ca, <0.00005 Ag and Ba each, <0.00004 Bi, 0.00003 Mn, 0.00002 Nb, <0.00002 Ge, In, Hg and Na each, 0.00001 Ti and <0.00001 Ag, Li, and P each (at.%); content of C by combustion analysis, H, N, and O by vacuum fusion analysis, and of others by semi-quantitative spectrographic analysis; material supplied by Girovalloy Corp.; consolidated into a disc-shaped ingot and drop-cast into a rod of 1.6 cm diam and 15 cm long; machined to 0.96 cm in diam and 7.7 cm in length; density 7.15 g cm ⁻³ ; grain diam 440 to 840 μ m; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 58$; measurement error $\pm 0.38\%$; smoothed values from table.
48*	10	Moore, J.P., et al.	1968	A	310-315	CrB	The above in the vicinity of the Néel temperature.
49*	10	Moore, J.P., et al.	1968	A	307-320	CrB	The above measured with a temperature gradient of 0.014 k m ⁻¹ .

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
50*	10	Moore, J.P., William, R.K., and McElroy, D.L.	1968	A	80-400	CrA	99.98 ⁺ pure; 0.0070 C, 0.0030 Fe, <0.0020 Au and Mg each, 0.0015 S, 0.0014 O, 0.0010 Mn, <0.0010 Pt, 0.0006 Ca, 0.0005 H and Si each, 0.0003 B, N and U each, 0.0002 Ba, Cu, and Pb each, <0.0002 Ta, 0.0001 Ag and As each, <0.0001 Hg, P, Pd, Ru, Sb, Sn, Te, Ti, U, W, and Zr each, <0.00008 Cd, 0.00005 Ti, <0.00005 Mo, Ni, and Sr each, 0.00004 K and Nb each, <0.00004 Bi, <0.00003 Co, <0.00002 Ge, In, and Na each, and <0.00001 Li and Rn each (at.%); same methods of analysis as above; crystals supplied by Chromalloy Corp.; sealed in vacuum jacket after cleaned and compacted; hot extruded to a rod of 1.6 cm diam and 60 cm long (by BMI); machined to a rod of 0.96 cm diam and 7.7 cm long; density 7.19 g cm ⁻³ ; average grain diam 63 μ m; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 280$; measurement error $\pm 0.38\%$; smoothed values from table.
51*	10	Moore, J.P., et al.	1968	A	308-320	CrA	The above in the vicinity of the Néel temperature.
52*	10	Moore, J.P., et al.	1968	A	307-320	CrA	The above measured with a temperature gradient of 0.014 K m ⁻¹ .
53*	9	Laubitz, M.J. and Matsumura, T.	1970	D	300-319		The above specimen on loan from Oak Ridge National Lab.; annealed at 1100 K for 4 days and 1200 K for 1 day; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 380$; specimen immersed in oil bath maintained to within 1×10^{-3} K of desired temperature; measured with an ac (7 Hz) bridge with an absolute accuracy of 0.2%, and a precision of 0.03%; smoothed values from curve, which is reported to be based on 68 data points and to deviate from the measured values by amounts less than the precision of the measurements.
54*	9	Laubitz, M.J. and Matsumura, T.	1970	D	301-318		The above measured in a thermal conductivity apparatus, with temperature difference along specimen less than 0.01 K.
55*	9	Laubitz, M.J. and Matsumura, T.	1970	D	301		The above after being cooled from 320 K.
56*	9	Laubitz, M.J. and Matsumura, T.	1970	D	301-318		The above specimen measured during a thermal conductivity experiment with temperature difference along specimen less than 0.25 K.
57*	9	Laubitz, M.J. and Matsumura, T.	1970	D	301		The above after being cooled from 320 K.
58*	9	Laubitz, M.J. and Matsumura, T.	1970	D	301-315		The above specimen measured during a thermal conductivity experiment with temperature difference along specimen less than 0.5 K.
59*	9	Laubitz, M.J. and Matsumura, T.	1970	D	318		The above after being cooled from 320 K.
60*	9	Laubitz, M.J. and Matsumura, T.	1970	D	302-319		The above specimen measured during a thermal conductivity experiment with temperature difference along specimen less than 1.0 K.
61*	9	Laubitz, M.J. and Matsumura, T.	1970	D	302		The above after being cooled from 320 K.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
62	34	Moore, J.P., Williams, R.K., and Graves, R.S.	1977	A	293-1008	CrA'	Same sample material as the above; average grain diam 60 μm ; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 280 - 380$, depending on thermal cycling; measurement error $\pm 0.38\%$; corrected for thermal expansion.
63*	34	Moore, J.P., et al.	1977	A	300-320	CrA'	The above in the vicinity of the Néel temperature; corrected for thermal expansion.
64	34	Moore, J.P., et al.	1977	A	371-1172	CrA	From the same stock as the above; thermally cycled during brazing of specimen to heater and nickel heat sink; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 380$; corrected for thermal expansion.
65	34	Moore, J.P., et al.	1977	A	518-1319	CrB	Same specimen material as for Data Set 47; density 7.19 g cm^{-3} ; grain size 400 to 800 μm ; arc-cast, $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 58$; specimen found to have a hairline casting void along its axis, causing approximately 1% change in the measured values; corrected for thermal expansion.
66*	60	Söchtig, H.	1940		79-319	CrI	From Dr. Kroll of Luxemburg; cut from a rolled plate; approximate dimensions: 0.12 cm thick, 0.21 cm wide and 1.37 cm long; resistivity values calculated from reported resistances and reported $\rho(273 \text{ K}) = 19.7 \times 10^{-8} \Omega\text{m}$.
67*	60	Söchtig, H.	1940		20.8-373	CrII	Electrolytic chromium; approximate dimensions: 0.21 cm wide, 0.23 cm thick, and 0.58 cm long; resistivity values calculated from reported resistances and reported $\rho(273 \text{ K}) = 21.1 \times 10^{-8} \Omega\text{m}$.
68*	61	Meaden, G.T., Rao, K.V., and Loo, H.Y.	1969		278-323		99.999 pure; 0.0010 C, 0.0009 O ₂ , 0.0003 Ca, 0.0002 Fe, 0.0001 Al, Cu, and Mg each, and 0.0008 H ₂ ; residual resistance ratio 178.
69*	11	Meaden, G.T. and Sze, N.H.	1969		301-317		99.999 pure; 0.0010 C, 0.0008 O, 0.0003 Ca and N ₂ each, 0.0002 Fe, 0.0001 Al, Cu, and Mg each, and 0.0008 H ₂ ; cast iodide chromium machined to 3.8 mm in diam and 65 mm in length; unannealed; $\rho(295 \text{ K})/\rho(4.2 \text{ K}) = 178$; measured with increasing temperature (specimen is apparently the same as the above).
70*	11	Meaden, G.T. and Sze, N.H.	1969		300-318		The above specimen measured with decreasing temperature.
71*	12	Meaden, G.T., Rao, K.V., and Loo, H.Y.	1969		100-145		99.999 pure; 0.0010 C, 0.0008 O ₂ , 0.0003 Ca, 0.0002 Fe, 0.0001 Al, Cu, and Mg each; residual resistivity ratio 178 (specimen is apparently the same as the above).
72*	13	Meaden, G.T. and Sze, N.H.	1969		101-140	Cr-0	99.999 pure; unannealed; grain diam 0.25 mm; residual resistivity ratio 178 (specimen is apparently the same as the above).
73*	13	Meaden, G.T. and Sze, N.H.	1969		116-124	Cr-0	The above in the vicinity of the spin-flip transition.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
74*	13	Meaden, G.T. and Sze, N.H.	1969		101-140	Cr-I	Similar to the above except annealed at 1473 K for 1 h.
75*	13	Meaden, G.T. and Sze, N.H.	1969		116-125	Cr-I	The above in the vicinity of the spin-flip transition.
76*	13	Meaden, G.T. and Sze, N.H.	1969		101-140	Cr-50	Similar to the above except annealed for 75 h at temperatures above 1273 K, 50 h of which is at 1473 K; grain diam 2 to 4 μ m; grain orientation is random as determined by x-ray Laue photography; residual resistivity ratio 295.
77*	13	Meaden, G.T. and Sze, N.H.	1969		117-125	Cr-50	The above in the vicinity of the spin-flip transition (discrepancies between these two data sets are due to graph-reading errors).
78*	28	Rao, K.V., and Tee, K.T.	1970		202-329		Pure chromium sample; annealed in vacuum (10^{-6} Torr) for 75 h above 1273 K of which 50 h is at 1473 K; grain diam 2 to 4 μ m; residual resistivity ratio 295 (specimen is apparently the same as the above).
79	15	Coff, J.F.	1970	A	2.4-343	CrII	99.92 pure; 0.005 Fe, 0.004 Mn, 0.003 Cu, 0.002 Mg, and balance mostly S, P, Ni and Mn; electrolytic; melted with argon arc, cast in oxygen-free copper boat; annealed twice at 1173 K for 24 h; ground to approximate dimensions of 4 mm wide, 4 mm thick and 50 mm long; polycrystalline; $\rho(297\text{ K})/\rho(4\text{ K}) = 88$; average residual resistivity $0.145 \times 10^{-6} \Omega\text{m}$; measurement error 1%.
80*	15, 16	Coff, J.F.	1970 1968	A	1.2-286	CrI	Similar to the above except $\rho(297\text{ K})/\rho(4\text{ K}) = 72$ and average residual resistivity (1.2 K to 12.9 K) = $0.1834 \times 10^{-6} \Omega\text{m}$.
81*	62	Moore, J.P., Williams, R.K., and McElroy, D.L.	1969	A	90-360		The above specimen; $\rho(296\text{ K})/\rho(4.2\text{ K}) = 70.5$; smoothed values from table.
82	43	Anderson, J.M., Stewart, A.D., and Ramsay, I.	1970	A	330-1973		Single crystal produced by the iodide process; supplied by Material Research Corp.; 0.0025 interstitial impurities and 0.0015 substitutional impurities, quoted by manufacturers; 1 mm thick, 1 mm wide and 5 mm long; cut by a combination of diamond saw, spark planning and electropolishing techniques; measurement done in a high purity hydrogen atmosphere; reported errors in resistivity value 0.002%, and in temperature 0.2%; values corrected for thermal expansion using expansion data of B.N. Vasyutinski, G.N. Katurazov and G.N. Finkel, Soviet Phys.-Phys. Met. Metallog., 12, 141 (1961); rapid increase of resistivity above 1750 K was reported to be due to evaporation of material.
83	43	Anderson, J.M., et al.	1970	A	391-1905		Similar to the above, except supplied by Koch-Light Lab. and containing 0.0021 interstitial and 0.0005 substitutional impurities.
84	43	Anderson, J.M., et al.	1970	A	1272-1724		"Crystal 3"; no specimen detail reported; measured with decreasing temperature.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
85*	43	Anderson, J.M., Stewart, A.D., and Ramsay, I.	1970	A	1482-1740		The above specimen; measured with increasing temperature.
86*	43	Anderson, J.M., et al.	1970	A	1469-1693		The above specimen; measured with decreasing temperature again.
87	36	Anderson, J.M., Stewart, A.D., and Rameay, I.	1972	A	285-324		0.0012 O ₂ and Fe each, 0.0010 Si, 0.00009 N ₂ , 0.00002 H ₂ , <0.000001 C and 0.0002 others (at.%); single crystal prepared from Koch-Light crystalline, cut with diamond saw, spark planned and electropolished; 1 mm thick, 1 mm wide and 7 mm long; specimen axis parallel to the [100] direction; annealed at 1970 K for 3 h in hydrogen; T _N 311.5 K.
88*	36	Anderson, J.M., et al.	1972	A	282-324		The above specimen deformed at a strain rate of $5 \times 10^{-5} s^{-1}$; T _N 309 K.
89*	36	Anderson, J.M., et al.	1972	A	77		0.03 O ₂ , 0.02 H ₂ , 0.0030 C and 0.0010 N ₂ (at.%); polycrystal from vacuum melted ingot; average grain size 1 mm.
90*	63	Marcinkowski, M.J. and Lipsitt, H.A.	1961		199-414		Pure; plastically deformed: 96% reduction in area at 623 K.
91*	63	Marcinkowski, M.J. and Lipsitt, H.A.	1961		283-323		The above specimen in the vicinity of the Néel temperature.
92	63	Marcinkowski, M.J. and Lipsitt, H.A.	1961		196-425		The above specimen recrystallized by annealing at 1323 K for 1/2 h.
93*	63	Marcinkowski, M.J. and Lipsitt, H.A.	1961		284-323		The above specimen in the vicinity of the Néel temperature.
94*	17	Arajs, Calvin, R.V., and Marcinkowski, M.J.	1962	A	298-315		0.055 O ₂ and <0.001 N ₂ ; single crystal; 0.235 cm thick, 0.254 cm wide and 0.900 cm long; long axis of specimen aligned to within 1 degree of the [100] direction; measured at temperatures in the order: 297.4, 305.3, 306.1, 307.2, 313.5, 314.5, 312.4, 311.4, 310.3, 308.8, 309.5, and 309.6 K.
95*	17, 18	Arajs, S., et al.	1962	A	310-328		The above specimen measured after being left overnight at 309.6 K; measurements between 309.6 K and 312.7 K were done over a period of 8 h, changing temperature slowly; the rest of the data points were obtained the following day.
96*	17, 18	Arajs, S., et al.	1962	A	312		The above specimen, cooled; "leaving the crystal at this temperature overnight did not change the value of the resistivity".
97*	17, 18	Arajs, S., et al.	1962	A	78		The above specimen; heated to 373 K and cooled rapidly to 78 K, and kept at 78 K for two days.
98*	17, 18	Arajs, S., et al.	1962	A	78-140		The above specimen.
99*	17, 18	Arajs, S., et al.	1962	A	4.2-330		The above specimen, cooled to 78 K, left overnight and cooled to 4.2 K; measurements in the temperature ranges, 4 to 115 K, 130 to 155 K, 155 to 215 K, and 215 to 295 K were done in successive days.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	P. f. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
100	19	Arajs, S. and Dunwyre, G.R.	1965	A	4-297		0.0013 O, 0.0010 C and Si each, 0.0001 Ca, Mg, N, and Ni each, and 0.00005 H, others not detected; contents of O, H, and N determined by vacuum fusion, of C by combustion, and the rest by spectroscopic methods; arc-melted polycrystalline ingot 0.7 x 1.6 x 9.0 cm supplied by Chromalloy Corp.; cut by surface grinding machine to 0.478 ± 0.001 cm thick, 0.476 ± 0.001 cm wide and 0.5 cm long; $\rho(273 \text{ K})/\rho(4.2 \text{ K}) = 180$; TN 313.0 ± 0.2 K; duration of measurement 8 h; specimen was left at 290.5 K for 15 h without detectable change in resistivity (due to small size of graph, data points below 80 K are selected values; hence this curve does not represent all the measurements reported by the authors).
101*	19	Arajs, S. and Dunwyre, G.R.	1965	A	300-318		The above specimen in the vicinity of the Néel temperature; duration of measurement 8 h.
102*	20	Arajs, S., DeYoung, T.F., and Anderson, E.E.	1970	A	9-1035		From the same stock as the above.
103	24	Cox, J.E. and Lucke, W.H.	1967	+	299-1281		99.999 iodide chromium; melted 15 to 18 times in 2/3 atmosphere of ionization grade; 99.999 pure argon; spark eroded to cylindrical form; centerless-ground to 1/8 in. diam and 2 1/2 in. long; wrapped in molybdenum foil and annealed at 1523 K in 1/2 atmosphere of ionization grade helium; water quenched and etched; measured by a method by Dauphinee, T.M. and Mooser, E., Rev. Sci. Instr., 26, 660, 1955.
104*	25	Taylor, M.A.	1962		77-359		99.99 pure; 0.01 O, ~0.0001 N, C, S and Sn each, and <0.0001 others; electrolytic; supplied by the Aeronautical Res. Lab., Melbourne; smoothed values from curve.
105*	64	Taylor, M.A. and Smith, C.H.L.	1962	A	20-273		99.99 pure; supplied by the Aeronautical Res. Lab., Melbourne; 1 mm wide, 1 mm thick, and 10 mm long; cut with carborundum slitting wheel; annealed at 1073 K for 50 h in vacuum; measurement error 1%.
106*	47	Powell, R.W. and Tye, R.P.	1956	A	94-1707		99.985 pure; <0.01 N and <0.005 O (as Cr ₂ O ₃); electrodeposited chromium 1.28 cm O.D., 0.63 cm I.D. and 18.05 cm long; prepared from chromium flakes supplied by Johnson Matthey Co.; enclosed in alumina tube and heat-treated at 443, 486, 678, 818, 1133, 1327, and 1683 K, with the five last treatments done in vacuum; reductions of 0.010 cm in O.D. and I.D. are observed after final treatment; initial density 6.975 x 10 ³ kg m ⁻³ , after final treatment 7.15 x 10 ³ kg m ⁻³ .
107*	47	Powell, R.W. and Tye, R.P.	1956	A	273-333		The above in the vicinity of the Néel temperature.
108*	26	Muhela, J. and Müller, J.	1964	+	100-339		Cylindrical specimen 1 mm in diam and 50 cm long; measured by a compensation method.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
109*	23	DeVries, G.	1959		165-356		Prepared by Prof. Fast, Eindhoven.
110*	65	Zinov'ev, V.E., Krentsis, R.P., and Gel'd, P.V.	1969		300-1800		0.05 total impurity; residual resistance ratio = 65; values calculated from reported equations: $\rho = 4.8 \times 10^{-2} T$, $T < 800$ K and $\rho = 4.8 \times 10^{-2} T + 3.1 \times 10^{-6} (T-800)^2$, $T > 800$ K.
111*	66	Maystrenko, L.G. and Polovov, V.M.	1977	A	89-400		Polycrystalline; appropriate dimensions 1 mm thick, 1 mm wide, and 10-15 cm long; annealed at 1273 K for 24 h in helium; furnace cooled; measurement error 15%.
112	35	Stebler, B.	1970	A	296-344		99.996 pure, 0.0002 N ₂ ; single crystal; 7 mm thick, 8 mm wide and 25 mm long; specimen axis ~ 8 degrees of arc from the [100] direction; measured with increasing temperature; resistivity values calculated from reported $\Delta\rho/\rho_0$, where ρ_0 is the resistivity at 273.2 K, taken to be $11.687 \times 10^{-6} \Omega\text{m}$.
113	33	Stebler, B.	1970	A	287-350		The above specimen measured with decreasing temperature.
114	33	Stebler, B.	1970	A	305-316		The above specimen in the vicinity of the Néel temperature; measured with increasing temperature.
115	33	Stebler, B.	1970	A	305-316		The above specimen in the vicinity of the Néel temperature; measured with decreasing temperature.
116	37	Trego, A.L. and Mackintosh, A.R.	1968		277-320		0.0072 C, 0.0030 N, 0.0016 O and 0.0001 H (at.%); Iodide Cr supplied by Chromalloy Corp.; arc melted and arc zone melted; single crystal 2 mm square cross section and 1.5 cm long, cut by spark erosion technique from 0.5 in. diam and 6 in. long ingot; electropolished in orthophosphoric acid; annealed in vacuum at 1273 K for 50 h; sample length parallel to crystal [001] axis; resistivity values calculated from reported resistance R(T) and R(320 K), with $\rho(320 \text{ K})$ taken to be $12.906 \times 10^{-6} \Omega\text{m}$.
117	37	Trego, A.L. and Mackintosh, A.R.	1968		272-310		The above specimen measured after cooled through the Néel temperature in a longitudinal magnetic field of 55 kG.
118	37	Trego, A.L. and Mackintosh, A.R.	1968		275-311		The above specimen but measured after cooled through the Néel temperature in a transverse magnetic field (either in [010] or [100] direction) of 55 kG.
119	37	Trego, A.L. and Mackintosh, A.R.	1968		270-310		The above specimen, but measured after cooled through the Néel temperature in a longitudinal magnetic field of 40.5 kG; resistivity value calculated from reported $\Delta\rho/\rho(TN)$, with $\rho(TN)$ taken to be $12.70 \times 10^{-6} \Omega\text{m}$.
120	37	Trego, A.L. and Mackintosh, A.R.	1968		270-313		The above specimen except strength of magnetic field is 28 kG.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
121	37	Trego, A.L. and Mackintosh, A.R.	1968		275-307		The above specimen except strength of magnetic field is 15.6 kG.
122	37	Trego, A.L. and Mackintosh, A.R.	1968		285-310		The above specimen but cooled through the Néel temperature in a transverse magnetic field of 47 kG.
123	37	Trego, A.L. and Mackintosh, A.R.	1968		274-308		The above specimen except strength of magnetic field is 40.5 kG.
1-4	37	Trego, A.L. and Mackintosh, A.R.	1968		274-310		The above specimen except strength of magnetic field is 28 kG.
125	37	Trego, A.L. and Mackintosh, A.R.	1968		275-307		The above specimen except strength of magnetic field is 15.6 kG.
126	67	Akiba, C. and Mitsui, T.	1972	A	294-320		99.997 pure; iodide chromium supplied by A.D. Mackay Inc.; 0.77 mm thick, 0.7 mm wide and 10 mm long; single crystal, spark cut and chemically polished; specimen axis within 13 degrees from the c-axis; <100> direction; $R(293\text{ K})/R(4.2\text{ K}) = 630$ without magnetic field cooling; specimen in single magnetic domain state (single Q) prepared by heating to 329 K, applying a longitudinal magnetic field of 74 kG, then cooled to 273 K, and magnetic field reduced to zero; resistivity values calculated from reported resistance, and with $\rho(320\text{ K})$ taken to be $12.906 \times 10^{-6} \Omega\text{m}$.
127	67	Akiba, C. and Mitsui, T.	1972	A	301-320		The above specimen except magnetically cooled with a transverse field of 74 kG.
128	73	Semenenko, E.E. and Tutov, V.I.	1969		1.5-6.8		Monocrystalline whisker specimen, 0.10-0.12 mm in diam and ~ 8 mm long; $R(4.2\text{ K})/R(300\text{ K}) = 8 \times 10^{-3}$; resistivity values calculated from reported resistance and $\rho(300\text{ K})$, taken to be $12.650 \times 10^{-6} \Omega\text{m}$.
129	68	Semenenko, E.E. and Tutov, V.I.	1972		4.7-329		Primary impurity is iron; only resistance as a function of temperature reported; resistivity calculated by assuming $\rho(328.6\text{ K}) = 13.177 \times 10^{-6} \Omega\text{m}$.
130	68	Semenenko, E.E. and Tutov, V.I.	1972		280-330		The above specimen in the vicinity of the Néel temperature.
131	68, 74	Semenenko, E.E. and Tutov, V.I.	1972, 1966		4.4-16.1		Similar to the above; $R(1.5\text{ K})/R(300\text{ K}) = 8 \times 10^{-3}$; resistivity values calculated from reported $R/R(300\text{ K})$ with $\rho(300\text{ K})$ taken to be $12.650 \times 10^{-6} \Omega\text{m}$.
132	68	Semenenko, E.E. and Tutov, V.I.	1972		1.7-14.8		Similar to the above; $R(1.5\text{ K})/R(300\text{ K}) = 7.6 \times 10^{-3}$.
133	68	Semenenko, E.E. and Tutov, V.I.	1972		2.0-7.1		Similar to the above; $R(1.5\text{ K})/R(300\text{ K}) = 6.8 \times 10^{-3}$.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
134	40	Kostina, T.I., Ekonomova, L.N., and Kondorskii, E.I.	1970		103-130		Single crystal cut by electric spark method; 0.1 mm thick, 0.15 mm wide and 4 mm long; R(293 K)/R(4.2 K) = 500; TN 311 + 2 K; longitudinal axis of sample parallel to the [110] direction; resistivity values calculated from reported R/R(77 K), with $\rho(77 \text{ K})$ taken to be $0.737 \times 10^{-8} \Omega \text{ m}$.
135	40	Kostina, T.I., et al.	1970		285-326		The above specimen.
136	40	Kostina, T.I., et al.	1970		106-131		The above specimen after magnetically annealed by an external field in the [100] direction of magnitude 34 kOe.
137	40	Kostina, T.I., et al.	1970		103-119		Similar to the above except external field is in the [110] direction.
138	40	Kostina, T.I., et al.	1970		121-153		Similar to the specimen of data set 134, except longitudinal axis of sample is parallel to the [100] direction.
139	40	Kostina, T.I., et al.	1970		287-329		The above specimen.
140	40	Kostina, T.I., et al.	1970		101-125		The above specimen after magnetically annealed by an external field in the [100] direction of magnitude 34 kOe.
141	40	Kostina, T.I., et al.	1970		111-133		Similar to the above except magnetic field is in the [110] direction.
142	41	Muir, W.B. and Ström-Olsen, J.O.	1971		76-326		Single crystal, 1 mm thick, 1 mm wide and 7 mm long; cut by spark erosion technique from vapor transport grown polycrystal ingot containing many large single crystals, supplied by Battelle Memorial Institute; annealed at 1470 K in argon for 50 h; strain free (found by x-ray technique; specimen axis parallel to the <100> direction; R(300 K)/R(4.2 K) = 350; in single magnetic domain state by cooling from 343 to 273 K in a longitudinal magnetic field of 60 kOe; measuring current parallel to the spin density wave vector Q; resistivity values calculated from reported R/R(320 K) with $\rho(320 \text{ K})$ taken to be $12.906 \times 10^{-8} \Omega \text{ m}$.
143	41	Muir, W.B. and Ström-Olsen, J.O.	1971		76-328		The above specimen in the multidomain state.
144	41	Muir, W.B. and Ström-Olsen, J.O.	1971		296-320		The above specimen, single domain state, in the vicinity of the Néel temperature; measuring current parallel to Q.
145	41	Muir, W.B. and Ström-Olsen, J.O.	1971		298-318		The above specimen, in the multidomain state after the above measurement.
146	41	Muir, W.B. and Ström-Olsen, J.O.	1971		300-319		The above specimen, magnetically cooled to the single domain state again after the above measurement; measuring current again parallel to Q.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
147	41	Muir, W.B. and Ström-Olsen, J.O.	1971		295-312		The above specimen, in the multidomain state after the above measurement.
148	41	Muir, W.B. and Ström-Olsen, J.O.	1971		299-320		The above specimen, in single domain state after cooling from 343 to 273 K in a transverse magnetic field of 60 kOe; measuring current perpendicular to Q.
149	41	Muir, W.B. and Ström-Olsen, J.O.	1971		299-310		The above in the multidomain state.
150	69	Borovik, E.S. and Volotakaya, V.G.	1959		2.4-78		Vacuum distilled chromium; needle shaped; "0.35 mm across" and 8 mm long; "appear to be single crystal;" resistivity value calculated from reported $R/R(273\text{ K})$, with $\rho(273\text{ K})$ taken to be $11.687 \times 10^{-6} \Omega\text{m}$.
151	70	McWhan, D.B. and Rice, T.M.	1967		4.3-232	Sample 2	Battelle Iochrome; single crystal; $R(298\text{ K})/R(4.2\text{ K}) = 140$; measured under a pressure of 26.5 kbar; AgCl used as pressure transmitting medium; resistivity values calculated from reported $R/R(1\text{ atm}, 298\text{ K})$, with $\rho(1\text{ atm}, 298\text{ K})$ taken to be $12.319 \times 10^{-6} \Omega\text{m}$.
152	70	McWhan, D.B. and Rice, T.M.	1967		60-223	Sample 2	The above measured under a pressure of 45.7 kbar.
153	70	McWhan, D.B. and Rice, T.M.	1967		32.5-223	Sample 2	The above measured under a pressure of 64.9 kbar; data points below 30 K cannot be resolved from graph, and are not reported here.
154	70	McWhan, D.B. and Rice, T.M.	1967		188-276	Sample 2	From the same ingot as the above specimen; $R(298\text{ K})/R(4.2\text{ K}) = 275$ and 165 before and after pressure experiment respectively; measured under a pressure of 26.3 kbar; AgCl used as pressure transmitting medium; resistivity values calculated by the same method as for the above specimen.
155	70	McWhan, D.B. and Rice, T.M.	1967		59-262	Sample 2	The above measured under a pressure of 45.9 kbar.
156	70	McWhan, D.B. and Rice, T.M.	1967		70-262	Sample 2	The above measured under a pressure of 65.9 kbar.
157	27	Suzuki, T.	1966		216-331		99.99 pure; electrolytic, supplied by Johnson and Matthey Co.; 0.5 mm thick, 0.5 mm wide and 20 mm long; degassed at 773 K; electropolished in a solution of 90% acetic acid and 10% perchloric acid; resistivity values calculated from reported $[\rho(T) - \rho(300\text{ K})]/\rho(300\text{ K})$, with $\rho(300\text{ K})$ taken to be $12.650 \times 10^{-6} \Omega\text{m}$.
158	32	Ishikawa, Y., Ikeda, S., and Akiba, C.	1975		299-320		99.997 pure; iodide chromium from A.D. Mackay Inc.; single crystal, 0.7 mm thick, 0.7 mm wide and 10 mm long; specimen axis along [100] direction; resistivity value calculated from reported resistance values and $\rho(370\text{ K})$, taken to be $12.906 \times 10^{-6} \Omega\text{m}$.

* Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
159	71	Bridgman, P.W.	1933		193-347		Swaged rod supplied by P.H. Bracc, Westinghouse Electric and Manufacturing Co., spectroscopic examination by Martin Graban show only a "doubtful trace of magnesium"; resistivity values calculated from reported $R(T)/R(273\text{ K})$ with $\rho(273\text{ K})$ taken to be $11.687 \times 10^{-8} \Omega\text{m}$.
160	31	Rapp, Ö., Benediktsson, G., Aström, H.O., Arajs, S., and Rao, K.V.	1978	A	300.2-315.0	Cr(1)	Specimen material same as for Data Set 100; approximate dimensions 0.1 mm thick, 1 mm wide and 20 mm long; spark cut from arc-melted ingot, etched in HCl, placed in silica tube, flushed with helium, evacuated to about 0.1 Torr and encapsulated; annealed at 1250 K for 100 h, and water quenched; etched again in HCl to the suitable dimensions; data reported as ratio of ρ to the ρ at the Néel temperature. value at Néel temperature not reported; measured with increasing temperature at $\sim 1\text{ K hr}^{-1}$; resistivity values calculated from reported resistance ratios and an assumed $\rho(300.23\text{ K}) = 12.655 \times 10^{-8} \Omega\text{m}$; because of graph reading difficulties, not all data points are included.
161	31	Rapp, Ö., et al.	1978	A	300.1-313.5	Cr(2)	Similar to the above except approximate dimensions 1 mm thick, 1 mm wide and 20 mm long, and annealed at 1250 K for 24 h and furnace-cooled in 12 h; T_N determined from a power law fit to $\rho^{-1} \times dp/dT$ 310.79 K; resistivity value calculated from reported resistance ratio and an assumed $\rho(300.07\text{ K}) = 12.651 \times 10^{-8} \Omega\text{m}$.
162	31	Rapp, Ö., et al.	1978	A	304.2-313.4	Cr(3)	Similar to the above except approximate dimensions 1 mm thick, 1 mm wide and 15 mm long, and annealed at 1250 K for 100 h and furnace-cooled in 24 h; "thermally cycled before measurement;" T_N determined by same method as above 310.77 K; resistivity values calculated from reported resistance ratio and an assumed $\rho(304.15\text{ K}) = 12.733 \times 10^{-8} \Omega\text{m}$.
163*	72	Clinard, F.W. and Kempster, C.P.	1968	A	4-300		0.3 O, <0.002 N and H each by chemical analysis; <0.03 Zn and <0.01 K by spectrographic analysis; polycrystalline specimen; 0.25 in. in diam and 1 in. long; annealed; resistivity values calculated from reported equations: $\rho = 0.1 + 1.58 \times 10^{-6} T^{3.62}$ for $T < 109.5\text{ K}$, and $\rho = -1.08 + \left(\frac{C}{M\theta}\right) \times \left(\frac{T}{\theta}\right)^5 \int_0^{\theta/T} \frac{x^5 dx}{(e^x - 1)(1 - e^{-x})}$, with $\theta = 357\text{ K}$ and $C = 1.43 \times 10^{-6}$ (M = atomic weight).

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	p	T	p	T	p	T	p	T	p
<u>DATA SET 10(cont.)*</u>									
1803	120.3	281.3	9.96	82.7	0.967	200	8.16	310.7	13.015
1813	120.8	300.7	10.69	91.8	1.27	204	8.51	311.4	12.964
1823	121.2	325.0	10.93	105	1.93	210	8.86	312.0	12.950
1833	121.6	344.4	11.42	123	2.86	219	9.37	312.5	12.950
1843	122.0	378.4	12.87	133	3.72	226	9.74	313.7	12.983
1853	122.5	402.6	13.12	160	5.26	233	10.21	319.5	13.167
1863	122.9	485.1	16.27	192	7.63	238	10.44	324.4	13.339
1873	123.3	509.4	16.76	214	8.49	243	10.72	328.2	13.465
1883	123.8	538.5	18.22	237	10.25	251	11.09	<u>DATA SET 22*</u>	
1893	124.2	577.3	19.67	271	12.05	262	11.65	291.5	12.721
1903	124.7	630.6	21.86	293	12.65	268	11.98	295.2	12.825
1913	125.2	698.5	24.77	298	12.83	279	12.44	299.1	12.922
1923	125.6	771.3	27.69	305	12.97	287	12.72	301.3	12.966
1933	126.0	853.7	31.57	308	12.96*	294	12.93	304.5	13.003
1943	126.5	941.0	35.70	313	12.93*	303	13.12	305.5	13.024
1953	126.9	941.0	35.70	314	12.95	314	13.14	308.1	13.026
<u>DATA SET 13*</u>									
		317	13.00*	321	13.30	326	13.47	<u>DATA SET 25*</u>	
		320	13.07	331	13.65	331	13.65	295	14.5
		4.2	0.255	340	13.93	340	13.93	308	15.6
<u>DATA SET 14*</u>									
		200	8.05	<u>DATA SET 18*</u>		200	8.05	325	16.1
		204	8.33	200	8.05	204	8.33	357	16.9
		209	8.61	209	8.61	209	8.61	382	18.6
		215	8.91	215	8.91	215	8.91	477	23.2
		219	9.16	219	9.16	219	9.16	668	33.2
		224	9.47	224	9.47	224	9.47	745	37.3
		230	9.84	230	9.84	230	9.84	810	40.6
		239	10.28	239	10.28	239	10.28	<u>DATA SET 26</u>	
		242	10.51	242	10.51	242	10.51	288.6	12.515
		251	10.93	251	10.93	251	10.93	292.5	12.655
		256	11.23	256	11.23	256	11.23	296.0	12.761
		259	11.37	259	11.37	259	11.37	298.6	12.824
		267	11.79	267	11.79	267	11.79	301.6	12.891
		272	12.02	272	12.02	272	12.02	303.1	12.918
		278	12.26	278	12.26	278	12.26	304.9	12.939
		283	12.47	283	12.47	283	12.47	305.5	12.945
		290	12.72	290	12.72	290	12.72	306.7	12.945
		310	13.23	310	13.23	310	13.23	308.0	12.937
		316	13.37	316	13.37	316	13.37	309.8	12.937
		321	13.54	321	13.54	321	13.54	310.4	12.924
		324	13.88	324	13.88	324	13.88	310.6	12.907
		336	14.05	336	14.05	336	14.05	311.0	12.827
		339	14.16	339	14.16	339	14.16	320.5	13.124
		344	14.35	344	14.35	344	14.35	328.2	13.367
		300.8	13.014	<u>DATA SET 20*</u>		300.8	13.014	<u>DATA SET 23*</u>	
		301.9	13.033	93.5	2.6	93.5	2.6	288.6	12.515
		304.1	13.037	133.2	5.1	133.2	5.1	292.5	12.655
		305.5	13.069	166.4	7.8	166.4	7.8	296.0	12.761
		307.2	13.069	196.7	9.8	196.7	9.8	298.6	12.824
		308.5	13.062	222.2	11.0	222.2	11.0	301.6	12.891
		309.9	13.037	252.5	13.0	252.5	13.0	303.1	12.918
				291.2	14.3	291.2	14.3	304.9	12.939
				322.4	14.9	322.4	14.9	305.5	12.945
				347.0	16.0	347.0	16.0	306.7	12.945
				367.8	16.9	367.8	16.9	308.0	12.937
				377.2	17.1	377.2	17.1	309.8	12.937
				389.5	17.6	389.5	17.6	310.4	12.924
				<u>DATA SET 21*</u>		<u>DATA SET 21*</u>		310.6	12.907
				300.8	13.014	300.8	13.014	311.0	12.827
				301.9	13.033	301.9	13.033	320.5	13.124
				304.1	13.037	304.1	13.037	328.2	13.367
				305.5	13.069	305.5	13.069		
				307.2	13.069	307.2	13.069		
				308.5	13.062	308.5	13.062		
				309.9	13.037	309.9	13.037		

* Not shown in figure.

TABLE J. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

DATA SET 37*		DATA SET 37(cont.)*		DATA SET 38(cont.)*		DATA SET 39(cont.)*		DATA SET 40(cont.)*		DATA SET 42*	
T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
242.8	11.24	320.8	13.25	317.2	13.18	311.7	12.74	310.0	12.38	243.0	10.40
247.0	11.39	322.7	13.30	321.0	13.31	314.3	12.85	312.1	12.48	249.1	10.60
251.0	11.56	324.2	13.37	325.1	13.49	316.0	12.90	316.2	12.66	252.1	10.68
252.3	11.61	327.2	13.50	329.1	13.63	318.3	13.00	318.9	12.78	255.3	10.77
255.1	11.70	329.3	13.59			321.1	13.12	321.1	12.89	259.1	10.85
256.1	11.75	331.4	13.68	DATA SET 39*		323.8	13.23	323.6	13.00	264.0	10.95
259.3	11.86			243.0	11.02	325.9	13.31	325.3	13.08	266.1	10.99
262.1	11.98	DATA SET 38*		246.4	11.16	327.4	13.39	326.4	13.14	267.0	10.99
263.8	12.04	242.8	11.24	251.3	11.32	330.0	13.49			268.0	10.99
264.6	12.08	247.0	11.39	254.2	11.44	331.2	13.56	DATA SET 41*		269.1	10.99
267.4	12.18	251.0	11.56	258.3	11.61			243.0	10.78	270.2	10.97
269.1	12.23	252.3	11.60	260.2	11.68	DATA SET 40*		244.3	10.82	271.0	10.96
271.6	12.31	254.9	11.71	262.1	11.75	243.0	10.80	246.6	10.92	272.7	10.91
273.5	12.37	255.9	11.73	264.2	11.84	244.3	10.82	248.3	10.99	273.1	10.88
275.4	12.43	259.1	11.87	266.1	11.90	246.8	10.91	250.0	11.05	273.6	10.86
278.0	12.51	262.1	11.98	269.9	12.03	248.3	10.99	253.0	11.16	273.8	10.80
280.5	12.57	263.6	12.04	273.1	12.14	250.0	11.05	253.6	11.20	273.8	10.76
284.1	12.67	264.6	12.09	275.0	12.20	252.7	11.15	256.6	11.32	274.4	10.74
287.1	12.75	267.4	12.17	277.1	12.26	253.6	11.20	260.0	11.43	275.0	10.74
289.2	12.78	268.9	12.22	278.0	12.28	256.6	11.32	264.2	11.57	275.9	10.77
290.9	12.80	271.6	12.31	279.0	12.31	260.2	11.43	268.0	11.66	278.0	10.82
292.2	12.81	273.3	12.38	281.2	12.37	264.2	11.56	271.4	11.71	280.1	10.91
293.3	12.83	275.4	12.43	282.9	12.41	268.0	11.66	273.1	11.77	282.3	10.96
294.1	12.84	278.0	12.51	284.1	12.43	271.2	11.71	275.0	11.77	285.0	11.06
294.9	12.84	280.5	12.66	285.6	12.43	273.1	11.72	275.9	11.79	287.1	11.14
296.0	12.81	283.1	12.66	288.6	12.46	274.1	11.72	277.1	11.79	290.1	11.25
296.9	12.81	284.1	12.67	290.1	12.46	275.0	11.72	279.7	11.77	293.5	11.37
297.7	12.81	286.9	12.78	291.1	12.46	275.9	11.75	281.4	11.75	295.8	11.47
298.8	12.79	288.8	12.80	291.6	12.46	277.1	11.75	282.4	11.68	298.8	11.58
299.0	12.76	290.1	12.83	292.4	12.46	278.0	11.73	283.1	11.61	300.9	11.66
299.4	12.74	291.8	12.84	293.3	12.43	279.1	11.73	283.7	11.54	304.1	11.77
299.8	12.69	293.0	12.87	294.1	12.38	280.3	11.70	284.6	11.52	309.0	11.95
300.3	12.66	294.1	12.87	294.3	12.32	281.0	11.68	285.0	11.54	314.1	12.17
300.5	12.61	294.9	12.89	294.5	12.27	281.6	11.63	286.5	11.60	320.6	12.46
300.7	12.56	296.0	12.89	294.7	12.23	282.9	11.58	287.8	11.65	323.0	12.59
301.3	12.55	297.1	12.89	295.2	12.20	283.3	11.53	289.9	11.73		
302.2	12.56	298.6	12.84	296.4	12.20	283.3	11.48	293.1	11.83	DATA SET 43*	
303.2	12.59	299.8	12.75	296.9	12.23	283.9	11.46	296.7	12.00	243.0	10.30
304.1	12.61	300.5	12.70	297.1	12.26	285.2	11.46	300.9	12.15	247.5	10.44
305.1	12.64	300.9	12.65	298.1	12.26	286.5	11.53	306.0	12.33	251.5	10.55
307.7	12.71	303.3	12.60	300.1	12.34	289.0	11.60	312.1	12.60	255.7	10.60
309.0	12.78	302.2	12.62	303.7	12.45	293.1	11.73	316.0	12.76	257.4	10.69
313.8	12.94	303.9	12.69	305.1	12.50	295.0	11.82	321.1	12.98		
315.5	13.02	305.1	12.74	307.1	12.57	297.9	11.93	326.4	13.14		
317.7	13.08	308.1	12.84	309.2	12.65	303.7	12.13			259.5	10.73
319.6	13.17	312.4	12.99	310.9	12.70	307.1	12.28			262.1	10.76

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ				
<u>DATA SET 43(cont.)*</u>													
264.0	10.76	1372	101.2	307.4	13.20*	311.9	12.901	306.9	12.812				
264.9	10.76	1447	113.0	308.2	13.31*	313.6	12.895	307.1	12.800				
265.9	10.74	1531	112.7	309.9	13.40*	315.0	12.911	307.9	12.807				
267.0	10.74	1605	119.1	313.4	13.52*	316.5	12.953	310.1	12.786				
268.0	10.72	1669	119.1	<u>DATA SET 47*</u>						311.4	12.777		
269.3	10.69	1724	140.9	<u>DATA SET 48*</u>						312.9	12.774		
270.2	10.64	1869	140.6	310.4	12.937	<u>DATA SET 51*</u>							
270.4	10.60	1919	140.6	311.8	12.880	307.9	12.788	<u>DATA SET 52*</u>					
270.8	10.54	1976	148.6	312.6	12.853	310.3	12.795	<u>DATA SET 53(cont.)*</u>					
271.0	10.49	<u>DATA SET 46</u>						<u>DATA SET 54*</u>					
271.9	10.50	5.1	0.20	313.1	12.847	311.8	12.722	<u>DATA SET 55*</u>					
272.5	10.53	79.6	1.03	313.8	12.871	314.6	12.696	<u>DATA SET 56*</u>					
274.0	10.58	89.0	1.26	314.6	12.896	314.6	12.759	<u>DATA SET 57*</u>					
275.0	10.62	94.1	1.47	306.6	12.961	316.9	12.805	<u>DATA SET 58*</u>					
277.6	10.69	100.9	1.76	309.5	12.937	320.5	12.945	<u>DATA SET 59*</u>					
280.1	10.80	107.8	2.08*	310.4	12.937	<u>DATA SET 60*</u>							
282.7	10.88	112.0	2.32*	311.8	12.880	<u>DATA SET 61*</u>							
286.3	11.01	116.3	2.58*	312.6	12.853	<u>DATA SET 62*</u>							
291.4	11.20	125.7	3.17	313.1	12.847	<u>DATA SET 63*</u>							
293.5	11.30	132.5	3.77	313.8	12.871	<u>DATA SET 64*</u>							
298.2	11.46	147.8	4.55	314.6	12.896	<u>DATA SET 65*</u>							
303.3	11.63	157.2	5.28*	<u>DATA SET 66*</u>									
308.3	11.85	167.4	6.01*	<u>DATA SET 67*</u>									
312.2	12.03	180.2	6.80	<u>DATA SET 68*</u>									
316.2	12.18	190.4	7.54*	<u>DATA SET 69*</u>									
321.5	12.42	199.0	8.09*	<u>DATA SET 70*</u>									
325.9	12.64	206.6	8.97	<u>DATA SET 71*</u>									
<u>DATA SET 44</u>													
1124	46.2	211.8	8.94*	<u>DATA SET 72*</u>									
1206	56.7	218.6	9.62*	<u>DATA SET 73*</u>									
1276	61.1	223.7	9.65	<u>DATA SET 74*</u>									
1379	72.1	233.9	10.35*	<u>DATA SET 75*</u>									
1466	90.2	245.9	11.03*	<u>DATA SET 76*</u>									
1605	95.3	257.8	11.67	<u>DATA SET 77*</u>									
1761	112.5	268.0	12.26*	<u>DATA SET 78*</u>									
1938	166.3	278.3	12.76*	<u>DATA SET 79*</u>									
<u>DATA SET 45(cont.)*</u>													
1120	68.6	286.8	13.11	<u>DATA SET 80*</u>									
1160	72.4	291.9	13.28*	<u>DATA SET 81*</u>									
1239	80.9	297.1	13.37*	<u>DATA SET 82*</u>									
1284	93.7	297.9	13.37*	<u>DATA SET 83*</u>									
<u>DATA SET 46(cont.)*</u>													
306.6	12.961	300.5	13.34	<u>DATA SET 84*</u>									
309.5	12.937	303.1	13.31*	<u>DATA SET 85*</u>									
311.74	12.643	304.8	13.28*	<u>DATA SET 86*</u>									
<u>DATA SET 47(cont.)*</u>													
306.9	12.812	<u>DATA SET 49(cont.)*</u>						300.00	12.619	300.95	12.698		
307.1	12.800	<u>DATA SET 50</u>						300.71	12.637	304.74	12.762		
307.9	12.807	80	0.860	<u>DATA SET 53*</u>						308.52	12.794		
310.1	12.786	90	1.225	<u>DATA SET 54*</u>						310.99	12.762		
312.21	12.646	100	1.630	<u>DATA SET 55*</u>						312.22	12.672		
312.33	12.651	120	2.605	<u>DATA SET 56*</u>						314.05	12.720		
312.92	12.665	140	3.760	<u>DATA SET 57*</u>						317.79	12.835		
313.70	12.689	160	5.295	<u>DATA SET 58*</u>						301.04	12.692		
314.41	12.710	180	6.575	<u>DATA SET 59*</u>						301.16	12.704		
315.29	12.736	200	7.545	<u>DATA SET 60*</u>						304.91	12.766		
316.01	12.759	220	9.100	<u>DATA SET 61*</u>						308.79	12.795		
317.07	12.793	240	10.300	<u>DATA SET 62*</u>						311.27	12.740		
317.78	12.814	260	11.385	<u>DATA SET 63*</u>						312.63	12.678		
318.38	12.835	280	12.270	<u>DATA SET 64*</u>						314.27	12.728		
318.91	12.850	300	12.880	<u>DATA SET 65*</u>						317.96	12.844		
319.15	12.859	304	12.946	<u>DATA SET 66*</u>						<u>DATA SET 51*</u>			
<u>DATA SET 48</u>													
300.95	12.698	306	12.958	<u>DATA SET 49*</u>						<u>DATA SET 52*</u>			
304.74	12.762	308	12.958	<u>DATA SET 50</u>						<u>DATA SET 53(cont.)*</u>			
308.52	12.794	310	12.931	<u>DATA SET 51*</u>						<u>DATA SET 54*</u>			
310.99	12.762	312	12.900	<u>DATA SET 52*</u>						<u>DATA SET 55*</u>			
312.22	12.672	314	12.898	<u>DATA SET 53*</u>						<u>DATA SET 56*</u>			
314.05	12.720	316	12.940	<u>DATA SET 54*</u>						<u>DATA SET 57*</u>			
317.79	12.835	320	13.080	<u>DATA SET 55*</u>						<u>DATA SET 58*</u>			
<u>DATA SET 49</u>													
308.45	12.751	340	13.765	<u>DATA SET 56*</u>						<u>DATA SET 59*</u>			
308.69	12.752	360	14.470	<u>DATA SET 57*</u>						<u>DATA SET 60*</u>			
309.10	12.753	380	15.200	<u>DATA SET 58*</u>						<u>DATA SET 61*</u>			
309.34	12.752	400	15.935	<u>DATA SET 59*</u>						<u>DATA SET 62*</u>			
309.57	12.752	<u>DATA SET 60*</u>						<u>DATA SET 63*</u>		<u>DATA SET 64*</u>			
309.69	12.751	<u>DATA SET 61*</u>						<u>DATA SET 65*</u>		<u>DATA SET 66*</u>			
309.93	12.749	<u>DATA SET 62*</u>						<u>DATA SET 66*</u>		<u>DATA SET 67*</u>			
310.05	12.748	<u>DATA SET 63*</u>						<u>DATA SET 67*</u>		<u>DATA SET 68*</u>			
310.16	12.746	<u>DATA SET 64*</u>						<u>DATA SET 68*</u>		<u>DATA SET 69*</u>			
310.40	12.743	<u>DATA SET 65*</u>						<u>DATA SET 69*</u>		<u>DATA SET 70*</u>			
310.52	12.740	<u>DATA SET 66*</u>						<u>DATA SET 70*</u>		<u>DATA SET 71*</u>			
310.63	12.735	<u>DATA SET 67*</u>						<u>DATA SET 71*</u>		<u>DATA SET 72*</u>			
310.87	12.725	<u>DATA SET 68*</u>						<u>DATA SET 72*</u>		<u>DATA SET 73*</u>			
311.16	12.696	<u>DATA SET 69*</u>						<u>DATA SET 73*</u>		<u>DATA SET 74*</u>			
311.45	12.662	<u>DATA SET 70*</u>						<u>DATA SET 74*</u>		<u>DATA SET 75*</u>			
311.57	12.648	<u>DATA SET 71*</u>						<u>DATA SET 75*</u>		<u>DATA SET 76*</u>			
311.57	12.645	<u>DATA SET 72*</u>						<u>DATA SET 76*</u>		<u>DATA SET 77*</u>			
311.68	12.644	<u>DATA SET 73*</u>						<u>DATA SET 77*</u>		<u>DATA SET 78*</u>			
311.74	12.643	<u>DATA SET 74*</u>						<u>DATA SET 78*</u>		<u>DATA SET 79*</u>			

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 58*</u>													
301.43	12.710	299.7	12.630	1123	53.64	283.6	12.30	313.4	12.859	311.8	12.920		
301.54	12.712	299.4	12.680	1175	56.88	287.3	12.44	313.6	12.868	311.9	12.909		
301.60	12.709	304.8	12.799	1224	60.41	294.6	12.70	313.7	12.875	312.0	12.906		
305.21	12.769	305.3	12.847	1273	63.07	297.1	12.75	314.4	12.892	312.0	12.902		
309.11	12.794	305.8	12.841	1319	65.72	299.3	12.80	314.6	12.892	312.1	12.898		
311.51	12.704	306.4	12.835	<u>DATA SET 66*</u>				301.0	12.81	315.1	12.909	312.3	12.892
312.81	12.685	308.1	12.844	78.9	2.51	302.4	12.85	315.6	12.900	312.3	12.889		
314.59	12.736	308.7	12.877	273.2	19.70	305.6	12.86	316.6	12.937	312.5	12.884		
<u>DATA SET 59*</u>													
309.1 12.862 307.1 12.86 307.1 12.86 309.6 12.83 312.6 12.878 312.7 12.873 312.9 12.867 313.0 12.860 313.3 12.857 313.3 12.854 313.6 12.854 313.8 12.856 314.0 12.867 314.3 12.874 314.3 12.885 314.7 12.885 315.1 12.882 315.4 12.911 315.6 12.920 315.9 12.932 316.4 12.939 316.4 12.939 316.6 12.931 317.0 12.939 317.3 12.960 317.5 12.966 317.8 12.875													
<u>DATA SET 60*</u>													
301.98	12.718	311.9	12.731	<u>DATA SET 69*</u>				300.3	12.888	313.3	12.857		
302.10	12.719	312.3	12.776	301.2	12.900	313.3	12.78	302.1	12.922	313.6	12.854		
302.21	12.720	311.6	12.773	302.8	12.940	313.8	12.90	302.8	12.946	314.0	12.867		
305.91	12.776	311.6	12.773	303.4	12.946	313.4	12.90	303.4	12.946	314.0	12.867		
309.62	12.792	311.9	12.731	303.9	12.957	314.3	13.08	303.9	12.957	314.3	12.874		
311.99	12.675	312.0	12.770	<u>DATA SET 67*</u>				304.9	12.963	315.1	12.882		
313.39	12.699	312.5	12.743	20.8	0.36	305.6	12.964	304.9	12.963	315.1	12.882		
315.09	12.752	313.8	12.746	77.7	1.83	306.0	12.964	305.6	12.975	315.4	12.911		
318.72	12.867	314.5	12.808	90.1	2.57	306.0	12.964	305.6	12.975	315.4	12.911		
<u>DATA SET 61*</u>													
302.10	12.714	315.1	12.803	192.3	14.78	306.0	12.964	306.0	12.978	315.6	12.920		
<u>DATA SET 62</u>													
293	12.88	317.5	12.809	273.2	21.10	306.0	12.964	306.0	12.978	315.9	12.932		
325	13.19	317.7	12.865	278.3	21.51	306.5	12.972	306.5	12.978	316.4	12.939		
432	17.92	318.5	12.913	285.4	21.93	307.0	12.971	307.0	12.986	316.4	12.939		
475	19.40	319.1	12.925	285.4	21.93	307.0	12.971	307.0	12.986	316.6	12.931		
507	21.17	320.0	12.942	291.5	22.23	307.4	12.968	307.4	12.986	316.6	12.931		
553	22.94	320.3	12.910	295.2	22.41	308.3	12.961	308.3	12.988	317.0	12.939		
625	25.90	<u>DATA SET 64</u>				308.6	12.967	308.6	12.978	317.0	12.939		
636	26.49	371	15.25	20.8	0.36	309.5	12.968	309.5	12.970	317.3	12.960		
671	27.97	371	15.25	77.7	1.83	309.5	12.968	309.5	12.970	317.5	12.966		
677	28.56	391	16.14	90.1	2.57	309.5	12.968	309.5	12.970	317.8	12.875		
775	33.28	484	19.69	192.3	14.78	309.5	12.968	309.5	12.970				
777	33.87	484	19.69	273.2	21.10	309.5	12.968	309.5	12.970				
890	39.77	726	30.92	278.3	21.51	309.5	12.968	309.5	12.970				
947	43.02	824	35.94	285.4	21.93	309.5	12.968	309.5	12.970				
1008	46.56	873	38.60	285.4	21.93	309.5	12.968	309.5	12.970				
<u>DATA SET 65</u>													
371 15.25 391 16.14 484 19.69 579 24.13 726 30.92 824 35.94 873 38.60 976 44.20 1077 50.10 1172 55.71													
<u>DATA SET 66</u>													
371 15.25 391 16.14 484 19.69 579 24.13 726 30.92 824 35.94 873 38.60 976 44.20 1077 50.10 1172 55.71													
<u>DATA SET 67</u>													
20.8 0.36 90.1 2.57 192.3 14.78 273.2 21.10 278.3 21.51 285.4 21.93 291.5 22.23 295.2 22.41 300.4 22.51 305.7 22.57 310.2 22.43 312.5 22.37 316.8 22.52 319.6 22.64 325.0 23.11 327.8 23.21 373.2 29.58													
<u>DATA SET 68</u>													
278.4 12.07 518 21.47 875 39.48 968 44.79													
<u>DATA SET 69</u>													
300.6 12.899 301.2 12.913 302.2 12.932 303.1 12.945 303.6 12.949 304.7 12.953 305.0 12.961 305.6 12.958 306.0 12.964 306.5 12.972 307.0 12.971 307.4 12.968 308.3 12.961 308.6 12.967 309.5 12.968 309.9 12.974 310.3 12.961 310.3 12.964 310.5 12.997 310.8 13.000 311.3 12.971 311.4 12.979 311.7 12.920 312.0 12.918 312.3 12.910 312.5 12.899 312.8 12.879 313.0 12.859													
<u>DATA SET 70*</u>													
300.3 12.888 301.2 12.900 302.1 12.922 302.8 12.940 303.4 12.946 303.9 12.957 304.5 12.960 304.9 12.963 305.5 12.975 306.0 21.977 306.4 12.979 307.0 12.985 307.3 12.986 307.8 12.984 308.0 12.986 308.4 12.979 308.8 12.981 309.0 12.978 309.2 12.986 309.5 12.989 309.8 12.988 309.9 12.978 310.3 12.970 310.3 12.961 310.5 12.965 310.8 12.964 310.9 12.982 311.2 12.982 311.4 12.979 311.7 12.920 312.0 12.918 312.3 12.910 312.5 12.899 312.8 12.879 313.0 12.859													
<u>DATA SET 71*</u>													
100.0 1.78 102.6 1.89 104.9 1.99 107.6 2.10 109.2 2.18 110.2 2.23 111.3 2.28 112.2 2.32 113.0 2.36 113.9 2.41 114.5 2.45 115.9 2.51 116.5 2.55 117.2 2.59													

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

DATA SET 79(cont.)			DATA SET 81(cont.)*			DATA SET 82(cont.)			DATA SET 86*			DATA SET 90(cont.)*			DATA SET 91(cont.)*																																																																																																																																																																																																																																																																																															
T	ρ	T	T	ρ	T	T	ρ	T	T	ρ	T	T	ρ	T	T	ρ																																																																																																																																																																																																																																																																																														
94.9	1.56	180	6.830	82.6	1764	1469	70.7	239.7	239.7	11.54	311.0	13.98	118	2.64*	190	7.480	84.4	1829	84.4	1520	11.78	313.9	14.05	140	4.03	200	8.130	86.1	1846	1565	75.5	249.9	249.9	12.09	318.7	14.15	162	5.46	210	8.805	89.2	1900	1570	77.4	254.6	254.6	12.31	323.0	14.33	201	8.01	220	9.460	93.1	1939	1603	79.2	259.3	259.3	12.48	250	11.3	230	10.125	97.1	1967	1633	81.3	265.9	265.9	12.80	282	12.2	240	10.755	99.3	1973	1693	85.1	269.6	269.6	12.99	343	13.3	250	11.325					274.3	274.3	13.14	<u>DATA SET 80*</u>			<u>DATA SET 83</u>			<u>DATA SET 87*</u>			<u>DATA SET 92</u>																																																																																																																																																																																																							
1.15	0.183	260	11.895										195.7	9.29	2.40	0.186	270	12.385								203.2	9.72*	2.70	0.178	280	12.840											207.9	9.96*	3.16	0.187	290	13.230	12.8	391	285.4	14.16	290.3	290.3	13.62	215.3	10.33*	4.26	0.187	300	13.680	16.3	498	289.1	14.27	295.0	295.0	13.67	224.7	10.93*	4.70	0.182	310	14.140	20.1	595	292.9	14.38	300.7	300.7	13.74	228.4	11.13	7.0	0.183	320	14.600	24.7	692	298.3	14.47	305.4	305.4	13.84	232.1	11.34*	12.4	0.186	330	15.190	29.3	800	303.4	14.52	308.2	308.2	13.87	239.6	11.68*	19.6	0.197	340		36.9	967	308.6	14.47	310.1	310.1	13.89	243.4	11.88*	21.3	0.198	350		42.4	1075	311.6	14.40	312.0	312.0	13.94	248.0	12.14*	25.9	0.205	360		43.2	1092	313.5	14.40	315.8	315.8	13.99	255.5	12.43*	32.9	0.227			51.6	1231	318.7	14.56	319.6	319.6	14.13	259.3	12.60*	48.0	0.313			60.6	1371	323.6	14.77	325.2	325.2	14.33	262.1	12.72	62.1	0.530			66.1	1469			330.8	330.8	14.47	265.9	12.87*	77.3	1.01			71.6	1564			334.6	334.6	14.67	273.3	13.23*	82.0	1.14			75.0	1628			340.2	340.2	14.86	275.2	13.31*	90.6	1.55			82.5	1730			346.8	346.8	15.06	280.9	13.43*	115	2.05			90.0	1850			350.5	350.5	15.27	286.5	13.57*	168	6.27			97.3	1905			358.1	358.1	15.47	289.3	13.67*	221	9.60							363.7	363.7	15.76	295.0	13.79*	235	11.10							369.3	369.3	15.93	299.7	13.84*	286	13.00							375.9	375.9	16.19	304.5	13.84	<u>DATA SET 81*</u>			<u>DATA SET 84</u>			<u>DATA SET 89*</u>			<u>DATA SET 91*</u>		
90	1.495	1025	42.6	58.1	1272	308.6	16.53	375.9	375.9	16.19	304.5	13.84	100	1.935	1047	43.2	61.4	1325	16.56	381.5	381.5	16.39	307.3	13.79*	110	2.415	1165	49.5	65.3	1387	16.79	386.5	386.5	16.70	310.2	13.70*	120	2.935	1264	55.6	69.1	1443	17.06	391.7	391.7	16.83	312.1	13.75*	130	3.515	1460	67.5	72.5	1498		397.5	397.5	17.07	313.0	13.79*	140	4.190	1652	77.4	76.5	1560		405.9	405.9	17.33	315.8	13.84*	150	4.870	1822	82.4	79.8	1664		411.5	411.5	17.60	320.5	13.99*	160	5.515	2025	88.4	83.0	1724		414.4	414.4	17.70	325.2	14.13*	170	6.170	2264	80.8	85.5	1774					335.6	14.47*	<u>DATA SET 85*</u>			<u>DATA SET 90*</u>			<u>DATA SET 91*</u>																																																																																																																																																																																											
		1740	80.8	86.7	1724								198.5	9.48			1740	86.7								337.5	14.50*			2000	87.9											337.5	14.52*			2264	87.9											338.4	14.62*			2460	87.9											341.2	14.72*			2660	87.9											343.1	14.77*			2860	87.9											348.7	14.96			3060	87.9											358.1	15.37*			3260	87.9											375.9	16.07			3460	87.9											383.4	16.29*																																																																																																																																																			

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ		
<u>DATA SET 92(cont.)</u> *												
400.3	17.04	310.1	12.40	56.0	1.26	300.1	12.86	204	8.20	123.6		
405.0	17.24*	310.4	12.38	59.4	1.32	301.4	12.88	216	9.02*	3.22		
408.7	17.46*	310.8	12.38	65.4	1.49	302.7	12.90	226	9.69*	3.65		
414.3	17.72*	311.2	12.38	72.7	1.63	303.9	12.90	237	10.39*	4.50		
424.7	18.13	311.7	12.39	78.7	1.85	305.2	12.91	248	11.01	4.93		
<u>DATA SET 93*</u>												
283.8	13.57	312.3	12.40	79.8	1.90	306.0	12.92	260	11.69*	5.79		
288.1	13.68	312.7	12.41	89.2	2.26	306.9	12.92	271	12.26*	6.64		
293.3	13.79	313.0	12.41	101	2.76	308.2	12.92	282	12.77	7.50		
296.3	13.84	313.6	12.44	110	3.23	309.0	12.91	291	13.05*	8.14		
297.7	13.86	314.8	12.47	114	3.40	209.5	12.88	297	13.24*	9.00		
305.0	13.82	316.1	12.50	133	4.47	209.8	12.85	<u>DATA SET 101</u>				
305.9	13.78	318.5	12.58	136	4.64	309.9	12.80	300.3	13.29	10.28		
306.4	13.75	321.8	12.68	143	5.08	310.1	12.76	301.4	13.31	10.93		
307.4	13.70	323.8	12.77	149	5.33	310.4	12.74	302.5	13.33	11.57		
308.8	13.80	327.5	12.94	151	5.49	311.1	12.74	302.5	13.36	12.21		
309.8	13.73	<u>DATA SET 96*</u>										
310.7	13.79	312.22	12.45	152	5.60	311.7	12.76	303.3	13.37	12.85		
313.6	13.89	<u>DATA SET 97*</u>										
318.0	13.98	78	1.635	156	5.84	311.9	12.67	304.3	13.38	13.07		
323.2	14.15	<u>DATA SET 98*</u>										
<u>DATA SET 94*</u>												
297.4	12.64	77.6	1.77	205	8.69	312.2	13.27	305.3	13.37	13.07		
305.3	12.78	85.9	2.10	217	9.02	312.2	13.27	306.3	13.38	13.50		
306.1	12.75	93.3	2.51	222	9.35	310.2	13.36	308.6	13.38	13.93		
307.2	12.72	106.3	3.01	227	9.62	311.0	13.34	309.6	13.37	14.14		
308.8	12.66	116.8	3.56	233	10.23	311.3	13.31	312.2	13.27	14.36		
309.5	12.60	119.5	3.73	239	10.51	311.9	13.29	312.2	13.27	14.79		
309.6	12.58	123.4	3.89	244	10.75	312.2	13.27	312.2	13.27	15.22		
310.3	12.49	127.2	4.06	249	11.00	311.9	13.29	311.9	13.29	15.64		
311.4	12.54	130.5	4.28	254	11.25	312.2	13.27	312.2	13.27	16.07		
312.4	12.57	140.4	4.83	259	11.47	313.0	13.27	313.0	13.27	16.72		
313.5	12.57	<u>DATA SET 99*</u>										
314.5	12.62	4.19	1.04	271	11.94	313.4	13.28	313.4	13.28	17.36		
<u>DATA SET 95*</u>												
309.6	12.57	10.3	1.04	282	12.39	314.2	13.30	314.2	13.30	17.79		
309.8	12.52	17.0	1.02	290	12.61	315.9	13.35	315.9	13.35	18.65		
309.8	12.49	24.2	1.03	293	12.69	317.4	13.40	317.4	13.40	19.50		
309.8	12.47	33.2	1.08	296.5	12.82	317.8	13.41	317.8	13.41	20.36		
310.1	12.45	41.0	1.10	298.2	12.84	<u>DATA SET 102*</u>					511.1	20.79
		47.1	1.15	299.2	12.86	8.5	0.21	562.2	22.93	562.2		
						21.3	0.21	592.0	24.43	592.0		
						34.1	0.21	604.7	25.29	604.7		
						46.9	0.22	630.3	26.36	630.3		
						64.0	0.43	664.3	27.86	664.3		
						85.3	1.08	685.6	28.93	685.6		
						93.8	1.50	719.7	30.43	719.7		
						106.6	1.93	749.5	31.93	749.5		
						115.1	2.36	787.8	33.86	787.8		
								813.3	35.57	813.3		
								847.4	37.29	847.4		

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 113(cont.)</u>									
317.7	12.658	277.1	12.109	284.8	12.954	302.3	12.594	294.6	12.764
322.5	12.809	282.4	12.307	289.8	12.916	305.1	12.611	294.8	12.768
327.6	12.948	286.4	12.460	292.1	12.901	307.2	12.634	295.0	12.774
332.4	13.087	292.4	12.610	294.9	12.883	309.9	12.659	295.3	12.776
338.0	13.276	294.0	12.669	297.4	12.863	<u>DATA SET 123</u>			
350.2	13.667	300.5	12.789	299.9	12.857	274.4	12.471	295.8	12.786
<u>DATA SET 114</u>									
304.5	12.633	305.4	12.837	302.4	12.833	279.7	12.502	296.0	12.789
305.4	12.648	307.8	12.829	304.9	12.810	279.7	12.502	296.3	12.795
307.1	12.665	310.4	12.789	307.2	12.786	284.6	12.522	296.5	12.800
308.1	12.662	311.8	12.632	310.0	12.751	289.7	12.551	296.8	12.804
309.0	12.662	312.8	12.661	<u>DATA SET 120</u>					
309.7	12.625	317.5	12.818	269.9	12.879	292.2	12.562	297.0	12.855
310.1	12.620	320.1	12.906	274.9	12.855	295.0	12.574	297.2	12.816
<u>DATA SET 117</u>									
310.8	12.600	284.8							
311.1	12.497	289.8							
311.7	12.497	292.3							
312.1	12.514	294.8							
312.6	12.520	297.4							
312.6	12.523	299.9							
313.0	12.540	302.4							
313.1	12.577	304.9							
315.1	12.620	307.7							
316.1	12.646	310.0							
<u>DATA SET 118</u>									
305.0	12.519	271.5							
306.0	12.539	277.4							
306.8	12.554	277.4							
308.0	12.551	282.4							
309.0	12.542	290.0							
309.1	12.534	294.5							
309.9	12.517	294.5							
310.5	12.500	300.5							
311.0	12.383	305.4							
311.4	12.372	308.0							
311.7	12.377	310.4							
312.1	12.360	<u>DATA SET 121</u>							
312.7	12.395	274.9							
313.0	12.403	279.9							
314.2	12.437	284.7							
315.3	12.475	289.8							
316.1	12.503	294.8							
<u>DATA SET 119</u>									
305.0	12.519	275.2							
306.0	12.539	277.6							
306.8	12.554	280.7							
308.0	12.551	287.3							
309.0	12.542	293.1							
309.1	12.534	297.3							
309.9	12.517	302.1							
310.5	12.500	306.8							
311.0	12.383	308.0							
311.4	12.372	310.4							
311.7	12.377	310.9							
312.1	12.360	284.8							
312.7	12.395	289.7							
313.0	12.403	292.2							
314.2	12.437	295.0							
315.3	12.475	297.5							
316.1	12.503	300.0							
<u>DATA SET 122</u>									
305.0	12.519	274.9							
306.0	12.539	279.9							
306.8	12.554	284.7							
308.0	12.551	289.8							
309.0	12.542	294.8							
309.1	12.534	299.6							
309.9	12.517	304.7							
310.5	12.500	307.2							
311.0	12.383	<u>DATA SET 125</u>							
311.4	12.372	274.7							
311.7	12.377	279.5							
312.1	12.360	284.6							
312.7	12.395	289.7							
313.0	12.403	294.5							
314.2	12.437	299.6							
315.3	12.475	304.6							
316.1	12.503	307.4							
<u>DATA SET 126(cont.)</u>									
306.1	12.891	294.6							
306.4	12.889	294.8							
306.8	12.888	295.0							
307.1	12.886	295.3							
307.4	12.882	295.5							
307.6	12.879	295.8							
307.8	12.877	296.0							
308.1	12.875	296.3							
308.3	12.873	296.5							
308.6	12.867	296.8							
308.9	12.862	297.0							
309.2	12.855	297.2							
309.3	12.851	297.5							
309.5	12.847	297.7							
309.6	12.842	297.9							
309.7	12.837	298.2							
309.9	12.835	298.4							
310.0	12.830	298.7							
310.0	12.826	299.0							
310.2	12.819	299.2							
310.2	12.815	299.4							
310.3	12.810	299.6							
310.3	12.807	299.9							
310.5	12.803	300.1							
310.5	12.796	300.4							
310.6	12.792	300.9							
310.6	12.790	301.2							
310.7	12.787	301.7							
310.7	12.785	301.9							
310.7	12.784	302.1							
310.7	12.780	302.4							
310.7	12.776	302.7							
310.7	12.775	302.9							
310.8	12.766	303.0							
310.8	12.762	303.4							
310.9	12.756	303.6							
310.9	12.752	303.8							
310.9	12.742	304.1							
310.9	12.739	304.4							
310.9	12.732	304.6							
311.0	12.724	304.8							
311.0	12.722	305.0							
311.0	12.721	305.3							
311.0	12.716	305.5							
311.0	12.711	305.6							
311.0	12.711	305.8							

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 133(cont.)													
3.30	0.37245	121.4	1.141	313.9	6.676	311.3	12.816	299.4	12.725	311.8	12.687	DATA SET 146(cont.)	
4.20	0.34839	123.9	1.307	317.6	6.722	313.0	12.622	299.9	12.739	311.9	12.661		
4.79	0.18398	126.3	1.426	320.0	6.814	320.8	12.958	301.0	12.777	311.9	12.648		
5.70	0.39417	128.8	1.736	322.5	6.929	326.4	13.151	301.7	12.777	312.9	12.661		
6.45	0.79188	130.0	1.973	328.6	7.067	DATA SET 143		302.4	12.790	316.7	12.790		
7.07	1.26000	130.6	2.140	DATA SET 140		76.1	0.929	303.2	12.816	318.5	12.854		
DATA SET 134													
102.6	1.731	DATA SET 137		101.3	2.378	90.1	1.407	305.4	12.841	DATA SET 147			
105.3	1.801	103.2	1.825	103.8	2.473	99.0	1.794	307.9	12.841	295.2	12.609		
106.6	1.895	105.1	1.966	105.0	2.544	123.6	3.239	309.0	12.829	297.1	12.674		
108.6	1.941	107.7	2.106	106.3	2.639	175.6	6.376	310.0	12.816	299.0	12.725		
111.8	1.968	109.6	2.224	108.8	2.758	225.3	9.512	310.6	12.803	300.1	12.751		
113.8	2.034	111.6	2.317	111.3	2.972	275.0	12.003	311.2	12.790	303.0	12.803		
116.4	2.151	113.5	2.411	115.0	3.091	311.3	12.764	311.5	12.738	305.2	12.829		
119.1	2.291	116.9	2.598	116.9	3.115	312.4	12.570	311.7	12.661	307.5	12.841		
120.4	2.409	118.8	2.785	122.5	3.281	320.8	12.958	312.7	12.635	309.6	12.816		
124.3	2.572	DATA SET 138		122.5	3.281	327.5	13.151	313.4	12.661	311.8	12.700		
130.2	2.852	121.3	1.667	123.8	3.376	DATA SET 144		313.8	12.674	DATA SET 148			
DATA SET 135													
285.0	9.153	124.0	1.739	125.0	3.566	296.1	12.880	314.6	12.712	299.4	12.545		
288.8	9.226	126.6	1.836	DATA SET 141		297.5	12.906	315.2	12.725	300.8	12.583		
291.3	9.367	128.6	1.932	110.6	2.651	299.9	12.930	316.1	12.764	302.1	12.622		
295.1	9.438	129.9	1.957	111.3	2.740	302.4	12.958	316.5	12.777	303.4	12.661		
300.1	9.533	131.8	2.005	113.9	2.793	303.1	12.971	316.9	12.803	304.7	12.700		
303.8	9.580	133.8	2.005	115.2	2.865	304.2	12.971	317.4	12.816	306.1	12.725		
308.8	9.580	135.8	2.101	117.1	3.007	304.9	12.971	318.0	12.841	307.2	12.738		
308.8	9.532	139.1	2.078	117.1	3.007	305.5	12.971	DATA SET 146		308.6	12.751		
312.6	9.461	140.4	2.294	118.4	3.102	306.5	12.958	309.9	12.751	309.9	12.751		
315.1	9.532	141.7	2.391	119.7	3.197	307.4	12.958	310.6	12.751	310.6	12.751		
317.6	9.627	145.0	2.511	122.3	3.269	308.3	12.945	300.0	12.919	311.1	12.738		
322.6	9.722	148.3	2.559	123.6	3.411	309.2	12.919	301.9	12.932	311.3	12.725		
326.3	9.793	150.2	2.632	126.2	3.625	311.3	12.880	302.9	12.945	311.4	12.712		
DATA SET 136													
105.5	0.975	152.9	2.704	130.0	3.791	312.3	12.622	303.8	12.958	311.8	12.648		
107.3	0.975	132.6	4.028	130.7	3.886	314.1	12.687	304.8	12.958	311.9	12.635		
109.8	0.999	DATA SET 139		132.6	4.028	314.9	12.725	306.5	12.971	312.3	12.635		
111.6	0.999	287.3	6.609	DATA SET 142		315.9	12.751	307.7	12.958	313.1	12.648		
114.7	1.046	290.9	6.655	76.1	0.929	316.6	12.790	309.0	12.932	313.5	12.674		
116.5	1.094	295.8	6.701	99.0	1.988	317.7	12.829	310.3	12.893	314.7	12.712		
119.6	1.117	299.4	6.746	124.7	3.536	319.6	12.893	310.9	12.854	317.1	12.816		
DATA SET 145													
303.1	6.793	303.1	6.793	175.0	6.814	DATA SET 145		311.5	12.790	320.0	12.906		
307.9	6.792	307.9	6.792	225.4	10.041	297.8	12.687	311.5	12.790				
312.7	6.723	312.7	6.723	275.6	12.390	298.6	12.712	311.6	12.751				
DATA SET 146													
298.6 12.712													

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 149</u>											
299.1	12.725	163.6	5.273	32.5	0.0370	233.2	6.96	232.4	6.58	216.1	6.79
301.8	12.790	168.9	5.494	40.0	0.062	237.6	7.18	236.6	6.82	228.4	7.87
304.4	12.829	174.3	5.716	44.4	0.062	242.9	7.43	240.9	7.07	240.0	8.83
307.0	12.854	178.5	5.852	49.8	0.123	247.1	7.61	247.3	7.23	252.3	9.83
309.6	12.829	183.9	6.049	55.2	0.172	251.4	7.83	252.6	7.48	263.9	10.65
<u>DATA SET 150</u>											
193.5	6.209	193.5	6.209	59.5	0.259	256.7	8.06	256.9	7.70	274.0	11.40
198.9	6.344	198.9	6.344	63.8	0.370	262.0	8.28	262.2	7.92	280.5	11.87
202.6	6.270	202.6	6.270	69.1	0.480	266.3	8.50	<u>DATA SET 156</u>			
207.6	6.209	207.6	6.209	74.5	0.591	271.6	8.75	70.2	0.505	286.3	12.13
213.0	6.320	213.0	6.320	83.1	0.788	275.8	9.02	76.6	0.727	289.2	12.31
218.3	6.517	218.3	6.517	84.1	0.899	<u>DATA SET 155</u>				292.1	12.43
221.5	6.677	221.5	6.677	88.4	1.12	59.4	0.320	76.6	0.727	297.8	12.52
221.0	6.936	221.0	6.936	93.7	1.26	63.7	0.394	83.0	0.949	300.0	12.56
232.2	7.096	232.2	7.096	98.0	1.39	69.1	0.505	89.4	1.17	306.5	12.60
<u>DATA SET 151 (cont.)</u>											
4.31	0.0616	59.5	0.283	102.3	1.56	76.1	0.727	93.7	1.37	308.6	12.60
9.73	0.0616	64.8	0.394	113.0	1.84	83.0	0.949	103.3	1.61	310.8	12.51
15.1	0.0616	70.2	0.505	118.4	1.86	89.4	1.17	113.0	1.72	320.9	12.99
20.6	0.0616	74.5	0.616	122.7	1.92	92.6	1.39	118.4	1.79	331.0	13.47
24.9	0.0616	78.8	0.788	129.1	2.22	99.0	1.53	128.1	2.08	<u>DATA SET 158</u>	
34.6	0.0862	82.0	0.702	133.4	2.30	99.0	1.53	133.4	2.30	298.5	12.712
45.4	0.148	84.1	0.899	144.1	2.66	102.3	1.84	138.7	2.48	299.6	12.720
60.0	0.123	88.4	1.12	149.4	2.83	108.6	2.08	144.1	2.66	300.6	12.742
68.7	0.172	93.7	1.39	154.7	3.06	113.9	2.33	149.4	2.83	301.7	12.765
55.1	0.259	98.0	1.39	159.0	3.24	118.1	2.55	152.6	3.02	302.8	12.789
59.4	0.370	107.5	1.946	164.3	3.41	123.5	2.71	157.9	3.19	304.0	12.809
63.7	0.456	113.9	2.193	168.6	3.63	128.8	2.86	163.3	3.41	305.0	12.827
70.2	0.616	118.2	2.415	177.1	4.04	134.1	3.10	168.6	3.63	306.0	12.843
73.4	0.727	123.5	2.599	183.5	4.26	138.4	3.29	173.9	3.83	307.4	12.845
77.6	0.899	129.9	2.883	187.7	4.48	143.7	3.46	179.2	4.07	307.8	12.853
84.0	1.121	133.1	3.129	194.2	4.68	149.1	3.51	184.6	4.26	308.4	12.853
89.4	1.281	138.4	3.289	198.4	4.89	154.9	3.57	188.8	4.48	308.8	12.853
93.7	1.528	144.8	3.462	202.7	5.06	159.0	3.46	194.2	4.64	309.3	12.849
100.0	1.749	149.1	3.622	209.1	5.31	163.1	3.73	198.4	4.84	309.7	12.845
104.3	1.971	153.4	3.683	217.6	5.73	163.1	3.73	203.8	5.06	309.9	12.845
108.5	2.242	158.8	3.622	222.9	5.94	173.8	4.15	208.0	5.28	310.5	12.837
113.8	2.550	164.2	3.794	<u>DATA SET 154</u>				212.3	5.47	311.0	12.831
120.2	2.821	173.8	4.152	188.2	6.07	179.1	4.35	217.6	5.69	311.2	12.823
124.4	3.043	180.2	4.349	192.5	6.18	183.4	4.56	222.9	5.89	311.5	12.805
128.6	3.351	184.5	4.570	197.9	6.23	187.6	4.82	227.2	6.11	312.0	12.787
134.0	3.622	188.5	4.824	202.6	6.39	194.0	4.98	232.5	6.27	312.2	12.763
139.3	3.868	192.5	5.174	207.9	6.51	198.3	5.09	236.8	6.47	312.3	12.723
144.6	4.139	197.9	5.359	208.8	6.07	202.7	5.21	241.0	6.71	312.6	12.719
148.8	4.447	202.6	5.581	213.1	6.18	207.9	5.58	247.4	6.94	312.9	12.719
156.2	4.780	212.1	5.778	218.6	6.38	213.1	5.75	251.7	7.13		
159.4	5.002	216.4	5.975	222.8	6.58	218.6	5.97	257.0	7.32		
		222.8	6.164	227.7	6.79	221.7	6.16	262.3	7.54		
		227.7	6.164	227.7	6.79	227.1	6.38				

* Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF CHROMIUM Cr (continued)

T D

DATA SET 162 (cont.)

312.20	12.624
312.31	12.626
312.42	12.628
312.58	12.634
312.69	12.638
312.80	12.642
312.95	12.644
313.07	12.650
313.22	12.653
313.37	12.658
313.44	12.662

DATA SET 163*

4.2	0.100
7	0.101
10	0.102
15	0.107
20	0.113
25	0.126
30	0.146
40	0.209
50	0.314
60	0.470
70	0.690
80	0.983
90	1.36
109.5	2.37
125	3.35
150	4.92
175	6.48
200	8.01
225	9.51
250	10.99
273	12.33
300	13.90

* Not shown in figure.

3.2. Cobalt

There are 55 sets of experimental data available for the electrical resistivity of cobalt with purity higher than 99.9%. The information on specimen characterization and measurement condition for each of the data sets is given in table 5. The data sets are tabulated in table 6 and shown partially in figures 3 and 4.

Since cobalt is a transition element and is ferromagnetic, its electrical resistivity is expected to resemble those of nickel and iron. As it can be surmised from the size of the available data, the electrical resistivity of cobalt is not investigated as extensively as either that of iron or of nickel. Nonetheless, some features of the behavior of the electrical resistivity of iron and nickel have also been reported for cobalt, such as the T^2 variation in the temperature dependence of the electrical resistivity at low temperatures. The coefficient of this T^2 term becomes larger when measured in an applied magnetic field than when measured in the absence of an applied field [75].

Judging from the impurity analyses reported by some of the authors, cobalt specimens of purity higher than 99.999% are available commercially: Laubitz and Matsumura [76] (data set 43), White and Woods [77,21] (data sets 27, 28, 39), Kierspe et al. [78] (data set 16). However, there are wide disagreement between the reported residual resistivity ratios, even for specimens from the same manufacturer and having nearly the same impurities, as illustrated by the specimens of data sets 27 and 39. Since cobalt is nearly as strongly magnetic as iron (the spontaneous magnetization of 20.8 kG vs. 22 kG for iron), effects due to magnetic structure of specimens and due to the measuring current densities, etc. are expected to be significant. Unfortunately, there is very scarce information on these effects for cobalt specimens. In addition, the morphology of cobalt may have significant influence also. Even though the temperature of the α - β phase transformation of cobalt is greater than 700 K, this cph-fcc transformation is very sluggish, and the high-temperature fcc phase has been reported to persist at lower temperatures [79]. It is not unlikely for a specimen to contain a mixture of these two phases, depending on its thermal and mechanical history [76].

Amongst the available data for high-purity polycrystalline cobalt specimens, Laubitz and Matsumura [76] (data set 43) reported the highest residual resistance ratio ($\rho_{273\text{ K}}/\rho_{4\text{ K}}$) of 140 ± 10 . The impurity analysis reported also indicated

that their specimen was one of the purest. For comparison, the residual resistance ratio of a whisker specimen by Marker et al. [75] (data set 19) was reported to be 388. There are available only a few data sets giving the electrical resistivity of high-purity cobalt over a temperature range extending from ~ 4 to 300 K: White and Woods [77] (data sets 27, 28), and Price and Williams [80] (data set 26). Aside from these data sets, White and Woods [21] (data set 39) reported data from 126 to 273 K, Semenenko et al. [81] (data set 3), Ols en-Bar [82] (data sets 4, 5), and Radhakrishna and Nielsen [83] (data set 9) reported data for low temperatures (< 20 K) only. In addition, Loegel and Gautier [84] (data sets 47, 48) reported the temperature dependent part of the resistivity of specimens of unspecified purity for temperatures below 80 K. Most of these authors reported a T^2 dependence for the temperature dependent part of the resistivity for temperatures below 10 K. The coefficient of this T^2 component was reported to be $1.6 \times 10^{-11} \Omega \text{ m K}^{-2}$ by White and Woods [77] and $\leq 1.0 \times 10^{-11} \Omega \text{ m K}^{-2}$ by Radhakrishna and Nielsen [83]. Loegel and Gautier [84] reported, together with a T^5 component, a coefficient of $1.06 \times 10^{-11} \Omega \text{ m K}^{-2}$ for temperatures up to 30 K. Semenenko et al. [81] reported an additional T component for temperatures 1.4-4.2 K; however, Radhakrishna and Nielsen [83] concluded from their data that the T component, if present at all, was not significant. Marker et al. [75] also reported the T^2 dependence for their whisker specimen with a coefficient of $1.5 \times 10^{-11} \Omega \text{ m K}^{-2}$ in the temperature range 1.1-4.2 K.

The present analysis of the electrical resistivity of cobalt at low temperatures follows the same method as employed in the analysis of that of iron and nickel, i.e., by fitting the resistivity data to the expression

$$\rho = \rho_0 + \alpha T^2 + A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^2} dx \quad (8)$$

However, because of the small number of available data sets and because of the apparent large deviation of the electrical resistivity of cobalt from the Mattheissen's rule, the coefficients α and A cannot be determined simultaneously with small uncertainties. Therefore, the value of α is taken to be $1.00 \times 10^{-11} \Omega \text{ m K}^{-2}$, a value close to the mean of the coefficients reported by Radhakrishna and Nielsen [83] and by Loegel and Gautier [84]. Using a Debye

temperature of 445 K approximately as the value for θ_R , the value of A was determined from eq (8) with the data of White and Woods [77] (data sets 27, 28). A value of $70 \times 10^{-8} \Omega \text{ m K}^{-2}$ was obtained. With these values of α and A, the resistivity values calculated from eq (8) agree to within 2% with the experimental data for temperatures below ~ 25 K. For higher temperatures, the calculated values do not agree well with the experimental data, and therefore in the temperature range ~ 35 to 90 K, the recommended values were obtained by interpolating the low-temperature values calculated from eq (8) and the data of Laubitz and Matsumura [76] (data set 43).

Two data sets are available covering a very wide temperature range (~ 80 -1700 K): by Laubitz and Matsumura [76] (data set 43) and by Kierspe et al. [78] (data set 16). Except for temperatures below ~ 200 K, where the latter data set appears to be in error, the agreement between these two data sets is within $\sim \pm 3\%$. The recommended values from 90 to 1700 K are therefore based on these, with more weight given to that of Laubitz and Matsumura, especially for temperatures below 250 K. In this temperature range, cobalt undergoes two transitions: one polymorphic at ~ 715 K, from cph(α) to fcc(β), and one ferromagnetic-paramagnetic at ~ 1395 K. The polymorphic transformation is martensitic and is very sluggish, due to the small associated free energy change. Thus the temperature range in which this transformation occurs has been reported to vary from about 660 K [85] (data set 14) to about 740 K [86] (data set 1), and thermal hysteresis is generally reported. The careful study of Laubitz and Matsumura [76] on a specimen which had been x-ray analyzed to contain no detectable fcc phase at room temperature showed that the range of transformation was about 703-710 K upon heating and about 686-693 K upon cooling (data sets 44, 45). The resistivity of the β phase is generally reported to be lower than that of the α phase. Kierspe et al. [78] did not report details of the transformation, even though their data appeared to have a strange behavior at the transformation, which occurred at $\sim 720 \pm 5.0$ K. These authors reported a temperature coefficient that shows a decrease ($\sim 30\%$) at the transition, instead of the usual positive-negative-positive change in the temperature coefficient indicated by a number of the other works (see, e.g., Laubitz and Matsumura [76] (data sets 44, 45), Powell [85] (data sets 11-14), and Fraser et al. [86] (data sets 1, 2)). A possible reason for the behavior of the data of Kierspe et al. was that their specimen might have been heated or cooled at too fast a rate.

The α - β phase transition temperature of 715 K indicated in figures 3 and 4 is based actually on specific heat measurements. At temperatures above the α - β transition, the temperature coefficient reported by Kierspe et al. [78] showed a gradual rise to a flat maximum at ~ 1150 K, and decreased gradually again. It became almost constant at temperatures above ~ 1500 K. There was no sharp δ -function like maximum as in the cases of nickel and iron at the Curie temperature. This behavior of their data is consistent with the data of Laubitz and Matsumura [76] (data set 43), which appeared to have a change of slope at ~ 1250 K. At temperatures above the Curie temperature, the data of Seydel and Fucke [87] (data set 42) are in good agreement with those of Laubitz and Matsumura [76] and of Kierspe [78] and are also taken into account.

There are eight data sets for the electrical resistivity of molten cobalt [87-94] (data sets 7, 18, 40-42, 50-52). Of these, the data of Güntherodt et al. [92] (data set 41) and of Seydel and Fucke [87] (data set 42) agree to within $\pm 1\%$. In addition, their data for the solid phase at the melting point agree to within $\pm 1.5\%$ of the recommended value. The recommended values for the molten state are therefore based on their data. The linear temperature dependence of the electrical resistivity of molten cobalt was reported also by Ono and Yagi [89] (data set 18), and by Kita et al. [93] (data sets 50,51).

The recommended values both uncorrected and corrected for thermal expansion of the material are presented in table 4, while only the uncorrected values (except those for the liquid state) are shown in figures 3 and 4 along with the experimental data. The values are for polycrystalline cobalt of purity 99.99% or higher; however, those values for temperatures below 200 K are applicable only to cobalt having a residual resistivity of $0.0370 \times 10^{-8} \Omega\text{m}$. The estimated uncertainty in the recommended values is about $\pm 5\%$ for the solid state and $\pm 7\%$ for the molten state.

As mentioned earlier, the electrical resistivity of cobalt appears to deviate from the Matthiessen's rule fairly large. For specimens with somewhat higher residual resistivities, the application of Matthiessen's rule is likely to underestimate the electrical resistivity by up to a few percent. For example, for the specimens of White and Woods [77] (data sets 27,28) which have residual resistivities of about $0.09 \times 10^{-8} \Omega\text{m}$, Matthiessen's rule appears to be applicable for temperatures below ~ 15 K with resulting error less than -1% , but the error increases with temperature to -2% at ~ 20 K, -5% at $\sim 35\%$, and -6% at 200 K and higher. For the specimen of Price and Williams [80] (data set 26) which has

a residual resistivity of $\sim 0.13 \times 10^{-8} \Omega\text{m}$, the errors are approximately +2% at ~ 20 K, -6% from ~ 35 to 60 K, -5% at 100 K, and -4% from 200 to 300 K. Unfortunately, there are no available data sets for specimens of higher residual resistivity covering more or less continuously from low to room temperatures, so that a more extensive comparison could be made. The earlier measurement by McLennon et al. [50] (data set 30) on a specimen of residual resistivity $0.45 \times 10^{-8} \Omega\text{m}$ indicates that the use of Mattheissen's rule yields an error of only -1% at 20.6 K, but the error jumps to -20% at 83 K and reduces to $\sim -10\%$ at 293 K. The more recent measurement by Wilkes [95] (data set 46), whose specimen has a residual resistivity of $\sim 0.17 \times 10^{-8} \Omega\text{m}$, shows that the errors are -1% at ~ 77 K, +1% at ~ 200 K, and +1% at ~ 300 K. It is interesting to note that the total impurity content of this specimen, $\sim 0.08\%$, is more than ten times higher than that of the specimen of White and Woods [77] (data sets 27, 28). However, it has been determined by Laubitz and Matsumura [76] that the specimen of Wilkes contains approximately 33% of the fcc phase at room temperature. It is evident that the phase constitution of a specimen has significant influence on the resistivity of cobalt, especially below the α - β transition. The presence of the fcc phase below 700 K is likely to lower the resistivity. On the other hand, the low-temperature cph phase is not likely to be stable at temperatures much higher than 700 K so that the higher temperature resistivity of cobalt of reasonable purity should not deviate by more than two or three percent from the recommended values.

The recommended values uncorrected for thermal expansion given in table 4 can be represented approximately by the following expressions to within $\pm 0.5\%$.

1-35 K:

$$\rho = 0.0370 + 1.00 \times 10^{-5} T^2 + 70 \left(\frac{T}{445} \right)^5 \int_0^{445/T} \frac{x^5 e^x}{(e^x - 1)^2} dx \quad (19)$$

35-90 K:

$$\rho = 8.20 \times 10^{-2} - 5.261 \times 10^{-3} T + 1.477 \times 10^{-4} T^2 - 8.559 \times 10^{-8} T^3 \quad (20)$$

90-700 K:

$$\rho = -9.98 \times 10^{-1} + 1.865 \times 10^{-2} T + 4.237 \times 10^{-6} T^2 + 3.777 \times 10^{-8} T^3 \quad (21)$$

715-1250 K:

$$\rho = 31.71 - 1.0987 \times 10^{-1} T + 1.7872 \times 10^{-4} T^2 - 5.098 \times 10^{-8} T^3 \quad (22)$$

1250-1400 K:

$$\rho = -117.61 + 1.8101 \times 10^{-1}T + 2.042 \times 10^{-8}T^2 - 1.773 \times 10^{-8}T^3 \quad (23)$$

1400-1767 K:

$$\rho = -342.15 + 7.2544 \times 10^{-1}T - 4.1393 \times 10^{-4}T^2 + 8.201 \times 10^{-8}T^3 \quad (24)$$

1767-3000 K:

$$\rho = 94.80 + 1.128 \times 10^{-2}T \quad (25)$$

It should be stressed that these expressions do not necessarily suggest any theoretical justification, and should be treated, most appropriately, as numerical aids only. It should also be understood that giving these expressions does not imply a recommendation for the temperature derivative of the electrical resistivity.

TABLE 4. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF COBALT^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.0370	0.0370	1100	59.26	60.05
4	0.0372	0.0372	1200	69.14	70.18
7	0.0375	0.0374	1300	78.79	80.11
10	0.0381	0.0380	1400	87.20	88.83
15	0.0396	0.0395	1500	91.46	93.34
20	0.0426	0.0425	1600	94.81	96.94
25	0.0481	0.0480	1700	97.76	100.15
30	0.0581	0.0580	1767	99.75(β)	102.32(β)
40	0.102	0.102	1767		114.7 ^b (ℓ)
50	0.178	0.178	1800		115.1 ^b
60	0.280	0.279	1900		116.2 ^b
70	0.408	0.407	2000		117.4 ^b
80	0.563	0.562	2100		118.5 ^b
90	0.742	0.740	2200		119.6 ^b
100	0.947	0.945	2300		120.7 ^b
150	2.02	2.02	2400		121.9 ^b
200	3.20	3.20	2500		123.0 ^b
250	4.52	4.52	2600		124.1 ^b
273	5.18	5.18	2700		125.2 ^b
293	5.78	5.78	2800		126.4 ^b
300	6.00	6.00	2900		127.5 ^b
350	7.67	7.67	3000		128.6 ^b
400	9.56	9.57			
500	14.11	14.15			
600	19.88	19.97			
700	27.09(α)	27.25(α)			
715	25.89(β)	26.07(β)			
800	32.09	32.36			
900	40.43	40.83			
1000	49.58	50.15			

^a The values are for polycrystalline cobalt of purity 99.99% or higher, but those below 200 K are applicable only to cobalt having a residual resistivity of $0.0370 \times 10^{-8} \Omega \text{m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation, while dotted line indicates solid phase transition.

^b α : cph; β : fcc.
Provisional value.

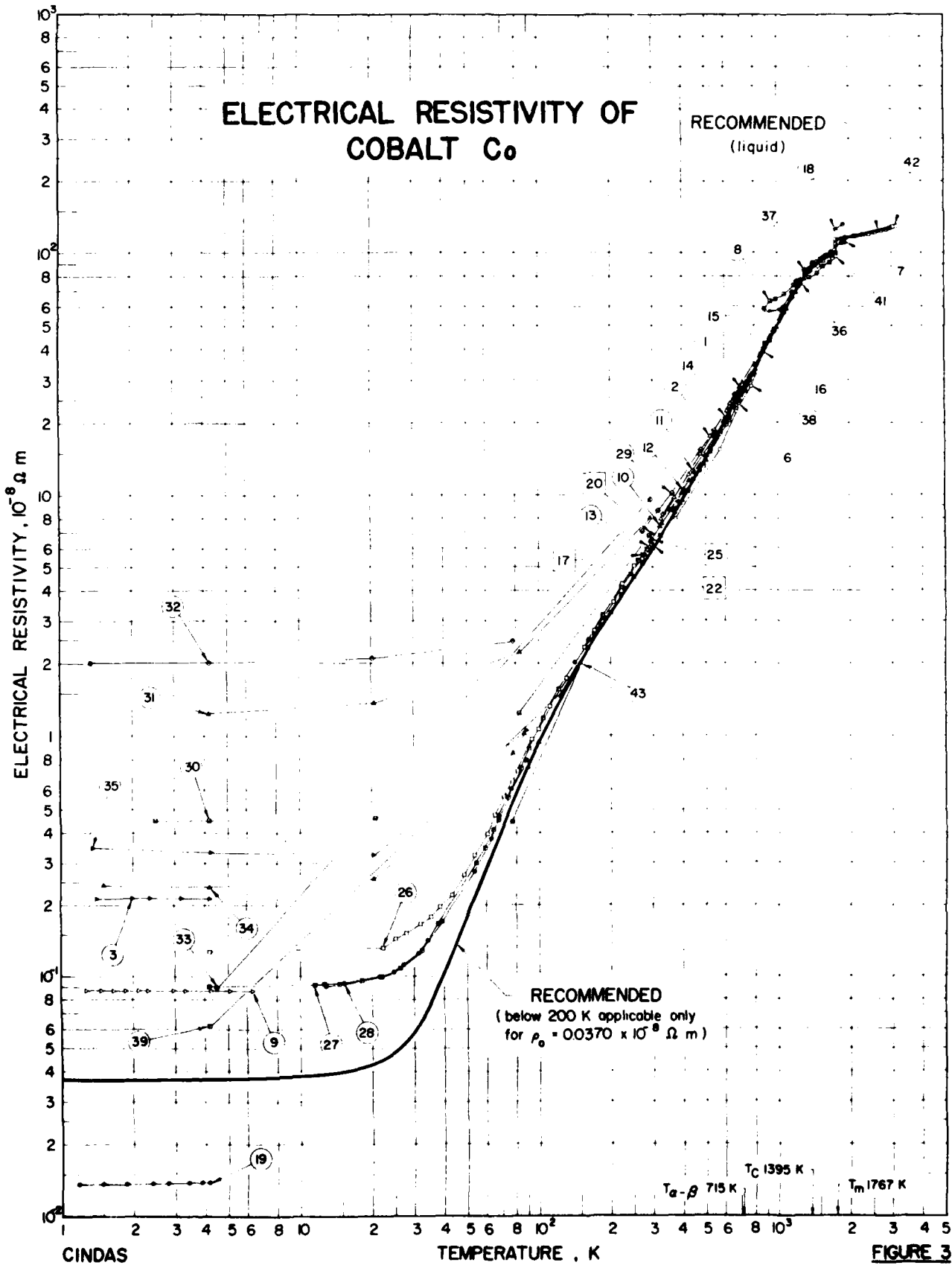


FIGURE 3

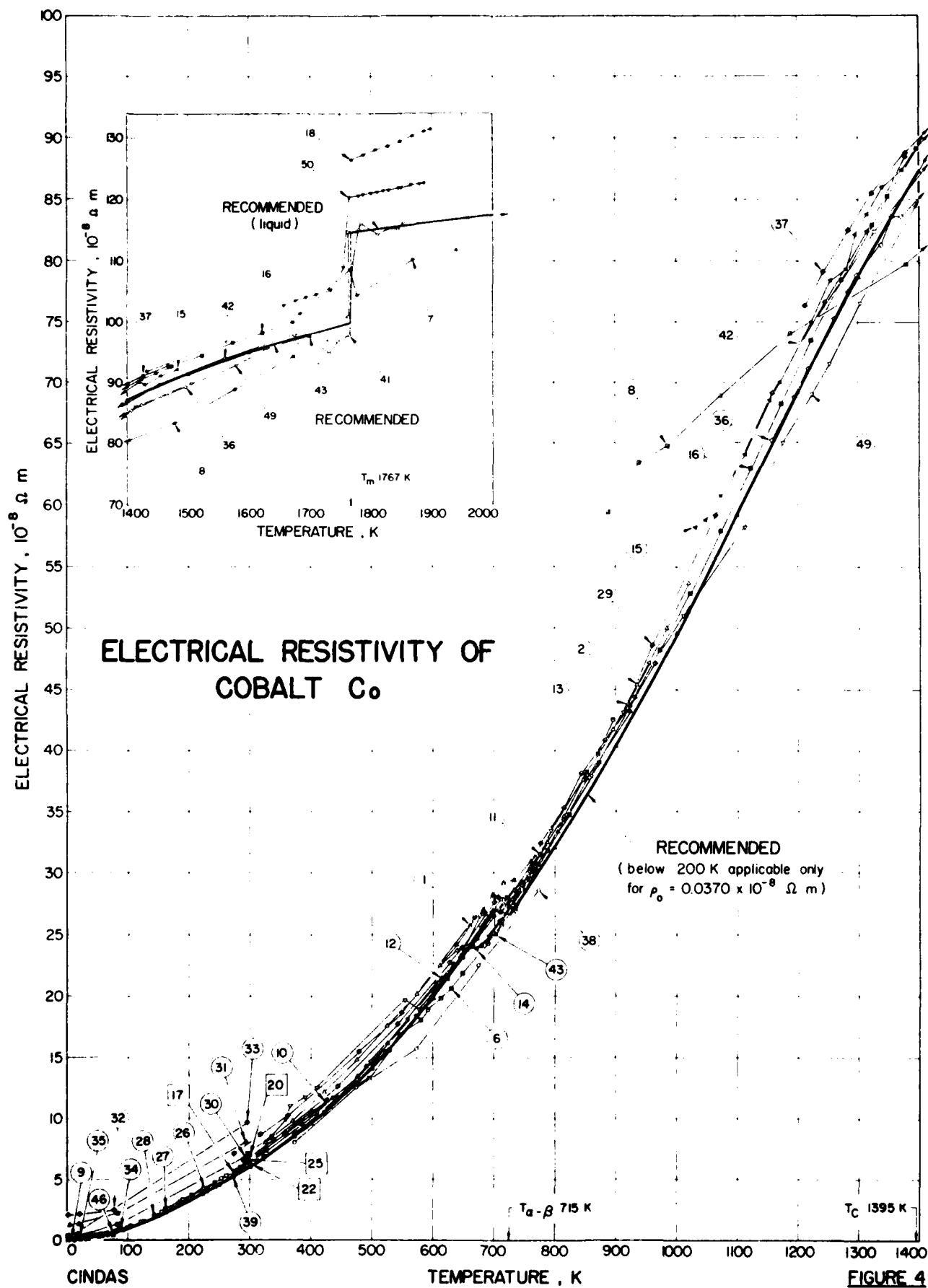


TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	86	Fraser, R.W., Evans, D.J.I., and Mackiv, V.N.	1964		293-1020		99.9 pure; 99.5 Co + Ni, 0.07 Ni, 0.005 Fe, 0.001 Cu, 0.005 S, and 0.008 C by chemical analysis; <0.001 Ag, Al, B, Be, Ca, Cr, Hg, Mg, Mn, Mo, Pb, Sb, and Zn each by spectrographical analysis; 8.99 x 10 ⁻⁶ O ₂ , <2.4 x 10 ⁻⁶ N ₂ , and 1.9 x 10 ⁻⁶ H ₂ for specimen hot rolled to 0.127 cm (0.050 in.); 1.47 x 10 ⁻³ O ₂ , 1.19 x 10 ⁻³ N ₂ , and 4.3 x 10 ⁻⁶ H ₂ for specimen hot rolled to 0.127 cm (0.050 in.) and cold rolled to 0.0762 cm (0.030 in.); 1.31 x 10 ⁻³ O ₂ , 4.8 x 10 ⁻⁶ N ₂ , and 3.9 x 10 ⁻⁶ H ₂ for specimen annealed at 1273 K after hot and cold rolling; gas impurities determined by gas analysis; strip specimens from Sherrit Gordon Mines; prepared from powder; rolled and compacted, sintered in hydrogen, hot rolled and cold rolled to desired thickness, and annealed at 1203 K for 1 h; density 8.85 x 10 ³ kg m ⁻³ ; Rockwell hardness 45T 50-70; α -8 transformation temperature 778 K upon heating and 668 K upon cooling; TC 1394 K; data extracted from heating curve.
2	86	Fraser, et al.	1964		325-1020		The above specimen; data extracted from cooling curve.
3	81	Semenenko, E.E., Sudovtsov, A.I., and Volkenshtein, N.V.	1964	A	1.4-4.2		99.9984 pure; specimens prepared by electric-spark cutting from rod 5 mm in diam; cross section of \sim 0.30 x 0.25 mm ² and \sim 0.35 mm ² long; residual resistivity ratio, R(273 K)/R(0 K) 26.19; values calculated from reported R(T)/R(273 K) with $\rho(273 K) = 5.57 \times 10^{-9} \Omega m$, taken from Bridgman, P.W., Proc. Am. Acad. Arts Sci., 79, 149, 1940; measured with terrestrial magnetic field compensated by means of Helmholtz coils.
4*	82	Olsén-Bär, M.	1956	G	4.2-20.2		Spectroscopically pure wire; obtained from Johnson Matthey Co.; 0.1 mm in diam and 3 to 5 cm long; annealed for several hours in high vacuum at approximately two thirds of the melting temperature by passing a current through it; Debye temperature = 385 K; values calculated from reported $\rho(T)/\rho(90 K)$, with $\rho(90 K) = 0.744 \times 10^{-9} \Omega m$, taken from Data Set 43.
5*	82	Olsén-Bär, M.	1956	G	4.2-20.4		Same as above.
6	96	Chevemushkina, A.V. and Vasil'eva, R.P.	1965		374-895		Cobalt samples consist of an orthogonal parallelepiped with dimensions 3 x 6 x 120 mm; magnetized along the long axis of a solenoid producing magnetic fields up to 3000 Oe; data extracted from figure.
7	88	Eliutin, V.P., Turov, V.D., and Maurakh, M.A.	1965	R	1433-1940		98.5-99.0% pure; electrolytic cobalt; data extracted from figure.
8	97	Kovenskiy, I.I. and Samsonov, G.V.	1963	+	888-1673		99.82 Co, 0.12 C, 0.008 Ni, 0.004 Fe, 0.002 Cu, and 0.001 Mn and S each; measured by a direct-heating method.
9	83	Radhakrishna, P. and Nielsen, N.	1965	D	1.3-6.4		Pure; polycrystalline wire from Johnson Matthey Co.; 1 mm in diam and \sim 6.6 cm long; annealed at 1313 K for 3 h at a pressure less than 4 x 10 ⁻⁵ PA; samples were demagnetized and the earth's field were compensated; data extracted from figure.

* Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
10	85	Powell, R.W.	1964	A, +	293-427		99.97% pure; 0.951 cm in diam and 4.346 cm long; supplied by Metallurgy Division of the National Physical Laboratory; data obtained from figure; resistivity measured by passing a measured current both forward and reverse through the sample; and using thermocouples as potential leads.
11	85	Powell, R.W.	1964	A, +	290-965		Similar to the above specimen except heated to 973 K.
12	85	Powell, R.W.	1964	A, +	290-923		Similar to the above specimen except cooled from 973 to 293 K.
13	85	Powell, R.W.	1964	A, +	290-955		The above specimen heated up to 973 K again.
14	85	Powell, R.W.	1964	A, +	384-756		The above specimen except cooled from 973 to 293 K.
15	98	Schröder, K. and Giannuzzi, A.J.	1969		1032-1483		99.7 pure; samples in wire form; annealed in an inert gas atmosphere consists of 92% He and 8% H for 2 h at ~150 K above the Curie temperature; values calculated from reported resistivity ratio, $\rho(T)/\rho(T_C) = 87.45 \times 10^{-8} \Omega \cdot m$ taken from Data Set 16.
16	78	Kierpe, W., Kohlhass, R., and Gonska, H.	1967	+	77-1673		99.999 ⁺ pure, <0.0001 Ag, Al, Ca and Cu each, 0.0003 Fe and Si each, 0.0002 Mn and 0.0001 Mg; supplied by Koch-Light Laboratories Ltd., England; a non-compensated Thomson bridge is used; values from table.
17	99	Bennett, M.R. and Wright, J.G.	1972	V, +	273	Bulk Co	99.998 ⁺ pure with major impurities of 0.002 C, 0.0001 to 0.02 O ₂ , 0.004 N ₂ , and 0.0001 H ₂ ; amorphous specimen with a cross section of $\sim 5 \times 10 \text{ mm}^2$; supplied by Koch Light Laboratories Ltd.; resistance measured by a standard four-terminal technique using thick silver lands evaporated on to the substrate prior to mounting in the ultra-high vacuum system; a 0.03% Fe in gold-chrome thermocouple from Johnson Matthey Co. was damped to the surface of the substrate for temperature measurement from 4 to 500 K.
18	89, 90	Ono, Y. and Yagi, T. Ono, Y.	1972 1977	R	1768-1898		99.9 ⁺ pure; in molten state in a vacuum induction furnace; values calculated from the reported equation $\rho(\mu\Omega\text{-cm}) = \alpha T(^{\circ}\text{C}) + \beta$ with $\alpha = 0.0384$ and $\beta = 69.07$.
19	75	Marker, D.L., Reichardt, J.W., and Coleman, R.V.	1971		1.1-4.2	Co DIM 20	Cobalt whisker grown by the hydrogen reduction of CoBr ₂ at 673-773 K in a reducing atmosphere of argon; residual resistance ratio R(295 K)/R(4.2 K) = 388; data extracted from figure of R(T)/R(295 K) as function of square of temperature; reference value of $\rho(295 \text{ K}) = 5.8 \mu\Omega\text{cm}$, taken from White and Woods, Phil. Trans. Roy. Soc., London, <u>A251</u> , 273, 1959; used to calculate resistivity.
20	100	Plewe, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 13	99.999 pure Co; 0.0250 N, 0.0190 O, <0.0010 C, <0.0003 Si, Mg and Fe each; determined by emission spectrograph and vacuum fusion analysis; samples consist of 0.025 cm thick strips cold rolled to 0.011 cm, cut to ~20 cm in length by 0.25 cm in width; supplied by Johnson Matthey Co.; the center of the hysteresis gap found at 690 K; resistivity was monitored by superimposing an ac voltage on the ramp and measuring differentially the ac voltage drop across the specimen against an 0.010 Ω standard using a phase sensitive detection; reproducibility of measurement 0.1%.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
21*	100	Pleves, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 13	Similar to the above specimen except heated to 573 K for 46 h after cold rolled, representing a fully recovered structure.
22	100	Pleves, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 13	The above specimen after 100 heat cycles carried out in a diffusion-pumped vacuum system which could be evacuated to 1.33×10^{-6} PA.
23*	100	Pleves, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 11	Similar to the above specimen except only cold rolled.
24*	100	Pleves, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 11	The above specimen heated to 1073 K for 10 minutes representing a fully recrystallized structure.
25	100	Pleves, J.T. and Bachmann, K.J.	1973	D	299	Specimen No. 11	The above specimen after 100 heat cycles carried out in a diffusion-pumped vacuum system which could be evacuated to 1.33×10^{-6} PA.
26	80	Price, D.C. and Williams, G.	1973	V	4.2-292		99.9985% pure; dimensions of 10 cm x 0.2 cm x 0.15 cm; supplied by Johnson Matthey Co.; annealed in vacuo for 2 h at 1173 K and then quenched; ideal resistivity ρ_i were reported from 22 to 292 K; data from table, uncorrected for thermal expansion; total resistivity calculated from data of ideal resistivity by the relation $\rho(T) = \rho_i(T) + \rho(4.2 K)$; temperatures stabilized and measured to better than 0.5%; area to length ratio determined to within 0.5%.
27	77	White, G.K. and Woods, S.B.	1957	G	4.2-286	Col. a	Pure; 0.0002 Si, <0.0005 Fe, <0.0001 Al, and <0.0001 Mg and Cu each; by spectrographic analysis; rod specimen 5 to 8 cm long and 2 mm in diam; supplied by Johnson Matthey Co.; annealed in vacuum for 2 h at 973 K; residual resistance ratio $\rho(295 K)/\rho(4.2 K) = 65.36$; total resistivity calculated using $\rho = \rho_i + \rho_0$; ideal resistivity ρ_i extracted from figure; measurement error $\sim 1\%$.
28	77	White, G.K. and Woods, S.B.	1957	G	4.2-279	Col. b	The above specimen remounted in a second cryostat and resistivity and thermal conductivity determined together; residual resistance ratio $\rho(295 K)/\rho(4.2 K) = 64.52$.
29	101	Tsoukalas, I.A.	1974		273-1378		99.99 (nominal) pure; polycrystalline; dimension 1 x 5 x 0.1 cm ³ ; martensitic transformation at 660 K form hcp to fcc; values extracted from figure.
30	50	McLennan, J.C., Niven, C.D., and Wilhelm, J.O.	1928		2.5-293	Cobalt(aged)	Pure; supplied by Belga American Trading Corp., New York; cut into strip and annealed in vacuum for 4 h at a dull red heat; values extracted from table.
31	50	McLennan, J.C., et al.	1928		4.2-293	Cobalt(unaged)	Similar to the above specimen, unannealed.
32	102	Meissner, W.	1928		1.4-273	CoI(27)	Specimen annealed for 2.5 h at 600 K; 50 mm in length and 2.5 x 0.5 mm cross section; resistance ratio reported; reference value of $\rho(273) = 5.57 \mu\Omega \text{ cm}$, taken from Bridgman, P.W., Proc. Am. Acad. Arts Sci., 79, 149, 1940, used to calculate resistivity from resistance ratio.

* Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

Date Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
33	103	Horak, J.A. and Blewitt, T.H.	1972	A	4.5-295		Polycrystalline; ≈ 5 cm long, 0.025 cm diam; annealed, values extracted from table.
34	104	Meissner, W. and Voigt, B.	1930	+	1.5-273	Co 2	Sample in sintered polycrystalline form; 0.5 mm in diam and 57.6 mm in length; specimen obtained from Heraeus, von A.E.G.; sample annealed at 773 K in vacuum for 2.5 h; measured by compensation method; relative resistance data reported; reference value of $\rho(273) = 5.57 \mu\Omega$ cm, taken from Bridgman (Proc. Am. Acad. Arts Sci., 79, 149, 1940), used to calculate resistivity.
35	104	Meissner, W. and Voigt, B.	1930	+	1.3-273	Co 3	Pure; 0.05 Cr, 0.01 Mn and <0.05 Fe; dimension 12.5 mm in length and 2.5 mm thick; obtained from Kahlb; sample melted in vacuum; measured by compensation method; reference value of $\rho(273) = 5.57 \mu\Omega$ cm, taken from Bridgman (Proc. Am. Acad. Arts Sci., 79, 149, 1940), used to calculate resistivity.
36	105	Jain, S.C., Narayan, V., and Goel, T.C.	1969	B, +	1158-1496	Specimen 1	99.998 pure; specimen in rod form 15 cm long and 0.5 cm in diam; supplied by Koch Light Laboratories, United Kingdom; sample heated by electric current; the potential drop across the length of the uniform-temperature region was measured with a Tinsley ac/dc coordinate potentiometer type 4580; current through the sample determined by measuring the potential difference across a non-inductive standard resistor type 660 of 0.001Ω in series with the specimen; values extracted from graph.
37	105	Jain, S.C., et al.	1969	B	1213-1468	Specimen 2	Similar to the above specimen except contains a total impurity concentration of about 0.001% of Si, Ni, Cu, Fe, Mg and Ag; sample supplied by Johnson Matthey Co.
38	106	Kirichenko, P.I.	1969		373-773		99.7 pure; 3 mm in diam and 30 cm long; annealed in vacuum for 24 h at 1273, oven-cooled; measured in a vacuum of 4×10^{-4} mmHg; measurement error: 1-1.5% in resistivity and 0.1 K in temperature; minima of resistivity versus temperature reported to occur at 655 K upon heating and 655 K upon cooling; values extracted from figure.
39	21	White, G.K. and Woods, S.B.	1959	C	4.2-273	Co 2	99.999 pure; 0.0002 Si, 0.0005 Fe, ≈ 0.0001 Al, and <0.0001 Mg and Cu; wire specimen 0.05 mm in diam and about 6 to 8 cm long, from Johnson Matthey Co. (JM9484); annealed in vacuum at 973 K; Debye temperature reported to be 380 K; residual resistance ratio $R(295 K)/R(0 K) = 90.9$; values calculated from reported ideal resistivity, extracted from graph, and reported $\rho_0 = 0.062 \times 10^{-8} \Omega$ m from Table 1.
40*	91	Levin, E.S., Ayushina, G.D., and Gel'd, P.V.	1972	R	1923		99.98 pure; electrolytic.

* Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
41	92	Güntherodt, H.J., Hauser, E., Künsi, H.U., and Müller, R.	1975	A	1720-1854		99.999 pure from Johnson Matthey Co.; measured by a four probe method in which the sample material was enclosed in an alumina tube with four protrusion serving as current and potential contacts.
42	87	Seydel, U. and Fucks, W.	1977	→	1205-3108		99.99 pure, 0.0007 Fe, 0.0005 Si, 0.0003 Cu, 0.0002 Ag and Ni, 0.0001 Al, Ca, Mg, and Sn each, <0.0001 Bi, Cr, and Mn each; measured by an exploding wire technique; measurement error 4%; smoothed values from curve; values corrected for thermal expansion.
43	76	Laubitz, M.J. and Matsumura, T.	1973	A	90-1700		99.999 pure, 0.00070 C, 0.00060 Ni, 0.00050 O ₂ , 0.00016 Fe, 0.00010 K, 0.00008 N ₂ , 0.00007 Na, 0.00004 Mo, 0.00003 Al, 0.00002 Ga and S each, 0.00001 P, 0.000007 Cl, 0.000006 Ca, Cr, and Cu each, 0.000002 Mg, and 0.000008 Ag and Pd each (at.Z), by semi-quantitative mass spectrographic analysis; from Metals Research Ltd., England; material originally rod shape $\sqrt{2}$ cm in diam and 20 cm long; polycrystalline; annealed at 1500 K for 4 h in a vacuum of 5×10^{-6} Torr; cooled at 100 K hr ⁻¹ , except in the range 710 to 670 K, where it is cooled at 0.5 K hr ⁻¹ ; residual resistance ratio 140 ± 10 ; density 8.831×10^3 kg m ⁻³ at 293 K; grain size ~ 0.1 cm; specimen trimmed to a nominal diam of 2 cm and length of 20 cm with no machining of the region on which measurements were made; one specimen then measured from 300 to 1250 K; residual resistivity ratio after measurement 120; a second specimen, 1 cm in diam and 10 cm long were cut from the first and measured from 90 to 370 K; no change in residual resistivity ratio, and in ice point resistivity were detected; the second specimen measured again from 1000 to 1750 K; grain size increases to 0.5-1.0 cm after high temperature measurement; residual resistivity ratio unchanged; but specimen geometry is changed; smoothed values from table, these values are reported to be averages in the temperature range where the two measurements overlap; values uncorrected for thermal expansion; values above 1300 K had been adjusted by +0.8% by the authors to avoid discontinuity in resistivity values between the large and the small specimens; measurement error reported 0.5%; smoothed values from table.
44*	76	Laubitz, M.J. and Matsumura, T.	1973	A	680-710		The smaller of the above specimens, measured after the high temperature measurements; measured while heating.
45*	76	Laubitz, M.J. and Matsumura, T.	1973	A	685-715		The above measured while cooling.
46	95	Wilkes, K.E.	1968	A	78-300		99.92 pure, 0.040 Fe, 0.012 Ni, 0.004 C, 0.001 Ca, Cu and Si each, 0.0008 S, 0.0003 Al and Mn each, 0.0002 Mg, and 0.0001 Pb; rod specimen 1.000 cm in diam and 10.05 cm long; supplied by Centre D'information on Cobalt, Brussels, Belgium; density 8.805×10^3 kg m ⁻³ at 296 K; values from table; residual resistivity ratio measured by M.J. Laubitz and T. Matsumura, Can. J. Phys., 51(2), 1247, 1973, to be 31, and reported to change to 48 after being "carefully reannealed".

* Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

Date Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
47*	84	Loegel, B. and Gautier, F.	1973		6.8-79		"Pure cobalt," no other details reported; only temperature dependent part of resistivity reported.
48*	84	Loegel, B. and Gautier, F.	1973		5.6-33		Similar to the above.
49	107	Zinov'yev, V.F., Krentsis, R.P., Perova, I.N., and Gel'd, P.V.	1968	A, R	296-1730		99.95 pure; 0.2 mm thick, 8 mm wide and 8 mm long; ground from rolled stock; annealed at 1200 K for 7 h under a pressure of 1×10^{-5} mmHg; $\rho(288 \text{ K})/\rho(4.2 \text{ K}) = 86$; α - β transition reported at 703 K; measured by potentiometric method below 1330 K, and by rotating field method above 1400 K; hysteresis at α - β transition reported but resistivity values given for heating only; uncertainty in temperature measurement 10-15 degrees.
50	93	Kita, Y., Ohguchi, S., and Morita, Z.	1978	+	1658-1888		0.137 Fe, 0.09 Ni, 0.017 Si, 0.012 Mn, 0.011 S, and 0.008 C; measured in a vacuum of 10^{-4} Torr, with a four probe method in which the electrodes are of the same material as the specimen; data points are taken at temperatures in the following sequence: 1786, 1811, 1829, 1850, 1870, 1888, 1881, 1867, 1846, 1820, 1801, 1781, 1765, 1755, 1735, 1711, 1694, 1678 and 1658 K; values from table supplied by authors.
51*	93	Kita, Y., et al.	1978	+	1764-1895		Same as the above, a second melt; temperature sequence: 1795, 1809, 1823, 1843, 1863, 1878, 1895, 1883, 1869, 1854, 1836, 1819, 1801, 1782, and 1764 K.
52*	94	Samarin, A.M.	1962	R	1767-2000		Measured by the rotating field method; apparatus calibrated with iron; using resistivity value reported by R.W. Powell, Philos. Mag., 44, 772, 1953; resistivity value calculated from reported conductivity $(1.12-0.228 \times 10^{-3} T(C)) \times 10^4 \Omega^{-1} \text{ cm}^{-1}$; upper temperature limit issued to be 2000 K.
53*	108	Shimank, H.	1914	A	20.2-273		Wire specimen 1-2 m long; resistivity values calculated from reported $R(T)/R(273 \text{ K})$ ratio, with $\rho(273 \text{ K})$ taken to be $5.178 \times 10^{-6} \Omega \text{ m}$.
54	109	Thomas, J.G. and Mendoza, E.	1952		1.2, 4.2		99.95 pure, from New Metals and Chemical Ltd., 0.13 mm in diam; drawn; x-ray show hcp structure; resistivity value calculated from reported $R(T)/R(273 \text{ K})$ with $\rho(273 \text{ K})$ taken to be $5.178 \times 10^{-6} \Omega \text{ m}$.
55	109	Thomas, J.G. and Mendoza, E.	1952		0.06-4.2		Similar to the above except annealed for 3 h in vacuo at 1273 K; contains a small amount of fcc structure.

* Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF COBALT Co

[Temperature, T, K; Electrical Resistivity, ρ , $10^{-6} \Omega m$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ										
DATA SET 1																											
293	6.3	983	50.0	4.16	0.1499	1765	108.6	293	6.46	290	6.45	644	23.18														
325.15	8.0	1020	53.7	4.80	0.1500	1773	104.3	294	6.27*	701	28.00	650	23.58*														
370	9.8	DATA SET 2																									
422	12.2	DATA SET 3																									
476	14.8	1.41	0.212755	6.10	0.1502	1868	110.3	324	7.43	DATA SET 4*																	
525	17.6	1.61	0.212750*	5.22	0.1501	1940	111.8	327	7.63	DATA SET 5																	
573	20.2	1.78	0.212765*	5.75	0.1501	DATA SET 6																					
611	22.5	1.99	0.212773	6.62	0.1504	DATA SET 7																					
638	24.2	2.22	0.212817*	7.24	0.1504	DATA SET 8																					
660	25.7	2.22	0.212817*	8.54	0.1506	DATA SET 9																					
682	27.0	2.36	0.212815	9.08	0.1507	DATA SET 10																					
698	28.2	2.60	0.212812*	9.54	0.1507	DATA SET 11																					
715	29.2	2.80	0.212831*	10.15	0.1508	DATA SET 12																					
733	29.4	3.19	0.212855	10.83	0.1511	DATA SET 13																					
761	30.9	3.38	0.212868*	11.56	0.1512	DATA SET 14																					
788	32.5*	3.57	0.212893*	12.43	0.1519	DATA SET 15																					
814	34.6*	3.77	0.212912*	20.38	0.1547	DATA SET 16																					
847	37.6*	4.00	0.212944*	DATA SET 17																							
895	41.7*	4.18	0.212948	DATA SET 18																							
983	50.0*	DATA SET 19																									
1020	53.7*	DATA SET 20																									
DATA SET 21																											
325.15	8.0	4.17	0.1578	374	8.8	1.28	0.087146	370	9.3	1.49	0.087151	290	6.06														
370	9.8	4.96	0.1578	391	9.3	1.49	0.087151	391	9.3	1.67	0.087151	327	7.35														
422	12.2	5.24	0.1576	443	11.7	1.67	0.087151	443	11.7	1.87	0.087162	410	10.51														
476	14.8	5.86	0.1579	496	13.3	1.87	0.087162	496	13.3	2.12	0.087168	488	14.71														
525	17.6	6.52	0.1581	529	15.6	2.12	0.087168	529	15.6	2.31	0.087176	498	14.71														
573	20.2	7.41	0.1581	546	16.8	2.31	0.087176	546	16.8	2.49	0.087187*	626	22.07														
611	22.5	8.21	0.1581	580	18.0	2.49	0.087187*	580	18.0	2.74	0.087200*	644	23.39*														
638	23.7	9.08	0.1566	592	18.9	2.74	0.087200*	592	18.9	2.96	0.087212	679	25.67														
664	24.4	10.50	0.1585	612	19.8	2.96	0.087212	612	19.8	3.16	0.087223*	695	26.76														
683	25.7	11.29	0.1589	630	20.6	3.16	0.087223*	630	20.6	3.34	0.087238	701	27.14														
700	26.7	11.66	0.1584	649	21.8	3.34	0.087238	649	21.8	3.56	0.087248*	705	27.75*														
716	27.7	12.26	0.1595	690	24.3	3.56	0.087248*	690	24.3	3.75	0.087263*	711	27.83														
733	28.6	12.40	0.1584	712	25.8	3.75	0.087263*	712	25.8	3.96	0.087281*	714	27.88*														
761	30.5	13.35	0.1585	733	27.1	3.96	0.087281*	733	27.1	4.22	0.087302	718	28.00														
788	32.5	14.11	0.1589	752	28.6	4.22	0.087302	752	28.6	4.61	0.087337*	721	27.51														
814	34.6	14.88	0.1594	764	30.3	4.61	0.087337*	764	30.3	4.82	0.087355*	721	27.73														
847	37.6	16.08	0.1600	789	31.8	4.82	0.087355*	789	31.8	5.11	0.087381	725	27.59*														
895	41.7	16.75	0.1604	815	34.6*	5.11	0.087381	815	34.6*	5.28	0.087400*	730	27.88														
915	45.4	18.85	0.1612	851	38.2	5.49	0.087423*	851	38.2	5.69	0.087445*	735	28.13														
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DATA SET 95																											

* Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

T	p	T	p	T	p	T	p	T	p	T	p	T	p	T	p
<u>DATA SET 14</u>															
384	9.44	1428	89.72	273	5.6	299	6.90	131.8	1.7419	4.2	0.0907*	273	7.1	4.2	0.0907*
458	12.39	1455	91.04	<u>DATA SET 18</u>		<u>DATA SET 21*</u>		141.9	1.9810*	12.91	0.0922	317	8.7	12.91	0.0922
539	18.15	1458	91.21*	<u>DATA SET 19</u>		<u>DATA SET 24*</u>		156.3	2.3355	15.31	0.0933	362	10.2	15.31	0.0933
605	21.05	1461	91.82*	1.18	0.013658	299	6.58	171.8	2.7038	18.07	0.0958*	479	15.5	18.07	0.0958*
626	22.14*	1483	92.17	1.31	0.013668	<u>DATA SET 25</u>		182.0	2.9524*	21.53	0.0991	549	18.7	21.53	0.0991
645	23.24*	<u>DATA SET 16</u>		1.49	0.013675	<u>DATA SET 26</u>		191.4	3.2178	24.83	0.1048	668	26.4	24.83	0.1048
652	23.66	77	0.44	1.60	0.013680	<u>DATA SET 27</u>		206.5	3.5710	27.42	0.1122	684	26.8	27.42	0.1122
659	23.94*	173	2.65	1.78	0.013684	4.2	0.0902	225.9	4.2580	31.05	0.1250	698	26.8	31.05	0.1250
665	24.02	223	3.84	1.97	0.013698	11.56	0.0914	253.5	5.0140	34.36	0.1418	711	26.8	34.36	0.1418
667	24.00*	273	2.65	2.08	0.013705	12.74	0.0921	261.2	5.2917	39.81	0.1707	745	28.7	39.81	0.1707
669	23.92*	323	5.25	2.19	0.013711	14.59	0.0926	271.0	5.4120	63.10	0.2786	777	32.4	63.10	0.2786
671	23.89	373	6.81	2.28	0.013715	16.07	0.0943*	292.4	6.1254*	76.38	0.4726	815	35.3	76.38	0.4726
673	23.85*	423	8.52	2.40	0.013732	18.20	0.0958	<u>DATA SET 28</u>		88.31	0.7970	<u>DATA SET 29</u>		88.31	0.7970
674	23.87*	473	10.50	2.50	0.013728	22.03	0.0998	299	6.91	143.5	2.032	273	7.1	143.5	2.032
676	23.91*	523	15.45	2.59	0.013740	26.24	0.1088	<u>DATA SET 20</u>		279.3	5.688	273	7.1	279.3	5.688
677	24.02*	573	18.38	2.69	0.013746	28.91	0.1172*	299	6.90	<u>DATA SET 26(cont.)</u>		273	7.1	<u>DATA SET 27</u>	
680	24.22	623	21.48	2.80	0.013758	31.8	0.1660	299	6.90	131.8	1.7419	273	7.1	131.8	1.7419
684	24.63*	673	24.72	2.89	0.013765	35.3	0.1788	299	6.90	141.9	1.9810*	273	7.1	141.9	1.9810*
709	26.16	698	26.55	3.00	0.013772	38.7	0.1965	299	6.90	156.3	2.3355	273	7.1	156.3	2.3355
725	27.29	728	28.00	3.09	0.013780	43.1	0.2208	299	6.90	171.8	2.7038	273	7.1	171.8	2.7038
734	28.00	773	30.65	3.19	0.013790	48.5	0.2662	299	6.90	182.0	2.9524*	273	7.1	182.0	2.9524*
756	29.45	823	34.72	3.29	0.013797	53.5	0.3203	299	6.90	191.4	3.2178	273	7.1	191.4	3.2178
<u>DATA SET 15</u>															
1032	58.24	873	39.05	3.50	0.013822	60.7	0.3954	299	6.90	206.5	3.5710	273	7.1	206.5	3.5710
1049	58.85	923	43.66	3.59	0.013834	65.3	0.4726	299	6.90	225.9	4.2580	273	7.1	225.9	4.2580
1074	60.87	973	48.20	3.69	0.013843	71.6	0.5660	299	6.90	253.5	5.0140	273	7.1	253.5	5.0140
1114	64.10	1023	52.85	3.79	0.013859	73.3	0.5974*	299	6.90	261.2	5.2917	273	7.1	261.2	5.2917
1133	68.56	1073	57.92	3.89	0.013867	77.4	0.6547	299	6.90	271.0	5.4120	273	7.1	271.0	5.4120
1154	68.56	1123	62.95	3.99	0.013875	81.3	0.7177	299	6.90	292.4	6.1254*	273	7.1	292.4	6.1254*
1171	70.05	1173	68.20	4.09	0.013888	84.7	0.7791*	299	6.90	<u>DATA SET 20</u>		273	7.1	<u>DATA SET 27</u>	
1223	74.94	1223	73.48	4.18	0.013898	89.7	0.8671*	299	6.90	299	6.90	273	7.1	299	6.90
1231	76.69	1323	82.97	4.18	0.013898	93.8	0.9612*	299	6.90	131.8	1.7419	273	7.1	131.8	1.7419
1255	78.44	1348	85.27	4.18	0.013898	100.0	1.0694	299	6.90	141.9	1.9810*	273	7.1	141.9	1.9810*
1280	79.32	1373	87.45*	4.18	0.013898	105.2	1.1966*	299	6.90	156.3	2.3355	273	7.1	156.3	2.3355
1283	80.63*	1398	89.17	4.18	0.013898	113.2	1.3298	299	6.90	171.8	2.7038	273	7.1	171.8	2.7038
1295	82.12	1423	90.60	4.18	0.013898	122.2	1.563	299	6.90	182.0	2.9524*	273	7.1	182.0	2.9524*
1316	83.86	1448	91.62	4.18	0.013898	122.2	1.563	299	6.90	191.4	3.2178	273	7.1	191.4	3.2178
1321	83.86*	1473	92.65	4.18	0.013898	122.2	1.563	299	6.90	206.5	3.5710	273	7.1	206.5	3.5710
1340	86.05	1523	94.35	4.18	0.013898	122.2	1.563	299	6.90	225.9	4.2580	273	7.1	225.9	4.2580
1370	87.45	1573	96.52	4.18	0.013898	122.2	1.563	299	6.90	253.5	5.0140	273	7.1	253.5	5.0140
1373	87.45	1623	98.17	4.18	0.013898	122.2	1.563	299	6.90	261.2	5.2917	273	7.1	261.2	5.2917
1407	88.85	1673	99.95	4.18	0.013898	122.2	1.563	299	6.90	271.0	5.4120	273	7.1	271.0	5.4120
1420	90.07			4.18	0.013898	122.2	1.563	299	6.90	292.4	6.1254*	273	7.1	292.4	6.1254*
1424	89.72*			4.18	0.013898	122.2	1.563	299	6.90	299	6.90	273	7.1	299	6.90

* Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	$\rho - \rho_0$		
<u>DATA SET 30(cont.)</u>													
83	1.24	1261	75.2	1720	95.8	800	32.056	690.8	24.786	41.1	0.0701		
293	6.85	1284	77.4	1732	94.8	900	40.377	691.0	24.561	44.1	0.0807		
<u>DATA SET 31</u>													
		1298	78.8	1746	96.8	1000	49.562	691.4	24.481	46.1	0.0998		
		1338	81.3	1765	97.8	1100	59.259	692.0	24.348	50.6	0.118		
		1356	83.6	1769	109.3	1200	69.116	693.0	24.308	50.6	0.132		
4.2	1.24	1372	83.6	1780	115.1	1250	73.995	694.0	24.348	78.6	0.507		
20.6	1.37	1407	85.4	1784	116.1	1300	78.78	694.8	24.467	<u>DATA SET 48*</u>			
83	2.22	1423	86.3	1811	114.3	1400	87.17	696.0	24.493	5.61	0.000189		
293	8.07	1434	86.5*	1828	115.3	1500	91.37	698.0	24.625	6.44	0.000545		
<u>DATA SET 32</u>													
		1470	88.1	1844	115.3	1600	94.86	700.2	24.757	7.40	0.000659		
		1488	89.6*	1854	116.3	1700	97.62	702.0	24.903	7.57	0.000834		
		1496	89.2	<u>DATA SET 44*</u>								8.13	0.000628
<u>DATA SET 33</u>													
1.35	2.0230	<u>DATA SET 42</u>											
4.22	2.0498	1205	73.2	1205	73.2	679.8	25.212	679.8	25.212	8.51	0.000795		
20.44	2.1004	1327	82.4	1327	82.4	684.8	25.595	684.8	25.595	9.56	0.000916		
78.24	2.4653	1424	79.1	1424	89.7	688.2	25.846	688.2	25.846	11.5	0.00127		
273.2	5.57	1284	82.5	1561	93.9	696.7	26.493	696.7	26.493	18.3	0.00368		
<u>DATA SET 34</u>													
4.5	0.0896	1323	85.5	1675	97.8	697.9	26.599	697.9	26.599	21.0	0.00512		
295	9.677	1379	88.8	1763	100.8	699.1	26.652	699.1	26.652	22.0	0.00604		
<u>DATA SET 35</u>													
1.51	0.2401	1427	91.1	1763	114.7	700.3	26.744	700.3	26.744	23.1	0.00801		
4.20	0.2373	1468	92.7	1959	117.0	701.1	26.823	701.1	26.823	24.7	0.00923		
20.42	0.2578	<u>DATA SET 38</u>											
77.78	0.8444	373	8.00	2295	120.7	702.3	26.929	702.3	26.929	28.4	0.0196		
86.92	1.0188	473	12.60	2442	122.2	703.1	26.955	703.1	26.955	30.4	0.0273		
273.2	5.57	573	15.70	2584	123.9	703.5	26.942	703.5	26.942	32.6	0.0314		
<u>DATA SET 36</u>													
1.36	0.3476	673	22.50	2718	125.6	703.9	26.862	703.9	26.862	<u>DATA SET 47*</u>			
4.22	0.3387	773	28.50	2869	127.2	704.5	26.478	704.5	26.478	6.75	0.000357		
20.41	0.3258	2982	128.4	2982	128.4	705.3	26.107	705.3	26.107	7.41	0.000357		
78.30	0.8845*	3057	129.3	3057	129.3	706.0	25.763	706.0	25.763	7.94	0.000521		
88.16	1.0589	3108	129.8	3108	129.8	706.4	25.630	706.4	25.630	9.56	0.000983		
<u>DATA SET 37</u>													
4.2	0.062	<u>DATA SET 43</u>											
126	1.512	90	0.744	90	0.744	707.2	25.550	707.2	25.550	10.7	0.00108		
20.41	0.3258	100	0.939	100	0.939	708.0	25.510	708.0	25.510	11.0	0.00113		
77.78	0.8444	125	1.461	125	1.461	708.8	25.497	708.8	25.497	11.0	0.00113		
86.92	1.0188	150	2.018	150	2.018	710.0	25.497	710.0	25.497	12.6	0.00161		
<u>DATA SET 38</u>													
1.36	0.3476	200	3.214	200	3.214	714.2	25.516	714.2	25.516	14.2	0.00186		
4.22	0.3387	250	4.062	250	4.062	715.5	25.516	715.5	25.516	15.5	0.00335		
20.41	0.3258	284	4.402	284	4.402	716.6	25.542	716.6	25.542	21.5	0.00783		
78.30	0.8845*	250	4.582*	300	5.995	717.6	25.515	717.6	25.515	23.6	0.0117		
88.16	1.0589	273	5.182	400	9.542	718.6	25.462	718.6	25.462	27.1	0.0155		
<u>DATA SET 39</u>													
115.8	65.2	500	14.118	500	14.118	719.4	25.462	719.4	25.462	29.7	0.0243		
1195	68.8	600	19.872	600	19.872	720.0	25.382	720.0	25.382	34.2	0.0389		
1218	71.1	700	26.590	700	26.590	720.0	25.289	720.0	25.289	36.6	0.0469		
		900	33.4	900	33.4	720.0	25.051	720.0	25.051	38.3	0.0567		
<u>DATA SET 40*</u>													
1923	109	701	25.015	701	25.015	<u>DATA SET 45(cont.)</u>					296	7.1	
<u>DATA SET 41</u>													
<u>DATA SET 42</u>													
<u>DATA SET 43(cont.)</u>													
<u>DATA SET 44*</u>													
<u>DATA SET 45(cont.)</u>													
<u>DATA SET 46</u>													
<u>DATA SET 47*</u>													
<u>DATA SET 48*</u>													
<u>DATA SET 49</u>													
<u>DATA SET 50</u>													

* Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF COBALT Co (continued)

T	ρ	T	ρ
<u>DATA SET 49 (cont.)</u>			
1224	69.0	1883	122.8
1253	71.5	1895	123.05
1304	76.5	<u>DATA SET 52*</u>	
1414	86.0	1767	128.3
1506	89.9	1800	129.6
1579	92.7	1900	133.5
1627	95.6	2000	137.7
1730	98.5	<u>DATA SET 53*</u>	
<u>DATA SET 50</u>			
1658	102.9	20.2	2.02
1678	103.5	81.8	2.38
1694	104.0	198.4	3.89
1711	104.5	273.09	5.18
1735	105.25	<u>DATA SET 54*</u>	
1755	109.0	1.20	0.828
1765	120.4	4.2	0.839
1781	120.75	<u>DATA SET 55*</u>	
1788	120.85	0.06	0.157
1801	121.1	1.20	0.157
1811	121.25	1.36	0.156
1820	121.5	4.2	0.156
1829	121.6	<u>DATA SET 51*</u>	
1846	121.95	1764	120.5
1850	122.0	1782	120.8
1867	122.35	1795	121.15
1870	122.4*	1801	121.2
1881	122.6	1809	121.4
1888	122.75	1819	121.55
<u>DATA SET 51*</u>			
1764	120.5	1823	121.65
1782	120.8	1836	121.9
1795	121.15	1843	122.05
1801	121.2	1854	122.25
1809	121.4	1863	122.4
1819	121.55	1869	122.5
1823	121.65	1878	122.7
1836	121.9		
1843	122.05		
1854	122.25		
1863	122.4		
1869	122.5		
1878	122.7		

* Not shown in figure.

3.3. Iron

The electrical resistivity of iron has been studied extensively. There are 223 sets of experimental data available for iron specimens of purity 99.8% or higher. The information on specimen characterization and measurement condition for each of the data sets is given in table 8. The data are tabulated in table 9 and shown partially in figures 5 and 6.

Because of magnetic effects, the residual resistivity of iron has been studied with intense interest. Berger and De Vroomen [111] suggested that the residual resistivity measured in the absence of an applied magnetic field is not an indication of the purity of an iron specimen. Fujii and Morimoto [112] suggested that the magnetic contribution to the residual resistivity is $\sim 0.02 \times 10^{-8} \Omega \text{m}$; a value also agreed upon by Volkenshteyn and Yakina [113]. Thus, even for zone-refined iron, the residual resistance ratio, RRR, has low values of ≤ 400 as compared with values of a few thousand or even higher for other pure metals. The more recent results of Takaki and Igaki [114] seem to indicate that even for iron that has been highly purified by anion exchange separation, floating-zone melting and hydrogen treatment, the residual resistance ratio has a limiting value of about 500. It is worth noting, however, that the quantity RRR_H , the residual resistance ratio obtained as the minimum value from resistivity measurement as a function of an applied longitudinal magnetic field, of some of their specimens are greater than 2000. The same group of authors [115] later reported RRR_H values of over 10,000 on specimens that had been electrolytically polished to remove silicon contamination on the surface layer (of thickness $\sim 100 \mu\text{m}$). The resistivity of iron is also dependent on the density of the measuring current [114]. The results of Takaki and Igaki [114] on specimens of various purity showed that the current density dependence is negligible for specimens having RRR (or RRR_H) values of ≤ 100 . For RRR values higher than 200, or RRR_H values of ≥ 1000 , the resistivity may still be increasing with measuring current at current density values of $> 6 \times 10^6 \text{ A m}^{-2}$. The limiting RRR value of 500 quoted above was for a measuring current density of $\sim 5 \times 10^5 \text{ A m}^{-2}$ [114]. This result appears to agree with that of Glaeser et al. [116], even though the magnetic field and current density dependences reported by these two works show some discrepancies. The recommended value for the electrical resistivity of iron at the temperature of 1 K is based on these

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references and subject to the condition that the measuring current is $<1 \times 10^6 \text{ A m}^{-2}$ in the absence of an applied magnetic field. It should represent the value for the electrical resistivity of iron purified by modern electron-beam zone-refining techniques.

The electrical resistivity of iron is dependent on other factors. The most notable factors are the external magnetic field and the magnetic domain structure of the specimen itself. The effects of the former have been investigated mostly at 4 K [112,116-120], and those of the latter have been investigated chiefly for single crystal whisker specimens [121-125]. These effects, together with the effect of the measuring current, are inter-related and are not fully investigated. A resolution of these effects is clearly beyond the scope of this work. Hence, only the conditions "in the absence of an external applied magnetic field" and "with a measuring current of $<1 \times 10^6 \text{ A m}^{-2}$ " are specified. The latter condition is chosen such that the (transverse) magneto-resistance due to the self-induction of the measuring current would not adversely affect the resistivity value.

The electrical resistivity of iron at low temperatures has been reported to contain a T^2 component; see, for example, White and Woods [21], Volkenshtein and Yakina [113], Semenenko and Sudovtsov [126], Fert and Campbell [127], Price and Williams [80], and Janos et al. [128]. However, Kondorskii et al. [129] reported $T^{1.5}$ and T^5 components. More recently, Isshiki and Igaki [115] reported that the temperature dependent part of the resistivity, when measured in a longitudinal magnetic field of 60 K A m^{-1} , can be fitted to a T^2 term and a Bloch-Grüneisen term with $\theta_R = 467 \text{ K}$ for temperatures 1 to 250 K.

A similar analysis [115] on the data of Kemp et al. [130] (data set 31), Kemp et al. [131] (data set 11), White and Woods [21] (data set 86), Fert and Campbell [127] (data set 119), Schwerer et al. [132] (data set 107), Volkenshtein and Yakina [113] (data set 209), and of Hust and Giarratano [133] (data set 47) indicates that, even without an applied longitudinal magnetic field, the electrical resistivity of iron can be represented by eq (8) for temperatures up to about 100 K. The values of A and θ_R in the Bloch-Grüneisen term were taken to be $58.1 \times 10^{-8} \Omega \text{ m}$ and 467 K [115], respectively. The coefficient of the quadratic term varies approximately from 1.1 to $3.4 \times 10^{-13} \Omega \text{ m K}^{-2}$. There seems to be some correlation between the values of the coefficient and the residual resistivities, i.e., a specimen with a low residual resistivity seems

to have a low coefficient (data sets 11, 86, 209) and vice versa (data sets 31, 47). However, with only a few data sets available, a definite relationship between the residual resistivity and the coefficient in the quadratic term cannot be established. Furthermore, the reported T^2 dependence of the electrical resistivity extends probably to temperatures below 1 K [126], whereas most of the authors report data to 2 K or higher. One way of circumventing this problem is to assume that the resistivity can be fitted to a residual term plus a quadratic term below 20 K (the T^5 or the Bloch-Grüneisen term is negligible compared with the other two terms at these temperatures), and the residual term, taken to be the value at 1 K, is obtained by extrapolation. The recommended values for the electrical resistivity of iron below 100 K are obtained by this procedure. The values of the quantities A and θ are taken to be the same as those given by Isshiki and Igaki [115], i.e., $58.1 \times 10^{-8} \Omega \text{m}$ and 467 K, respectively. The value of the coefficient of the T^2 term is $1.3 \times 10^{-13} \Omega \text{m K}^{-2}$. This was obtained both by a graphical method where the logarithm of the quantity

$$\left(\rho_{\text{measured}} - \rho_0 - A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^5} dx \right)$$

is plotted against $\log T$, and also by numerically fitting the same quantity to a quadratic function in temperature. This value is in agreement with that suggested by White and Woods [21], but is somewhat below the value of $2.2 \times 10^{-13} \Omega \text{m K}^{-2}$ given by Isshiki and Igaki [115] for their highly purified specimens measured in an applied magnetic field of 60 KAm^{-1} .

At temperatures above ~ 100 K, there is a slight change in the temperature dependence of the electrical resistivity. A log-log plot of the quantity

$$\rho_{\text{measured}} - \rho_0 - A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^2} dx$$

against T for the data of White and Woods [21] (data set 86), Richter and Kohlhaas [134] (data set 43), Moore et al. [135] (data set 17), Fulkerson et al. [136] (data set 16), Hust and Giarratano [133] (data set 47), Kohlhaas and Richter [137] (data set 34), and of Dewar and Fleming [138] (data set 53) shows a decrease from the T^2 line starting at ~ 100 K. This departure from the T^2 line is at a maximum of about $0.05 \times 10^{-8} \Omega \text{m}$ at ~ 140 K. At higher temperature,

the temperature dependence becomes stronger, with a temperature dependence that approaches a T^3 function. It is interesting to note that the same plot on the data of Kemp et al. [130] (data set 31) shows a slight increase at ~ 100 K. It is not obvious whether this behavior is purely a magnetic effect or an impurity effect. The specimens of Isshiki and Igaki [115] (data sets 219-223) are supposed to be purged of metallic impurities and to contain approximately 0.001 at.% C, 0.0007 at.% O, and <0.0001 at.% N, whereas the specimen of White and Woods [21] (data set 86) contains about 0.03% of mostly metallic impurities. The electrolytic iron specimen of Hust and Giarratano [133] (data set 47) contains about 0.1% also of mostly metallic impurities. The specimen of Moore et al. [139] (data set 15) contains $\leq 0.01\%$ Ni, $\leq 0.01\%$ Si, and lesser amounts of other impurities, and the specimen of Fulkerson et al. [136] (data set 16) contains $\leq 0.02\%$ Si, 0.014% C, $<0.01\%$ Ni, and lesser amounts of other impurities. On the other hand, the magnetoresistance of iron is positive at room temperatures (see, for example, Kornetzki [140], Schindler and La Roy [118]), and negative at helium temperatures (see, for example, Fujii and Morimoto [112], Glaeser et al. [116], and Arajcs et al. [119]). From the only available data by Shirakawa [141] and by Matuyama [142] on electrolytic iron at intermediate temperatures, the magnetoresistance of iron changes sign at ~ 77 K. For the lack of definite conclusion, this behavior is ignored at the present, and the resistivity value is assumed to follow the relation

$$\rho = \rho_0 + \alpha T^2 + A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^2} dx \quad (8)$$

up to 200 K. This assumption may result in a maximum probable error of only -1% or $-0.03 \times 10^{-8} \Omega \text{m}$ at ~ 150 K.

At temperatures above 150 K, and up to the Curie temperature, there are a number of data sets that agree with each other to within $\pm 2\%$: Moore et al. [139] (data set 15), Fulkerson et al. [136] (data set 16), Wallace et al. [143] (data set 21), Pallister [144] (data set 22), Jaeger et al. [145] (data set 24), Kohlhaas et al. [137] (data set 33), Kierspe et al. [78] (data set 39), Richter and Kohlhaas [134] (data set 43), Powell et al. [146] (data set 77), and Lauchbury and Saunders [147] (data set 217). Among these sources, Fulkerson et al. [136] (data set 16) reported also the resistivity at 4 K, and Moore

et al. [135] (data set 17) reported the residual resistivity ratio. Greater weight is given to the data of Moore et al. Most of these data sets appear to have resistivity values slightly higher than $9 \times 10^{-8} \Omega\text{m}$ at 273 K, except for data sets 17 and 22. Judging from the ice point resistivity values and from the temperature variation of the solute resistivities of various elements in iron (see, for example, Schwerer and Cuddy [148]), it appears that the specimen of Moore et al. [135] (data set 17) is the purest among these groups, and its resistivity values at 90 K are within 2% of the recommended values based on the analysis of the available low temperature data described earlier. The recommended values within the range of ± 100 K of the ice point are, therefore, based on data set 17, and in the higher temperature range up to 900 K they are based on the data sets mentioned above. Both graphical and numerical methods in curve fitting are employed. It is found that the electrical resistivity of iron can be represented by a cubic polynomial at temperatures from about 200 to 900 K.

For temperatures higher than 900 K and up to the Curie temperature, the electrical resistivity of iron increases more rapidly with temperature. Even though there are detailed accounts on the temperature derivative of the resistivity, the agreement between them is not good (see, for example, Lauchbury and Saunders [147] and Seehra et al. [149]), the values of $d\rho/dT$ from these two accounts differ by $\sim 40\%$ at 1030 K). The value of the Curie temperature has been reported to be 1040.3 ± 1 K (Seehra et al. [149]), 1038-1043 K (Fulkerson et al. [136]), 1051-1055 K (Morris [150]), 1044 ± 2 K (Arajs and Colvin [151]), 1037 K (Richter and Kohlhaas [134]), 1027 K (Kohlhaas and Richter [137]), 1042.7 K (Lauchbury and Saunders [147]), and 1036 K (Wallace et al. [143]). The last authors also reported a Curie temperature of 1042 K from their specific heat measurements. In view of the wide spread of the reported values, a Curie temperature of 1043 K is adopted from the AIP Handbook [152]. The recommended values below this temperature are based on the data of Lauchbury and Saunders [147] (data set 217), with slight adjustments so that they would merge smoothly with the recommended values at lower temperatures. The recommended value at the Curie temperature, $101.1 \times 10^{-8} \Omega\text{m}$, is within 0.3% of the values given by Lauchbury and Saunders [147].

At temperatures above the Curie point, the differences between reported resistivity values from the above references become greater, even though they

are generally still within 2% of each other. The recommended values from the Curie temperature to the α - γ transition are based mainly on the results of Fulkerson et al. [136] (data set 16), Wallace et al. [143] (data set 21), and also of Powell et al. [146] (data sets 75-77). There are other detailed accounts on the resistivity at temperatures close to the α - γ transition: Kohlhaas and Richter [137] (data sets 37-38), Arajs and Colvin [151] (data sets 57-58), and Bullock [153] (data sets 98-99). However, there are wide discrepancies among these data sets. The data of Bullock and of Kohlhaas and Richter are too high and too low, respectively. The specimen of Arajs and Colvin showed some unexplained behavior, its residual resistivity ratio changed by +13% after the high temperature measurement, and the resistivity value at room temperature appeared to be too high for a zone-refined specimen. The onset of the α - γ transition, A_{c3} point, has been reported at 1188-1189 K (Moore et al. [139]), 1189 K (Kohlhaas and Richter [137]), 1189.7 K (Richter and Kohlhaas [134]), and has been inferred from graphical illustrations at about 1182 K (Powell et al. [146] data set 75), 1183 K (Arajs and Colvin [151] data set 57), 1187 K (Kohlhaas and Richter [137] data set 37), and 1186 K (Bullock [153] data set 98). Because of the lack of general agreement, the transition temperature is taken to be 1185 K, a value deduced from thermal expansion data [42].

As mentioned earlier, there are detailed reports on the behavior of the electrical resistivity at the α - γ transition. Not surprisingly, all these reports show that the transition occurs over a finite temperature range: ~ 4 K according to Powell et al. [146], Arajs and Colvin [151], and Bullock [153], and ~ 5 K according to Kohlhaas and Richter [137]. All reported a hysteresis effect: in the transition region, the resistivity values measured at decreasing temperatures were lower than those measured at increasing temperatures. The temperature for the onset of the γ - α transition upon cooling, the A_{r3} point, is also reported to be somewhat lower than the A_{c3} point. The latter three groups of authors also reported that the resistivity of α -iron at temperatures about one degree below the A_{r3} point after being cooled from the γ phase is higher than that of the α -iron after being heated from lower temperatures. In view of these and of the lack of such evidence from the data of Powell et al. [24] (data sets 75-77), this behavior is ignored at the present time. It is probable that such behavior is dependent upon specimen purity and its thermal history and mechanical history as well. Even though recommended values are

given at a single transition temperature (for both the α and the γ phases), the transition for a given specimen may be expected to occur over a small temperature range of approximately 1180-1190 K. Its resistivity below 1185 K may be somewhat ($\leq 0.5 \times 10^{-8} \Omega\text{m}$) lower than the recommended value and vice versa above 1185 K.

There are relatively fewer data sets available for temperatures above the α - γ transition. The data sets which are considered reliable and from which the recommended values are derived are generally for temperatures less than ~ 1300 K. For higher temperatures, close to the γ - δ transition, the data of Cezairliyan and McClure [154] (data sets 62-63) appear to merge well with extrapolations of the data of Fulkerson et al. [136] (data set 16) and of Wallace et al. [143] (data set 21). The recommended values from the α - γ transition to the γ - δ transition are based on these data sets. At temperatures immediately above the α - γ transition, the data of Powell et al. [146] (data sets 75-76) are also taken into account.

There are only a few data sets for the electrical resistivity of δ -iron. The recommended values are based on the results of Cezairliyan and McClure [154] (data sets 62 and 64), Güntherodt et al. [92] (data set 206), and Powell [155] (data set 138). The slight upturn in the resistivity value at temperatures close to the melting point is based on the latter two data sets. This upturn seems to be substantiated by the data of Baum et al. [156] (data set 114) and of Kita et al. [93] (data sets 212-214). There exists only two reports on the change of resistivity value at the γ - δ transition, by Cezairliyan and McClure [154] (data sets 62-65), and by Kierspe et al. [78] (data set 41). Both indicate a slight increase in resistivity from the γ to the δ phase. No hysteresis has been reported.

There are in excess of 10 data sets available for the electrical resistivity of molten iron. Some of these: Güntherodt et al. [92] (data set 206), Kita et al. [93] (data sets 211-213), Baum et al. [156] (data set 114), Arsentiev et al. [157] (data sets 215-216), Ono and Yagi [89] (data set 115), and Eliutin et al. [88] (data set 92) cover the transition from the solid to the molten state. The last authors reported a decrease in resistivity upon melting for an impure specimen ($\sim 99.0\%$ purity), in agreement with an earlier measurement on Armco iron by Mokrovskii and Regel [158], whose result has been widely quoted. However, the more recent measurements on purer specimens reported

in the references mentioned above all show an increase. The majority of the reported data show a linear dependence on temperature, and the data of Seydel and Fucke [87] (data set 205) which were obtained by a pulse-heated exploding wire technique show that the linear dependence is applicable up to 3000 K. The recommended values are based on the data of Güntherodt et al. [92] (data set 206) and Kita et al. [93] (data sets 212-214), both of which are obtained by steady state methods. Values above 1900 K are extrapolated according to a linear temperature dependence. The value at 3000 K is about 7% higher than that of Seydel and Fucke [87]. A few of the available data sets are obtained by the rotating field method: Ono and Yagi [89] (data set 115), Levin et al. [91] (data set 112), Baum et al. [156,159] (data sets 114,121), and Samarin [94] (data set 137). The reported data among this group show relatively large variations, but are still within 4% of the recommended values.

The recommended values both uncorrected and corrected for thermal expansion of the material are presented in table 7, while only the uncorrected values (except those for the liquid state) are shown in figures 5 and 6 along with the experimental data. These values at temperatures above 200 K are for iron of purity 99.99% or higher, while those below 200 K are applicable only to highly purified zone-refined iron having a residual resistivity of $0.0200 \times 10^{-8} \Omega\text{m}$. The estimated uncertainty in the recommended values is $\pm 5\%$ below 100 K, $\pm 3\%$ from 100 to 200 K, and $\pm 2\%$ above 200 K up to the melting point. The uncertainty at temperatures immediately above the melting point is about $\pm 5\%$, increasing to $\pm 10\%$ at the highest temperatures.

For slightly less pure iron having different residual resistivity, its electrical resistivity values can be calculated from the recommended values using the Matthiessen's rule, which will not introduce serious errors. For example, using Matthiessen's rule for the specimen of Moore et al. [135] (data set 17) gives discrepancies of +1.7% (compared with the measured data) at 90 K, -2% at 260 K, and <0.1% at 400 K. That for the specimen of White and Woods [21] (data set 86) gives discrepancies of -0.5% at 11 K, -1.3% at 22 K, +1.6% at 98 K, -1.8% at 178 K, and -1.4% at 273 K. And that for the specimen of Fulkerson et al. [136] (data set 16) gives -1% at 77.5 K, +0.6% at 194 K, and -1% at 273 K. Thus, it does appear that using the Matthiessen's rule and the recommended values will give resistivity values for a specimen with residual resistivity lower than $0.4 \times 10^{-8} \Omega\text{m}$ to within $\pm 2\%$, subject to the uncertainties

in the recommended values specified in the preceding paragraph. The applicability of Matthiessen's rule, to within the possible error of $\pm 2\%$, also seems to be confirmed for a more commonly available material, the electrolytic iron. For example, using the rule and comparing with the data of Hust and Giarratano [133] (data set 47) for SRM Iron-1265 gives discrepancies of -1% at 100 K, -0.9% at 200 K, and -1.5% at 280 K. Since deviations from the Matthiessen's rule for dilute iron alloys are positive (see, for example, Bass [160]), its application in calculating the electrical resistivity of an iron specimen lower in purity is likely to result in an underestimate, especially around room temperature where the deviation from Matthiessen's rule approaches a maximum. At high temperatures, the relative error introduced by the application of Matthiessen's rule should diminish with increasing temperature.

The recommended values uncorrected for thermal expansion given in table 7 can be represented approximately by the following expressions to within $\pm 0.1\%$.

1-200 K:

$$\rho = 0.02 + 58.1 \left(\frac{T}{467} \right)^5 \int_0^{467/T} \frac{e^x x^5}{(e^x - 1)^5} dx + 1.3 \times 10^{-5} T^2 \quad (26)$$

200-900 K:

$$\rho = -1.120747752 + 2.261529506 \times 10^{-2} T + 3.913892564 \times 10^{-5} T^2 + 2.952608182 \times 10^{-8} T^3 \quad (27)$$

900-1020 K:

$$\rho = -513.9789758 + 1.70577020 T - 1.804410343 \times 10^{-3} T^2 + 7.034546961 \times 10^{-7} T^3 \quad (28)$$

1020-1143 K in the vicinity of the Curie temperature, T_C :

$$\rho = 101.13 - 2.984305206 \times 10^{-1} (T_C - T) + 1.905714112 \times 10^{-3} (T_C - T)^2 - 1.3355493086 \times 10^{-5} (T_C - T)^3 \quad (29)$$

1043-1070 K:

$$\rho = 101.13 + 2.020277018 \times 10^{-1} (T - T_C) - 5.420505212 \times 10^{-3} (T - T_C)^2 + 1.294364953 \times 10^{-4} (T - T_C)^3 - 1.243596538 \times 10^{-6} (T - T_C)^4 \quad (30)$$

1070-1185 K:

$$\rho = -1309.064808 + 3.419572717 T - 2.773870769 \times 10^{-3} T^2 + 7.595259559 \times 10^{-7} T^3 \quad (31)$$

1185-1667 K:

$$\rho = 55.0861977 + 5.289665269 \times 10^{-2}T + 9.77621850 \times 10^{-12}T^2 - 4.0798019 \times 10^{-9}T^3 \quad (32)$$

1667-1811 K:

$$\rho = -11976.94918 + 21.0981583 T - 1.226701138 \times 10^{-2}T^2 + 2.378811917 \times 10^{-6}T^3 \quad (33)$$

1811-3000 K:

$$\rho = 135.2 + (T-1811) \times 1.545 \times 10^{-2} \quad (34)$$

It should be emphasized that these expressions do not necessarily suggest any theoretical justification, and should be treated, most appropriately, as numerical aids only. It should also be understood that giving these expressions does not imply a recommendation for the temperature derivative of the electrical resistivity.

TABLE 7. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF IRON^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{ m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.0200	0.0200	800	57.14	57.56
4	0.0202	0.0202	900	72.46	73.11
7	0.0206	0.0206	1000	90.80	91.76
10	0.0213	0.0212	1043	101.1 ^b	102.2 ^b
15	0.0232	0.0231	1100	107.0	108.3
20	0.0262	0.0261	1150	110.1	111.5
25	0.0313	0.0312	1185	111.9(α)	113.4(α)
30	0.0396	0.0395	<u>1185</u>	<u>111.0(γ)</u>	<u>112.1(γ)</u>
40	0.0733	0.732	1200	111.5	112.6
50	0.145	0.145	1300	114.9	116.4
60	0.268	0.267	1400	117.9	119.7
70	0.449	0.448	1500	120.7	122.8
80	0.690	0.689	1600	123.0	125.4
90	0.964	0.962	1667	124.4(γ)	127.0(γ)
100	1.28	1.28	<u>1667</u>	<u>124.6(δ)</u>	
150	3.16	3.16	1700	125.4	
200	5.20	5.19	1800	127.9	
250	7.44	7.44	1811	128.6(δ)	
273	8.57	8.57	1811		135.2 ^c (δ)
293	9.61	9.61	1900		136.6 ^c
300	9.98	9.98	2000		138.2 ^c
400	16.08	16.10	3000		153.6 ^c
500	23.66	23.72			
600	32.92	33.06			
700	44.02	44.27			

^a The values are for iron of purity 99.99% or higher, but those below 200 K are applicable only to iron having a residual resistivity of $0.0200 \times 10^{-8} \Omega \text{ m}$. Columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Dotted lines separating tabular values indicate solid phase transitions and solid line indicates solid to liquid state transformation.

α : bcc; γ : fcc; δ :bcc.

^b Value at the Curie temperature.

^c Provisional value.

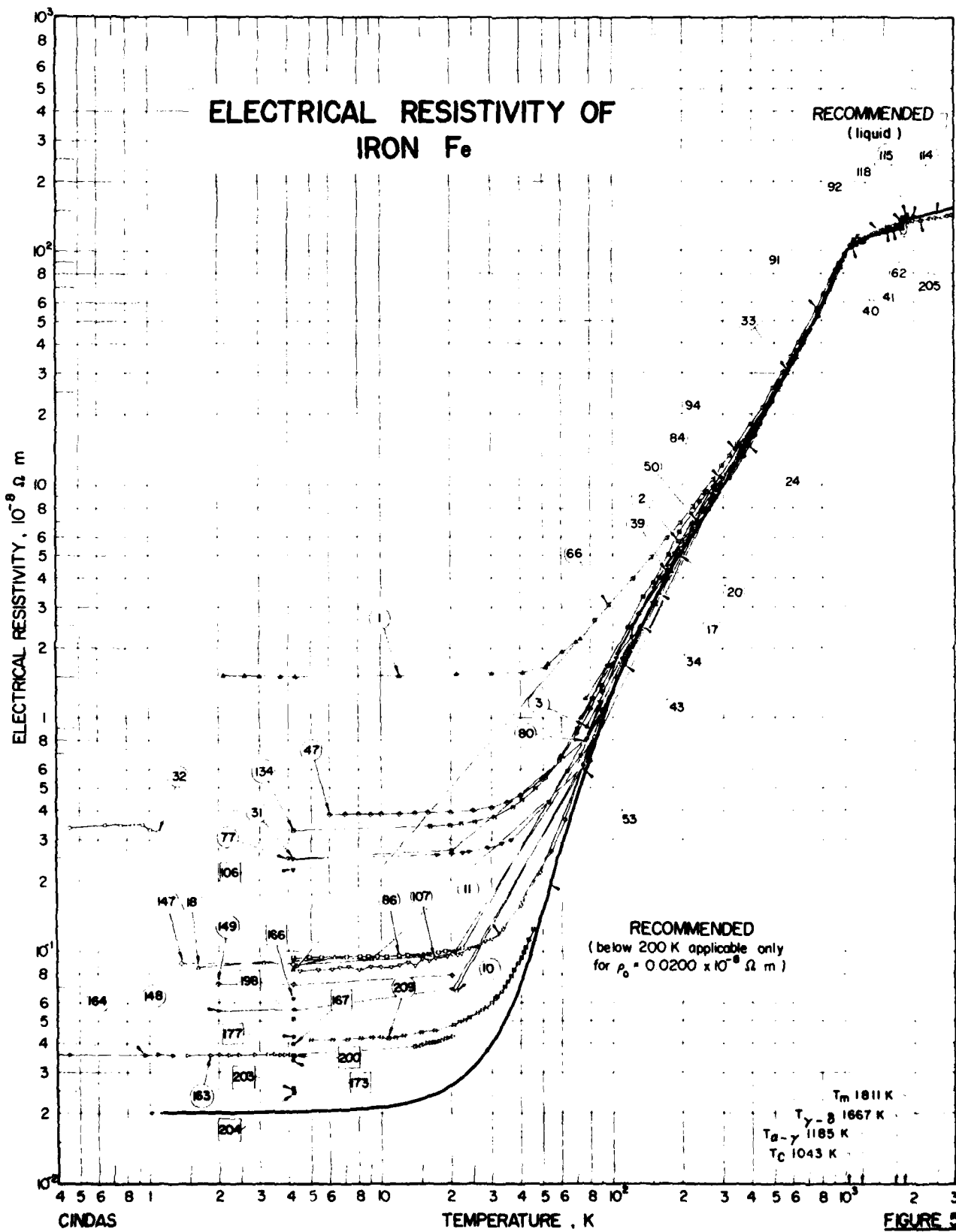


FIGURE 5

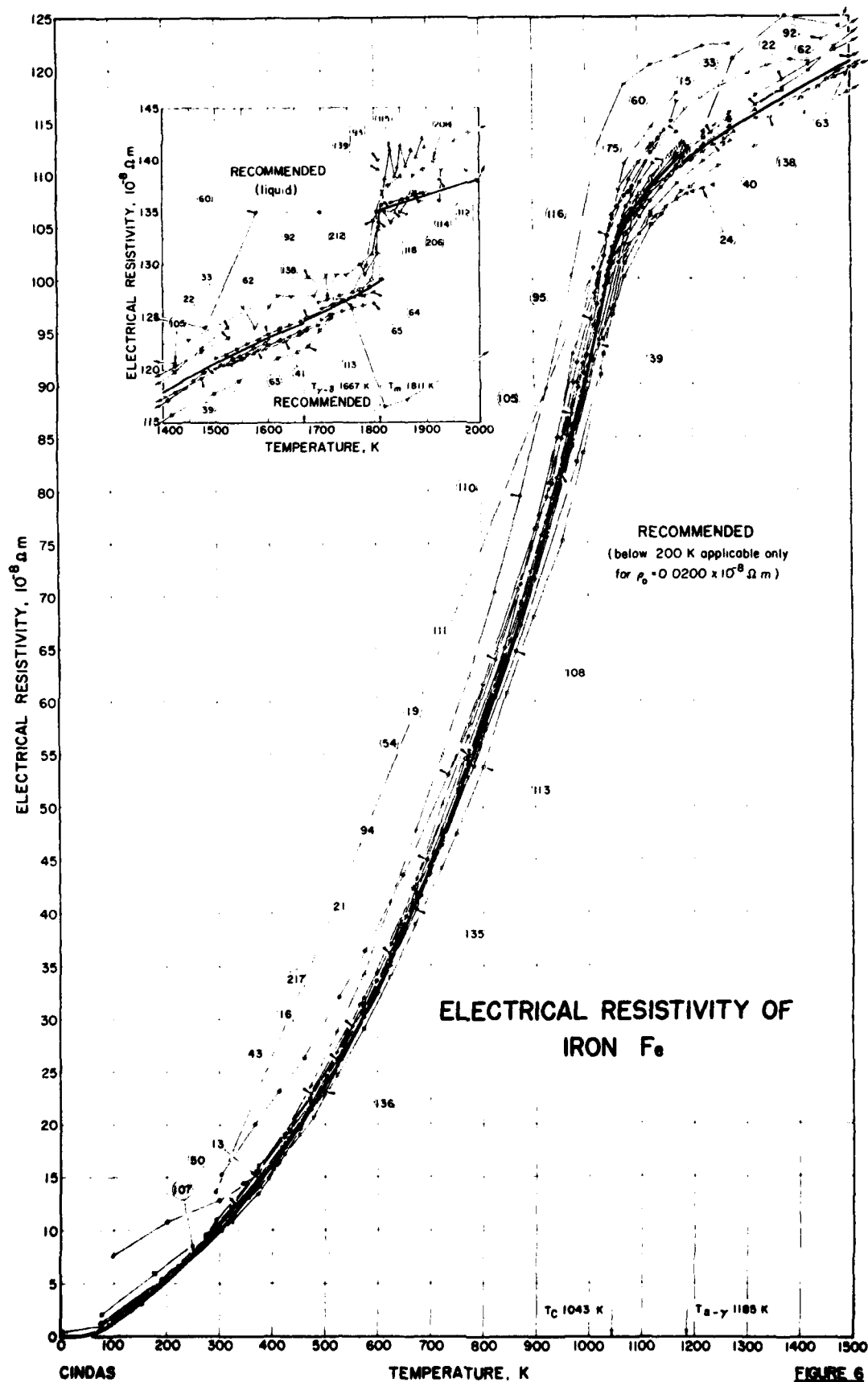


TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	129	Kondorskii, E.I., Gaikina, O.S., and Chernikova, L.A.	1958		2.1-73		"Chemically pure"; annealed in vacuum at 1173 K for 1 h; furnace cooled; residual resistivity $1.458 \times 10^{-6} \Omega \cdot \text{m}$.
2	161	Broom, T.	1952	B	90-373		99.99 Fe, 0.005 Ni, and 0.002 Cu (spectrographic analysis); wire specimen 0.056 cm (0.022 in.) in diam; drawn from 0.183 cm (0.072 in.) in diam; annealed at 873 K for 2 h; furnace cooled.
3	95	Wilkes, K.E.	1968	A	78-299	18 AF 3	99.96 Fe, 0.007 Cu, 0.0058 C, 0.004 Mn, 0.004 Si, 0.003 S, 0.0023 N, 0.002 Cr, 0.001 Al, 0.001 P, and 0.0008 O; 3.167 cm (1.247 in.) in diam and 10.44 cm long; prepared by National Physical Lab., England; density 7.872 g cm^{-3} at 297 K; resistivity values corrected for thermal expansion.
4*	162	Adrock, F. and Bristow, C.A.	1935	A	323-423	Batch 5	0.0045 C, 0.002 Mn, 0.0015 S, 0.001 P, 0.0001 H, 0.001 N, 0.0006 Ni, 0.0005 O, 0.0002 Si, traces of Al and Mg; prepared by chemical reduction of ferrous chloride; cold-rolled from 3 cm bar to 1 cm in diam; annealed at 1223 K for 2 h; density at 292 K, $7.871 \pm 0.002 \text{ g cm}^{-3}$; smoothed values from table.
5*	163	Eucken, A. and Dittich, K.	1927		80,273	Electrolytic Iron; 1	Coarse-grained
6*	163	Eucken, A. and Dittich, K.	1927		80,273	Electrolytic Iron; 2	Fine-grained
7*	163	Eucken, A. and Dittich, K.	1927		80,273	Electrolytic Iron; 3	Obtained from Firma Heraeus.
8*	164	Grüneisen, E. and Goens, E.	1927		21-273	Iron 2	Technically pure; polycrystalline; electrolytically precipitated; unannealed.
9*	164	Grüneisen, E. and Goens, E.	1927		21-273	Iron 3	Electrolytic; repeatedly hammered; annealed at 773 K for 1 h.
10	164	Grüneisen, E. and Goens, E.	1927		21-273	Iron 1	Twice electrolytically refined; hammered and annealed.
11	131	Kemp, W.R.G., Klemens, P.G., and Tainish, R.J.	1959	A	4.2-293		Doubly refined electrolytic iron; 2.4 x 1.7 x 30 mm; cut from a precipitated plate; annealed at 1223 K; compressed, and reannealed in vacuum at 1223 K; sample material believed to be the same as Data Set 10 above.
12*	165	Powell, R.W. and Tye, R.P.	1967	A	323-523	Pure iron Sample No. 1	0.025 Ni, 0.01 Cu, 0.01 Mo, 0.007 Cr, 0.005 C, 0.004 O, 0.004 S, 0.004 V, 0.003 P, 0.001 Mn, <0.001 Si, 0.0006 N, and 0.000048 H; type 1, 1.27 cm in diam and 15 cm long; supplied by Metals Research, submitted for test by Tube Investments Ltd.

* Not shown in figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	165	Powell, R.W. and Tye, R.P.	1967	A	323-1073	Pure iron Sample No. 2	0.0055 Ni, 0.0053 Si, 0.0038 Al, 0.0035 S, 0.002 Co, 0.0017 P, 0.0014 C, 0.001 Cr, <0.001 Mn, 0.0008 O, 0.0007 N, and 0.000016 H; short rod 1.27 cm diam prepared by Metallurgy Division of National Physical Lab.; machined from a disk.
14*	165	Powell, R.W. and Tye, R.P.	1967	A	323-523	Purefree iron Sample No. 6	0.08 Si, 0.03 C, 0.015 P, 0.01 Mn, and 0.01 S; 2.54 cm in diam and 20 cm long; supplied by Low Moor Best Yorkshire Iron Ltd.; 1) the above specimen measured at increasing temperatures; 2) the above specimen measured at decreasing temperatures.
15	139	Moore, J.P., Fulkerson, W., and McElroy, D.L.	1964	A	73-1273	High purity iron	0.001-0.01 Ni, 0.001-0.01 Si, 0.003 S, 0.003 C, 0.0025 O, 0.0011 P, 0.0001-0.001 Al, 0.0001-0.001 Ca, 0.0001-0.001 Cu, 0.0005 N, and 0.0001 H; atomic percent, data here rounded off; prepared by arc melting Armo iron stock in pure inert atm to produce pancake shaped billets, rolled into sheets and cut to make feed stock for electron beam melting, then cast into 10.16 cm (4 in.) in diam and 15.24 cm (6 in.) long billet; trimmed off outside edges; rod specimen 38.10 cm (0.15 in.) in diam and 7.62 cm (3 in.) long ₁ cut from center portion of billet; measured in vacuum at 10 ⁻⁵ to 10 ⁻⁷ Torr; data corrected for thermal expansion except points at 73, 189, and 273 K.
16	136	Fulkerson, W., Moore, J.P., and McElroy, D.L.	1966	A	4.0-1273	High purity iron	99.95 Fe, 0.002-0.02 Si, 0.014 C, 0.00095-0.0095 Ni, 0.0088 O, <0.0056 H, 0.0052 S, 0.00021-0.0021 Al, 0.002 P, 0.002 N, 0.00014-0.0014 Ca, and 0.00009-0.0009 Cu, in atomic percent; obtained by electron beam melting of Armo iron; homogeneous to ±0.19; rod specimen 38.10 cm (0.15 in.) in diam and 7.62 cm (3 in.) long; free of voids; immersion density 7.881 g cm ⁻³ ; smoothed data extracted from table; data corrected for thermal linear expansion; resistivity measured with current densities of 6.8, 11 and 11 A cm ⁻² at 4, 77.5 and 194.1 K respectively.
17	135	Moore, J.P., McElroy, D.L., and Barisoni, M.	1966	A	90-40	Grade 1	Cylindrical specimen machined from electron beam zone-refined iron ₂ (3-pass); produced by Materials Research Corp.; density 7.824 g cm ⁻³ ; electrical resistivity ratio $\rho(273 K)/\rho(4.2 K) = 201$; smoothed data from table.
18	166	McDonald, W.J., Jr.	1962	A	1.63-80.7		Cut from a zone-refined ingot; prepared at BMI; machined into a rectangular parallel piped 0.157 x 0.157 x 0.381 cm (0.062 x 0.062 x 1.5 in.).
19	167	Bungardt, K. and Spyra, W.	1965		293-1373		0.03 Ni, 0.015 C, 0.007 S, and traces Al, Mo, P, and Si; cylindrical specimen.
20	168	Bohm, R. and Wachtel, E.	1969		196-406		0.005 N, 0.004 C, and 0.003 O; cylindrical specimen 10 mm in diam.

* Not shown in figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
21	143	Wallace, D.G., Sidles, P.H., and Danielson, G.C.	1959	B	298-1323	High purity iron	0.03 C, 0.01 N and 0 each, <0.0005 Ni, 0.0001 Cu, <0.0001 each of Mg, Si, Ag, and Na; wire specimen 0.0254 to 0.0346 cm in diam and 4 to 7 cm long; from Johnson and Matthey Co.; drawn and annealed at 1273 K for 1 h, cooled at a rate of 40 Kh ⁻¹ while measurement was made; with some specimens, measurements here made upon reheating, and data "accurately reproduced"; smoothed values from table, representing data for two specimens; uncorrected for thermal expansion; values do not reflect a discontinuity of 0.6% upon cooling at 1173 K.
22	144	Pallister, P.R.	1949	A	273-1548		⁺ 99.99% Fe; 1 cm in diam and 30 cm long; density 7.87 ± 0.005 g cm ⁻³ ; uncorrected for thermal expansion; data extracted from table.
23*	169	Bäcklund, N.G.	1961	A	90-290	Pure iron Data Set 1	Three types of specimens: 1) "very pure iron" wire; 2.5 mm in diam; from Phillips Research Lab., 2) spectroscopically standardized pure iron wire; 5 mm in diam; from Johnson and Matthey Co., 3) pure iron wires; 1.0 and 2.0 mm in diam; from Heraeus Inc.; all specimens annealed at 773.2 K for 10 h; data reported as average of all three types.
24	145	Jaeger, F.M., Rosenbohm, E., and Zuthoff, A.J.	1938		293-1243	Pure iron	Pure; 0.25 mm in diam and 925 mm long; data corrected for thermal expansion.
25*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #2	0.002 S, <0.002 Cu, 0.001 C and Si each, <0.001 Be, and <0.0005 P; 2 mm in diam and about 1 meter long; ingots produced by recrystallization of ferric nitrate, conversion to ferric oxide, reduction to sponge iron, and melting under hydrogen and in a vacuum, forged, cold-rolled, swaged, and drawn; annealed in vacuum for 15 min at 1123 K.
26*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #7	0.002 O and S each, <0.002 Cu, 0.001 Si, and <0.001 C; dimensions, fabrication method, and heat treatment same as the above specimen.
27*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #14	0.002 O and Si each, 0.002 Cu, 0.001 C and S each; dimensions, fabrication method, and heat treatment same as the above specimen.
28*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #11	0.004 O, <0.002 Cu, and 0.001 S; dimensions, fabrication method, and heat treatment same as the above specimen.
29*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #19	0.004 O, 0.002 S, <0.002 Cu, and <0.001 C; dimensions, fabrication method, and heat treatment same as the above specimen.
30*	170	Cleaves, H.E. and Hiegel, J.M.	1942		293	Ingot #6	0.004 O, <0.002 Cu, 0.001 S and Si each, and <0.001 Be and C each; dimensions, fabrication method, and heat treatment same as the above specimen.
31	130	Kamp, W.R.G., Klemens, P.G., and White, G.K.	1956	A	4.2-293	JM 5092	99.99 Fe, 0.005 Ni, 0.0002 Cu, 0.0001 Ag, and traces Mg and Mn; 2 mm in diam rod; supplied by Johnson and Matthey Co.; annealed at 1023 K (750 C) for 4 h in vacuum; resistivity values calculated from reported ideal electrical resistivity and $\rho_0 = 0.248 \cdot 10^{-8} \Omega \cdot m$.

* Not shown in figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
32	171	Yoshida, I.	1965		0.5-1.1		No details given.
33	137	Kohlhaas, R. and Richter, F.	1962	B	293-1523	Fe _μ	0.064 O, 0.0027 C, 0.002 S, 0.001 Mn, N, and Si each; 0.5 cm in diam and 25 to 30 cm long; turned from square bar, transition temperatures: Ac ₂ = 1027 K (754 C), Ac ₃ = 1189 K (916 C) and Ar ₃ = 1175 K (902 C); smoothed data from table.
34	137	Kohlhaas, R. and Richter, F.	1962	B	90-291	Fe _μ	The above specimen.
35*	137	Kohlhaas, R. and Richter, F.	1962	B	969-1080	Fe _μ	The above specimen at temperatures about the Curie point; measured while heating.
36*	137	Kohlhaas, R. and Richter, F.	1962	B	972-1079	Fe _μ	The above specimen at temperatures about the Curie point; measured while cooling.
37*	137	Kohlhaas, R. and Richter, F.	1962	B	1176-1198	Fe _μ	The above specimen at temperatures about the α-γ transition; measured while heating.
38*	137	Kohlhaas, R. and Richter, F.	1962	B	1164-1195	Fe _μ	The above specimen at temperatures about the α-γ transition; measured while cooling.
39	78	Vierspe, W., Kohlhaas, R., and Gonska, H.	1967	B	73-1715		0.0060 S, 0.0050 C, O, and Si each, 0.0016 M, 0.0010 N, and P each; wire specimen from Prof. W.A. Fischer, Max-Planck-Institute for Iron Research; smoothed values from figure.
40	78	Kierspe, W., et al.	1967	B	1103-1283		The above specimen at temperatures about the α-γ transition.
41	78	Kierspe, W., et al.	1967	B	1553-1713		The above specimen at temperatures about the γ-δ transition.
42*	208	Kohlhaas, R. and Kierspe, W.	1965		83-353		0.0027 C, 0.002 S, 0.001 Mn, N, and Si each, and trace of Cr.
43	134	Richter, F. and Kohlhaas, R.	1964		93-1273		0.012 O, 0.008 P, 0.007 C, and Al each, 0.004 S, and 0.002 N; disk specimen 63 mm outer diam; annealed for several hours at 1193 K (900 C); Curie point 1037 K (764 C); α-γ transition: Ac ₃ = 1189.7 K (916.7 C), Ar ₃ = 1186.3 K (913.3 C).
44*	172	Jaeger, W. and Desselhorst, H.	1900		291-373	Eisen I	0.1 C, metallic impurities not determined; 1.3007 cm in diam and 27.0 cm long; density 7.84 g cm ⁻³ .
45*	173	Lorenz, L.	1881		273-373		No details reported.
46*	174	Brown, H.M.	1928	A	312		0.0794 cm ² x 10 cm.

* Not shown in figure.

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ELECTRICAL RESISTIVITY OF CHROMIUM COBALT IRON AND
NICKEL(U) THERMOPHYSICAL AND ELECTRONIC PROPERTIES

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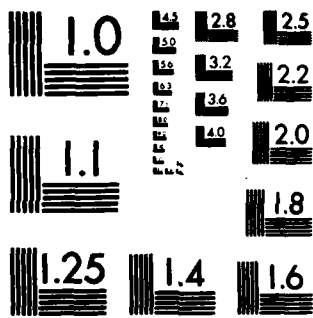
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TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
47	133	Hust, J.G. and Giarratano, P.J.	1975		4-280	NBS electrolytic iron	99.9 ⁺ Fe, 0.041 Ni, 0.0080 Si, 0.0072 Cr, 0.007 Co, 0.0067 C, 0.0059 S, 0.0058 Cu, 0.0057 Mn, 0.005 Mo, 0.0025 P, 0.0007 Al, 0.0006 Ti and V each, 0.0002 As, 0.00013 B, 0.00002 Pb; chemical composition certified by NBS, SRM 1265; grain size 0.05 μ m for annealed condition; rod specimen 3.6 mm in diam and 23 cm long, apparently annealed; density 7.867 \pm 0.005 g cm ⁻³ , for annealed condition; Rockwell hardness B24, for annealed condition; mean residual resistivity ratio $\rho(273\text{ K})/\rho(4\text{ K})$ is 23; residual resistivity 0.385 $\times 10^{-8}$ Ω m, average $\rho(273\text{ K})$ 8.71 $\times 10^{-8}$ Ω m; data uncorrected for thermal expansion; data smoothed with linear least squares methods to the expression $\rho = \sum b_i (T_i)^i - 1$, where $b_1 = 1.52995843 \times 10^{-5}$, $b_2 = 6.57221842 \times 10^{-5}$, $b_3 = 1.28083500 \times 10^{-4}$, $b_4 = -1.50027718 \times 10^{-4}$, $b_5 = 1.17942860 \times 10^{-4}$, $b_6 = -6.57740194 \times 10^{-5}$, $b_7 = 2.67952461 \times 10^{-5}$, $b_8 = -8.08115141 \times 10^{-6}$, $b_9 = 1.80581011 \times 10^{-6}$, $b_{10} = -2.95519976 \times 10^{-7}$, $b_{11} = 3.44469418 \times 10^{-8}$, $b_{12} = -2.70952664 \times 10^{-9}$, $b_{13} = 1.28939040 \times 10^{-10}$ and $b_{14} = -2.80388287 \times 10^{-12}$, the b 's presented on p. 18 of Hust, J.G. and Sparks, L.L., "Thermal Conductivity Standard Reference Materials from 4 to 300 K: II OSRM Iron-1265," NBS Report 9771, 35 pp., 1970; data presented here are from table on p. 24 and 25, and are different from values calculated from the above polynomial.
48*	138	Dewar, J. and Fleming, J.A.	1893	B	76-471	Iron A	0.25 Mn, 0.01 S, very free from C, Si, and P; wire specimen 0.02657 cm mean diam and 100 cm long; from Armstrong's works, sent by Colonel Dyer of Elswick Ordnance Works; resistance 0.4223, 1.1909, 1.5086, 1.9104, 2.0737, 2.4167, 2.8368, 3.4091, and 4.1935 Ω at 76.1, 191.3, 229.3, 274.55, 291.65, 325.25, 363.7, 412.0, and 470.5 K, respectively; temperature below 273 K (0 C) measured by platinum resistance thermometer; data uncorrected for thermal expansion, length and mean diameter measured at 288 K; data extracted from table.
49*	138	Dewar, J. and Fleming, J.A.	1893	B	54,76	Iron A	Longer specimen cut from the same piece as Data Set 48; resistance 2.983 and 3.834 Ω at 53.8 and 76.1 K, respectively; data uncorrected for thermal expansion, length measured at 288 K; data extracted from text; temperatures measured by platinum resistance thermometer.
50	138	Dewar, J. and Fleming, J.A.	1893	B	76-469	Iron H.W.	High degree of purity; wire specimen 0.023078 cm mean diam and 100 cm long; from Messrs. Hopkins and Williams; very soft and well annealed, cold worked under the hammer and drawn without heating, into a very uniform wire; resistance 0.2918, 1.2713, 1.7137, 2.1791, 2.3940, 2.9546, 3.4976, 4.2536, and 5.1395 Ω at 76.1, 191.3, 234.0, 273.85, 291.40, 333.40, 371.25, 418.6 and 469.3 K, respectively; mean temperature coefficient between 273 and 373 K, 0.00625; data uncorrected for thermal expansion, length and mean diameter measured at 288 K; data extracted from table; temperature below 273 K (0 C) measured by platinum resistance thermometer.

* Not shown in figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
51*	138	Devar, J. and Fleming, J.A.	1893	B	51-76	Iron H.W. Coil (a)	Longer coil cut from the same wire as the above specimen, 2600 cm long; resistance 3.839, 4.154, and 7.091 Ω at 51.0, 54.0 and 76.1 K, respectively; data uncorrected for thermal expansion, length measured at 288 K; data extracted from table; temperature measured by platinum resistance thermometer.
52*	138	Devar, J. and Fleming, J.A.	1893	B	51-76	Iron H.W. Coil (b)	Still longer coil cut from the same wire as the above specimen; wire specimen 3700 cm long; resistance 5.222, 5.257, and 9.875 Ω at 50.5, 50.8, and 76.1 K, respectively; data uncorrected for thermal expansion, length measured at 288 K; data extracted from table; temperature measured by platinum resistance thermometer.
53	138	Devar, J. and Fleming, J.A.	1892	B	76-370		Pure soft iron; wire specimen had probable dimensions of 0.0076 (0.003 in) in diam and 50 or 100 cm long; from Messrs. Griffin and Co.; annealed; mean diameter of wire measured to nearest 0.000254 cm (0.0001 in); measurement of resistance repeated several times; mean observed specific resistance reported; data uncorrected for thermal expansion; data extracted from table.
54	175	Honda, K. and Simidu, T.	1917	A	303-1174	Swedish iron	Cylindrical specimen, 0.5 cm in diam and 20 cm long.
55*	151	Arajs, S. and Colvin, R.V.	1964	A	300-1291		0.00300 O, 0.0015 Ni, 0.0012 Co, 0.0005 C and Ge each, 0.0004 Cr, 0.0003 N, 0.00015 Cu, 0.00008 Zn, 0.00004 Ga, 0.00003 Sb, 0.00025 Ti, 0.00002 V, 0.000009 As, <0.000004 Mn, and <0.000007 Others; zone refined; 0.1 x 0.3 x 2.0 cm; $\rho(4.2 \text{ K})/\rho(298 \text{ K}) = 3.76 \times 10^{-3}$ before high temperature test and 4.30×10^{-3} after test; Curie temperature 1044 \pm 2 K; measured with a current density of $\sim 12.9 \times 10^4 \text{ A m}^{-2}$; corrected for thermal expansion; data extracted from figure.
56*	151	Arajs, S. and Colvin, R.V.	1964	A	1018-1068		The above specimen in the neighborhood of Curie temperature; corrected for thermal expansion; data from figure.
57*	151	Arajs, S. and Colvin, R.V.	1964	A	1151-1197		The above specimen measured through α - γ transition; temperature increasing; uncorrected for thermal expansion; data from figure.
58*	151	Arajs, S. and Colvin, R.V.	1964	A	1150-1186		The above specimen measured while cooling through α - γ transition; temperature decreasing; uncorrected for thermal expansion; data from figure.
59*	176	Kondorskii, E.I. and Sedov, V.L.	1960	A	4-2		Technically pure; 0.59 cm in diam and 11.2 cm long; vacuum annealed at 1273 K for 8 h; oven cooled; measured under saturation magnetization 1751 g.
60	177	Ibragimov, Sh.Sh.	1962	A	293-1698	Iron	0.06 Si, 0.04 C, 0.02 Mn and Cr each; annealed at 1033 K.
61*	178	Butler, E.H., Jr. and Pugh, E.M.	1940		313-343		Electrolytical iron; annealed in hydrogen.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
62	154	Cezairliyan, A. and McClure, J.L.	1973		1500-1660	γ iron 1	Tubular specimen 6.3 mm O.D., 0.5 mm thick and 102 mm long; fabricated from rods by electro-erosion technique; γ-δ transition reported at 1682 K; melting point 1808 K; specimen heated to measuring temperature in one second by passing current through; uncorrected for thermal expansion; smoothed data from table.
63	154	Cezairliyan, A. and McClure, J.L.	1973		1500-1660	γ iron 2	Similar to the above specimen (one of this or the above specimen has an electrical resistivity value of $10.2 \times 10^{-9} \Omega \text{m}$ at 273 K).
64	154	Cezairliyan, A. and McClure, J.L.	1973		1700-1800	δ iron 1	The same specimen as for Data Set 62.
65	154	Cezairliyan, A. and McClure, J.L.	1973		1700-1800	δ iron 2	The same specimen as for Data Set 63.
66	179	Micolai, G.	1908	B	84-673		0.5 mm in diam and 5 mm long; from Firma C.A.F. Kahlbaum.
67*	180	Wruck, D. and Wert, C.	1955	V	293		99.95 pure; 0.04 O, little metallic impurity.
68*	180	Wruck, D. and Wert, C.	1955	V	93		Same as above; foil polycrystal, 0.008 cm x 0.2 cm x 4 cm; Run II.
69*	180	Wruck, D. and Wert, C.	1955	V	93		Same as above; Run II.
70*	180	Wruck, D. and Wert, C.	1955	V	93		Same as above; Run IV.
71*	180	Wruck, D. and Wert, C.	1955	V	77		Same as above except wire specimen; 0.0762 cm in diam and 15 cm long; grain size ~0.6 cm; decarbonized.
72*	180	Wruck, D. and Wert, C.	1955	V	77		Same as above.
73*	180	Wruck, D. and Wert, C.	1955	V	77		Same as above.
74*	181	Rosenberg, H.H.	1955	A	1.8-77	JM 4975 (Run 2)	99.99 pure (excluding gases); from Johnson and Matthey Co.; polycrystalline; 0.202 cm in diam and 2.89 cm long; annealed in vacuum for several hours.
75	146	Powell, R.W., Tye, R.P., and Woodman, M.J.	1961	A	1088-1196	18 AF 3	0.007 Cu and Ni each, 0.0058 C, 0.004 Mn and Si each, 0.003 S, 0.0023 N, 0.002 Cr, <0.001 Al, <0.001 P, 0.0008 O, and <0.00005 H; measured during heating of the sample at a rate of 0.25 K min^{-1} , data uncorrected for thermal expansion.
76*	146	Powell, R.W., et al.	1961	A	1160-1191	18 AF 3	The above specimen measured during cooling.
77	146	Powell, R.W., et al.	1961	A	4-1073	18 AF 3	The above specimen measured at lower temperatures; smoothed values from table.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
78*	123	Taylor, G.R., Iain, A., and Coleman, R.V.	1963	A	77-191	Specimen #25	One of iron whisker specimens 100-300 μ in diam and about 2 cm long.
79*	123	Taylor, G.R., et al.	1963	A	77-300		Same as the above; resistivity values calculated from reported resistance ratios $R(T)/R(77K)$ and $\rho(77K)$ given in the above data set; average of seven specimens.
80	182	Soffer, S., Dierssen, J.A., and Pugh, E.M.	1965	A	76-300		<0.0185 metallic impurities; zone-refined; 0.1 cm x 1 cm x 10 cm.
81*	183	Yoshikawa, A. and Okamoto, M.	1967	A	77, 273		Zone refined iron (total impurity less than 0.001Z); grain <70 μ in diam; decarburized at 973 K for 7 days in wet hydrogen stream; ultrasonically cleaned; chemically polished with HF + H ₂ O ₂ ; heat treated at 773 K for 14 days in hydrogen purified by zirconium hydride; annealed at 773 K for 30 min; then cooled at 15 K hr ⁻¹ ; room temperature value calculated from reported resistance ratio (ρ_{273K}/ρ_{77K}).
82*	184	Takamura, S., Maeta, H., and Okuta, S.	1968	A	4.2, 293		99.996 pure, 0.0008 C, 0.007 N and 0 each, from Materials Research Corp.; wire specimen 0.12 mm in diam; as received condition; value at room temperature calculated from reported resistivity ratio and residual resistivity.
83*	184	Takamura, S., et al.	1968	A	4.2, 293		Same as the above specimen except annealed at 773 K for 1 h in vacuum and then at 973 K for 1 h.
84	184	Takamura, S., et al.	1968	A	4.2, 293		Similar to the above.
85*	184	Takamura, S., et al.	1968	A	4.2, 293		Similar to the above.
86	21	White, C.K. and Woods, S.B.	1959	A	4.2-295	Fe 2	99.97 pure, ~0.004 Si, <0.004 Co, Cu, and Ni each, ~0.003 Mo and Mn each, <0.00015 N, and traces Pb and Zr; 0.05 mm x 1 mm x 6-8 cm; from Vacuum Metals Co.; zone-melted in wet hydrogen to obtain large crystals; electrically annealed at 973 K to remove hydrogen; resistivity calculated from reported $\rho_i(T)$, $\rho(4.2K)/\rho(295K) = 9.61 \times 10^{-3}$, and $\rho_i(295K) = 9.82 \times 10^{-3} \Omega m$.
87*	185	Smith, A.W., Gregory, J.H., and Lynn, J.T.	1946	B	293		"Chemically pure"; wire specimen 0.1019 cm in diam and 15.2 cm long.
88*	185	Smith, A.W., et al.	1946	B	293		"Chemically pure"; wire specimen 0.0823 cm in diam and 15.5 cm long.
89*	185	Smith, A.W., et al.	1946	B	293		"Chemically pure"; wire specimen 0.0406 cm in diam and 5.0 cm long.
90*	185	Smith, A.W., et al.	1946	B	293		"Chemically pure"; wire specimen 0.0201 cm in diam and 3.4 cm long.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Refs. Cited	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
90	Shapiro, V.R., Abej'abi, Sh.Sh., Sundaresan, R.I., Percy, L.R., and Gul'd, P.V.	1973	A	288-1167		0.03 impurities; well annealed iron supplied by Johnson and Matthey Co.; average grain size 2 μ m; measured in vacuum ($\sim 10^{-5}$ mmHg) under quasistatic condition with heating rate not more than 1 K/min.
92	Elkstein, V.P., Turov, V.D., and Kuznetsov, N.A.	1965	B	1448-2094		99.0 pure; by carbonyl method or electrolytically; liquid state obtained by melting in graphite crucible either in a helium atm or in vacuum.
93	Gilchrist, B.J. and Gibson, B.G.	1973	C	1808		99.998 pure, from Johnson and Matthey Co., in liquid state; temperature = 1808 K assumed.
94	Thomson, I.A.	1974	A	344-1015		99.99 ⁺ pure; from Goodfellow Metals Ltd., England; 1.0 cm x 2.0 cm x 4.0 cm; annealed at 1473 K for 24 h under vacuum.
95	Ballock, G.	1956	V	928-1150		0.027 Mn, 0.02 C, 0.018 P, 0.017 S, 0.007 Si, 0.005 N and traces Ni; Armco iron manufactured by the basic O.H. technique; inclusions not detectable by microscope; 1.3 cm in diam and about 15 cm long; measured with a current of 20-30 A, and in vacuo; this curve represents coincident values during both heating and cooling (rate 1-1.5 K/min).
96*	Ballock, G.	1956	V	1004-1221		The above specimen while cooling.
97*	Ballock, G.	1956	V	1003-1220		The above specimen while heating.
98*	Ballock, G.	1956	V	1174-1198		The above specimen heating at a rate of 0.25 K min ⁻¹ .
99*	Ballock, G.	1956	V	1163-1186		The above specimen cooling at a rate of 0.25 K min ⁻¹ .
100*	Kaufman, L., Clougherty, E.V., and Weiss, R.J.	1963		325-1425		Specimen same as used in Kaufmann, L., Leyenaar, A., and Harvey, J.S., Progress in Very High Pressure Research, p. 89, Wiley, New York, 1961; swaged; α - γ transition 1183 K; resistivity calculated from reported magnetic resistivity: ρ (magnetic) = ρ - 0.029T (1-0.002P); values from table.
101*	Morris, D.M.	1897	B	273-1036	Specimen A	Charcoal iron; from Messr. Jos. Sankey and Sons; ring shape specimen of cross sectional area 0.131 cm ² and mean ring diameter 2.35 cm; density 7.775 g cm ⁻³ ; measured during heating.
102*	Morris, D.M.	1897	B	273-1323	Specimen A	The above specimen; measured during cooling after heated to 1323 K; Curie temperature 1068 K.
103*	Morris, D.M.	1897	B	289-1158	Specimen A	The above specimen after reheating to 1193 K; measured during cooling.
104*	Morris, D.M.	1897	B	273-1099	Specimen B	0.075 impurities, including C, P and Si, and traces of Mn; Swedish transformer iron from Messr. Jos. Sankey and Sons; ring shape specimen with cross sectional area 0.143 cm ² and mean ring diam 2.23 cm; density 7.461 g cm ⁻³ ; measured during cooling after annealing at 1113 K; Curie temperature 1055 K.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
105	150	Morris, D.M.	1897	B	273-1423	Specimen B	The above specimen measured during cooling after annealing at 1423 K Curie temperature 1051 K.
106	190	Arajs, S., Schwerer, F.C., and Fisher, R.H.	1969	A	4.2		99.9 ⁺ pure; electrolytic; about 5 mm in diam and 50 mm long.
107	132	Schwerer, F.C., Conroy, J.W., and Arajs, S.	1969	A	4.5-300		0.002 C and 0.001 N, "high purity iron standard"; 0.508 cm in diam and 5 cm long; annealed at 1273 K for 1 week under vacuum; machined, re-annealed at 1123 for 2 h; data at T > 50 K calculated from reported residual resistivity (0.08 x 10 ⁻⁹ Ωm) and smoothed ideal resistivity (from graph); measured with a current of 0.1 A.
108	141	Shirekawa, Y.	1939		78-1123		0.05 P and Si each, 0.04 C, 0.02 Co and Mn each, 0.01 Al and 0.003 S; electrolytic from Nippon-Deukai-Seitetsusho; 0.0617 cm in diam and 5.25 cm long; annealed at 1273 K for 1 h under vacuum with specimen axis in the east-west direction; slow-cooled; lead wires of nickel soldered by pure silver; reannealed at 1123 K for 1 h under vacuum and slow-cooled; measurement done with sample axis in east-west direction.
109*	191	Meyer, A.R.	1911	V	273-1273		Chemically pure; ~0.015 SiO ₂ , 0.004 Cu and Ni each, <0.001 Mn and trace Si; from Kahlbaum; impurities analyzed by Physikalisches-Technisches Reichsanstalt; measured by AC voltage-current method; smoothed values from table; except for value at 293 K which is measured separately by a DC voltage-current method.
110	191	Meyer, A.R.	1911	V	273-1273		99.94 pure charcoal iron from Araco; other information same as above.
111	191	Meyer, A.R.	1911	V	273-1173		<0.008 S, 0.007 P and traces C, Cu, Mn, and Si; from Langbein-Pfunhauser-Werke; other information same as above.
112	91	Levin, E.S., Ayushina, G.D., and Gel'd, P.V.	1972	R	1923		99.988 pure, carbonyl iron class V-3; measurements carried out in aluminum or zirconium oxide crucibles covered with lapped lids in purified helium at a pressure of 760 mmHg; pure tungsten used as comparison standard.
113	192	Gumenyuk, V.S. and Lebedev, V.V.	1959		309-1718		High purity iron obtained by vacuum distillation; total impurity 0.02%, estimated from residual resistivity; specimen 3-6 mm in diam and 50-100 mm long; annealed at 1373 K for 4 h in high vacuum; measured in a vacuum of 10 ⁻³ -10 ⁻⁴ mmHg.
114	156	Reum, B.A., Gel'd, P.V., and Tygunov, G.V.	1967	→	1551-2018		99.99 pure carbonyl iron; remelted in a hydrogen atm and degassed in vacuum in the molten state; measured by "refluence method" of Baum et al., Izv. Akad. Nauk SSSR, Neorg. Materialy, 1, 1289(1965) in pure helium at a pressure of 780 mmHg; tungsten used as comparison standard.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
115	89	Ono, Y. and Yagi, T. Ono, Y.	1972 1977	R R	1773-1898		99.9% pure; liquid state; contained in a 10 mm I.D. recrystallized alumina crucible at a pressure of 0.05-0.1 mmHg; density data of Saito et al. (Bull. Res. Inst. Min. Dress. Metall., Tohoku Univ., 25, 67, 109, 1969) used for calculating specimen volume.
116	149	Seehra, M.S., Capan, V.L., and Silinsky, P.	1974	A	1000-1087		0.02 Mn, 0.01 Cr, and <0.01 Co; single crystal; 0.2 x 0.3 x 1.5 cm; polished and etched in HCl + 10% H ₂ O ₂ ; Curie temperature 1040 ± 1 K; uncorrected for thermal expansion; data in table form supplied by the first author.
117	193	Dubini, E., Esin, O.A., and Vetolin, N.A.	1969		1873		"High purity"; measured in purified helium.
118	194	Lebedev, S.V., Savvatimskii, A.I., and Smirnov, Yu.B.	1974	→	1809		<0.2 C; in liquid state; measured by an exploding wire technique, wire heated by an almost rectangular shape pulse (10 ⁻⁶ s); current density ~4 x 10 ¹⁰ A/m ² ; voltage and current measured by pulse oscillogram.
119*	127	Fert, A. and Campbell, I.A.	1976	A	2.4-63		"Pure"; obtained by zone melting; residual resistivity ratio 25; only temperature dependent part of resistivity reported.
120*	127	Fert, A. and Campbell, I.A.	1976	A	2.4-78		Same as the above.
121*	159	Baum, B.A., Tyagunov, G.V., Gal'd, P.V., and Khasin, G.A.	1971	R	1573, 1873		Specimen contained in either alumina or zirconia crucible, measured in an atm of helium.
122*	195	Tanaka, K. and Watanabe, T.	1972	A	77, 298	JM	0.0166 N, 0.0048 C, 0.0036 O, 0.0003 Si, 0.0002 Cu and Mn each, and <0.0001 Al and P each; grain diam 50 ± 20 μm; 0.5 mm in diam and 170 mm long; from Johnson and Matthey; heated at 1246 K for 48 h in wet hydrogen; and 2 h in dry hydrogen; cold rolled from 5 mm to 2 mm in diam; annealed at 823 K for 1 h in dry hydrogen; drawn to wire; annealed at 923 K for 3 h in vacuum; carbon or nitrogen in solution <0.0002.
123*	195	Tanaka, K. and Watanabe, T.	1972	A	77, 298	RE	0.0060 S, 0.0040 C, Mn and Si each; 0.0030 P and 0.0020 Cu; re-electrolytic iron supplied by Denko Co.; melted by induction heating and cast in vacuum; surface layer removed; hot-swaged into rod of 7 mm in diam; wire specimen prepared in similar manner as above.
124*	195	Tanaka, K. and Watanabe, T.	1972	A	77, 298	TI	0.0400 Ti; "C, N, O and B atoms in solid solution extremely low"; prepared from re-electrolytic iron by alloying with Ti; specimen preparation same as the above except for no annealing in wet hydrogen.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
125*	196	Wagenblast, H., Schwerer, F.C., and Borak, J.A.	1971	A	4.2		0.005 interstitial C; specimen prepared from vacuum melted iron with <0.1 at.% impurities; drawn and swaged to 0.6 mm in diam and 13.3 cm long wire; annealed at 1023 K for 15 min; annealed 1058 K for 5 h in wet hydrogen and furnace cooled in dry hydrogen to reduce carbon and nitrogen to 0.004 and 0.0004 at.%, respectively; carbonized by heating at 998 K for 16 h in a hydrogen-methane mixture; quenched in brine.
126*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.022 interstitial C.
127*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.005 interstitial C.
128*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.098 interstitial C.
129*	196	Wagenblast, H., et al.	1971	A	4.2		0.012 interstitial N; specimen preparation similar to the above except nitrogenized by heating at 748 K in a hydrogen-ammonia mixture.
130*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.057 N.
131*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.104 N.
132*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.138 N.
133*	196	Wagenblast, H., et al.	1971	A	4.2		Similar to the above except containing 0.158 N.
134	80	Price, D.C. and Williams, G.	1973	A	4.2-276		99.9985 pure; 0.15 x 0.2 x 10 cm; supplied by Johnson and Matthey Co.; prepared by cold rolling rod stock between Melinex sheets; etched; annealed in vacuum at 1173 K for 2 h; quenched; resistivity calculated from reported $\rho(4.2 \text{ K}) = 0.3300 \times 10^{-9} \Omega \text{ m}$ and temperature dependent part of the resistivity, ρ_T ; $\rho = \rho_T + \rho(4.2 \text{ K})$.
135	197	Vasil'eva, R.P. and Kadyrov, Ya.	1974		373-773		No details reported.
136	148	Schwerer, F.C. and Cuddy, L.J.	1970	V	4.2-1200		$\sim 1.8 \text{ mm}$ in diam; zone-refined iron; swaged; average of two specimens; $\rho(4.2 \text{ K}) \sim 0.04 \times 10^{-9} \Omega \text{ m}$; smoothed values from graph.
137*	94	Samarin, A.M.	1962	R	1811-2000		Measured by the rotating field method in a helium atm; apparatus calibrated against the resistivity value of molten iron reported in Data Set 139; resistivity value calculated from reported conductivity: $[1.47 - 0.50 \times 10^{-3} T(C)] \times 10^4 \text{ dm}^{-1} \text{ cm}^{-1}$; upper temperature limit assumed 2000 K.
138	155	Powell, R.W.	1953	*	279-1793		"High purity" iron; measured under vacuum; resistivity above 1623 K are smoothed values from graph.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
139	155	Powell, R.W.	1953	+	1811		In molten liquid state; resistivity measured by immersing a specially constructed alumina tube in molten iron; current and potential contacts made by tungsten rods through separate holes, in the middle of the wall of the tube, which are open to the axial hole through small up-turned channels; reported value mean of 24 measurements with two heating rates of the specimen and with two different durations of the measuring current.
140*	108	Schismak, H.	1914	B	20-273	Fe I	High grade pure electrolytic iron; from Kalbaum; 1-2 m long; drawn by Heraeus of Hanau.
141*	108	Schismak, H.	1914	B	20-273	Fe II	Same as the above except annealed in nitrogen atm.
142*	198	Holborn, L.	1919		80-784	Fe I ₁	Electrolytic iron, from vacuum melted iron supplied by Firma W.C. Heraeus; wire specimen 0.2 mm in diam; heated for several min at 773 K.
143*	198	Holborn, L.	1919		81-761	Fe I ₂	Same as the above except annealed at 573 K for 3 h.
144*	198	Holborn, L.	1919		80-572	Fe II	0.004 Co, Cu and Ni each, 0.001 Mn, traces of C, O and Si; "Nitrateisen" made from iron nitrate by Firma C.A.F. Kalbaum; drawn from 5 mm to 0.2 mm in diam; annealed at 653 K for 3 h.
145*	104	Meissner, M. and Voigt, B.	1930	+	1.4-273	Fe 1	Specimen same as for Data Set 142; 0.2 mm in diam and 55 mm long; distance between potential contacts 50 mm; tempered; measured by compensation method; resistivity calculated from reported resistance ratio, ice point resistance (0.149 Ω) and sample dimensions.
146*	104	Meissner, M. and Voigt, B.	1930	+	1.4-273	Fe 2	Specimen same as for Data Set 144; 0.2 mm in diam and 59.7 mm long; distance between potential contacts 56.6 mm; tempered; measurement method and resistivity calculation same as above.
147	104	Meissner, M. and Voigt, B.	1930	+	1.4-273	Fe 2*	Same as the above specimen except 60 mm long and distance between potential contacts 56.3 mm.
148	104	Meissner, M. and Voigt, B.	1930	+	2.0-273	Fe 3	Specimen same as for Data Set 10; tempered; 1.0 mm in diam and 33.0 mm long; distance between potential contacts 30.0 mm; measurement method and resistivity calculation same as above.
149	104	Meissner, M. and Voigt, B.	1930	+	2.0-273	Fe 4	Electrolytic iron from Firma Heraeus; 1.0 mm in diam and 58.2 mm long; stretched; course grained; distance between potential contacts 53.4 mm; measurement method and resistivity calculation same as above.
150*	104	Meissner, M. and Voigt, B.	1930	+	1.4-273	Fe 5	Specimen obtained from Dr. Kreuzer; 0.1 mm in diam and 58.6 mm long; distance between potential contacts 54.2 mm; measurement method and resistivity calculation same as above.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
151*	Meisner, M. and Voigt, B.	1930	→	1.4-273	Fe 6	Same as the above specimen except 58.4 mm long and annealed at 573 K for 3 h; distance between potential contacts 54.4 mm.
152*	Meisner, M. and Voigt, B.	1930	→	1.4-273	Fe 7	Electrolytic (5 times) iron from Firma Siemens and Matske; 0.3 mm in diam and 58.7 mm long; distance between potential contacts 54.0 mm; measurement method and resistivity calculation same as above.
153*	Meisner, M. and Voigt, B.	1930	→	1.3-273	Fe 10	The above specimen annealed at red-hot, 15 h and etched; 57.5 mm long; distance between potential contacts 51.7 mm.
154*	Meisner, M. and Voigt, B.	1930	→	1.4-273	Fe 8	Same as the above specimen except annealed at 573 K for 3 h; distance between potential contacts 54.4 mm.
155*	Meisner, M. and Voigt, B.	1930	→	1.4-273	Fe 9	Same as the above specimen etched; 57.8 mm long; distance between potential contacts 55.1 mm.
156*	Trussell, C.H., Christopher, J.R., and Coleman, R.V.	1970	A	0.3-1.2		<100> iron whisker; measured in a magnetic field of 570 Oe.
157*	Matthiessen, A. and Voigt, G.	1864		273		Hard-drawn; resistivity value calculated from reported ratio of resistivities of silver and iron, with $\rho(\text{silver})$ assumed to be $1.468 \times 10^{-9} \Omega \cdot \text{m}$.
158*	Potter, R.H.	1937		20-1130		99.96 pure; chief impurities are O and Si; from Messrs. Adam Hilger; U-shape specimen 2 mm in diam and 8 cm long.
159*	Ribbeck, F.	1926	→	273-1273		0.07 Mn, 0.014 P, and traces of Si, Cu, S and Cr; electrolytic; 0.3-0.4 cm ² x 10 cm; measured by compensation method with current 2-3 A.
160*	Bhagat, S.M., Anderson, J.R., and Wu, H.	1967		84-297		<111> iron whiskers; about 0.2-0.4 mm wide and 8 mm long; grown by hydrogen reduction of FeCl ₃ either at room temperature using hydrogen saturated with water vapor or at 1023 K with a hydrogen flow rate of 300 ml/min; electropolished; measured in a longitudinal magnetic field of 2 kG.
161*	Mumery-Tassy, G.	1950		194-1208		No details reported.
162*	Sudovtsov, A.I. and Semenov, E.E.	1957	A	1.2-4.2		99.98 pure; polycrystalline specimen in the form of thin ribbons from Hilger; $R(4.2\text{K})/R(273\text{K}) = 3.928 \times 10^{-2}$; resistance at 273 K, 0.5091 Ω ; $R(T)/R(273\text{K}) = 3.92930 \times 10^{-2}$ with T extrapolated to 0 K.
126	Semenko, E.E. and Sudovtsov, A.I.	1962	A	1.3-20.3		>99.99 pure; grain size $\sim 0.1 \mu\text{m}$; $\sim 0.1 \text{ mm}$ "transverse dimension" 38 mm long; needle-shaped specimen grown by distillation in vacuum; $R(T)/R(273\text{K}) = 3.9606 \times 10^{-2}$ with T extrapolated to 0 K; measured under condition where the earth's magnetic field is compensated by Helmholtz coils; specimen demagnetized with a 50 cps magnetic field of decreasing amplitude after each reversal in measuring current.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
164	204	Semenenko, E.E., Sudovtsov, A.I., and Shvets, A.D.	1962	A	0.4-1.3		The above specimen measured at lower temperatures.
165*	116	Glaesser, W., Imgrund, H., and Wever, H.	1967		4.2		0.008 total impurities; pure iron from Firma HEK. Lübeck.
166	116	Glaesser, W., et al.	1967		4.2		The above material 1 time zone-melted.
167	116	Glaesser, W., et al.	1967		4.2		The above material 5 times zone-melted.
168*	116	Glaesser, W., et al.	1967		4.2		The above material 10 times zone-melted.
169*	116	Glaesser, W., et al.	1967		4.2		The above material co-worked.
170	116	Glaesser, W., et al.	1967		4.2		The above material recrystallized at glowing.
171*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-2	99.999 pure; from United Mineral and Chemical Co.; 0.5 mm in diam; distance between potential leads 3.5 cm; annealed at about 673 K under vacuum (7×10^{-9} Torr); measured under zero applied magnetic field.
172*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-2	The above specimen; resistivity value obtained by extrapolating its magnetoresistance to zero magnetic induction.
173*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-2	Same as Data Set 171 except with a different set of potential contacts.
174*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-2	Same as above; resistivity value obtained by extrapolating the magnetoresistance to zero magnetic induction.
175*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-1	Same specimen material as above; about 4 mm in diam; twice electron beam, float-zone-refined under 10^{-5} Torr; distance between potential leads 4 cm; measured in zero applied magnetic field.
176*	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-1	Same as above; resistivity value obtained by extrapolating the magnetoresistance to zero magnetic induction.
177	118	Schindler, A.I. and LaRoy, B.C.	1966	A	4.2	Fe-3	"High purity" single crystal from Materials Research Corp.; 2 mm in diam; distance between potential contacts 1.5 cm; measured under zero applied magnetic field.
178*	98	Schröder, K. and Giannuzzi, A.	1969		879-1126		Thermocouple grade; annealed in Ar at about 1190 K for 2 h.
179*	124	Beitchman, J.C., Trussel, C.W., and Coleman, R.V.	1970	A	0.3-1.2	T-7	Single crystal; specimen axis in a <111> direction; measured in a magnetic field of 950 Oe.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
180*	124	Beitchman, J.G., Trussel, C.W., and Coleman, R.V.	1970	A	0.4-1.2	T-7	The above specimen measured in a longitudinal magnetic field of 1230 Oe.
181*	124	Beitchman, J.G., et al.	1970	A	0.3-1.2	T-7	The above specimen measured in a longitudinal magnetic field of 1520 Oe.
182*	124	Beitchman, J.G., et al.	1970	A	1.0-4.1	T-7	The above specimen measured in a longitudinal magnetic field of 1150 Oe.
183*	124	Beitchman, J.G., et al.	1970	A	1.4-4.3	B-1	Single crystal; specimen axis in a <111> direction; measured in a longitudinal magnetic field of 1200 Oe.
184*	124	Beitchman, J.G., et al.	1970	A	4.7-21	B-1	The above specimen, measured at higher temperatures.
185*	205	Swartz, J.C. and Cuddy, L.J.	1970	V	4.2		Zone-refined iron; 0.13-0.40 mm in diam and 5-10 cm long; resistivity value calculated from reported $\rho(295K)/\rho(4.2K) = 180$, with $\rho(295K)$ taken to be $10.19 \times 10^{-8} \Omega m$.
186*	119	Arajs, S., Oliver, B.F., and Michalak, J.T.	1967	A	4.2	I	99.9966 pure; 0.0019 C, 0.0011 O and 0.0004 others (at. %); interfacial grain area 7.0 mm^{-2} ; 1 mm in diam and about 80 mm long; produced by oxidation zone refining (oxygen activity ~ 1).
187*	119	Arajs, S., et al.	1967	A	4.2	II	0.0019 C, 0.0018 O and 0.0042 others (at. %); polycrystalline; interfacial grain area 14.3 mm^{-2} , 80 cm long.
188*	119	Arajs, S., et al.	1967	A	4.2	II	Same as the above except interfacial grain area 16.5 mm^{-2} .
189*	112	Fujii, T. and Morimoto, I.	1968	A	4.2	Fe I	0.0300 C (determined by vacuum combustion method), 0.0100 N and 0 each (determined by vacuum fusion method), and 0.0015 total metallic impurity; polycrystalline material obtained from Johnson and Matthey Co.; formed into a bar 5 mm in diam and 20 cm long; swaged into cylindrical rod 2.7 mm in diam and 50 cm long; annealed at 1163 K; chemically polished in a 500-1-500 solution of $\text{H}_2\text{O}-\text{HF}-\text{C}_2\text{H}_5\text{OH}$, removing a surface layer of 0.1 mm; resistivity value calculated from reported $\rho(295K)/\rho(4.2K)$, with $\rho(295K)$ taken to be $9.91 \times 10^{-8} \Omega m$.
190*	112	Fujii, T. and Morimoto, I.	1968	A	4.2	Fe I	Same specimen material as the above, prepared by a method similar to the above but exact treatment not given; resistivity calculated by some method as above.
191*	112	Fujii, T. and Morimoto, I.	1968	A	4.2	Fe II	0.0300 C, 0.0100 O, and 0.0015 N (determined by the same methods as for Data Set 189), and 0.0015 total metallic impurity; from the same specimen material as the above; zone-refined (1 pass at 3 mm min ⁻¹ in dry H_2); other preparations same as the above except annealed at 1123 K for 20 h in a vacuum of 2×10^{-6} mm before chemical polishing; resistivity calculated by same method as above.
192*	112	Fujii, T. and Morimoto, I.	1968	A	4.2	Fe III	0.0100 O, 0.0080 N, 0.0030 C (determined by same methods as for Data Set 189), and 0.0015 total metallic impurity; prepared from the same material and by a similar method as the above except decarbonized at 1023 K for 200 h in wet H_2 ; resistivity calculated by same method as above.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
193*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Prepared from the same specimen material and by a similar method as the above, but exact method not given; resistivity calculated by the same method as above.
194*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
195*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
196*	112	Fujii, T. and Morimoto, I.	1968	A	4.2	Fe IV	0.0030 C, <0.0005 N, 0.0004 C (determined by the same methods as for Data Set 189); and 0.0015 total metallic impurity; zone-refined (1 pass at 0.3 mm min ⁻¹ in dry H ₂); other preparations same as above; resistivity calculated by same method as above.
197*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Prepared from the same specimen material and by a similar method as the above; but exact method not given; resistivity calculated by the same method as above.
198	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
199*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
200*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
201	112	Fujii, T. and Morimoto, I.	1968	A	4.2		0.0020 C, 0.0001 O, 0.015 total metallic impurity and trace N (determined by same method as for Data Set 189); zone-refined (1 pass at 0.3 mm min ⁻¹ and 5 passes at 1 mm min ⁻¹ in dry H ₂); other preparations and resistivity calculation same as the above.
202*	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Prepared from the same specimen material and by a similar method as the above, but exact method not given; resistivity calculated by same method as above.
203	112	Fujii, T. and Morimoto, I.	1968	A	4.2		Same as the above.
204	112	Fujii, T. and Morimoto, I.	1968	A	4.2		<0.0010 C, 0.015 total metallic impurity and traces N and O (determined by the same method as for Data Set 189); zone-refined (2 passes at 0.3 mm min ⁻¹ in wet H ₂ , 5 passes at 1 mm min ⁻¹ in vacuum and 2 passes at 1 mm min ⁻¹ in dry H ₂); other preparations same as the above; resistivity calculated by same method as above.
205	87	Seydel, U. and Pucke, W.	1977	+	1007-2997		99.99 pure; 0.0003 Ca and Si each, 0.0002 Al, Cu, and Mg each, and 0.0001 Ag, Cr, Mn, and Ni each (chemical analysis); measured by an exploding wire technique; measurement error 4%; smoothed values from curve; values corrected for thermal expansion.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
206	92	Güntherodt, H.J., Hauser, E., Künzi, H.U., and Müller, R.	1975	+	1726-1915		99.999 pure from Johnson and Matthey Co.; measured with potential method in which the sample material was enclosed within an alumina tube with four protrusions serving as current and potential contacts.
207*	110	Holder, T.K.	1976		90-400		99.99 pure, 0.0025 C and N each, 0.0007 O and 0.0001 H; Material Research Corp. MARZ grade 3 pass zone refined iron; smoothed values from table; $\rho(273.15K)/\rho(4K) = 189$.
208	207	Ershov, G.S., Kasatkin, A.A., and Gavrilin, I.V.	1974		1828-2065		99.997 ⁺ pure; measured by a contact method in a helium atm with specimens inside a vertical alumina crucible; liquid metal column 40-50 mm long.
209	113	Volkenshtein, N.V. and Yakina, V.P.	1971	A	4.2-46	Fe-4	Polycrystalline specimen from Johnson and Matthey Co.; 0.1 mm thick, 3.0 mm wide and 15 mm long; vacuum (10^{-6} mmHg) annealed at 1273 K for 1 h, demagnetized; measuring current density 3.3 A mm^{-2} .
210*	113	Volkenshtein, N.V. and Yakina, V.P.	1971	A	4.5-494		The above measured in an applied longitudinal magnetic field of 1.1 kOe.
211*	113	Volkenshtein, N.V. and Yakina, V.P.	1971	A	4.4-46.1		The above measured in an applied transverse magnetic field of 1.1 kOe.
212	93	Kita, Y., Ohguchi, S., and Morita, Z.	1978	+	1695-1895		0.008 Ni, 0.006 S, 0.005 Si, <0.005 Cu and Cr each, 0.003 Mn and P each, and 0.002 C; measured with a four probe method in which the electrodes are made of the same material as the specimen, in a vacuum of 10^{-4} Torr; data points are taken at temperatures in the sequence: 1833, 1854, 1864, 1880, 1895, 1872, 1855, 1835, 1816, 1799, 1786, 1759, 1735, 1713, 1695 K; values from table supplied by authors; values corrected for thermal expansion.
213*	93	Kita, Y., et al.	1978	+	1676-1919		Same as the above; a second melt; temperature sequence: 1823, 1842, 1857, 1874, 1893, 1905, 1919, 1900, 1875, 1858, 1836, 1817, 1803, 1798, 1776, 1760, 1741, 1720, 1699 and 1676 K.
214*	93	Kita, Y., et al.	1978	+	1673-1973		Same as the above; a third melt; temperature sequence: 1823, 1843, 1866, 1876, 1889, 1905, 1915, 1937, 1915, 1896, 1869, 1850, 1833, 1814, 1829, 1845, 1863, 1878, 1893, 1910, 1894, 1879, 1864, 1850, 1841, 1817, 1802, 1777, 1764, 1748, 1728, 1708, and 1673 K.
215*	157	Arzentiev, P.P., Filippov, S.I., and Litvitskii, B.S.	1970	+	1693-1874		Specimen produced from electrolytic powder of composition: 0.23 C, 0.015 O, 0.012 S, 0.005 P and Si each, and trace Mn; melted in a hydrogen atm; electrical resistivity reported is the same as that reported for a 0.005 C specimen; measured with a potential method with tungsten electrodes; experimental chamber evacuated before heating and then filled with pure Fe; measured while heating.

* Not shown on either figure.

TABLE 8. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
216*	157	Arsentiev, P.P., Phillipov, S.I., and Litsitskii, B.S.	1970	+	1683-1874		The above; measured while cooling.
217	147	Lauchbury, M.D. and Saunders, N.H.	1976	A	373-1128		<0.03 Mn and <0.01 Cu, Si, Ca, and Mg each; cylindrical specimens 1.5 mm or 2 mm in diam and ~20 mm long, machined from 5 mm diam polycrystalline rods from Johnson and Matthey Co.; annealed at 1250 K for several hours in an argon atm; random measurement error 1%.
218	128	Janos, S., Kovacs, L., and Mlynek, R.	1972		9.9-28		Only temperature dependent part of resistivity reported; values from graph.
219	115	Ishiki, M. and Igaki, K.	1978	A	1.7-271		High purity, prepared by floating zone refining and heated treated at 1073 K for 24 h in wet hydrogen described by authors in Trans. Jpn. Inst. Metals, 18, 413, 1977; specimen then electropolished in 95% acetic acid and 5% perchloric acid from a diam of 500 μm to 150 μm ; about 10 cm in length; measured in a longitudinal applied magnetic field of 60 KA m^{-1} ; values from graph.
220	115	Ishiki, M. and Igaki, K.	1978	A	1.7-301		Similar to the above except specimen diam reduction from 500 μm to 180 μm .
221	115	Ishiki, M. and Igaki, K.	1978	A	1.7-268		Similar to the above except specimen diam reduction from 500 μm to 190 μm .
222	115	Ishiki, M. and Igaki, K.	1978	A	1.6-164		Similar to the above.
223	115	Ishiki, M. and Igaki, K.	1978	A	1.6-292		Similar to the above except specimen diam reduction from 500 μm to 350 μm .

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe
 [Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \cdot m$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ			
<u>DATA SET 1</u>																
2.1	1.46*	77.78	0.925	4.2	0.092	873	71.0	194.1	5.31	380	14.77	<u>DATA SET 17 (cont.)</u>				
2.6	1.46*	194.8	5.26	15.2	0.097	973	87.5	273	9.04	400	16.09					
3.0	1.46*	273.3	8.97*	20.8	0.100	1073	107.2	323	11.72	<u>DATA SET 18</u>						
3.7	1.46*	298.8	10.33*	26.1	0.106								1.63	0.0859		
4.3	1.46*												3.14	0.0899		
4.9	1.46*												4.33	0.0875		
5.5	1.46*												6.30	0.0955		
6.2	1.46*												9.02	0.0935		
7.1	1.46*												80.7	0.659		
8.3	1.46*												<u>DATA SET 19</u>			
10.4	1.46												293	10.4		
12.3	1.46												373	14.4*		
15.0	1.47												473	22.0		
17.0	1.47												573	31.4		
18.6	1.44												673	42.2		
21.4	1.44												773	54.8		
23.7	1.44												873	69.4		
25.4	1.47												973	86.1		
27.7	1.48												1073	106.0		
30.0	1.51												1173	111.8		
33.1	1.51												1273	115.0*		
36.4	1.51												1373	118.0		
38.9	1.50												<u>DATA SET 20</u>			
41.5	1.51												4.2	0.0428		
44.0	1.51												90	0.97		
49.5	1.59												100	1.27*		
51.9	1.61												120	1.95		
53.9	1.68												140	2.67		
56.8	1.80												160	3.47		
58.9	1.84												180	4.31		
61.2	1.89												200	5.20		
62.9	1.92												220	6.11		
66.7	2.02												240	7.04		
69.2	2.08												260	8.00		
72.6	2.14												273	8.61		
<u>DATA SET 2</u>																
90.2	1.09												280	8.99		
194.7	5.78												300	10.01*		
273.2	9.06*												320	11.09		
373.2	14.73												340	12.25		
<u>DATA SET 3</u>																
21.2	1.060												360	13.48		
83.2	1.917												<u>DATA SET 21</u>			
273.2	9.95												298.2	10.37		
<u>DATA SET 4*</u>																
323	18.7												323.2	11.86*		
373	22.0												373.2	14.69*		
473	25.9												423.2	18.08*		
523	30.0												473.2	21.89*		
573	34.6												523.2	26.20*		
673	45.0												<u>DATA SET 22</u>			
773	57.1												4	0.40		
<u>DATA SET 5*</u>																
80	0.826												77.5	1.01		
273	10.3												<u>DATA SET 23</u>			
<u>DATA SET 6*</u>																
323	11.7												<u>DATA SET 24</u>			
373	14.7												4.2	0.0428		
423	17.9												90	0.97		
473	21.6												100	1.27*		
523	25.6												120	1.95		
<u>DATA SET 7*</u>																
323	11.9												140	2.67		
373	14.9												160	3.47		
423	18.2												180	4.31		
473	21.8												200	5.20		
523	25.8												220	6.11		
573	30.3												240	7.04		
673	41.0												260	8.00		
773	53.3												273	8.61		
873	67.9												280	8.99		
973	85.2												300	10.01*		
1073	104.2												320	11.09		
<u>DATA SET 8*</u>																
21.2	0.1437												340	12.25		
83.2	0.929												360	13.48		
273.2	9.11												<u>DATA SET 25</u>			
<u>DATA SET 9*</u>																
21.2	1.060												4	0.40		
83.2	1.917												77.5	1.01		
273.2	9.95												<u>DATA SET 26</u>			
<u>DATA SET 10</u>																
21.2	0.0681												4	0.40		
83.2	0.778												77.5	1.01		
273.2	8.71												<u>DATA SET 27</u>			

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ		
<u>DATA SET 21(cont.)</u>		<u>DATA SET 22(cont.)</u>		<u>DATA SET 24(cont.)</u>		<u>DATA SET 29*</u>		<u>DATA SET 33</u>		<u>DATA SET 35(cont.)*</u>	
573.2	30.97	1273	115.8	973.2	87.44*	293.2	9.72	293.2	10.00*	983.4	86.5
623.2	36.16	1323	117.5	983.2	89.76*	<u>DATA SET 30*</u>		323.2	11.82*	994.0	88.6
673.2	41.82	1373	119.0	993.2	92.00	20.2	0.261	373.2	14.85*	1001.1	90.2
723.2	47.87	1423	120.4	1003.2	93.96*	22.6	0.263	423.2	18.16*	1005.1	91.0
773.2	54.39	1473	121.8	1013.2	95.65	24.0	0.266	473.2	21.97	1011.0	92.0
823.2	61.39	1523	123.1	1023.2	97.17*	28.8	0.276	523.2	26.20	1014.1	92.9
873.2	68.87*	<u>DATA SET 23*</u>		1033.2	98.50*	33.3	0.290	573.2	30.89	1016.0	93.3
923.2	76.96	90	1.08	1043.2	99.67	36.6	0.300	623.2	35.97*	1018.7	93.8
973.2	85.85	195	5.15	1053.2	100.70*	53.1	0.439	673.2	41.55	1019.8	94.2
1023.2	96.03	290	9.95	1063.2	101.60	64.4	0.593	723.2	47.49	1020.8	94.3
1033.2	98.57	<u>DATA SET 24</u>		1073.2	102.36*	73.6	0.689	773.2	53.94	1021.8	94.5
1036.2	99.41	293.2	9.760*	1083.2	103.03	88.9	1.18	823.2	60.92	1023.5	95.0
1043.2	100.84	323.2	10.80	1093.2	103.66*	105	1.81	873.2	68.51	1025.1	95.4
1053.2	102.23	373.2	13.4	1103.2	104.25	122	2.24	923.2	76.55	1026.3	95.6
1073.2	104.33*	433.2	17.20	1113.2	104.83*	122	2.45	973.2	85.91*	1027.6	95.9
1123.2	108.10*	443.2	18.83*	1123.2	105.40*	131	3.64	998.2	90.95	1028.9	96.3
1173.2	110.78	453.2	19.66	1133.2	105.92	151	5.10	1023.2	96.31*	1031.6	96.9
1223.2	112.79	463.2	20.49*	1143.2	106.41*	186	6.87	1048.2	102.2	1032.8	97.2
1273.2	114.49	473.2	21.31*	1153.2	106.85	221	10.1	1073.2	105.5*	1034.2	97.6
1323.2	116.04	483.2	22.14	1163.2	107.23*	282	10.1	1098.2	107.8	1036.7	98.1
<u>DATA SET 22</u>		493.2	22.98*	1173.2	107.55	293.2	10.0*	1123.2	109.6*	1038.1	98.6
273	8.86*	513.2	24.69*	1183.2	107.83*	<u>DATA SET 32</u>		1173.2	112.3*	1042.9	99.7
293	9.81*	523.2	25.58*	1193.2	108.07	0.452	0.339	1223.2	113.4	1045.9	100.5
323	11.54*	533.2	26.58*	1203.2	108.26	0.643	0.346	1273.2	115.0	1047.7	100.8
373	14.53*	543.2	27.58*	1213.2	108.43*	0.752	0.339*	1323.2	116.6	1050.8	101.2
423	17.85*	553.2	28.58*	1223.2	108.57	0.788	0.350	1373.2	118.2	1054.7	101.8
473	21.55*	563.2	29.58*	1233.2	108.70*	0.849	0.350	1423.2	119.9	1061.7	102.7
523	25.65*	573.2	30.30*	<u>DATA SET 25*</u>		0.860	0.349	1473.2	122.1	1079.8	104.6
573	30.2*	583.2	31.30*	293.2	9.69	0.860	0.339	<u>DATA SET 34</u>		<u>DATA SET 36*</u>	
623	35.3	593.2	32.30*	<u>DATA SET 26*</u>		0.643	0.346	90	1.30	974.6	85.4
673	40.95*	603.2	33.30*	293.2	9.72	0.752	0.339*	133	2.48	984.2	88.0
723	47.0	613.2	34.30*	<u>DATA SET 27*</u>		0.849	0.350	152	3.19	994.3	89.9
773	53.7	623.2	35.30*	293.2	9.71	0.860	0.339	172	4.06	1002.4	91.5
823	60.9*	633.2	36.30*	<u>DATA SET 28*</u>		0.860	0.349	192	5.15	1012.3	93.6
873	68.7*	643.2	37.30*	293.2	9.70	0.892	0.338	211	6.01	1018.9	95.0
923	76.85*	653.2	38.30*	<u>DATA SET 29*</u>		0.936	0.350	231	6.95	1022.2	95.7
973	85.9*	663.2	39.30*	293.2	9.71	0.944	0.353	251	7.95	1025.6	96.6
1023	96.0*	673.2	40.30*	<u>DATA SET 30*</u>		0.952	0.338	271	8.92	1028.0	97.3
1073	104.9	683.2	41.30*	293.2	9.70	1.001	0.333	<u>DATA SET 35*</u>		1031.1	98.2
1123	111.6	693.2	42.30*	<u>DATA SET 31*</u>		1.031	0.328	977.6	83.5	1035.1	99.1
1173	118.7	703.2	43.30*	293.2	9.70	1.102	0.326	<u>DATA SET 36*</u>		1039.2	100.1
1223	126.3	713.2	44.30*	<u>DATA SET 32*</u>				1044.9	101.4	1044.9	101.4
		723.2	45.30*	<u>DATA SET 33*</u>				1059.8	103.4	1059.8	103.4
		733.2	46.30*	<u>DATA SET 34*</u>				1078.8	105.3	1078.8	105.3
		743.2	47.30*	<u>DATA SET 35*</u>							
		753.2	48.30*	<u>DATA SET 36*</u>							
		763.2	49.30*	<u>DATA SET 37*</u>							
		773.2	50.30*	<u>DATA SET 38*</u>							
		783.2	51.30*	<u>DATA SET 39*</u>							
		793.2	52.30*	<u>DATA SET 40*</u>							
		803.2	53.30*	<u>DATA SET 41*</u>							
		813.2	54.30*	<u>DATA SET 42*</u>							
		823.2	55.30*	<u>DATA SET 43*</u>							
		833.2	56.30*	<u>DATA SET 44*</u>							
		843.2	57.30*	<u>DATA SET 45*</u>							
		853.2	58.30*	<u>DATA SET 46*</u>							
		863.2	59.30*	<u>DATA SET 47*</u>							
		873.2	60.30*	<u>DATA SET 48*</u>							
		883.2	61.30*	<u>DATA SET 49*</u>							
		893.2	62.30*	<u>DATA SET 50*</u>							
		903.2	63.30*	<u>DATA SET 51*</u>							
		913.2	64.30*	<u>DATA SET 52*</u>							
		923.2	65.30*	<u>DATA SET 53*</u>							
		933.2	66.30*	<u>DATA SET 54*</u>							
		943.2	67.30*	<u>DATA SET 55*</u>							
		953.2	68.30*	<u>DATA SET 56*</u>							
		963.2	69.30*	<u>DATA SET 57*</u>							

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ		
DATA SET 37*											
1165.5	110.311	1183.6	110.589	1103	105.8	253	7.9	0.390	53.8		
1167.3	110.345	1184.4	110.603	1123	107.3	273	8.8	0.390	76.1		
1169.1	110.387	1186.2	110.624	1143	108.7	293	9.87	0.392	DATA SET 50		
1171.9	110.437	1187.0	110.638	1163	109.8	323	11.6	0.399			
1172.9	110.463	1187.9	110.652	1183	110.7	373	14.7	0.410			
1173.9	110.481	1189.7	110.676	1203	111.4	423	18.1	0.426			
1174.6	110.494	1190.6	110.691	1223	112.0	473	21.4	0.450			
1176.2	110.520	1191.4	110.701	1243	112.7	523	26.0	0.484			
1177.3	110.544	1192.5	110.715	1263	113.3	573	30.1	0.528			
1178.2	110.567	1194.9	110.741	1283	114.0	623	35.0	0.585			
1179.0	110.589	DATA SET 39								60	0.654
1179.8	110.604	DATA SET 41								65	0.737
1180.8	110.624	73.2	1.0	1553	121.11	723	46.8	0.737			
1181.7	110.640	123.2	2.5	1573	121.52*	773	53.3	0.832			
1183.5	110.666	173.2	4.3	1593	121.92	823	60.1	0.938			
1183.5	110.680	223.2	6.6	1613	122.34	873	68.0	1.0			
1185.3	110.697	273.2	9.0	1633	122.76	923	76.0	1.188			
1186.1	110.696	323.2	11.8	1653	123.17	973	84.8	1.327	DATA SET 51*		
1187.0	110.696	373.2	14.4	1659	123.31	1023	94.2	1.476			
1188.0	110.689	423.2	17.9	1663	123.49	1073	102.1	1.632			
1188.9	110.689	473.2	21.5	1673	123.74	1123	106.3	1.969			
1192.3	110.687	523.2	25.5	1693	124.22	1173	109.0	2.330			
1193.2	110.698	573.2	30.6	1713	124.73	1223	111.1	2.707			
1194.0	110.709	623.2	35.8	DATA SET 42*						140	3.10
1195.0	110.719	673.2	40.9	83	1.22	291	11.96	3.50	DATA SET 52*		
1197.6	110.745	723.2	46.4	203	5.60	373	16.81	3.91			
DATA SET 38*											
1163.9	110.485	823.2	60.2	223	6.50	DATA SET 44*					
1165.5	110.519	923.2	75.6	248	7.65	DATA SET 45*					
1167.4	110.553	1023.2	84.2	273	8.96	273	9.64	4.32			
1168.3	110.578	1073.2	93.8	293	10.0	373	15.09	4.75			
1169.2	110.595	1123.2	103.3	313	11.3	DATA SET 46*					
1171.8	110.603	1173.2	108.1	333	12.5	DATA SET 48*					
1171.9	110.598	1223.2	110.4	353	13.6	312.07	11.99	5.18			
1172.7	110.582	1273.2	112.5	DATA SET 47*						200	5.61
1173.7	110.552	1323.2	114.0	93	1.1	DATA SET 53					
1174.6	110.522	1373.2	115.3	113	1.7	DATA SET 54					
1175.4	110.480	1423.2	116.6	133	2.4	6	0.387	6.67			
1176.4	110.476	1473.2	117.9	153	3.1	7	0.387	8.364			
1177.3	110.491	1523.2	119.2	173	3.1	8	0.385	10.512			
1178.2	110.514	1573.2	120.6	193	5.0	9	0.385	12.341			
1180.1	110.531	1623.2	121.6	213	5.9	10	0.385	14.498			
1180.9	110.546	1673.2	122.8	233	6.9	12	0.387	16.399			
1181.8	110.566	1715.2	125.2	DATA SET 49*						280	9.43
DATA SET 43(cont.)											
DATA SET 44*											
DATA SET 45*											
DATA SET 46*											
DATA SET 47*											
DATA SET 48*											
DATA SET 49*											
DATA SET 50											
DATA SET 51*											
DATA SET 52*											
DATA SET 53											
DATA SET 54											
DATA SET 55											

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 54(cont.)</u>									
663	45.3*	784	58.1	1042.2	103.89	1182.9	113.16	1580	121.92
735	53.1	793	59.8	1043.2	104.06	1185.7	112.69	1600	122.35
800	61.6	831	65.1	1044.3	104.23	1186.7	112.74	1623	122.76
900	76.4	836	66.3	1044.8	104.32	<u>DATA SET 59*</u>			
933	80.7	885	74.0	1046.8	104.60	1640	123.17	1660	123.56
973	90.2	894	75.2	1047.9	104.78	<u>DATA SET 64</u>			
982	92.1	901	76.7	1049.1	104.97	4.2	2.58		
1015	98.7	918	79.8	1052.6	105.45	<u>DATA SET 60</u>			
1019	100.0*	951	85.9	1054.9	105.78	293	13.55	1700	125.80
1024	100.7*	975	90.9	1057.6	106.09	1720	126.13	1740	126.43
1033	102.2	979	91.5	1059.1	106.33	1760	126.71	<u>DATA SET 68*</u>	
1045	105.8*	982	91.8	1059.9	106.32	1780	127.07	93	2.635
1062	108.6	987	92.8	1062.2	106.69	1800	127.38	<u>DATA SET 69*</u>	
1083	110.3*	1000	94.6	1064.7	107.01	<u>DATA SET 65</u>			
1090	111.3*	1050	106.8	1065.9	107.09	1700	124.87	93	1.647
1124	112.9	1054	107.5	1068.4	107.39	1720	125.22	<u>DATA SET 70*</u>	
1124	113.8*	1061	108.1	<u>DATA SET 57*</u>					
1135	114.5*	1067	108.9	1151.9	113.48	1760	125.83	93	2.25
1161	115.1	1076	110.0	1156.2	113.71	1780	126.08	<u>DATA SET 71*</u>	
1163	115.8*	1085	110.6	1162.8	114.07	1800	126.30		
1174	117.0	1100	111.3	1165.8	114.23	<u>DATA SET 66</u>			
<u>DATA SET 55*</u>									
300	10.8	1150	115.0	1173.7	114.63	84	2.653	77	0.7786
311	11.5	1213	115.0	1175.5	114.70	98	3.091	<u>DATA SET 72*</u>	
347	13.4	1229	115.7	1177.5	114.79	123	3.988	<u>DATA SET 73*</u>	
358	14.1	1247	116.2	1180.8	114.13	148	4.962	<u>DATA SET 74*</u>	
373	15.1	1291	117.9	1182.9	113.52	173	5.929		
389	16.3	<u>DATA SET 56*</u>							
434	19.6	1017.6	97.78	1185.2	112.97	198	6.937		
451	20.8	1018.9	98.04	1187.7	112.77	223	8.147		
478	22.9	1019.9	98.24	1189.5	112.84	248	9.368		
503	25.2	1022.8	98.83	1190.6	112.90	273	10.681		
510	25.8	1023.4	99.04	1194.6	113.05	298	12.063		
523	27.2	1023.6	99.09	1197.3	113.16	323	13.504		
562	31.0	1025.5	99.56	<u>DATA SET 58*</u>					
569	31.7	1028.6	100.42	1151.0	113.73	348	15.022		
584	33.0	1029.5	100.56	1161.7	114.30	373	16.630		
628	38.3	1031.1	100.99	1167.9	114.62	398	18.235		
636	38.8	1034.0	101.76	1171.5	114.82	423	20.012		
644	40.0	1036.0	102.27	1174.1	114.92	448	21.504		
675	43.6	1038.7	103.13	1175.7	114.74	473	23.928		
687	45.2	1039.5	103.26	1177.8	114.42	498	26.000		
725	50.1	1041.0	103.58	1180.4	113.78	523	28.196		
752	54.0								

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (contInued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ			
DATA SET 74(cont.)*												
16.8	0.155	1160	111.69	87.1	0.88	76	0.8	22.5	0.1048			
20.0	0.155	1167.4	111.99	90.9	1.06	41.1	1.8	98.0	1.27			
27.5	0.164	1169.5	111.97	96.4	1.22	168	4.1	105.5	1.53			
36.3	0.188	1173.0	112.07	98.7	1.30	230	6.9	111.1	1.71			
40.2	0.228	1174.6	111.92	101.6	1.39	300	10.7	115.2	1.81			
54.3	0.327	1175.9	111.82	111.8	1.79	DATA SET 81*			124.1	2.15		
65.9	0.503	1178.4	111.34	115.0	1.89	77	0.57	134.3	2.59			
76.6	0.734	1180.6	111.43	118.5	1.90	273	9.61	178.4	4.47			
		1182.8	111.61	124.6	2.22	199.3	5.43	199.3	5.43			
		1184.8	111.70	126.9	2.31	226.8	6.44	226.8	6.44			
		1186.4	111.77	138.0	2.75	251.6	7.71	251.6	7.71			
		1187.9	111.86	143.9	3.02	DATA SET 82*			273	8.77		
1102	107.16	1189.7	111.83	155.1	3.52	4.2	0.24	295	9.92			
1123.3	109.41	1191.3	111.85	157.8	3.62	293	9.60	DATA SET 87*				
1142.1	110.62	DATA SET 77			162.9	3.85						
1150.2	111.24	4.00	0.25	165.2	3.93	DATA SET 83*			293	10.87		
1158.7	111.64	20.2	0.26	168.4	4.04	DATA SET 88*			1448	122.8		
1160.9	111.75	77.2	0.82	170.6	4.15	4.2	0.090	1677	128.8			
1162.5	111.75*	273	9.1*	172.6	4.25	293	9.90	1755	126.1			
1164.5	111.86*	473	21.4	174.7	4.33	DATA SET 84			1814	116.5		
1165.1	111.86*	673	41.0	177.2	4.45	4.2	0.084	1862	117.2			
1166.4	112.03*	873	68.2	179.8	4.55	293	10.92	2046	122.6			
1167.3	112.11*	1073	106.4	182.8	4.69	DATA SET 85*			2094	129.9*		
1169.5	112.22	DATA SET 78*			189.8	5.00						
1171.4	112.29*	76.7	0.63	194.9	5.25	DATA SET 86			293	12.5		
1173.4	112.38	85.5	0.86	199.5	5.42	4.2	0.057	DATA SET 90*		1808	140.00	
1175.3	112.52*	91.6	1.03	204.7	5.70	293	9.69	DATA SET 91				
1176.5	112.60	102.6	1.37	209.6	5.90	DATA SET 86			288	10.8*		
1178.4	112.64*	112.7	1.74	219.7	6.32	4.2	0.0953	373	15.5	344	14.4	
1179.4	112.74*	118.1	1.85	224.6	6.52	4.9	0.0956	411	17.7	405	16.1	
1180.7	112.83	127.3	2.21	229.2	6.83	5.5	0.0957	462	21.6	457	20.3	
1181.8	112.98*	139.0	2.68	234.2	7.08	6.3	0.0957	537	27.7	533	26.8	
1182.7	112.98	156.6	3.35	239.3	7.21	7.1	0.0959	589	32.8	598	33.6	
1183.4	112.73	172.2	4.00	249.4	8.01	8.1	0.0961	626	37.1	694	45.0	
1184.1	112.37*	180.9	4.38	254.6	8.36	8.8	0.0963	690	45.9*	769	56.7	
1185.1	112.14	191.0	4.80	259.4	8.59	9.6	0.0966	793	59.9*	813	62.1*	
1186.8	111.84*	DATA SET 79*			269.7	8.81	10.6	0.0969	828	65.4	DATA SET 95	
1189.1	111.89*	77	0.63	274.6	9.04	11.9	0.0930	845	68.0*	928	80.7	
1191.3	111.94	82.0	0.78	279.3	9.31	13.0	0.0979	889	75.0	933	81.6	
1192.8	112.02*	84.6	0.84	284.3	9.65	15.2	0.0990	917	79.4*	937	82.4	
1193.9	112.06*			288.5	9.77	17.8	0.1007					
1196.1	112.12			292.5	10.02							
				294.5	10.12							
				299.5	10.41							

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

DATA SET 105(cont.)			DATA SET 107(cont.)			DATA SET 108(cont.)			DATA SET 110(cont.)			DATA SET 113(cont.)		
T	ρ		T	ρ		T	ρ		T	ρ		T	ρ	
776.5	57.85		5.4	0.084	1044	91.6	873	79.55	477	20.6	1397	114.8*		
826.5	64.85		6.2	0.084	1053	97.5	923	89.70	499	22.7	1417	115.8		
827	64.80		6.7	0.086	1063	100.2	973	103.3	524	24.6	1445	116.4*		
867	70.05		7.3	0.086	1073	101.5	1023	113.2	545	26.8*	1472	117.2*		
903.5	76.40		8.1	0.089	1093	103.6	1073	118.6	574	29.0	1497	117.8		
910	77.40		9.1	0.086	1123	106.0	1123	120.4	597	31.2*	1524	118.3*		
931	80.85		10.1	0.086	DATA SET 109*			1173	121.2	623	33.9	1544	119.0	
970	87.20		11.2	0.086	273	8.53	1223	122.0	647	36.6*	1573	119.6*		
998.5	92.05		12.1	0.088	293	19.91	1273	122.4	671	38.9	1597	120.4*		
1017	95.65		13.1	0.090	DATA SET 111			721	44.2	41.9*	1618	121.1		
1037.5	99.85		14.1	0.087	323	11.64	750	47.4	721	44.2	1645	121.6*		
1038.5	100.00		15.1	0.093	373	15.53	273	9.35	775	50.9*	1656	121.7		
1043	100.90		16.2	0.091	423	20.21	293	10.95	802	53.8	1668	122.8*		
1054	103.40		17.1	0.094	473	25.00	323	11.92*	829	57.4*	1676	122.2		
1060	104.60		19.3	0.095	523	29.87	373	15.49	845	60.7	1686	123.5*		
1065.5	105.60		21.4	0.097	573	35.41	423	18.81	869	64.1*	1697	123.8		
1086.5	108.40		22.2	0.098	623	40.75	473	22.76*	897	68.0	1718	125.0		
1097.5	109.60		24.1	0.103	673	46.80	523	26.79	925	71.4*	DATA SET 114			
1117	111.40		25.2	0.098	723	53.50	573	31.93	953	75.2	1551	126		
1131	112.50		26.1	0.105	773	59.85	625	37.07	976	79.8*	1575	124		
1144	113.40		28.2	0.110	823	67.55	723	50.30	1016	89.0*	1601	126		
1164.5	114.50		30.2	0.114	873	76.35	773	56.70	1028	91.0	1617	127		
1189.5	116.20		33.9	0.125	923	85.20	823	63.95	1031	92.3*	1642	127		
1199.5	116.60		34.3	0.127	973	96.65	873	71.05	1031	93.2*	1685	127		
1219.5	117.20		40.4	0.157	1023	104.8	923	79.40	1031	94.4	1710	129		
1245	118.20		40.9	0.163	1073	107.8	973	92.9	1044	94.4*	1711	127		
1256	118.35		49.0	0.217	1123	109.9	1023	105.1	1046	95.8*	1729	129		
1262.5	118.55		100	1.28	1173	111.1	1073	111.0	1049	96.6	1748	129		
1273.0	118.80		150	3.14	1223	112.8	1123	114.5	1073	100.2	1775	130		
1287.5	119.10		200	5.38	1273	113.6	1173	117.8	1096	102.8*	1784	129		
1309	119.65		250	7.74	DATA SET 110			1124	105.1	105.1	1806	131		
1331.5	120.05		300	10.33	273	9.57	DATA SET 112			1145	106.1	1806	135	
1256	118.35		49.0	0.217	323	12.06	1923	138	1165	107.0*	1817	137		
1262.5	118.55		100	1.28	373	15.39*	DATA SET 113			1175	107.5*	1821	135	
1273.0	118.80		150	3.14	423	19.09	304	10.0	1182	108.3	1833	134		
1287.5	119.10		200	5.38	473	23.67	327	11.5	1198	108.0*	1852	135		
1309	119.65		250	7.74	523	28.72	355	13.1	1225	109.9*	1877	137		
1331.5	120.05		300	10.33	573	34.15	376	14.1	1253	110.7	1886	136		
1344.5	120.35				623	40.95	402	15.7	1274	111.1*	1925	136		
1368.5	120.60				673	47.80	425	17.2*	1296	112.1*	1929	139		
1368.5	120.85				723	54.85	451	19.0	1321	112.6	1976	139		
1415	121.00				773	62.35			1347	113.5*	1996	138		
1423	120.60				823	70.30			1373	114.3*	2018	140		
DATA SET 106			DATA SET 108			DATA SET 110			DATA SET 112			DATA SET 113		
4.2	0.23		78	2.07	273	9.57	273	9.57	1923	138	1923	138		
DATA SET 107			DATA SET 108			DATA SET 110			DATA SET 112			DATA SET 113		
4.2	0.080		178	5.81	373	15.39*	373	15.39*	304	10.0	304	10.0		
4.5	0.083		273	10.1*	423	19.09	423	19.09	327	11.5	327	11.5		
			370	15.5	473	23.67	473	23.67	355	13.1	355	13.1		
			471	22.4	523	28.72	523	28.72	376	14.1	376	14.1		
			570	30.6	573	34.15	573	34.15	402	15.7	402	15.7		
			681	41.6	623	40.95	623	40.95	425	17.2*	425	17.2*		
			783	53.9	673	47.80	673	47.80	451	19.0	451	19.0		
			864	64.7	723	54.85	723	54.85						
			982	82.7	773	62.35	773	62.35						
			1033	92.9	823	70.30	823	70.30						

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	$\rho - \rho_0$	T	ρ	T	ρ
<u>DATA SET 115</u>									
1773	129.9	1023.4	98.589*	1054.4	106.326*	2.438	0.00009942	77	0.700
1797	131.0	1024.7	98.981*	1055.4	106.433*	3.062	0.0001421	298	9.80
1822	138.1	1026.0	99.409*	1056.3	106.504*	4.159	0.0002509	<u>DATA SET 125*</u>	
1832	140.0	1027.1	100.015*	1057.4	106.611*	5.000	0.0003850	4.2	0.236
1842	138.4	1027.2	99.694*	1059.0	106.825*	7.674	0.0007771	<u>DATA SET 126*</u>	
1847	139.2	1028.5	100.122*	1059.6	106.896*	10.00	0.001326	4.2	0.353
1852	141.3	1029.6	100.443*	1060.5	106.896	11.32	0.001720	<u>DATA SET 127*</u>	
1862	139.8	1030.4	100.764*	1061.2	106.932*	15.92	0.004041	4.2	0.553
1872	140.8	1031.0	100.978*	1062.1	106.968*	18.84	0.006088	<u>DATA SET 128*</u>	
1882	140.2	1032.1	100.263*	1063.3	107.003*	29.31	0.02007	4.2	0.726
1892	142.1	1032.6	100.512*	1064.0	107.039*	38.90	0.04770	<u>DATA SET 129*</u>	
1898	141.1	1033.4	101.833*	1065.2	107.075*	49.77	0.1184	4.2	0.367
		1034.0	102.012*	1066.0	107.146*	62.66	0.2776	<u>DATA SET 130*</u>	
		1034.4	102.226*	1067.0	107.253*	<u>DATA SET 120*</u>		4.2	0.641
		1035.0	102.368*	1068.0	107.324*	<u>DATA SET 121*</u>		<u>DATA SET 131*</u>	
		1035.7	102.618*	1069.1	107.396*	T	ρ	4.2	0.924
993.8	91.101	1035.7	102.618*	1069.9	107.431*	2.366	0.0001022	4.2	0.924
993.4	91.422*	1036.0	102.760*	1069.9	107.431*	2.951	0.0001554	<u>DATA SET 132*</u>	
996.4	91.636*	1036.7	102.974*	1070.3	107.503*	4.130	0.0002732	4.2	1.086
996.7	91.707*	1037.4	103.295*	1071.4	107.574*	4.989	0.0004004	<u>DATA SET 133*</u>	
997.7	91.883*	1038.2	103.519*	1072.3	107.610*	7.379	0.0008138	4.2	1.298
998.2	91.921*	1038.8	103.687*	1073.0	107.645*	11.02	0.001700	<u>DATA SET 134 (cont.)</u>	
998.8	92.171*	1039.2	103.830*	1073.6	107.681*	19.01	0.005909	25.4	0.3583
1001.4	92.812	1039.7	103.973*	1074.7	107.752*	30.90	0.01993	26.2	0.3549*
1003.4	93.347*	1040.0	104.187	1076.0	107.824*	39.45	0.04041	26.5	0.3553*
1003.5	93.419*	1040.3	104.294*	1077.0	107.895*	47.64	0.07789	30.9	0.37608
1004.4	93.561*	1040.8	104.508*	1077.9	107.930*	77.62	0.5514	33.6	0.3802
1005.5	92.527*	1041.3	104.650*	1078.7	108.037*	<u>DATA SET 122*</u>		37.2	0.4146
1005.7	93.704	1041.9	114.828*	1079.7	108.144*	77.62	0.5514	40.7	0.4484
1006.2	93.918*	1042.4	104.864*	1080.4	108.251	<u>DATA SET 123*</u>		46.8	0.4977
1007.1	94.239*	1042.8	105.007*	1081.4	108.323*	77	0.580	51.6	0.5521
1008.2	94.417*	1043.2	105.042*	1082.4	108.394*	298	9.80	59.0	0.6562
1008.8	94.666*	1043.7	105.114*	1083.2	108.430*	<u>DATA SET 134</u>		64.6	0.7590
1010.4	94.916*	1044.3	105.185*	1084.3	108.501*	4.2	0.3300	70.6	0.8995
1011.4	95.130*	1045.2	105.399*	1085.5	108.572*	4.2	0.3300	71.9	0.9374*
1012.2	95.415*	1045.7	105.542*	1086.5	108.608*	<u>DATA SET 135</u>		75.2	1.037*
1013.4	95.665*	1046.0	105.613*	1088.1	108.751	373	16.1	79.4	1.120
1014.0	95.914*	1046.7	105.684*	<u>DATA SET 117</u>		473	22.8	86.3	1.218
1015.7	96.271*	1047.3	105.791*	1873	110	82.2	1.218	90.6	1.359*
1016.5	96.521*	1048.2	105.862*	<u>DATA SET 118</u>		95.1	1.690	102.1	1.920*
1017.1	96.628*	1049.2	105.969*	1809	134	108.9	2.173*	117.5	2.461
1017.8	96.877*	1049.8	106.005*	<u>DATA SET 119</u>		126.8	2.915*	126.8	2.915*
1019.3	97.412*	1050.4	106.041	<u>DATA SET 120</u>		137.4	3.34	137.4	3.34
1020.5	97.697	1051.0	106.112*	<u>DATA SET 121</u>		150.0	3.85	150.0	3.85
1021.1	97.911*	1052.1	106.148*	<u>DATA SET 122</u>		166.3	4.66*	166.3	4.66*
1021.9	98.268*	1052.9	106.183*	<u>DATA SET 123</u>		175.4	5.067	175.4	5.067
1022.4	98.339*	1053.7	106.290*	<u>DATA SET 124</u>		181.6	5.548*	181.6	5.548*
				<u>DATA SET 125</u>		195.9	6.321	195.9	6.321
				<u>DATA SET 126</u>		218.3	7.336	218.3	7.336
				<u>DATA SET 127</u>		239.3	8.581	239.3	8.581
				<u>DATA SET 128</u>		252.3	9.253	252.3	9.253
				<u>DATA SET 129</u>		259.4	9.482	259.4	9.482
				<u>DATA SET 130</u>		276.1	10.136	276.1	10.136
				<u>DATA SET 131</u>		300.0	10.33	300.0	10.33
				<u>DATA SET 132</u>					
				<u>DATA SET 133</u>					
				<u>DATA SET 134</u>					
				<u>DATA SET 135</u>					

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 136</u>											
4.2	0.04	1511	120.5	273.0	8.57	1.98	0.0556	1.98	0.0556	77.79	1.288
100	1.3	1532	121.1	293.0	9.6100	4.21	0.0565	4.21	0.0565	85.19	1.520
200	5.4	1546	121.8	373.0	14.296	20.40	0.0692	20.40	0.0692	273.16	11.04
300	9.7	1572	122.1	373.0	14.296	78.20	0.6751	78.20	0.6751	<u>DATA SET 154*</u>	
400	15.6	1595	122.5	373.0	14.296	273.16	9.11	273.16	9.11		
500	23.1	1623	124.0	472.6	21.432	<u>DATA SET 149</u>		1.38	0.6696	1012.2	91.467
600	32.4	1673	125.0	580.8	31.015	1.98	0.0723	3.71	0.6696	1012.25	92.762
700	43.7	1723	126.1	671.4	36.259	4.21	0.0723	4.21	0.6712	1023.8	94.296
800	56.6	1773	127.1	760.7	51.759	20.40	0.0795	20.40	0.6967	1026.1	94.828
900	71.6	1793	127.9	<u>DATA SET 144*</u>		81.73	1.982	81.73	1.982	1028.9	95.539
1000	90.3	<u>DATA SET 140*</u>		80.3	0.962	90.46	2.391	90.46	2.391	1031.0	96.233
1043	99.7	20.4	-0.64286	194.9	4.9928	273.16	8.93	273.16	15.98	1034.3	96.887
1100	106.5	80.6	0.97180	273.0	8.57	<u>DATA SET 150*</u>		<u>DATA SET 155*</u>		1035.3	97.086
1185	111.6	198.3	5.1537	373	14.296	1.38	0.1604	1.38	0.8806	1036.6	97.345
1200	110.3	273.1	8.57	373	14.296	4.21	0.1614	3.71	0.8806	1037.1	97.444
<u>DATA SET 137*</u>											
1811	142.7	<u>DATA SET 141*</u>		572.4	30.215	20.40	0.1781	4.21	0.8806	1038.3	97.684
1900	152.3	20.4	-0.64286	<u>DATA SET 145*</u>		77.74	0.8610	20.40	0.9348	1039.9	98.003
2000	164.9	80.6	0.97180	1.38	0.2020	273.16	9.84	20.40	0.9348	1040.6	98.143
<u>DATA SET 138</u>											
279	9.7	198.3	5.1337	4.21	0.2022	81.73	2.755	81.73	2.755	1041.6	98.344
432	19.4	273.1	8.57	20.40	0.2136	273.16	22.58	273.16	22.58	1043.0	98.625
565	31.3	<u>DATA SET 142*</u>		78.85	0.8602	<u>DATA SET 151*</u>		<u>DATA SET 156*</u>		1044.2	98.866
668	47.3	80.3	0.963	87.42	1.094	1.38	0.1631	0.302	0.006210	1045.9	99.208
766	54.5	80.5	0.969	273.16	9.36	3.71	0.1641	0.400	0.006211	1047.5	99.531
983	89.7	194.6	4.980	273.16	9.36	4.21	0.1674	0.500	0.006212	1049.2	99.875
1050	101.5	273	8.57	273.16	9.36	20.40	0.1772	0.600	0.006215	1051.4	100.32
1075	104.3	293	9.6100	<u>DATA SET 146*</u>		77.74	0.8935	0.700	0.006217	1053.6	100.77
1107	106.8	373	14.296	1.38	0.1252	273.16	10.87	0.805	0.006221	1057.5	101.56
1123	108.0	373	14.296	3.71	0.1252	<u>DATA SET 152*</u>		0.908	0.006225	1061.6	102.40
1131	109.1	373	14.296	4.21	0.1341	1.38	0.6166	0.700	0.006221	1078.4	105.90
1148	109.4	373	14.296	83.90	0.8898	3.71	0.6166	0.805	0.006225	1080.4	106.32
1170	111.3	472.1	21.393	273.16	8.88	4.21	0.6176	1.004	0.006229	1093.9	107.28
1200	111.6	581.6	31.094	<u>DATA SET 147</u>		20.40	0.6425	<u>DATA SET 137*</u>		1095.2	109.47
1213	112.1	682.5	41.954	1.38	0.0896	81.73	1.421	273	8.73	1129.6	117.03
1260	112.9	784.2	54.956	4.21	0.0887	273.16	9.59	<u>DATA SET 159*</u>		273	8.57
1352	115.4	<u>DATA SET 143*</u>		20.40	0.0974	<u>DATA SET 158*</u>		273	8.73	293	9.61
1410	117.2	81.0	0.9836	83.90	0.828	20	0.180	90	1.046	373	14.296
1458	119.0	194.9	4.9928	273.16	8.70	273	8.572	273	8.572	473	21.5
1479	119.8	195.0	4.9969	<u>DATA SET 153*</u>		1.29	0.5123	566	29.179	573	30.3
				20.40	0.5235	4.21	0.5167	578.3	30.265	773	53.4
										873	66.1

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 159(cont.)*</u>											
973	85.2	193.8	5.06	4.17	0.03370	4.2	0.0599	879	70.79	0.310	0.005887
1073	105	292.8	9.6	4.19	0.03370	<u>DATA SET 166</u>		901	73.82	0.338	0.005890
1173	127	398.0	15.9	4.19	0.03371	<u>DATA SET 167</u>		903	74.84	0.423	0.005894
1273	152	491.8	23.0	<u>DATA SET 163</u>		4.2	0.0381	927	78.98	0.504	0.005896
		586.6	31.6	1.26	0.03399*	<u>DATA SET 168*</u>		928	77.97	0.601	0.005900
		684.6	42.2	1.46	0.03400	<u>DATA SET 169*</u>		951	81.51	0.699	0.005905
		784.6	55.0	1.64	0.03401	4.2	0.0368	951	83.13	0.807	0.005911
		887.8	70.5	1.84	0.03403	<u>DATA SET 170</u>		976	87.58	0.909	0.005915
		988.9	88.1	2.01	0.03404	<u>DATA SET 171*</u>		979	86.57	1.01	0.005922
		1085.8	107.5	2.21	0.03405	4.2	0.0390	1000	91.02	1.10	0.005928
		1207.7	135.4	2.44	0.03406	<u>DATA SET 172*</u>		1022	95.67	1.20	0.005938
				2.85	0.03408	<u>DATA SET 173*</u>		1045	97.39	<u>DATA SET 182*</u>	
				3.05	0.03411	4.2	0.0261	1045	101.13	1.02	0.005216
				3.25	0.03412	<u>DATA SET 174*</u>		1050	102.65	1.17	0.005230
				3.47	0.03414	4.2	0.0317	1072	103.25	1.17	0.005230
				3.67	0.03417	<u>DATA SET 175*</u>		1071	104.87	1.29	0.005239
				3.84	0.03417	4.2	0.0216	1071	105.68	1.38	0.005250
				4.07	0.03421	<u>DATA SET 176*</u>		1100	106.19	1.49	0.005259
				4.26	0.03423	4.2	0.0371	1126	107.30	1.59	0.005272
				14.01	0.03702	<u>DATA SET 177</u>				1.68	0.005285
				14.30	0.03719	<u>DATA SET 178*</u>				1.78	0.005297
				14.88	0.03737	4.2	0.02637			1.89	0.005310
				15.94	0.03805	<u>DATA SET 179*</u>				2.00	0.005329
				16.40	0.03857	4.2	0.0371			2.08	0.005342
				16.87	0.03931	<u>DATA SET 180*</u>				2.20	0.005363
				17.31	0.03882	4.2	0.0261			2.29	0.005376
				17.89	0.03925	<u>DATA SET 181*</u>				2.40	0.005396
				18.39	0.03959	4.2	0.0216			2.50	0.005411
				18.95	0.03994	<u>DATA SET 182*</u>				2.60	0.005430
				19.40	0.04028	4.2	0.0371			2.70	0.005450
				20.32	0.04096	<u>DATA SET 183*</u>				2.79	0.005465
						4.2	0.0354			2.90	0.005485
						<u>DATA SET 184*</u>				3.00	0.005503
						4.2	0.0371			3.10	0.005523
						<u>DATA SET 185*</u>				3.20	0.005542
						4.2	0.0395			3.20	0.005564
						<u>DATA SET 186*</u>				3.40	0.005585
						4.2	0.0396			3.51	0.005606
						<u>DATA SET 187*</u>				3.60	0.005629
						4.2	0.0176			3.70	0.005650
						<u>DATA SET 188*</u>				3.79	0.005673
						4.2	0.0354			3.89	0.005699
						<u>DATA SET 189*</u>				4.11	0.005745
						4.2	0.0354				

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 183*</u>											
1.41	0.012080	17.6	0.02097	1677	123.5*	360	13.446	40.1	0.0941	<u>DATA SET 209(cont.)</u>	
1.51	0.012093	19.05	0.02248	1736	124.3*	380	14.699	41.3	0.0996		
1.61	0.012093	20.7	0.02454	1816	125.3	400	16.009	42.3	0.1047		
1.72	0.012099	<u>DATA SET 197</u>		1816	131.3	<u>DATA SET 208</u>		43.0	0.1083		
1.82	0.012098	4.2	0.0516	2001	133.2	1828	137.5	43.7	0.1134		
1.89	0.012119	<u>DATA SET 198*</u>		2198	134.9	1854	138.4	44.9	0.1216		
2.05	0.012124	4.2	0.0566	2375	136.6	1877	138.4	45.7	0.1253		
2.22	0.012148	<u>DATA SET 199*</u>		2505	137.8	<u>DATA SET 210*</u>					
2.32	0.012165	4.2	0.0429	2699	139.4	1980	142.6				
2.39	0.012166	<u>DATA SET 200</u>		2825	140.3	2010	143.2*				
2.50	0.012183	4.2	0.0246	2964	141.5*	2033	143.9*				
2.52	0.012193	<u>DATA SET 201*</u>		2997	142.0	2065	144.5*				
2.63	0.012215	4.2	0.0384	<u>DATA SET 206</u>		<u>DATA SET 209</u>					
2.78	0.012238	<u>DATA SET 202*</u>		1726	126.7	5.0	0.0416				
2.99	0.012266	4.2	0.0402	1748	126.7	6.2	0.0419				
3.09	0.012296	<u>DATA SET 203</u>		1772	127.6	7.2	0.0420				
3.26	0.012329	4.2	0.0358	1775	127.6	8.2	0.0423				
3.41	0.012364	<u>DATA SET 204</u>		1797	128.5	9.1	0.0429				
3.62	0.012405	4.2	0.0491	1813	135.1	10.8	0.0425				
3.86	0.012458	<u>DATA SET 205</u>		1822	135.2*	11.6	0.0429				
4.08	0.012516	4.2	0.0336	1832	135.1*	12.6	0.0433				
4.31	0.012571	<u>DATA SET 206</u>		1838	135.5	15.0	0.0449				
<u>DATA SET 184*</u>											
4.7	0.01271	4.2	0.283	1852	135.9	17.1	0.0458				
5.0	0.01283	<u>DATA SET 207</u>		1884	136.9	20.7	0.0482				
5.3	0.01288	4.2	0.134	1873	136.1	21.8	0.0499				
5.8	0.01304	<u>DATA SET 208</u>		1901	136.3*	23.1	0.0507				
6.1	0.01315	4.2	0.109	1915	137.0	24.6	0.0520				
6.2	0.01329	<u>DATA SET 209</u>		90	0.907	26.2	0.0544				
6.6	0.01329	4.2	0.109	100	1.218	27.5	0.0561				
7.0	0.01346	<u>DATA SET 210</u>		120	1.907	28.6	0.0579				
7.5	0.01373	4.2	0.0926	140	2.654	30.0	0.0603				
8.1	0.01395	<u>DATA SET 211</u>		160	3.452	31.7	0.0650				
8.6	0.01421	4.2	0.0926	180	4.299	33.0	0.0681				
9.2	0.01448	<u>DATA SET 212</u>		200	5.172	34.2	0.0696				
9.6	0.01479	1007	94.3*	220	6.070	35.3	0.073				
10.3	0.01514	1044	102.8*	240	7.000	36.7	0.079				
11.1	0.01565	1098	107.6*	260	7.964	37.8	0.085				
11.95	0.01622	1182	112.6*	280	8.966	39.0	0.096				
12.81	0.01678	1182	113.6*	300	10.002	41.6	0.099				
13.96	0.01765	1295	116.4*	320	11.102	42.6	0.105				
15.0	0.01850	1421	119.1*	340	12.255	44.0	0.114				
15.6	0.01892	1501	120.4*	360	13.446	44.9	0.114				
16.4	0.01973	1627	122.5*	380	14.699	47.8	0.135				
		1677	123.0*	400	16.009	49.4	0.146				

* Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 211*													
4.4	0.037	1803	135.9	1915	137.6	453.2	20.18*	923.2	77.04*	1046.95	102.33*		
6.6	0.057	1817	136.1	1915	137.7	473.2	21.82*	933.2	78.77*	1048.2	102.58*		
8.2	0.057	1823	136.2	1937	138.05	483.2	22.69*	943.2	80.54*	1049.45	102.79*		
10.0	0.058	1836	136.4			493.2	23.58*	948.2	81.44*	1050.7	103.00		
11.1	0.059	1842	136.45	DATA SET 215*				503.2	24.51*	953.2	82.40	1053.2	103.40*
12.4	0.058	1857	136.65	1693	125.8	513.2	25.46*	958.2	83.30*	1055.7	103.77*		
14.4	0.059	1858	136.75	1773	127.1	523.2	26.41*	963.2	84.21*	1058.2	104.13*		
16.3	0.059	1874	136.9	1813	132.1	533.2	27.38*	968.2	85.12*	1063.2	104.98		
18.7	0.060	1875	137.0	1832	133.7	543.2	28.33*	973.2	85.92*	1068.2	105.63*		
23.2	0.064	1893	137.2	1874	137.6	553.2	29.29	978.2	86.87*	1073.2	106.37*		
26.6	0.067	1900	137.35	DATA SET 216*				563.2	30.22*	983.2	87.85*	1078.2	106.89
29.9	0.071	1905	137.4	1683	121.167	573.2	31.18*	988.2	88.84	1083.2	107.38*		
33.0	0.076	1919	137.6	1692	122.332	583.2	32.16*	993.2	89.56*	1088.2	107.85		
35.7	0.084			1703	121.577	593.2	33.16*	995.2	90.07*	1093.2	108.31*		
38.2	0.090			1713	122.358	603.2	34.18*	998.2	90.60*	1098.2	108.73*		
40.5	0.100			1722	121.998	613.2	35.22*	1000.2	91.12*	1103.2	109.14		
42.8	0.112			1732	121.986	623.2	36.28*	1003.2	91.64*	1108.2	109.53*		
46.1	0.130			1743	123.935	633.2	37.37*	1005.7	92.17*	1113.2	109.93*		
DATA SET 212													
1695	126.4	1764	128.2	1732	121.998	653.2	39.56	1010.7	93.27*	1123.2	110.63*		
1713	126.65	1777	133.45	1743	123.935	663.2	40.67*	1013.2	93.83*	1128.2	110.96		
1735	127.0	1802	136.2	1752	124.332	673.2	41.80*	1015.7	94.38*				
1759	127.35	1814	136.2	1762	123.960	683.2	42.96*	1018.2	94.95*				
1786	129.9	1817	136.4	1773	126.678	693.2	44.14*	1020.7	95.52*				
1799	134.1	1829	136.4	1783	126.306	703.2	45.35*	1023.2	96.11*				
1816	135.65	1833	136.45	1808	126.709	713.2	46.62*	1025.7	96.70*				
1833	135.85	1841	136.75	1808	134.402	723.2	47.86*	1028.2	97.31*				
1835	135.9*	1843	136.45	1812	134.792	733.2	49.11*	1030.7	97.93				
1855	136.2*	1845	136.65	1823	135.960	743.2	50.38*	1033.2	98.57*				
1864	136.35	1850	136.7	1833	134.819	753.2	51.63	1034.45	98.90*				
1872	136.5*	1853	136.95	1843	131.369	763.2	52.93*	1035.7	99.24				
1880	136.55	1864	137.1	1849	138.685	773.2	54.26*	1036.95	99.58*				
1895	136.8	1866	136.8	1853	131.382	783.2	55.61*	1038.2	99.92*				
DATA SET 213*													
1676	126.9	1876	137.0	1853	139.075	793.2	57.00*	1039.45	100.28				
1699	127.25	1876	137.2	1858	138.697	803.2	58.39*	1039.7	100.33*				
1720	127.55	1879	137.35	1864	133.704	813.2	59.83*	1040.2	100.47*				
1741	127.9	1889	137.45	1864	139.088	823.2	61.27*	1040.7	100.62*				
1760	128.15	1894	137.55	1869	138.326	833.2	62.70*	1041.2	100.79*				
1776	129.4	1896	137.45	1874	139.486	843.2	64.17*	1041.7	100.95*				
1798	135.5	1910	137.75			853.2	65.67	1043.2	101.39*				
DATA SET 217													
1676	126.9	1879	137.35	1693	125.8	863.2	67.20*	1043.7	101.56				
1699	127.25	1889	137.2	1773	127.1	873.2	68.76*	1044.2	101.70*				
1720	127.55	1893	137.45	1813	132.1	883.2	70.35*	1044.7	101.83*				
1741	127.9	1894	137.55	1832	133.7	893.2	71.99*	1045.2	101.95*				
1760	128.15	1896	137.35	1874	137.6	903.2	73.64	1045.7	102.09*				
1776	129.4	1905	137.45			913.2	75.33*	1046.2	102.21*				
1798	135.5	1910	137.75										

Not shown on either figure.

TABLE 9. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF IRON Fe (continued)

T	ρ	T	ρ	T	ρ	T	ρ		
DATA SET 218(cont.)*									
21.78	6.64	48.7	15.0×10^{-2}	1.72	0.233×10^{-2}	2.20	0.158×10^{-2}		
22.46	7.12	52.8	19.2	1.85	0.235	2.40	0.159		
22.91	7.44	58.8	26.6	2.12	0.237	2.59	0.162		
23.23	7.80	68.5	37.7	2.24	0.239	2.88	0.168		
23.45	8.07	73.7	51.5	2.41	0.241	3.13	0.171		
23.95	8.74	89.8	90.0	2.49	0.241	3.36	0.176		
24.40	9.06	137	249	2.70	0.243	3.68	0.177		
24.90	9.64	172	393	2.86	0.248	4.06	0.184		
25.26	9.87	213	567	3.07	0.248	4.45	0.189		
25.81	10.32	271	840×10^{-2}	3.30	0.252	5.68	0.223		
26.26	10.90	DATA SET 220*						39.5	6.15
26.76	11.58	1.70	0.645×10^{-2}	3.68	0.259	6.93	0.265		
27.35	12.30	1.88	0.645	3.92	0.264	8.23	0.309		
27.89	13.07	2.02	0.651	4.32	0.271	9.69	0.375		
DATA SET 219*									
1.73	2.68×10^{-2}	2.21	0.657	4.73	0.276	12.6	0.512		
1.88	2.70	2.48	0.656	5.77	0.300	14.6	0.632		
2.02	2.70	2.77	0.662	7.17	0.337	18.6	0.928		
2.19	2.70	2.97	0.662	8.28	0.370	24.5	1.45		
2.35	2.73	3.22	0.668	9.23	0.401	32.5	2.84		
2.64	2.73	3.40	0.674	10.8	0.469	39.7	5.30		
3.00	2.71	3.66	0.674	13.1	0.590	47.5	10.9		
3.40	2.73	3.83	0.673	14.9	0.696	60.0	21.2		
3.89	2.70	4.23	0.686	16.0	0.770	70.0	33.8		
4.18	2.73	4.79	0.685	21.3	1.20	82.6	60.8		
5.52	2.75	5.54	0.710	24.6	1.55	88.9	77.9		
6.38	2.78	7.13	0.757	27.2	2.00	114	151		
7.43	2.78	8.85	0.829	33.5	3.23	134	214		
8.35	2.83	10.5	0.867	37.4	4.80	164	335×10^{-2}		
10.0	2.88	14.2	1.06	47.4	9.92	DATA SET 223*			
11.0	2.93	18.9	1.36	52.4	13.6	1.59	0.0631×10^{-2}		
11.9	2.99	20.9	1.54	60.1	22.7	1.70	0.0631		
14.2	3.10	33.5	3.79	73.5	42.0	1.82	0.0643		
16.3	3.24	49.2	11.8	106	123	1.92	0.0643		
17.5	3.33	67.1	17.3	119	180	2.11	0.0661		
21.1	3.69	79.1	56.5	127	192	2.24	0.0673		
27.4	4.85	99.4	116	172	358	2.45	0.0692		
37.5	7.60	107	133	200	517	2.64	0.0718		
40.3	8.96	138	237	240	645	2.76	0.0738		
43.3	11.0×10^{-2}	165	345	268	782×10^{-2}	2.91	0.0765		
DATA SET 222*									
1.62 0.154×10^{-2}									
1.75 0.155									
1.88 0.158 10^{-2}									
2.06 0.158×10^{-2}									

* Not shown on either figure.

3.4. Nickel

There are more than 100 data sets available for the electrical resistivity of nickel. The information on specimen characterization and measurement condition for each of the data sets is given in table 11. The data are tabulated in table 12 and shown partially in figures 7 and 8.

Since nickel belongs to the same group in the periodic table as iron and is also ferromagnetic, the electrical resistivity of nickel is expected to resemble that of iron. For example, the solute resistivities of dilute nickel alloys are similar to those of dilute iron alloys in magnitude and in temperature dependence (Schwerer and Cuddy [148]). However, since nickel is not as strongly magnetic as iron (with a spontaneous magnetization of 6.4 kG as compared to 21.8 kG for iron), the magnetic effect on the electrical resistivity is not as strong in nickel as it is in iron. While the minimum in the longitudinal magnetoresistance at 4 K of a pure iron specimen occurs at ≥ 750 Oe ($\sim 60 \times 10^3 \text{ A m}^{-1}$) (for example, see Fujii and Morimoto [112]), it occurs at ~ 250 Oe for pure nickel (Wycisk and Feller-Kniepmeier [209] and Fujii [210]). Furthermore, for iron the resistivity at the minimum can be as low as one third of the value at zero applied magnetic field. For nickel, it was only about 18% lower [209,210].

The electrical resistivity of nickel has not been investigated as widely as that of iron, and there has been apparently lesser effort spent in its purification. In fact, among the data sources reporting the electrical resistivity of nickel, less effort was made to analyze the impurity content of the specimen than those reporting the electrical resistivity of iron. Nonetheless, there are a few data sets which show very good agreement on the residual resistivity of pure nickel: $\sim 0.0033 \times 10^{-8} \Omega \text{ m}$ at 2.32 K from White and Tainsh [210] (data set 31), $\sim 0.0031 \times 10^{-8} \Omega \text{ m}$ at 1.85 K from Ehrlich et al. [212] (data set 73), and $0.0033 \times 10^{-8} \Omega \text{ m}$ from Wycisk and Feller-Kniepmeier [209] (data set 96). The recommended values for the residual resistivity (at 1 K) is based on these data sets. The specimens of the first two sources were described as "high purity" and "pure", respectively. The specimen of the last source was 99.999% pure and was five-time electron beam zone-refined.

The temperature-dependent part of the electrical resistivity has been reported to contain mostly of a T^2 component at low temperatures (≤ 10 K): see,

for example, White and Woods [21] (data sets 33-34), Ehrlich and Rivier [213] (data set 10), Greig and Harrison [214] (data set 4), Fert and Campbell [215] (data set 48), Price and Williams [80] (data set 55), and Sudovtsov and Semenenko [203] (data sets 77,78). An analysis similar to that applied in treating the low-temperature data on the electrical resistivity of iron, i.e., plotting the quantity

$$\rho - \rho_0 = A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^2} dx$$

with values of A and θ_R equal to $39.1 \times 10^{-8} \Omega m$ and 456 K, respectively, gives very similar results. It increases as T^2 at temperatures below ~ 50 K. It then varies much less rapidly with temperature: with some data sets (e.g., data sets 55,57) there is a plateau at ~ 110 K, and with some (e.g., data sets 3,43) there is a minimum at around the same temperature. At temperatures above 150 K, it increases more rapidly and approaches a T^3 dependence. The coefficient of the T^2 term (for temperatures below about 50 K) varies between $\sim 1.5 \times 10^{-5} \Omega m K^{-2}$ (data set 48) and $\sim 3.5 \times 10^{-5} \Omega m K^{-2}$ (data set 57), and furthermore there is no discernible correlation between these coefficients and the residual resistivities of the specimens. However, the agreement between data sets with the lowest reported residual resistivities are good. The data set of White and Tainsh [211] (data set 31) yields a coefficient of $2.7 \times 10^{-5} \Omega m K^{-2}$ and a residual resistivity of $\sim 0.0033 \times 10^{-8} \Omega m$, that of Ehrlich and Rivier [213] (data set 10) yields $2.4 \times 10^{-5} \Omega m K^{-2}$ and $0.0031 \times 10^{-8} \Omega m$, respectively. The data set of Farrell and Greig [216] (data set 11) yields $2.6 \times 10^{-5} \Omega m K^{-2}$ and $0.0095 \times 10^{-8} \Omega m$, and that of Ehrlich et al. [212] (data set 73) yields $2.3 \times 10^{-5} \Omega m K^{-2}$ and $0.0031 \times 10^{-8} \Omega m$. The recommended values below 60 K are based on the above four data sets, with the values of the coefficient, $2.6 \times 10^{-5} \Omega m K^{-2}$ is also the mean of the above four values. It should be mentioned that the plateau or the minimum region in the quantity

$$\rho - \rho_0 = A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x}{(e^x - 1)^2} dx$$

at around 110 K could not be eliminated by an effort in adjusting the values of A and θ_R . As a consequence, the value of θ_R was taken to be 456 K and the value of A was chosen so that the range of applicability of the T^2 term could be extended to as high a temperature as practicable. As an illustration, the

data of Farrell and Greig [216] (data set 11) deviate from the T^2 line by $+0.005 \times 10^{-8} \Omega m$ at 50 K; this deviation increases to $-0.02 \times 10^{-8} \Omega m$ at 60 K.

In the temperature range from about 60 K to room temperature, a number of authors reported data sets which agree well with each other: White and Woods [21] (data sets 33,34), Farrell and Greig [216] (data set 11), Laubitz et al. [217] (data set 52). In particular, the last two data sets merge very well at 90 K. The recommended values in this temperature range is based on the above four data sets, with more weight given to the last two.

For temperatures from the ice point up to the Curie point, a number of data sets agree to within $\sim \pm 4\%$: Pallister [218] (data set 14), Powell et al. [219] (data sets 17,20), Kierspe et al. [78] (data set 37), Schroeder and Giannuzzi [98] (data set 51), Laubitz et al. [217] (data set 52), Potter [199] (data set 53), Schwerer and Cuddy [148] (data set 65), and Ahmad and Greig [220] (data sets 89,90). Of these, the data of Laubitz et al. and of Potter show particularly good agreement ($\pm 2\%$). The recommended values in this temperature range are based on these results, with more weight given to those of Laubitz et al. [13] (data set 52) and of Potter [199] (data set 53).

The Curie temperature of nickel has been reported to be 631 K by Zumsteg and Parks [221] (data set 91), 631 K by Standley and Reich [222] (data set 2), 630 K by Dutta-Roy and Subrahmanyam [223] (data set 3), ~ 630 K by Laubitz et al. [13] (data set 52), 632.7 K by Jackson and Saunders [224] (data set 103), and from 629.3 to 629.8 K, depending on specimen, by Potter [199] (data set 53). Judging from the resistivity data of Pallister [218] (data set 14), the Curie temperature is ~ 627 K, of Kirichenko and Mikryukov [225] (data set 27), ~ 631 K, of Schwerer and Cuddy [148] (data set 65), ~ 628 K, of Kaul [226] (data set 67), 620-640 K, of Shirakawa [141] (data set 76), ~ 633 K, and of Schroeder and Giannuzzi [98] (data set 51), ~ 638 K. Among these sources, only two, Potter [199] (data set 53) and Zumsteg and Parks [221] (data set 91), give in detail the change of electrical resistivity at around the Curie temperature. The agreement between these are very good: $\pm 0.5\%$ below and $\pm 1\%$ above the Curie temperature. The recommended values in the vicinity (± 25 K) of the Curie temperature are based on this reference, with more weight given to the result of Zumsteg and Parks at temperatures above the transition. The resultant values are within 0.3% of those calculated on the basis of the $d\rho/dT$ values reported by Jackson and Saunders [224].

At temperatures from the Curie point to about 1300 K, the following data sets fall into a band of width $\sim 2 \times 10^{-8} \Omega\text{m}$: Pallister [218] (data set 14), Bode [227] (data set 16), Powell et al. [219] (data set 20), Davis et al. [228] (data set 32), Laubitz et al. [217] (data set 52), and Potter [199] (data set 53). The recommended values in this temperature range are based on these data sets, with more weight given to the data of Laubitz et al. [217] (data set 52). Data set 52 is also used as basis for recommendation for lower temperatures.

Unfortunately, most of the data sets mentioned in the previous paragraph are for temperatures below 1300 K. For higher temperatures, the available data sets show large discrepancies. In addition, the resistivity values for lower temperature given in these sets are quite different from the recommended values (for example, data sets 37,72). However, for temperatures slightly below the melting point, the data of Güntherodt et al. [92] (data set 93) and of Kita et al. [93] (data sets 100-101) are within $\sim 0.5 \times 10^{-8} \Omega\text{m}$ of each other. Extrapolations, either graphically or numerically using a cubic expression, from recommended values for lower temperatures give values that are also within $0.5 \times 10^{-8} \Omega\text{m}$ of the values reported by these authors. The recommended values are, therefore, obtained from the numerical extrapolation.

At temperatures immediately above the melting point, the available data sets show a spread of about $6 \times 10^{-8} \Omega\text{m}$. Between the data of Güntherodt et al. [92] (data set 93) and of Kita et al. [93] (data sets 100-102), which well agree below the melting point, the difference is about $4 \times 10^{-8} \Omega\text{m}$. The recommended value for the liquid phase at the melting point is based on the results of Güntherodt et al. [92] (data set 93), Seydel and Fucke [87] (data set 92), and Mokrovskii and Regel [158] (data set 56), which agree to within $0.2 \times 10^{-8} \Omega\text{m}$. The temperature dependence of the electrical resistivity in the molten state has been generally reported to be linear, e.g., Kita et al. [93] (data sets 100,101), Güntherodt et al. [92] (data set 93), Seydel and Fucke [87] (data set 92), Samarin [94] (data set 87), Mokrovskii and Regel [158] (data set 56), Eliutin et al. [88] (data set 49), and Ono and Yagi [89] (data set 61). The recommended values are generated with a temperature coefficient of $0.011 \times 10^{-8} \Omega\text{m K}^{-1}$, which is slightly (6%) lower than that given by Kita et al. [93], and slightly higher ($\sim 1\%$) than that determined from the data of Güntherodt et al. [92] (data set 93).

The recommended values for the solid state both uncorrected and corrected for thermal expansion of the material and those for the liquid state corrected

for thermal expansion are presented in table 10, and the values except those corrected for thermal expansion of the solid are also shown in figures 7 and 8 along with the experimental data. These values at temperatures above 100 K are for nickel of purity 99.99% or higher, while those below 100 K are applicable only to highly purified zone-refined nickel having a residual resistivity of $0.00320 \times 10^{-8} \Omega\text{m}$. The estimated uncertainty in the recommended values is $\pm 5\%$ below 150 K, $\pm 3\%$ from 150 to 1300 K, $\pm 5\%$ from 1300 K to the melting point, and $\pm 10\%$ for the liquid state.

For slightly less pure nickel having different residual resistivity, its electrical resistivity values can be calculated from the recommended values using the Matthiessen's rule, which will not introduce serious errors. For example, the data of Ahmad and Greig [220] (data set 90) show that for a specimen with a residual resistivity less than $0.009 \times 10^{-8} \Omega\text{m}$, the application of Matthiessen's rule causes an error of about 2% at 40 K and about 1% at 260 K. Also the data of Greig and Harrison [214] (data set 4), Ahmad and Greig [220] (data set 89), Berger and Rivier [229] (data set 23), White and Woods [21] (data sets 33,34), and of Kemp et al. [130] (data set 43), which are for specimens with residual resistivities of the order of a few tenths of a $\text{n}\Omega\text{m}$, show that the application of Matthiessen's rule causes errors generally of about 3% for temperatures below 300 K. The most interesting comparison is made with the data of Rowlands [230] (data set 57), since his data extend from liquid-helium temperatures up to above the Curie temperature. For this data set, the errors are less than 1% below 20 K, 10% at ~ 60 K, $\sim 6\%$ from ~ 100 to ~ 300 K, and drop to $\sim 3\%$ from ~ 500 K to above the Curie temperature. This behavior is consistent with the solute resistivities for dilute nickel alloys (see, for example, Schwerer and Cuddy [148]). Thus, when the Matthiessen's rule is used for calculating the electrical resistivity of less-pure nickel with a residual resistivity less than $0.05 \times 10^{-8} \Omega\text{m}$, the values are likely to be lower by $\sim 3\%$ than the true values from 40 K to room temperature, and are likely to be lower by $< 1\%$ at temperatures below 40 K and above the Curie temperature. For specimens of even lower purity, with a residual resistivity of about $0.3 \times 10^{-8} \Omega\text{m}$, the probable errors are about 2% at high and at low temperatures, but may be as high as -10% at intermediate temperatures (40-300 K).

The recommended values for the solid state uncorrected for thermal expansion and those for the liquid state given in table 10 can be represented approximately by the following expressions to within $\pm 0.5\%$.

1-60 K:

$$\rho = 0.0032 + 2.5 \times 10^{-5}T^2 + 39.1 \left(\frac{T}{456} \right)^5 \int_0^{456/T} \frac{x^5 e^x}{(e^x - 1)^2} dx \quad (35)$$

60-150 K:

$$\rho = 0.4214558798 - 2.07384562 \times 10^{-2}T + 3.48017305 \times 10^{-4}T^2 - 8.609303313 \times 10^{-7}T^3 \quad (36)$$

150-500 K:

$$\rho = -1.355285714 + 2.103190475 \times 10^{-2}T + 1.141428571 \times 10^{-5}T^2 + 4.523809524 \times 10^{-6}T^3 \quad (37)$$

500-600 K:

$$\rho = -50.1320558 + 2.978166536 \times 10^{-1}T - 5.156360117 \times 10^{-4}T^2 + 3.824418489 \times 10^{-7}T^3 \quad (38)$$

600-630 K:

$$\rho = 28.71 - 1.2315000 \times 10^{-1}(T_C - T) + 5.749999984 \times 10^{-4}(T_C - T)^2 \quad (39)$$

631-670 K:

$$\rho = 28.71 + 9.060833333 \times 10^{-2}(T - T_C) - 1.809583333 \times 10^{-3}(T - T_C)^2 + 3.941666667 \times 10^{-5}(T - T_C)^3 - 3.541666667 \times 10^{-7}(T - T_C)^4 \quad (40)$$

670-1400 K:

$$\rho = -6.329325957 + 8.023011038 \times 10^{-2}T - 4.451156858 \times 10^{-5}T^2 + 1.201757591 \times 10^{-8}T^3 \quad (41)$$

1400-1728 K:

$$\rho = -9.255955877 + 7.140577598 \times 10^{-2}T - 2.771379283 \times 10^{-5}T^2 + 5.589224949 \times 10^{-9}T^3 \quad (42)$$

1728-3000 K:

$$\rho = 63.22 + 1.10 \times 10^{-2}T \quad (43)$$

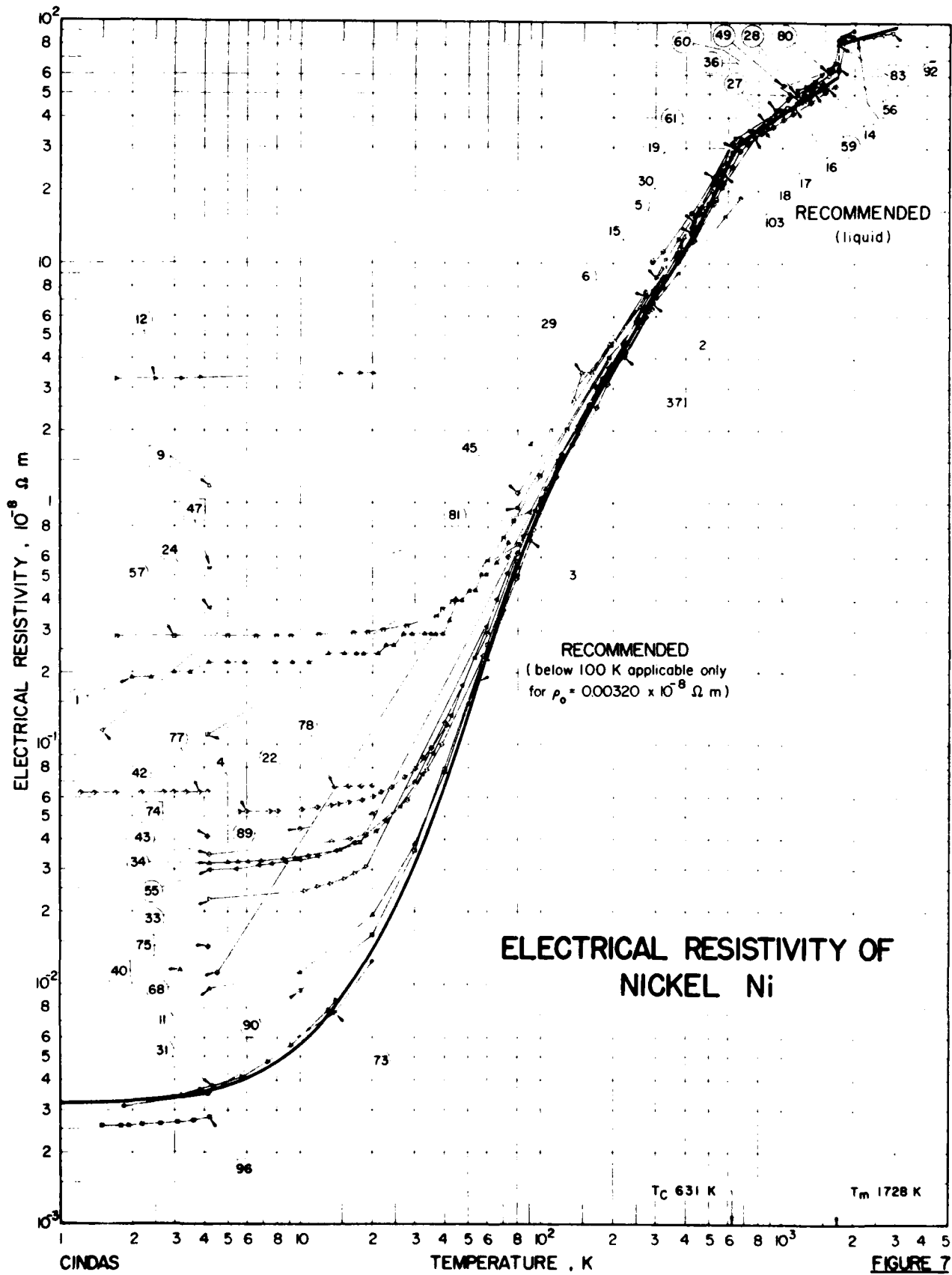
It should be emphasized that these expressions do not necessarily suggest any theoretical justification, and should be treated, most appropriately, as numerical aids only. It should also be understood that giving these expressions does not imply a recommendation for the temperature derivative of the electrical resistivity.

TABLE 10. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF NICKEL^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega\text{m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.00320	0.00320	630	28.71	28.86
4	0.00360	0.00359	670	31.06	31.24
7	0.00443	0.00442	700	32.14	32.34
10	0.00573	0.00572	800	35.52	35.80
15	0.00901	0.00899	900	38.58	38.95
20	0.0140	0.00140	1000	41.41	41.88
25	0.0212	0.0212	1100	44.06	44.65
30	0.0317	0.0316	1200	46.62	47.33
40	0.0678	0.0676	1300	49.15	50.00
50	0.135	0.134	1400	51.73	52.73
60	0.242	0.242	1500	54.36	55.54
70	0.377	0.376	1600	56.94	58.31
80	0.545	0.544	1700	59.50	61.07
90	0.741	0.739	1728	60.22(s)	61.85(s)
100	0.959	0.957	1728		82.23 ^b (l)
150	2.21	2.20	1800		83.02 ^b
200	3.67	3.67	1900		84.12 ^b
250	5.32	5.32	2000		85.22 ^b
273	6.16	6.16	2500		87.72 ^b
293	6.93	6.93	3000		90.22 ^b
300	7.20	7.20			
350	9.34	9.35			
400	11.78	11.80			
500	17.67	17.72			
600	25.54	25.66			

^a The values are for nickel of purity 99.99% or higher, but those below 100 K are applicable only to nickel having a residual resistivity of $0.00320 \times 10^{-8} \Omega\text{m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation.

^b Provisional value.



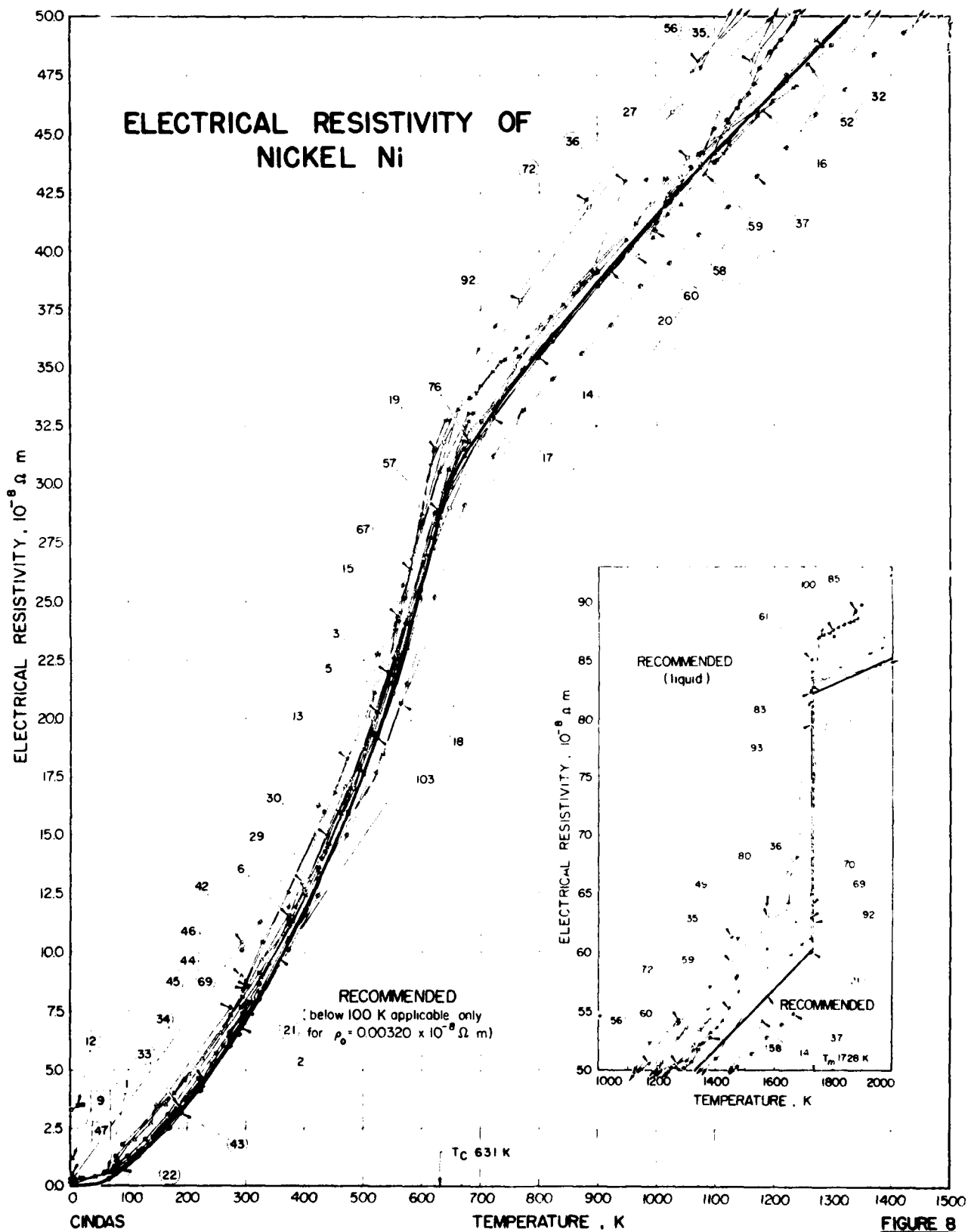


TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL NI

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	129	Kondorskii, E.I., Galkina, O.S., and Chernikova, L.A.	1958	A	2-73	Ni	Chemically pure; wire specimen 0.1 to 0.2 mm in diam and 150 to 160 mm long; annealed in vacuum at 1173 K for 1 h; slowly cooled at 100 K/h; residual resistivity $0.20 \times 10^{-8} \Omega \text{ m}$.
2	222	Standley, K.J. and Reich, K.H.	1955		293-473	Ni	Ingot heated in argon to about 1273 K for 12 h; slowly cooled; rolled to a sheet about 0.5 mm thick; discs of the required diameter punched out; polished on fine emery, annealed in vacuo and electrolytically polished, then annealed in vacuum; $T_C = 631 \text{ K}$.
3	223	Dutta-Roy, S.K. and Subrahmanyam, A.V.	1969	V	80-735		"Spectrographically pure," from Johnson Matthey Co.; $6 \times 0.3 \times 30 \text{ mm}$; annealed for 24 h at 1073 K in a vacuum furnace; cleaned in aqua regia; $T_C = 630 \text{ K}$.
4	214	Greig, D. and Harrison, J.P.	1965	G	5.6-41	JM 893; A	Pure; 0.0016 total impurity (mostly Fe and Si); polycrystalline; grain size $\sim 0.1 \text{ mm}$; from Johnson Matthey Co. (JM 893); annealed at 1023 K for 12 h; resistivity values calculated from reported ideal resistivity and ρ_0/L_0 ratio ($2.11 \pm 0.01 \text{ W}^2 \text{ cm}^2/\text{K}^2$).
5	231	Svensson, B.	1936	B	323-623		0.102 Fe, 0.036 Al and Si each; from Hilger of London; 1 mm in diam and 1 cm long; annealed at 1173 K; resistivity values calculated from measured resistance ratios and a $\rho(273 \text{ K})$ value of $6.58 \times 10^{-8} \Omega \text{ m}$ from Landolt-Börnstein: Physik-Chem. Tabellen 5 Auf. 5, 1050 (1923).
6	161	Broom, I.	1952	B	90-373		0.12 Mg, <0.05 Cr, Cu and Mn each, 0.03 C, and 0.01 Co; wire specimen 0.056 cm in diam; annealed at 873 K for 2 h, furnace cooled.
7*	232	Lavine, J.M.	1961	A	73-633	499 alloy	99.9 pure; from Driver Harris Co.; $T_C = 631 \text{ K}$.
8*	176	Kondorskii, E.I. and Sedov, V.L.	1960	A	4.2		Electrolytically pure; 5.9 mm in diam and 112 mm long.
9	176	Kondorskii, E.I. and Sedov, V.L.	1960	A	4.2		Technically pure; cylindrical specimen 5.9 mm in diam and 112 mm long; vacuum annealed at 1273 K for 8 h; furnace cooled.
10*	213	Ehrlich, A.C. and Rivier, D.	1968		1.6-19		"High purity"; polycrystalline plate, electropolished to a thickness of 0.19 mm; $\rho(293 \text{ K})/\rho(4.15 \text{ K}) = 2200$; only $\rho(T) - \rho(0)$ reported where $\rho(0)$ is the resistivity extrapolated to 0 K from data in the 1.75 to 4.15 K range, in which the resistivity is reported to be proportional to T^2 .
11	216	Farrell, T. and Greig, D.	1968	A	4.2-273		Pure; 3 mm in diam and 9 cm long; annealed for 15 h at 1123 K; resistivity values calculated from reported $\rho(0)$ and tabular values of $\rho(T)$.
12	233	Kondorskii, E.I., Galkina, O.S., and Chernikova, L.A.	1957	A	1.7-20	Ni	99.9 pure; wire specimen 0.1 to 0.2 mm in diam; supplied by Central Scientific Research Institute of Ferrous Metallurgy; cold drawn; annealed in neutral gas at 1173 K for 1 to 12 h; residual resistivity $1.54 \times 10^{-8} \Omega \text{ m}$.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL NI (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13	234	Kurbanniyazov, N., Cheremushkina, A.V., and Akmuradov, B.A.	1973	A	373-773		Pure; specimen of dimensions 3 x 6 x 100 mm; homogenize annealed in vacuum at 1273 K for 24 h, slow cooled in furnace; values from Graph.
14	218	Pallister, P.R.	1965	A	273-1550	E	99.84 Ni, <0.03 Fe, <0.01 Al, Co, Cr, Cu, Mg, Mn, Mo, Si, Sn, Ti, Zn, and Zr each, <0.005 Pb and <0.002 B; supplied by International Nickel Co. (Mond) Ltd.; annealed; measurements made in vacuum; smooth values from table; reported electrical resistivities based upon room temperature dimensions.
15	235	Sager, G.F.	1930	B	327-1016		Pure; wire specimen about 0.2 cm in diam and 35 cm long; electrolyzed from Mond anodes; vacuum melted under a pressure of 0.3 mmHg, chill cast, forged, hot rolled and cold drawn through a steel die plate; flash annealed, held at 1023 K for "considerable periods" and "later more thoroughly annealed"; current and potential leads of nickel silver soldered to specimen; density 8.74 g cm ⁻³ .
16	227	Bode, K.H.	1961	→	1098-1241		99.95 pure; wire specimen 1 mm in diam; vacuum melted, cast, polished, annealed for 12 h at about 1273 K; measured by compensation method.
17	219	Powell, R.W., Tye, R.P., and Hickman, M.J.	1965	A	293-1123	Sample 1	<0.03 Fe, <0.01 Al, Cr, Co, Cu, Mg, Mn, Mo, Si, Sn, Ti, Zn, and Zr each, <0.005 Pb, and <0.002 B; spectroanalyzed by International Nickel Co.; tubular specimen of 1.272 cm I.D., 1.908 cm O.D. and 20 cm long; supplied by the Castner Kellner Alkali Co.; density 8.61 g cm ⁻³ .
18	219	Powell, R.W., et al.	1965	A	373-773	Sample 2	"Very high purity"; electrolytic; tubular specimen of 0.634 cm I.D., 2.801 cm O.D., and 19 cm long; supplied by National Engineering Lab.; density 8.90 g cm ⁻³ .
19	219	Powell, R.W., et al.	1965	A	293-623	Sample 4	Commercial Ni; rod specimen 2.54 cm in diam and about 20 cm long; supplied by Explosives Research and Development Establishment.
20	219	Powell, R.W., et al.	1965	A	293-1323	Sample 5 JH Lab. No. 4497	"High spectrographic purity"; trace amounts of Al, Ca, Cu, Li, Mg, Si, Ag, and Na; rod 0.5 cm in diam and 15 cm long; supplied by Johnson, Matthey and Co.; density 8.91 g cm ⁻³ .
21	236	Martynuk, M.M. and Tsapkov, V.I.	1973	→	298, 1726		99.93 pure; specimen 0.3-1 mm in diam and 50 cm long; specimen heated in air by a 400 usec pulse of 1-4 kAmp; voltage and current measured by double beam pulse oscilloscope; resistivity at melting point determined from break points corresponding to the onset and end of fusion on the relative resistance curve; data not corrected at higher temperature.
22	229	Berger, L. and Rivier, D.	1962	B	4.2-292	NI 5011 (I)	Specimen 0.15 cm in diam and 5.2 cm long; supplied by Johnson, Matthey and Co.; annealed for 4 h at 1273 K in a vacuum of 10 ⁻⁵ mmHg; furnace-cooled at a rate of 150 K/h; $\rho(273 K)/\rho(4.2 K) = 60$.
23*	229	Berger, L. and Rivier, D.	1962	B	4.2-273	NI 5011 (II)	Specimen 0.19 cm in diam and 5.0 cm long; from the same stock as the above specimen; annealed for 10 h at 1573 K in hydrogen at 1573 K in a vacuum of 10 ⁻² mmHg for 2 h; $\rho(273 K)/\rho(4.2 K) = 298$.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL Ni (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
24	237	Krommeller, H. and Buck, O.	1964	A	4-273	No. 1	99.99 pure; from Johnson, Matthey and Co.; single crystal with one axis parallel to <111>; specimen 3.2 mm in diam and 120 mm long.
25*	238	Neimark, B.E. and Bykova, T.I.	1965		793-773	No. 1	99.87 (Ni + Co); tube 8.51 mm O.D. and 8.025 mm I.D.; smoothed values from table.
26*	238	Neimark, B.E. and Bykova, T.I.	1965		373-748	No. 2	Tube 12.96 mm O.D. and 11.025 mm I.D.; smoothed values from table.
27	225	Kirichenko, P.I. and Mikryukov, V.E.	1964		313-1172		99.999 ⁺ pure; 0.3 cm in diam and 30 cm long; forged from sheet; annealed in vacuum for 48 h at 1173 K; furnace cooled.
28	239	Jain, S.C., Goel, T.C., and Chandra, I.	1967	+	1152-1320		99.95 pure; filaments 0.05 cm thick, 1 cm wide, and 14 cm long; obtained from Johnson, Matthey and Co.; data from figure; experimental method same as Jain and Krishna, Proc. R. Soc. London, <u>A225</u> , 7, 1954.
29	240	Watson, T.W. and Robinson, H.E.	1964	V	110-803		99.85 Ni, 0.11 Co, 0.026 Cu, 0.006 Fe, 0.001 Al, <0.004 Si, <0.002 Ti, <0.001 Cr and Mg each, and <0.0005 Mn; electroformed nickel from International Nickel Co.; 2.54 cm in diam and 37 cm long; smoothed values from table.
30	240	Watson, T.W. and Robinson, H.E.	1964	V	384-680		The above specimen measured with decreasing temperature.
31	211	White, G.K. and Tainsh, R.J.	1967		2.3-14		"High purity"; specimen 0.1 cm x 0.1 cm x 7 cm; prepared by Bell Telephone Laboratories; annealed in vacuum of 10^{-7} Torr at 773 K; $\rho(273 \text{ K})/\rho(4 \text{ K}) = 2500$.
32	228	Davis, M., Densen, C.E., and Rendall, J.H.	1955		293-1273		0.01-0.2 O, 0.07 C, 0.016 Si, 0.013 Fe, 0.003 S, 0.0005 Mn, and 0.0003 Mg; grade A carbonyl nickel powder; supplied by Mond Nickel Co.; sintered and annealed; density 8.9 g cm ⁻³ ; Curie point 626 K.
33	21	White, G.K. and Woods, S.B.	1959	G	4.2-298	NI 2	99.997 pure, 0.0010 Fe, 0.0010 Si, 0.0003 Cr and Mg each, 0.0002 Ca, Cu, and Mn each, and 0.0001 Ag, from Johnson, Matthey and Co. (JM 10389); rod specimen 2 mm in diam and 6 to 8 cm long; vacuum annealed at 1073 K; resistivity values calculated from reported ρ_1 , $\rho(4.2 \text{ K})/\rho(295 \text{ K}) = 3.23 \times 10^{-3}$ and $\rho_1(295 \text{ K}) = 7.04 \times 10^{-8} \Omega \text{ m}$.
34	21	White, G.K. and Woods, S.B.	1959	G	4.2-252	NI 3	Similar to the above specimen except (1) specimen 0.63 mm in diam, (2) resistivity ratio $\rho(4.2 \text{ K})/\rho(295 \text{ K}) = 4.51 \times 10^{-3}$, (3) $\rho_1(295 \text{ K}) = 7.33 \times 10^{-8} \Omega \text{ m}$; because of slight uncertainty in ρ/A , ρ_1 was normalized to the value for the above specimen for which ρ_1 and A are more accurately known.
35	241	Reddy, B.K. and Goel, T.C.	1975	V	1163-1641		99.95 pure; tubular specimen 0.75 cm I.D., 0.3 mm wall thickness, and 18 cm long; obtained from Johnson, Matthey and Co.; specimen heated for about 1 h at 1630 K and cooled to room temperature repeatedly for 6 or 7 times.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL NI (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
36	97	Kovenskiĭ, I.I. and Samsonov, G.V.	1963	+	891-1673		99.86 Ni, 0.10 C, 0.01 Co, 0.008 Cu, 0.004 Fe, and 0.002 Si and S each; wire specimen; measured in argon atmosphere, specimen heated by passing an electric current through; smoothed values from figure.
37	78	Kierspe, W., Kohlhaas, R., and Gonska, H.	1967	B	73-1668		<0.0003 Si, 0.0002 Fe and Mg each, <0.0001 Al, Cu and Ag each; obtained from Koch-Light Laboratories Ltd.; smoothed values from figure.
38*	242	Birss, R.R. and Dey, S.K.	1961		78-1306		Smoothed values from graph of ρ vs. T/θ , with θ apparently equal to 370 K.
39*	243	Franklin Institute, Laboratories for Research and Development	1953		73-830		No details reported.
40	244	Coltman, R.R., Klabunde, C.E., and Redman, J.K.	1967		3-2		99.99 ⁺ nominal purity; 0.025 cm in diam and 5 cm long; annealed at 1223 K in air at a pressure of 8×10^{-6} Torr; furnace-cooled.
41*	179	Niccolai, G.	1908	B	84-673		Wire specimen 0.5 mm in diam and 8 m long wound on an insulating spool.
42	245	Sharma, J.K.N.	1967	D	1.5-293		99.995 pure; polycrystalline; wire specimen obtained from Johnson, Matthey and Co.; L/A ratio $2.88 \times 10^3 \text{ cm}^{-1}$.
43	130	Kemp, W.R.G., Klemens, P.G., and White, G.K.	1956	G	4.2-293		99.99 ⁺ pure, traces of Al, Ca, Cu, Si and Ag, and very faint traces of Li, Mg, and Na; 2 mm in diam; obtained from Johnson, Matthey and Co.; annealed in vacuum at 1023 K for 4 h; ideal electrical resistivity, ρ_1 , from figure; ρ_0 taken as $0.0347 \times 10^{-8} \Omega \text{ m}$, $\rho = \rho_1 + \rho_0$.
44	246	Masumoto, H.	1927		303		0.10 Fe, 0.037 C, 0.019 S, 0.013 Cu, 0.006 Si, and traces of Al, Co, Mn, and P; 5 mm in diam and 20 cm long; obtained from Mond & Co.; cast and machined; annealed at 1073 K for 40 min.
45	163	Bucken, A. and Ditttrich, K.	1927	V	80,273		Electrolytic.
46	247	Rubanenko, I.R. and Grossman, M.I.	1969		293		7 x 7 x 28 mm; measuring temperature assumed 293 K.
47	248	Mitchell, M.A., Klemens, P.G., and Reynolds, C.A.	1971	A	4-2		99.9 pure; single crystal, grown by a variation of the Bridgman method; specimen axis along <111> direction; annealed in vacuum at 1203 K for 48 h.
48*	215	Fert, A. and Campbell, I.A.	1968		4-79		Ideal resistivity reported only.

* Not shown in figure.

TABLE II. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL NI (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
49	88	Eljutin, V.P., Turov, V.D., and Maurakh, M.A.	1965	R	1013-1997	Nickel 270	98.5-99.0 pure; electrolytic; liquid state obtained by melting in graphite crucible either in an atmosphere of helium or in vacuum.
50*	249	Starr, C.D.	1969		811		100 nominal purity; temperature coefficient of resistivity $\alpha(298,378) = 0.00565$ and $\alpha(218,298) = 0.00461/\text{deg}$; data from table at 811 K only.
51*	98	Schroeder, K. and Giannuzzi, A.J.	1969		375-825		99.999 pure; wire specimen; annealed in an inert gas atmosphere (92 He, 8 Ar) for 2 h at ~ 150 K above the Curie temperature; resistivity values calculated from reported $\rho(T)/\rho(T_C)$, with $\rho(T_C) = 29.288 \times 10^{-6} \Omega \cdot \text{m}$, taken from data set 52.
52	217	Laubitz, M.J., Matsumura, T., and Kelly, P.J.	1976	A	90-1250		99.999 pure (nominal); 0.0016 C, 0.0014 Si, 0.0007 Fe, 0.0006 Cu, 0.0005 Al and O, 0.0003 F, 0.0002 K, Na, and S each, 0.0001 Ta, 0.0007 Cl, 0.00005 Ca, 0.00003 Ti and N ₂ each, 0.00002 Cr and Mn each, 0.00001 Pb, Mg and Ag each, 0.000004 V and 0.000002 B by mass spectrographic analysis; from Metals Research Ltd.; polycrystalline; specimen 2 cm in diam and 20 cm long; annealed in vacuum of 5×10^{-6} Torr at 1400 K for 2 h; slow cooled for measurements between 300 and 1250 K; machined to 1 cm in diam and 10 cm long, unannealed for measurements between 90 and 370 K; density $8.908 \pm 0.001 \text{ g cm}^{-3}$ at 293 K; T_C about 630 K; residual resistivity ratio 220 ± 10 ; smoothed values from table.
53*	199	Potter, H.H.	1937	V	77-1153		99.971 pure; 0.018 Fe and 0.010 C; obtained from Adam Hilger; specimen 2 mm in diam and 8 cm long, bent into U shape; resistance ratio R/R _{273K} reported; reference value of $\rho(273 \text{ K}) = 6.16 \times 10^{-6} \Omega \cdot \text{m}$ assumed; T_C reported to be 629.3 or 629.8 K, depending on specimen.
54*	250	Araj, S.	1961	V	0-1000		Pure; 0.01 Fe and Cu each, and traces of C, Co, S, and Si; wire specimen 1.5 mm in diam; enclosed in silica tubes evacuated to 10^{-5} mmHg; annealed for 68 h at 1400 K; quenched in saline solution at room temperature; original data reported graphically; extracted from the reported smooth curve.
55	80	Price, D.C. and Williams, G.	1973	A	4-300		99.998 pure; specimen $0.15 \times 0.2 \times 10$ cm; supplied by Johnson, Matthey and Co.; cold rolled between Melinex sheets, etched and annealed for 24 h in vacuo at 873 K; ideal resistivity reported graphically; total resistivity obtained by adding the reported residual resistivity $\rho(4.2 \text{ K}) = 0.0299 \times 10^{-6} \Omega \cdot \text{m}$ to the reported ideal resistivity.
56	158	Mokrovskii, N.P. and Regel, A.R.	1953	R	1073-1964		99.7 pure; specimen contained in corundum crucible ~ 12 mm in diam and 25 mm high; smoothed values from graph.
57	230	Rowlands, J.A.	1973	A	1.7-672		Pure; from Sherritt Gordon Mines, annealed; $\rho_0 = 0.28195 \times 10^{-6} \Omega \cdot \text{m}$; data in tabular form supplied by author.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL NI (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
58	251	Hong, H.Y.	1966	+	998-1381		99.92 pure; wire specimen 0.081 cm in diam and 61 cm long; supplied by British Driver-Harris Co. Ltd.; electrically heated in air at 1173 K for 5 min, oxide formed had olive-gree color; measured in vacuum of about 2×10^{-6} Torr; data extracted from figure.
59	251	Hong, H.Y.	1966	+	993-1479		Similar to the above specimen except electrically heated in air at 1173 K for 15 min.
60	251	Hong, H.Y.	1966	+	970-1408		Similar to the above specimen except electrically heated in air at 1173 K for 25 min.
61	89	Ono, Y. and Yagi, T.	1972	R	1728-1898		99.9 ⁺ pure; in liquid state; contained in a 10 mm I.D. recrystallized alumina crucible; density data of Saito et al. (Bull. Res. Inst. Min. Dress. Metall., Tohoku Univ., 25, 67, 109, 1969) used to calculate specimen volume; data given as the formula $\rho(10^{-8} \Omega m) = 0.0280 T(C) + 44.32$.
62*	252	Schindler, A.I., Smith, R.J., and Salkovitz, E.I.	1956	B	6-292		99.99 pure; material obtained from International Nickel Co.; specimen 2.0 mm in diam and 16.3 cm long; fabricated from spectrographic rod; vacuum annealed at 1073 K for 2 h then gradually cooled for 24 h.
63*	138	Devar, J. and Fleming, J.A.	1893	B	76-469		Pure; prepared by Mr. Mond, nickel tubes formed by passing vapor of nickel carbonyl through heated glass tube, portion of nickel tube cut into a very fine spiral on lathe; resistance ratio reported; data uncorrected for thermal expansion; data extracted from table; $\rho(273 K) = 12.323 \times 10^{-8} \Omega m$, Matthiesen's value as given in Everett's "Physical Units" used to convert resistance ratio to resistivity; temperatures at 76.1, 191.3, and 229.6 K measured by platinum resistance thermometer.
64*	253	Dewar, J. and Fleming, J.A.	1892	B	91-368		Pure, carbonyl nickel; wire specimen had probable dimensions of 0.0076 cm in diam and 50 to 100 cm long; from Johnson, Matthey and Co.; measurement of resistance repeated several times, mean observed resistivity reported; data uncorrected for thermal expansion; data extracted from table.
65*	148	Schweerer, F.C. and Cuddy, L.J.	1970	V	4-940		"High purity"; rod specimen 1.8 mm in diam; $\rho(4.2 K) = 0.024 \times 10^{-8} \Omega m$; measurement made "quasi-statically" with temperature decreasing at 1 C min ⁻¹ .
66	254	Kalinovich, D.F., Kovenski, I.I., Saulin, M.D., and Stateenko, V.M.	1972		657-1291		Pure; original data reported graphically.
67	226	Kaul, S.N.	1974	G	84-900		Values from table supplied by author.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL Ni (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
68	103	Morak, J.A. and Blewitt, T.H.	1972	A	4.5-295		Polycrystalline; wire 0.025 cm in diam and 5 cm long; data from table.
69	255	Borodovakaisa, L.N. and Lebedev, S.V.	1955	+	293, 1726		Wire specimen 0.015 cm in diam; measured by pulse heating of the wire with current density 6×10^4 to 5×10^5 Amp cm^{-2} ; voltage and current measured by pulse oscillogram.
70	255	Borodovakaisa, L.N. and Lebedev, S.V.	1955	+	293, 1726		Similar to the above except measured by heating the wire slowly in vacuum.
71	256	Köster, W. and Gähling, W.	1951	+	293		99.9 pure; Mond Nickel; vacuum melted in a high frequency oven; out-gassed; measured by compensation method.
72	257	Vedernikov, M.V. and Kolomoets, N.Y.	1961	A	295-1473	N-00	99.8 pure; electrolytic.
73	212	Ehrlich, A.C., Huguenin, R., and Rivier, D.	1967		1.9-20	N1 III	"Pure", polycrystal in the form of flat plates; from Johnson, Matthey and Co.; annealed at 1273 K; cooled at a rate of 3 K min^{-1} ; resistivity value calculated from reported $\rho(293 \text{ K})/\rho(T)$ ratio; $\rho(293 \text{ K}) = 6.93 \times 10^{-6} \Omega \text{ m}$.
74	212	Ehrlich, A.C., et al.	1967		4.2	N1 I	Similar to the above specimen except annealed at 1473 K and coded at a rate of 5 K min^{-1} .
75	212	Ehrlich, A.C., et al.	1967		4.2	N1 I	A different specimen cut from the same stock as the above; annealed at 1273 K and cooled at a rate of 5 K min^{-1} .
76	141	Shirakawa, Y.	1939	V	78-1123		0.02 Fe, 0.01 C, 0.003 S, 0.002 Si, and 0.001 P and Mn each; electrolytic nickel from Monson and Co.; 0.0608 cm in diam and 3.95 cm long; annealed at 1273 K for 1 h in vacuum, with specimen in the east-west direction; slowly cooled; lead wire of nickel soldered by pure silver; reannealed at 1123 K for 1 h in vacuum; slow cooled; measurement done with specimen in the east-west direction.
77	203	Sudovtsov, A.I. and Semeneako, E.E.	1956	A	1.2-4		Polycrystalline specimen in the form of this ribbon; from Hilger; $R(4.2 \text{ K})/R(273 \text{ K})$ reported to be 1.0148×10^{-2} ; resistivity value calculated from reported resistance ratio, with resistivity at 273 K taken to be $6.16 \times 10^{-6} \Omega \text{ m}$.
78	203	Sudovtsov, A.I. and Semeneako, E.E.	1962	A	14-20		The above specimen measured at hydrogen temperatures; specimen described at 99.94 pure; sealed in glass tube with helium gas; values calculated from reported $R(T)/R(273 \text{ K}) = 1.00986 \times 10^{-2} + 2.88 \times 10^{-6} T^2 + 4.85 \times 10^{-11} T^3$; with $\rho(273 \text{ K})$ taken to be $6.16 \times 10^{-6} \Omega \text{ m}$.
79*	258	Panakhov, T.M., Peninov, R.I., Muradov, T.I., and Ibragimov, A.I.	1974		339-1042		99.99 pure; electrolytic; measured in a vacuum of $\sim 10^{-6}$ mmHg.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL Ni (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
80	259	Tyagunov, G.V., Baum, B.A., and Kushnir, N.N.	1972	R	1573, 1973		99.98 pure.
81	198	Holborn, L.	1919		80-672		Wire specimen 0.5 mm in diam; resistivity values calculated from reported $R(T)/R(273\text{ K})$, with $\rho(273\text{ K})$ taken to be $6.16 \times 10^{-9}\ \Omega\text{ m}$.
82*	108	Schimank, H.	1914	A	20-273		Wire specimen 1-2 m long; from Hartmann and Braun; resistivity value calculated from reported $R(T)/R(273\text{ K})$, with $\rho(273\text{ K})$ taken to be $6.16 \times 10^{-9}\ \Omega\text{ m}$.
83	187	Güntherodt, H.J. and Künzi, H.U.	1973	C	1726		99.998 pure; from Johnson, Matthey and Co.; in liquid state; temperature = 1726 K assumed.
84*	260	Busch, G., Güntherodt, H.J., Künzi, H.U., Meier, H.A., and Schlapbach, L.	1970		1726		No details reported.
85	159	Baum, B.A., Tyagunov, G.V., Gel'd, P.V., and Khasin, G.A.	1971	R	1573, 1873		99.99 pure; zone refined; specimen contained in either an alumina or zirconia crucible; measured in an atmosphere of helium.
86*	193	Dubini, E., Esfin, O.A., and Vetrofin, N.A.	1969		1873		"High purity"; measured in purified helium.
87*	94	Samarin, A.M.	1962	R	1728-1900		Measured in an atmosphere of helium; rotating field apparatus calibrated against an iron specimen with resistivity value at melting taken from R.W. Powell, Philos. Mag. 44, 772, 1953; resistivity values calculated from reported conductivity = $(32.35 - 0.88 \times 10^{-3} T(C)) \times 10^4 (\text{ohm cm})^{-1}$; (this equation is apparently erroneous).
88*	261	Schwexer, F.C. and Silcox, J.	1968		16.4-56.3		No details reported; $\rho(273\text{ K})/\rho(4.2\text{ K}) \sim 1400$; values of $\rho(T)-\rho(1.4\text{ K})$ reported only.
89	220	Ahsad, H.M. and Grieg, D.	1974	A	10-873	Pure Ni(I)	"Spec-pure" nickel from Johnson, Matthey and Co.; 0.5 mm in diam and 10 cm long; annealed in vacuum at 1223 K for 24 h; $T_C = 631\text{ K}$; data below 260 K supplied by author; values from table.
90	220	Ahsad, H.M. and Grieg, D.	1974	A	10-260	Pure Ni(II)	Similar to the above; data supplied by author.
91*	221	Zumsteg, F.C. and Parks, R.D.	1970	V	623-650		99.999 pure; 0.005 cm thick, 0.05 cm wide and 50 cm long; swaged; annealed 1-30 days <i>in situ</i> before measurement; sample mounted on fiber-glass; resistivity values calculated from reported $R(T)/R(T_C)$ and $\rho(T_C)$ taken to be $28.70 \times 10^{-9}\ \Omega\text{ m}$.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL N1 (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
92	87	Seydel, U. and Fucke, W.	1977	*	508-2989		99.99 pure; 0.0015 Fe, 0.0003 Cu, 0.0002 Si, 0.0001 Ag, Al and Ca each, and <0.0001 Cr, Mg, Mn, and Sn each (Chemical analysis); measured by an exploding wire technique; measurement error 4%; smoothed values from curve.
93	92	Güntherodt, H.-J., Hauser, E., Künnz, H.U., and Müller, R.	1975	*	1723-1843		99.999 pure, from Johnson, Matthey and Co.; measured with a potential method in which the sample material was enclosed in an alumina tube with four protrusions setting as current and potential contacts.
94*	262	Yao, Y.D., Araj, S., and Anderson, E.E.	1975	A	4-300		0.0010 Fe, 0.0007 Al, 0.0005 Si, 0.0002 Ca, Cu, and Mg each, and <0.0001 Ag and Mn; from Johnson, Matthey and Co., R(4.2 K)/R(298 K) = 3.3×10^{-6} .
95*	209	Wycisk, W. and Feller-Kniepmeier, M.	1976	A	295-1390		99.999 pure, <0.0005 Si, 0.0003 Fe, <0.0001 Mg and Ag each, 0.00001 Co, and 0.00005 Ca, Cd, Cu and Pb each; 5 mm diam and 150 mm long rod from Gallard-Schlesinger Chemical Corp.; zone-refined 5 times in electron beam; rolled and drawn to 60 μ m diam wire with diamond tools; electrolytically cleaned after each rolling and drawing with a 7% acetic acid and 2% perchloric acid solution; annealed 1/2 to 1 h at 573-673 K in a vacuum of $<10^{-7}$ Torr; flushed with helium; then slowly lowered over a liquid helium bath, with copper guard roof the ends of which are immersed in liquid helium; heated to 1073 K for 40 min; then 1273-1373 for 5 min and 1473 K for 1 min; potential leads knotted and sintered to wire specimen at the highest temperature; 0(4.2 K) reported to be $0.0027 \times 10^{-6} \Omega \cdot m$ is a longitudinal magnetic field of 250 Oe; resistivity values calculated from reported R(T)/R(296 K), with $\rho(296 K)$ calculated from reported R(696 K)/R(4.2 K) = 1923.
96	209	Wycisk, W. and Feller-Kniepmeier, M.	1976	A	1.5-4.2		Similar to the above, except R(296 K)/R(4.2 K) = 1845; resistivity values calculated from reported $\Delta R/R(4.2 K)$; measured wire a current density of $\sim 3.5 \times 10^4 A \cdot cm^{-2}$, and in a magnetic field of 250 Oe.
97*	209	Wycisk, W. and Feller-Kniepmeier, M.	1976	A	4.2		Similar to the above, except measured without a longitudinal magnetic field, resistivity value calculated from reported R(296 K)/R(4.2 K), with $\rho(296 K) = 5.191 \times 10^{-6} \Omega \cdot m$ from data set 96.
98*	209	Wycisk, W. and Feller-Kniepmeier, M.	1976	A	4.2		Similar to the above, except single crystal 5 mm in diam and 10 cm long.
99*	209	Wycisk, W. and Feller-Kniepmeier, M.	1976	A	4.2		Similar to the above, except measured in a longitudinal magnetic field of 250 Oe.
100	93	Kita, Y., Ohguchi, S., and Morita, Z.	1978		1654-1882		0.08 Co, 0.007 Fe, 0.005 Si, 0.0025 Cu, $C \cdot 1/2$ Mg, 0.001 Al, and 0.0007 S; measured with a four probe method in which the electrodes are made of the same material as the specimen; measured in a vacuum of 10^{-6} Torr; data points taken at temperatures in the sequence: 1770, 1788, 1796, 1814, 1840, 1857, 1869, 1882, 1867, 1852, 1836, 1817, 1799, 1791, 1767, 1748, 1711, 1692, 1772, and 1654 K; values corrected for thermal expansion; values from table supplied by authors.

* Not shown in figure.

TABLE 11. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF NICKEL N1 (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
101*	93	Kita, Y., Ohguchi, S., and Morita, Z.	1978	+	1648-1872		Same as the above; a second melt; temperature sequence: 1755, 1780, 1798, 1818, 1838, 1856, 1872, 1885, 1836, 1821, 1805, 1793, 1769, 1756, 1740, 1693, 1667, and 1648 K.
102*	93	Kita, Y., et al.	1978	+	1720-1888		Same as the above; a third melt; temperature sequence: 1751, 1766, 1782, 1800, 1811, 1826, 1844, 1865, 1888, 1873, 1857, 1841, 1823, 1808, 1795, 1781, 1765, 1741, and 1720 K.
103	224	Jackson, P.J. and Saunders, N.H.	1968		293-673		99.999 pure (10 ppm metallic impurities); polycrystalline; annealed; data from table supplied by N.H. Saunders.
104*	263	Sherif, I.I., Ibrahim, A.F., Ghani Awad, A.A., Ammar, A.S., and Essail, S.A.	1976	+	373-873		Polycrystalline spectroscopic nickel supplied by the National Research Center, Cairo; either dumbbell-shaped specimen with long ends about 3 cm long and 0.9 cm in diam or wire specimen of gauge length 2.5 cm; measured by a four probe method in an over flushed with inert gas.
105*	210	Fujii, T.	1970	A	4.2		0.0050 C, 0.0007 O, 0.0004 H, 0.0003 N and Fe each, 0.0002 Si and Mg each, and <0.0001 Ag, Al, Ca and Cu each; supplied by Johnson, Matthey and Co.; carbon impurity determined by vacuum combustion method, nitrogen and oxygen impurities determined by vacuum fusion method with high purity silicon; metallic impurities determined by the supplier; 5 mm in diam and 20 mm long; annealed at 1273 K for 1 h; grain size reported to be 9 grams cm ⁻² ; resistivity value calculated from reported $\rho(295 \text{ K})/\rho(4.2 \text{ K})$ with $\rho(295 \text{ K})$ taken to be $7.004 \times 10^{-8} \Omega \cdot \text{m}$.
106*	210	Fujii, T.	1970	A	4.2		0.0040 C, 0.0010 O, <0.0002 H, and 0.0001 N; metallic impurities not determined; the above material after surface oxidation treatment to remove carbon in an air atmosphere for 2 h at 1173 K; "this treatment is aimed to remove carbon easily by volatilization in vacuum, with a chemical reaction to the forms of CO or CO ₂ from NiO in the process of molten containing excessive oxygen during zone melting"; 1 pass zone-refined in vacuum at 3 mm min ⁻¹ ; 1.5 mm in diam and 50 mm long, made by diameter controlled operation in zone refining process; gaseous impurities determined by the methods given above; grain size 4 grains cm ⁻² ; resistivity value calculated by the same method as above.
107*	210	Fujii, T.	1970	A	4.2		Similar to the above except containing 0.0020 C, <0.0002 O, <0.0001 H, and trace amount of N and 3 pass zone-refined in vacuum at 1 mm min ⁻¹ ; grain size 3-5 grain cm ⁻² .
108*	210	Fujii, T.	1970	A	4.2		Similar to the above except containing 0.0010 C, <0.0002 O and trace amounts of hydrogen and nitrogen and 5 pass zone-refined in vacuum at 1 mm min ⁻¹ ; grain size 4 grain cm ⁻² .
109*	210	Fujii, T.	1970	A	4.2		Similar to the above except oxygen content is reduced to 0.0001; 10 pass zone-refined in vacuum at 1 mm min ⁻¹ ; grain size 4 grain cm ⁻² .

* Not shown in figure.

TABLE 12. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF NICKEL NI
 [Temperature, T, K; Electrical Resistivity, ρ , $10^{-6} \Omega \text{ m}$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 1															
2.0	0.19	80	0.570	18.7	0.0592*	240	5.2	1.965	0.000108	3.24	0.329				
2.4	0.19	90	0.600	20.2	0.0606	245	5.3	2.537	0.000188	3.89	0.333				
3.0	0.20	98	0.776	20.5	0.0613*	260	6.1	2.560	0.000194	14.84	0.345				
3.5	0.20	120	1.34	21.7	0.0631	279	6.7	3.056	0.000271	15.92	0.345*				
4.2	0.22	141	1.72	22.3	0.0629*	298	7.6	3.131	0.000276	16.53	0.345				
5.0	0.22	159	2.29	22.3	0.0636*	321	8.4	3.552	0.000362	16.97	0.345*				
5.9	0.22	180	3.06	24.2	0.0661	329	8.8	3.587	0.000357	17.49	0.345*				
7.9	0.22	201	3.63	24.4	0.0667*	363	10.4	3.974	0.000450	17.99	0.346				
9.3	0.22	220	4.40	24.4	0.0692*	411	12.7	3.993	0.000448	19.42	0.346*				
10.9	0.22	249	5.35	25.1	0.0684*	475	16.6	13.79	0.00470	20.24	0.346				
13.2	0.24	272	6.31	27.4	0.0733	493	17.5	14.64	0.00520	DATA SET 13					
15.4	0.24	295	7.46	30.3	0.0793	508	18.8	15.54	0.00591	373	11.9				
17.3	0.24	304	7.84	30.3	0.0807	528	20.3	16.42	0.00658	473	18.3				
18.5	0.24	375	11.29	32.6	0.0884	548	21.8	16.62	0.00691	573	25.2				
21.1	0.24	429	13.97	35.2	0.0973	565	23.2	17.09	0.00711	673	32.5				
22.8	0.26	458	15.88	40.8	0.1222	574	23.9	17.53	0.00749	773	47.5*				
24.8	0.26	480	17.03	DATA SET 5		579	24.2	18.44	0.00832	DATA SET 14					
26.9	0.29	542	22.01	323.2	8.65	585	24.8	19.04	0.00887	273	6.36				
29.4	0.29	556	23.92	423.2	13.6	591	25.3	19.77	0.00961	293	7.01				
31.4	0.29	585	26.98	523.2	20.3	597	25.9	19.77	0.00961	300	7.31				
33.0	0.29	616	30.62	623.2	28.8	602	26.6	19.77	0.00961	400	11.65				
35.0	0.29	648	32.72	DATA SET 6		610	27.3	19.77	0.00961	500	17.6				
36.6	0.29	693.177	33.876	90	1.77	628	29.4	19.77	0.00961	550	21.1				
39.3	0.29	734	35.024	194.7	4.59	634	30.0	19.77	0.00961	600	25.25				
41.9	0.33	DATA SET 4		273	7.37	643	30.6	19.77	0.00961	627	28.15				
44.7	0.40	5.6	0.0523	373	11.56	DATA SET 7*		19.77	0.00961	700	32.0				
47.0	0.40	6.0	0.0523	DATA SET 8*		4.2	0.034	19.77	0.00961	800	35.45				
48.7	0.42	7.5	0.0526	4.2	0.034	DATA SET 9		19.77	0.00961	900	38.5				
50.7	0.44	8.1	0.0529	DATA SET 10*		4.2	1.18	19.77	0.00961	1000	41.25				
53.9	0.44	10.2	0.0533	74	0.4	1.612	0.0000777	19.77	0.00961	1100	43.8				
56.6	0.51	10.2	0.0535*	84	0.6	1.736	0.0000852	19.77	0.00961	1200	46.4				
59.2	0.51	11.8	0.0544	92	0.8	1.924	0.000105	19.77	0.00961	1300	48.75				
62.8	0.54	12.1	0.0544*	101	1.0	DATA SET 11		19.77	0.00961	1400	50.95				
65.3	0.57	13.5	0.0551	118	1.3	4.2	0.0095	19.77	0.00961	1500	54.25				
67.3	0.58	13.6	0.0555*	125	1.6	10	0.0115	19.77	0.00961	DATA SET 12					
69.9	0.62	14.2	0.0555*	145	2.4	20	0.0195	19.77	0.00961	300	8.77				
73.1	0.69	14.5	0.0564	173	3.0	30	0.0385	19.77	0.00961	327	10.42				
DATA SET 2															
293.2	6.8	15.1	0.0564*	173	3.0	40	0.0775	19.77	0.00961	DATA SET 15					
473.2	16.0	16.3	0.0569*	205	4.0	50	0.1495	19.77	0.00961	300	8.77				
		17.2	0.0585*	232	4.9	60	0.2295	19.77	0.00961	327	10.42				
		18.2	0.0586	DATA SET 12		70	0.3695	19.77	0.00961	DATA SET 15					
				1.612	0.0000777	80	0.5495	19.77	0.00961	300	8.77				
				1.736	0.0000852	90	0.7495	19.77	0.00961	327	10.42				
				1.924	0.000105	273	6.31	19.77	0.00961	DATA SET 15					

* Not shown in figure.

TABLE 12. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF NICKEL NI (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 53(cont.)*</u>													
551.5	21.25	600	27.0	100.2	1.052	27.85	0.3122	993	40.6	993	40.6	6	-0.08
617.5	27.06	637	30.8	105.7	1.177*	37	0.3411	1020	41.6	1020	41.6	65	0.41
621.5	27.48	700	33.4	113.5	1.368*	39	0.3654	1042	42.0	1042	42.0	77	0.63
624	27.80	800	36.8	122.2	1.566	43	0.3918	1083	43.3	1083	43.3	272	6.13
626.9	28.15	900	40.1	133.4	1.835*	60	0.5753	1110	44.9	1110	44.9	292	6.84
627.8	28.27	1000	43.3	143.2	2.107	70	0.7175	1143	46.4	1143	46.4	<u>DATA SET 53*</u>	
628.6	28.39	<u>DATA SET 55</u>		150.3	2.339*	77	0.8404	1168	46.9	1168	46.9	76.1	1.908
628.9	28.43	4.2	0.0299	174.2	3.019*	99.5	1.303	1204	48.7	1204	48.7	191.3	7.242
629.2	28.47	5.41	0.0306	182.0	3.300	129	2.020	1239	50.0	1239	50.0	229.6	9.456
629.4	28.50	5.70	0.0308*	191.4	3.699	167	3.073	1281	50.8	1281	50.8	273	12.323
630.5	28.59	6.07	0.0310*	204.2	4.165*	292	7.700	1305	50.8	1305	50.8	274.30	12.402
631.1	28.61	6.53	0.0311*	227.0	5.094*	297	7.770	1325	51.7	1325	51.7	291.9	14.653
633.4	28.79	6.75	0.0313	259.4	6.361*	326	8.874	1348	52.2	1348	52.2	324.9	16.185
635.2	28.89	7.33	0.0316*	267.9	6.509*	379	11.31	1382	53.2	1382	53.2	363.50	19.419
637.1	28.99	7.64	0.0316	291.1	7.266*	493	17.78	1422	54.2	1422	54.2	406.9	23.250
638.7	29.09	8.24	0.0319*	300.0	7.34*	519	19.32	1448	55.3	1448	55.3	468.5	29.730
641.1	29.24	8.79	0.0322	<u>DATA SET 56</u>									
642.3	29.32	9.44	0.0325	1073	48.1	561	22.59	1479	56.7	1479	56.7	<u>DATA SET 64*</u>	
643.8	29.41	9.95	0.0328	1173	50.6	578	24.12	1516	57.1	1516	57.1	91	1.900
651.8	29.80	11.0	0.0336*	1273	52.8	598	25.59	1673	62.7	1673	62.7	173	6.110
675.3	30.83	11.8	0.0341	1373	55.2	608	26.97	1722	65.0	1722	65.0	193	7.470
701.2	31.82	12.8	0.0346*	1473	57.8	628	28.89	1773	84.5	1773	84.5	274.4	12.350
724.8	32.66	14.1	0.0357	1573	60.3	646	30.63	1873	85.6	1873	85.6	293	13.494
740.1	33.17	15.6	0.0374*	1673	62.7	672	31.80	1968	86.1	1968	86.1	367.7	18.913
760.0	33.84	16.9	0.0390	<u>DATA SET 57</u>									
785.8	34.66	18.8	0.0411*	1723	82.5	998	40.9	998	40.9	998	40.9	<u>DATA SET 65*</u>	
815.3	35.57	20.6	0.0437	1723	82.5	1016	42.1	1016	42.1	1016	42.1	4.2	0.024
839.5	36.33	22.9	0.0480	1728	83.5	1040	42.7	1040	42.7	1040	42.7	100	1.0
898.6	38.06	26.1	0.0551	1773	84.5	1059	43.6	1059	43.6	1059	43.6	200	3.6
939.7	39.22	28.8	0.0627*	1873	85.6	1077	44.2	1077	44.2	1077	44.2	300	7.3
972.6	40.12	32.2	0.0756	1968	86.1	1099	45.2	1099	45.2	1099	45.2	400	11.9
1016.0	41.29	35.8	0.0923	<u>DATA SET 58</u>									
1044.0	42.04	38.7	0.1109*	998	40.9	1121	45.6	1121	45.6	1121	45.6	500	17.9
1102.0	43.51	42.9	0.1339	1016	42.1	1158	46.1	1158	46.1	1158	46.1	600	25.6
1128.2	44.17	47.9	0.1787	1040	42.7	1177	47.8	1177	47.8	1177	47.8	628	29.3
1143.1	44.52	53.1	0.2310	1059	43.6	1195	48.5	1195	48.5	1195	48.5	700	32.4
1153.1	44.75	59.6	0.3127	1077	44.2	1214	49.0	1214	49.0	1214	49.0	800	36.0
<u>DATA SET 54*</u>													
100	1.1	65.6	0.4036	8.0	0.2837	1250	50.2	1250	50.2	1250	50.2	1728	85.06
200	4.0	77.5	0.5813*	9.2	0.2844	1265	50.2	1265	50.2	1265	50.2	1800	87.08
300	8.0	81.5	0.6601*	12	0.2872	1294	50.2	1294	50.2	1294	50.2	1873	89.12
400	12.7	84.9	0.7338	16.6	0.2913	1318	50.7	1318	50.7	1318	50.7	1898	89.82
500	18.7	89.5	0.8178*	19.1	0.2944	1339	51.7	1339	51.7	1339	51.7	657	35.2
		94.8	0.9367	22.3	0.3002	1357	52.3	1357	52.3	1357	52.3	668	35.4

* Not shown in figure.

TABLE 12. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF NICKEL N1 (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ		
DATA SET 66(cont.)											
673	35.2*	160	2.50*	293	8.5*	373	11.35	18	0.0669		
683	36.0	180	3.10	1726	63	482	17.1*	20	0.0673		
696	37.1*	200	3.70*	DATA SET 70						DATA SET 84*	
704	37.4	220	4.38	1726	64	596	25.6	1726 85.1			
711	37.6*	240	5.10	1726	81.5	621	27.3	DATA SET 85			
733	38.2	260	5.95	1726	64	633	28.6	1573	58.6		
742	38.5*	280	6.75*	1726	81.5	642	29.1*	1873	89.4		
749	38.9*	300	7.65*	DATA SET 71						DATA SET 86*	
764	39.5	320	8.50*	293	7.61	666	31.3*	1873 125			
766	40.0*	340	9.45	DATA SET 72						DATA SET 87*	
777	39.5*	360	10.35*	293	7.61	702	32.7	1728 3.24			
787	39.5*	380	11.35*	DATA SET 73						1800 3.25	
795	40.2	400	12.40	1063	47.7	873	38.6	1900 3.26			
806	40.4*	420	13.50	1273	54.3	966	41.2				
817	41.6*	440	14.65	1473	58.0	1073	44.1				
855	43.0	460	15.90	DATA SET 74							
861	43.3*	480	17.15*	1.85	0.00308	1.23	0.0628				
873	43.5*	500	18.50	4.15	0.00346	1.23	0.0629*				
902	44.5	520	20.25*	14.1	0.00753	1.30	0.0628*				
945	45.2	540	22.00*	20.1	0.01237	1.40	0.0629				
952	45.4*	560	24.10	DATA SET 75							
962	45.6*	580	26.30*	4.15	0.0407	1.58	0.0629*				
997	47.5	600	28.75	DATA SET 76							
1019	47.5*	620	31.25	4.15	0.0141	1.62	0.0629*				
1036	48.0*	640	32.75	DATA SET 77							
1078	49.1	660	33.25	1.85	0.00308	1.73	0.0629				
1089	49.4*	680	33.75	4.15	0.00346	2.01	0.0629*				
1127	51.0	700	34.25	14.1	0.00753	2.20	0.0629				
1141	51.4*	720	34.75	20.1	0.01237	2.41	0.0629*				
1165	51.3	740	35.25	DATA SET 78							
1173	52.0*	760	35.75	14	0.0665	2.52	0.0629*				
1187	52.6*	780	36.25	16	0.0666	2.73	0.0630				
1200	52.8	800	36.75	DATA SET 79							
1211	53.0*	820	37.25	14	0.0665	3.00	0.0630*				
1222	53.4*	840	37.75	16	0.0666	3.18	0.0630*				
1270	54.8	860	38.25	DATA SET 80							
1273	54.7*	880	38.75	1573	63	3.52	0.0630*				
1281	54.9*	900	39.25	1973	87	3.55	0.0630*				
1291	55.2										
DATA SET 68*											
		4.5	0.0116							DATA SET 81	
		295	7.51							80.2 0.9510	
										80.9 0.9666	
										194.7 4.0486	
										373 9.1092	
										581.4 15.7036	
										671.8 18.7782	
DATA SET 69											
		78	1.28							DATA SET 82*	
		178	4.02							20.2 1.2726	
		273	7.36*							80.4 1.8063	
		294	8.13							195.4 4.1811	
		323	9.14							273.1 6.2200	
DATA SET 70											
										DATA SET 83	
		14	0.0665							1726 82.3	
		16	0.0666							10 0.044	
										20 0.0512	
										30 0.0698	
										40 0.1124	
										60 0.294	
										80 0.625	
										100 1.033	

* Not shown in figure.

TABLE 12. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF NICKEL N1 (continued)

T	ρ	T	ρ	T	ρ
<u>DATA SET 96(cont.)</u>					
2.88	0.002653	1693	61.4	573	23.03
3.09	0.002675	1740	87.35	623	27.86
3.3	0.002705	1755	87.45	673	31.18
3.60	0.002735	1756	87.55	<u>DATA SET 104*</u>	
3.80	0.002761	1769	87.63	373	8.54
3.97	0.002782	1780	87.7	422	10.67
4.21	0.002814	1793	87.95	471	13.60
<u>DATA SET 97*</u>					
1798	87.95	1805	88.05	522	17.19
1818	88.2	1821	88.25	571	20.56
1836	88.45	1838	88.45	617	24.27
1855	88.7	1856	88.65	666	26.74
1872	88.85	1872	88.85	718	28.76
<u>DATA SET 98*</u>					
4.2	0.003185	1872	88.85	770	31.01
<u>DATA SET 99*</u>					
4.2	0.003053	1872	88.85	821	33.26
<u>DATA SET 102*</u>					
4.2	0.002588	1720	87.15	873	35.06
<u>DATA SET 100</u>					
1741	87.4	1741	87.4	<u>DATA SET 105*</u>	
1751	87.45	1751	87.45	4.2	0.0233
1765	87.7	1765	87.7	<u>DATA SET 106*</u>	
1766	87.65	1766	87.65	4.2	0.0145
1781	87.85	1781	87.85	<u>DATA SET 107*</u>	
1792	87.8	1792	87.8	4.2	0.00834
1795	88.05	1795	88.05	<u>DATA SET 108*</u>	
1770	87.25*	1800	88.05	4.2	0.00584
1772	60.3*	1808	88.2	<u>DATA SET 109*</u>	
1788	87.4	1811	88.2	4.2	0.00424
1791	87.55*	1823	88.35	<u>DATA SET 101*</u>	
1796	87.55*	1826	88.35	293	7.00
1799	87.65	1841	88.6	323	8.22
1814	87.8	1844	88.6	373	10.48
1817	87.9*	1857	88.8	423	13.00*
1836	88.1	1865	88.85	473	15.86
1840	88.1*	1873	89.1	523	19.16
1852	88.3*	1888	89.3	<u>DATA SET 103</u>	
1857	88.3	293	7.00	293	7.00
1867	88.45*	323	8.22	323	8.22
1869	88.45	373	10.48	373	10.48
1882	88.65	423	13.00*	423	13.00*
<u>DATA SET 101*</u>					
1648	60.35	473	15.86	473	15.86
1667	60.8	523	19.16	523	19.16

* Not shown in figure.

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5. APPENDICES

5.1. Methods for the Measurement of Electrical Resistivity

At the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) of Purdue University, the experimental methods for the measurement of electrical resistivity have been classified into various categories according to a similar scheme used by CINDAS for the classification of methods for the measurement of thermal conductivity [264, pp. 13a-25a]. This classification scheme of CINDAS is presented below. Note that the letters in parentheses following the respective methods are the code letter used in the "Method Used" column of the Table of Measurement Information for indicating the experimental methods used by the various authors.

Methods for the Measurement of Electrical Resistivity

A. Steady-State Methods

1. Voltmeter and ammeter direct reading method (V) [265, p. 159; 266, pp. 244-5]
2. Direct-current potentiometer method (A) [267, pp. 151-8]
 - a. 4-probe potentiometer method
3. Direct-current bridge methods (B) [267, pp. 144-51]
 - a. Kelvin double bridge method
 - b. Mueller bridge method
 - c. Wheatstone bridge method
4. Van der Pauw method (P) [268, 269]
5. Galvanometer amplifier method (G) [270, pp. 159-62]

B. Non-Steady-State Methods

1. Periodic current method
 - a. Direct connection to sample
 - (1) Alternating-current potentiometer method (C) [267, pp. 161-2]
 - (2) Alternating-current bridge method (D) [267, p. 162]
 - b. No connection to sample
 - (1) Mutual inductance method (M) [271]
 - (2) Self-inductance method (S) [272]
 - (3) Rotating field method (R) [273]

2. Non-periodic current method

a. Direct connection to sample

(1) Transient (subsecond) method (T) [154]

b. No connection to sample

(1) Eddy current decay method (E) [273; 267, p. 103]

5.2. Conversion Factors for the Units of Electrical Resistivity

The recommended values and experimental data for the electrical resistivity tabulated in this work are in the units: $10^{-8} \Omega \text{m}$. Conversion factors for the units of electrical resistivity, which may be used to convert the values given in ($10^{-8} \Omega \text{m}$) to values in other units, are given below.

Conversion Factors for the Units of Electrical Resistivity

Units to be Converted to	Multiply the Value Given in ($10^{-8} \Omega \text{m}$) by
ohm-meter (Ωm)	1×10^{-8}
ohm-centimeter (Ωcm)	1×10^{-6}
ohm-inch ($\Omega \text{in.}$)	3.937×10^{-7}
ohm-foot (Ωft)	3.281×10^{-8}
microohm-centimeter ($\mu\Omega \text{cm}$)	1
abohm-centimeter ($\text{ab}\Omega \text{cm}$)	1×10^3
statohm-centimeter ($\text{stat}\Omega \text{cm}$)	1.113×10^{-18}
emu (= $\text{ab}\Omega \text{cm}$)	1×10^3
esu (= $\text{stat}\Omega \text{cm}$)	1.113×10^{-18}
ohm-circular mil per foot ($\Omega \text{cmil ft}^{-1}$)	6.015

Example: $1.000 \times 10^{-8} \Omega \text{m} = 3.937 \times 10^{-7} \Omega \text{in.}$

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